

SAMPLING AND ANALYSIS PLAN

REMEDIAL INVESTIGATION RICHARDSON FLAT

Site ID Number: UT980952840

February 20, 2001

Prepared for:

United Park City Mines Company P.O. Box 1450 Park City, UT 84060

Prepared by:

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Prepared by:	Jim Fricke	Date:	3/10/01
Reviewed by:	Resource Management Consultants	Date:	3/10/10
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1.0 INTRODUCTION

This Sampling and Analysis Plan (SAP) addresses the sampling and analysis of surface and ground water, soils, sediment and tailings materials during field activities conducted as part of a focused Remedial Investigation and Feasibility Study (FOCUSED RI/FS) at the Richardson Flat Tailings site (Site) near Park City, Utah. The Site is an inactive mill tailings impoundment owned by United Park City Mines Company (United Park), United Park is conducting the RI/FS pursuant to the Administrative Order on Consent (AOC) for a focused Remedial Investigation/Feasibility Study, dated September 28, 2000, U.S. EPA Docket No. [CERCLA-8-2000-19] The focused RI/FS Workplan (RMC, 2000) as referenced in this SAP was approved by the United States Environmental Protection Agency Region VIII (EPA) on September 28, 2000.

This SAP combines the relevant portions of a Field Sampling Plan (FSP) and Quality Assurance Project Plan (QAPP). As requested by EPA, the format of this plan follows the 16 elements of a QAPP as defined in EPA Requirements for Quality Assurance Project Plans for Environmental Data Operations (USEPA QA/R-5, 1998a). Section titles are followed by the corresponding QA/R-5 outline numbers in parentheses.

The SAP contains the Health & Safety Plan (HASP) that will be followed during Site activities by all visitors (See Appendix A).

This SAP, when necessary, can and may be amended should there be a need to do so. It will also provide procedures for sampling to be conducted as part of EPA oversight of the RI/FS.

2.0 PROJECT MANAGEMENT (A)

This SAP and subsequent revisions will be distributed to the following organizations and individuals:

Jim Christiansen U.S. EPA Region 8 999 18th Street Suite 300 Denver, CO 80202-2466 (2 copies)	Muhammad Slam Utah Division of Environmental Response & Remediation 168 North 1950 West 1st Floor Salt Lake City, UT 84116 (1 copy)
Betty Grizzle U.S. Fish & Wildlife Service Lincoln Plaza Suite 404 145 East 1300 South Salt Lake City, UT 84115 (1 copy)	Kevin Murray LeBoeuf, Lamb, Greene & MacRae 1000 Kearns Building 136 South Main Street Salt Lake City, UT 84101-1685 (1 copy)
Public Repository Park City Library 1255 Park Avenue Park City, UT 84060 (2 copies)	
Resource Management Consultants 8138 South State Street Suite 2a Midvale, UT 84047 (1 copy)	Global Environmental Strategies 5953 South Iola Way Englewood, CO 80111 (1 copy)

2.1 PROJECT/TASK ORGANIZATION (A4)

The RI/FS management team consists of United Park personnel with assistance from Resource Management Consultants (RMC) and other environmental consulting firms as needed. Figure 1 shows the chain-of-command for the project managers, engineers, and quality assurance officials responsible for managing the Richardson Flat Tailings Site FOCUSED RI/FS.

United Park's environmental Project Manager for the Site is Kerry Gee, who will be responsible for all project management and communication with the regulatory agencies. Jim Fricke of RMC, Salt Lake City, Utah, leads United Park's environmental project consultant team and will be the Site Manager, who will be responsible for implementation of the SAP. Todd Leeds, of RMC, is the Field Manager who will be

responsible for all field activities related to this document. Wesley McDonald, RMC, is the Site Safety Officer, who will be responsible for visitor sign in and ensure that all site visitors comply with the HASP.

The EPA Project Coordinator is Jim Christiansen, Region VIII, Denver, Colorado. The Utah Department of Environmental Remediation and Response (UDERR) Project Manager is Muhammad Slam. The EPA Project Coordinator and the UDERR Project Manager work cooperatively to oversee the work being performed at the Richardson Flat site.

Mr. Gee, as Project Manager, is responsible for the overall management and coordination of the following activities:

- Coordination with EPA/UDERR regarding the status of the project;
- Providing oversight of the subcontractors;
- Reviewing monthly status reports;
- Supervising production and review of deliverables;
- Tracking work progress against planned budgets and schedules;
- Informing EPA/UDERR of changes in the Workplan, SAP, HASP and/or other project documents;
- Notifying EPA/UDERR immediately of significant problems affecting the quality of data or the ability to meet project objectives;
- Procuring subcontractors to provide sampling and analytical support;
- Providing oversight of report preparation;
- Organizing and conducting a field planning meeting.

Mr. Fricke, as the Site Manager, is responsible for the following:

- Preparing monthly status reports;
- Coordinating with the laboratory regarding the analytical, data validation, and Quality Assurance/Quality Control (QA/QC) issues related to sample analysis;
- Reviewing analytical results and deliverables from subcontractors;
- Incorporating changes in the Workplan, SAP, HASP, and/or other project documents;
- Scheduling personnel and material resources;
- Implementing field aspects of the investigation, including this SAP and other project documents;

- Implementing the QC measures specified in the QAPP in this and other project documents;
- Implementing corrective actions resulting from staff observations, QA/QC surveillance, and /or QA audits;
- Providing oversight of data management;
- Coordinating and overseeing the efforts of the subcontractors providing sampling and analytical support;
- Scheduling and conducting field work;
- Notifying the subcontract analytical laboratory of scheduled sample shipments and coordinating work activities;
- Gathering sampling equipment and field logbooks, and confirming required sample containers and preservatives.
- Maintaining proper chain-of-custody forms and shipping of samples to the analytical laboratory during sampling events;
- Ensuring that sampling is conducted in accordance with procedures detailed in this SAP and that the quantity and location of all samples meet the requirements of the SAP; and
- Identifying problems at the field team level; resolving difficulties in consultation with the QA/QC staff; implementing and documenting corrective action procedures at the field team level; and providing communication between the field team and United Park management.

The roles and responsibilities of other field team members will be to assist the Site Manager with sampling activities, sample handling, and overall documentation.

Oversight activities including sampling to be conducted by EPA's on-site contractor will be coordinated between the EPA Project Coordinator and United Park's Project Manager. EPA's on-site contractor and the Site or Field manager will work together to coordinate sampling efforts.

2.1.1 Quality Assurance/Quality Control Organization

The Quality Assurance Official (QAO) is Gary Colgan, with Aquifer Science, who is responsible for the quality assurance/quality control of the data that are generated during implementation of the SAP. Mr. Colgan will report any QA/QC problems to the Site Manager. As the QAO, he will be responsible for the following:

- Reviewing and approving project specific plans;
- Directing the overall project QA/QC program;
- Maintaining QA/QC oversight of the project;
- Reviewing QA/QC sections in project reports, as applicable;
- Reviewing QA/QC procedures applicable to this SAP;
- Auditing selected activities of this project performed by RMC and subcontractors, as necessary;
- Initiating, reviewing, and following up on response actions to address QA/QC problems, as necessary;
- Consulting with the Site Manager and/or Project Manager, as needed, on appropriate QA/QC measures and corrective actions;
- Arranging performance audits of measurement activities, as necessary; and
- Providing written reports on QA/QC activity to the Project Manager and Site Manager.

2.2 PROBLEM DEFINITION/BACKGROUND (A5)

United Park is the current owner of a large parcel of property (the "Property"), comprising approximately 700 acres, located in Summit County, Utah. Figure 2.0 shows the general geographic location of the Property. The Site included a historic mine tailings impoundment, consisting of a large, geometrically closed basin formed by an earth embankment and a series of perimeter containment dikes and covers approximately 160 acres of the Property. The tailings impoundment resulted from decades of mining and milling silver-laden ore in the area around Park City known as the Park City Mining District. The Site is depicted in Figure 3.0.

The Site has remained unused since mining and milling operations ceased in 1982. Over the past fifteen years, the (EPA), the Utah Department of Environmental Quality (UDEQ) and United Park have been investigating the Site in order to characterize the Site and determine potential adverse impacts to human health and the environment associated with the Site. At the same time, United Park has been implementing a series of remedial measures at the Site intended to mitigate any potential adverse impacts on human health and the environment.

Remedial measures conducted by United Park include the following:

- Placement of clean cover soils over all exposed tailings,
- Reconstruction of the diversion ditch; and,
- Construction of a fence around the property perimeter.

Evaluation of these remedial measures relative to potential site risks is presented in the conceptual site model found in Section 2.2.4 of this document.

A detailed description of the Site, including a description of the Site operational history, existing closure measures and elements, regional geology and hydrogeology and surface water is set forth in Sections 2.0 to 2.5 of the FOCUSED RI/FS Work Plan.

2.2.1 General Goals and Objectives

The purpose of this SAP is to guide additional sampling of environmental media in support of the EPA approved Focused Remedial Investigation Feasibility Study Work Plan. The primary purposes of the remedial investigation for this site are to:

- examine existing data and remedial measures,
- estimate the degree of risk posed to human health and the environment, and
- determine what data are needed to further define the nature and extent of contamination contributing to that risk.

The SAP will identify sample types, number of samples to be collected, and establish sample collection and analytical procedures. The following data will be collected and plans prepared as part of the SAP process:

- 1. Collect soil, sediment, and water samples to further characterize Site conditions.
- 2. Provide data to fill data gaps due to seasonal variations in media such as groundwater and surface water.
- 3. Evaluate existing data to determine the need for further data collection.
- 4. Collect data of sufficient quality and quantity for EPA to conduct a streamlined risk evaluation.
- 5. Through this SAP, establish procedures for data collection and analysis.
- 6. Through this SAP, define Quality Assurance/Quality Control (QAQC) measures.
- 7. Through this SAP, provide a Site Health & Safety Plan for all workers and visitors.

2.2.1 Historical Data Summary

Previous data have been collected by EPA, United Park and Park City Ventures (PCV). The historic data cover a time period beginning in the early 1970's up to the present. Appendix E contains a listing of known site investigations and reports. Because past investigation activities by PCV, Noranda and United Park were performed without EPA oversight, the results from such investigations will be evaluated as part of, and incorporated as appropriate into, the focused RI/FS. A detailed description of the investigations previously conducted at the Site, which have included investigations of air quality, the extent of the tailings cover, tailings impoundment integrity and stability, and groundwater and surface water quality, are set forth in Sections 3.0 to 3.5 of the FOCUSED RI/FS Work Plan. Based on previous and current environmental studies and existing Site conditions, United Park has developed a draft Conceptual Site Model (CSM). The CSM has been developed with input from EPA, UDERR and the U.S. Fish and Wildlife Service (USFWS). Section 2.2.4 presents the CSM. The CSM will be used to determine what environmental samples are required, how the samples will be used in the risk assessment and derivation of clear SAP objectives. The CSM will be updated and modified as more information is obtained either through sampling or meetings with EPA and other stakeholders.

[Note: Although CERCLA guidance indicates that certain site descriptions that are already included in the FOCUSED RI/FS Work Plan need not be repeated, the descriptions set forth in the following Sections 2.2.1.1 to 2.2.2.4 are included to be more informative.]

2.2.2.1 Surface Water

Section 3.5 of the focused RI/FS Workplan presents a summary of historical surface water data for Silver Creek and the south diversion ditch. Surface water data have been collected on and near the Site since 1975 as part of permit requirements and investigations by EPA and United Park. The data generally show that metals concentrations measured in the south diversion ditch have declined particularly since the mid to late 1980's when the ditch was reconstructed and when the tailings were covered with clean soil (See, focused RI/FS Workplan, Table 3.1, Station N5). All exposed tailings have been covered with clean soils as part of voluntary remedial measures. Data collected in 1999 indicate that zinc concentrations measured at the outfall of the ditch

meet applicable water quality standards. Zinc concentrations exceed water quality criteria in the central portion of the diversion ditch and both upstream and downstream of the Site in Silver Creek. The downstream Silver Creek zinc concentrations are less than the upstream concentrations indicating that flow from the diversion ditch may be diluting the upstream zinc concentrations.

Table 3.3 in the focused focused RI/FS Workplan presents data collected in May and June of 1999. Additional data collected on the Site for the remainder of 1999 will be presented in the RI report. The 1999 data will be evaluated, following QA/QC criteria set forth in this SAP, to determine if the data, along with information collected in this FOCUSED RI/FS, can be used to guide decisions on what, if any, further remedial actions are required at the Site. If previously collected data do not meet the QA/QC criteria the data will only be used to guide additional sampling following the criteria set forth in this SAP.

Metals measured in 1999 include RCRA metals, copper and zinc. The surface water analyte list was determined based on metals present in tailing samples collected by E&E in 1984. Complete cation/anion analytes were also measured at select stations. (See, focused RI/FS Workplan, RMC 2000). Mercury detection limits were not adequate to determine if the water in the diversion ditch meets the chronic aquatic wildlife criteria (12 part per trillion (ppt) – dissolved). Measured mercury concentrations at all site sample locations were non-detectable at less than 0.0005 mg/l. In May of 2000, mercury samples were collected on Silver Creek upstream and downstream of Richardson Flat and on the diversion ditch, as part of the Upper Silver Creek Watershed surface water sampling (See, Upper Silver Creek Watershed Analytical Results Report, August 2000, RMC). The detection limit for these mercury analyses was low enough to be below the chronic wildlife criteria. Analytical results indicated that both Silver Creek and the diversion ditch met the appropriate water quality criteria for mercury.

2.2.2.2 Ground Water

Groundwater quality data have been collected in monitoring wells located on and near the Site by EPA, United Park and PCV. The focused RI/FS Workplan and attachments discuss the historic and recent groundwater data. Initial review of the historic groundwater quality data suggests that this data will be of more qualitative than quantitative use. The procedures for QA/QC control detailed in this SAP will be used to

determine the usability of the historic data. If the data do not meet QA/QC goals, the data will not be used in decision making directly. Rather, these data will be used to optimize the data gathering process and additional data that meet QA/QC requirements will be collected and used for decision making.

Historical groundwater data generally show that metals concentrations have steadily decreased in Site wells, with the exception of the upgradient well (RT-1) installed by EPA in 1985 and MW-5 located at the toe of the main embankment (See, Figure 3.3, focused RI/FS Workplan). EPA sampled RT-1 in 1985 and again in 1992, total and dissolved metals concentrations had increased for aluminum, antimony, arsenic, and barium over the time period.

Comparison of data collected from RT-1 in 1984 and 1992 reveals that water quality appears to have deteriorated at this location over time. Some dissolved metal concentrations have increased from 1984 to 1992. The 1992 data contains some anomalies that suggest either the sample was contaminated or there were some analytical errors. Dissolved metal concentrations are greater than total metal concentrations for antimony, copper, and silver. This suggests that there are sample or analytical errors or interference. The well is completed in two aquifers and there is mixing of water between the two water-bearing zones. During site visits in early 1999, it was apparent that vandals had damaged the wellhead integrity. It is not known if this was apparent in 1992. Thus, surface contamination may have impacted water quality.

A discussion of properly closing this well will take place in Section 3.2 of this SAP.

In 1973, PCV installed three monitoring wells (MW-1, MW-2 and MW-3) at the bottom of the main embankment as part of State of Utah requirements for the tailings impoundment operating permit. In 1976, PCV installed three additional wells (MW-4, MW-5, and MW-6). Figure 3.3 in the focused RI/FS Workplan shows the well locations. It appears that PCV buried monitoring well MW-2 in 1976 during installation of the three new wells. Thus, five groundwater monitoring wells are located near the toe of the embankment. The boring and well completion logs for these five wells can be found in Appendix D of the focused RI/FS Workplan and are summarized in Section 3.4 of the Workplan.

Table 3.2 in the focused RI/FS Workplan presents groundwater data collected by United Park from 1982 to 1987 and 1991 to 1998 from the embankment monitoring wells. Additional groundwater investigation is proposed as a part of the focused RI/FS Workplan. Details outlining sampling and analyses specifics are contained in this SAP.

In 1999, United Park hired Weston Groundwater Engineering, Inc. ("Weston") to conduct a supplemental hydrogeological investigation of the Site. This study represented the most extensive groundwater investigation conducted to date to better understand groundwater systems on the Property. Weston evaluated historical Site and regional data to derive a conceptual hydrogeological site model (See focused RI/FS Workplan, Appendix A). In the course of its investigation, Weston installed eleven additional piezometers throughout the Property (See focused RI/FS Workplan, Plate 1, Appendix A). Boring logs from the piezometer installation verified the existence of two aquifers associated with the Property. Water level data collected from the piezometers indicates that the two aquifers are confined and are separated from one another by a significant layer of stiff, clay-rich material. Water level data collected after the installation of the piezometers and subsequent water level measurements indicate that the water levels in the two aquifers vary seasonally, with higher water levels occurring in the spring.

The data reported by Weston was not available to earlier Site inspection teams and other agencies that previously evaluated the Site. Studies by Dames & Moore identified the presence of clays in the naturally occurring material at the Site. It was not until Weston's investigation that extent or the significance of the natural clay material underlying the Property was known. The existence of two to five feet of clay-rich topsoil and the presence of the large area of silt and clay that overly the upper aquifer represent a significant barrier to the vertical migration of any water from saturated tailings. Weston has collected monthly groundwater data from February of 1999 to February 2000. These data will be evaluated in a supplemental hydrogeological study and submitted in the RI report. Any additional data that needs to be collected for this study will be done so in accordance with this SAP. Groundwater monitoring proposed as a part of this SAP will be used to verify the Site hydrogeological conceptual model as presented by Weston.

2.2.2.3 Soils

During investigations conducted by EPA and United Park, limited soils data were collected to primarily characterize the tailings soil cover and determine the extent of wind

blown tailings. In 1984, EPA collected one soil sample to document background metals and tailings concentrations. Generally, the 1984 data demonstrated that the tailings contain elevated concentrations of metals, and that the soil/tailings contact contains elevated metals, but at lower concentrations than the tailings. In 1992, EPA collected samples of the soil cover to determine the extent and thickness of the cover. Based on the 1992 sampling efforts, the EPA data show that of the 29 sample locations 28 contained soil or vegetative cover and 9 contained no soil cover. Since 1992, United Park has covered all visibly exposed tailings on the impoundment.

2.2.2.4 Sediment

E&E collected four (4) sediment samples at the Site in 1992 in the wetland area between the main tailings embankment and Silver Creek. (See Final Report Richardson Flat Tailings, E&E, 1993) The data show that sediments in this area contain elevated concentrations of metals. Water from the south diversion ditch flows from east to west in the wetland area, while Silver Creek flows along the west side of the wetland area. The E&E data show that metals concentrations were elevated in the sediments; however, it appears that there is very little transfer of metals in the sediments to the water. Both historic and recent surface water data show that very little metal is being leached from the sediments. In 1985, E&E reported an increase in lead from 0.036 mg/l to 0.151 mg/l in upstream versus downstream Silver Creek surface water samples.

Additional sediment data will be collected in the diversion ditch as part of the focused RI/FS. The data collected from this sampling will be used to evaluate the long-term effectiveness of the wetland system to remove metals in the water, aid in the determination of the source of metal in water flowing in the diversion ditch and used in an ecological risk assessment to be conducted by the EPA. This data will also be applicable to evaluate long-term effectiveness of metal removal in the wetland area between the main embankment and Silver Creek.

2.2.3 Recent Environmental Conditions

The following conditions have been identified from field work and the chemical analysis of soil, surface, and groundwater samples previously collected at the site by EPA's contractor E&E, United Park and PCV from 1985 to the present time:

- An imported clean fill covers the tailings surface, depth of the cover ranges from less than 6 inches to several feet thick.
- Tailings beneath the soil cover contain the following metals at elevated concentrations: antimony, arsenic, cadmium, copper, lead, mercury, selenium and zinc (EPA, 1992).
- Surface water on and near the Site contains the following metals at low concentrations: arsenic, cadmium, copper, iron, lead, and zinc (focused RI/FS Workplan, 2000). Of the metals detected, only zinc and possibly mercury exceed water quality standards. However, Silver Creek is currently moving through a TMDL process for elevated levels of zinc and cadmium. Zinc levels upstream and downstream of the Site on Silver Creek exceed protection of aquatic wildlife criteria. Laboratory detection limits for mercury were low enough to verify that all samples collected in 1999 were non-detected for mercury at 0.0005 mg/l. The acute aquatic wildlife criteria for mercury is 0.00012 mg/l or 12 parts per trillion.
- Shallow groundwater at the downstream face of the main tailings embankment
 contains the following metals at low concentrations: aluminum, antimony, arsenic,
 barium, beryllium, cadmium, chromium, cobalt, copper, iron, lead, manganese,
 mercury nickel, selenium, silver, thallium, vanadium, and zinc. Of the metals
 detected, only lead exceeds the Utah Ground Water Quality Standard. (focused RI/FS
 Workplan, 2000).
- Data collected by E&E in 1992 show that sediments in the wetland below the main embankment contain the following metals at elevated levels: antimony, arsenic cadmium, lead, mercury, selenium and zinc. Surface water data from the wetland area, collected at the same time as the sediment data, show that only lead was elevated when compared to upstream concentrations. It is likely that the wetland sediments, that are rich in organic carbon, are binding the metals and not allowing significant mobilization of metals to occur.

There have been several investigations conducted on the Site by EPA, United Park and PCV over the past fifteen years; Appendix B contains a listing of known site investigation reports.

2.2.4 Conceptual Site Model

The CSM guides the evaluation of risk at a site. It represents our understanding at any moment of the potential for metals to move from a source to a human or ecological receptor. The CSM is a collection of potential specific pathways by which this exposure may occur. A complete exposure pathway consists of five necessary elements:

- A chemical source
- A release mechanism
- An environmental transport medium for the released chemical
- A physical point of contact (the exposure point) for human or ecological receptors
- A route of chemical uptake into the body

All five elements must be present for an exposure pathway to be complete (that is, for chemical exposure to occur). Incomplete pathways do not result in exposure to humans or biota, and therefore do not result in risk. Complete pathways may result in such a small exposure that risk is not significant. Where possible, the risk assessment process will evaluate both the completeness and the significance of the pathways. The conceptual site model helps to identify the data needed in the risk assessment process to complete this analysis.

In the following section, the potential sources, release mechanisms, exposure media, exposure points, and human and ecological receptors will be briefly described, and then potential pathways from source to receptor will be discussed one by one. Existing data will be reviewed to identify data gaps. The sampling plan described in this document is intended to begin to fill the data gaps and reduce uncertainty about the potential for risk from each exposure pathway.

2.2.4.1 Sources

The source of metals at the site is the historical placement of tailings from mining operations elsewhere. To better understand the potential release of metals, the original source has been conceptually divided into two parts: (1) the tailings that are contained within the tailings impoundment where they are covered by a clay soil cover or cap and contained within a containment dike system, and (2) the tailings that are covered in small, naturally low areas outside and to the south of the impoundment. The conceptual site

model for the impounded tailings is illustrated in Figure 8a and the model for tailings placed outside of the impoundment is in Figure 8b.

2.2.4.2 Release Mechanisms

Four primary release mechanisms have been identified: (1) direct contact with the tailings; (2) direct contact with soils that are the result of the mixing of the tailings with cover soils inside and outside of the impoundment, potentially due to root transport through cover soils; (3) infiltration by rainwater and snowmelt and subsequent leaching of metals into ground or surface water; and (4) historic wind erosion of the tailings before the cap was placed on them.

A secondary release mechanism may be the removal of metals in diversion ditch surface water through biological processes, resulting in metal deposition in the sediments. This is indicated by the apparent reduction in zinc concentrations as the water moves down the ditch to Silver Creek.

2.2.4.3 Exposure Media

The primary media to which humans or ecological receptors may be exposed as a direct result of release from the source include the tailings themselves; the cover soils mixed with tailings (if that has occurred); surface water; and any windblown tailings that are off site.

Metals have also been found in sediments in the wetland below the main embankment. (The origin of these metals is uncertain; they could be from Silver Creek tailings or impoundment tailings via the diversion ditch). There may be metals in the sediments in the diversion ditch possibly due to biochemical removal processes present in the diversion ditch wetlands. Data show that the metals concentrations in the diversion ditch surface waters decrease, , as the water moves down the ditch toward Silver Creek. Sediments in the ditch may be a secondary exposure medium. If the enhanced metals levels have occurred due to a biological reaction, it may be that these metals have a relatively low bioavailability for both ecological and human receptors.

Groundwater exposure does not occur for humans on site, since there are no drinking water wells. Potential for offsite exposure has not been evaluated yet. It is possible that metals are released from tailings (in or out of the impoundment) into groundwater and

then captured by downgradient wells. Ground water exposure may be a transport medium but not an exposure medium for ecological receptors in two very localized situations: (1) if ground water is affected by off-impoundment tailings and is intercepted by the south diversion ditch, or (2) if shallow ground water creating the seeps at the base of the embankment is affected.

Surface water in four main areas around the site (Silver Creek, the drainage ditch system, ponded water in the northwestern area of the impoundment, and the seeps discussed above) may be affected by transport of metals from the tailings. Parts of this sampling plan are intended to investigate whether (or how) this is occurring and, if so, which potential source (impounded tailings, or tailings outside of the impoundment) is responsible.

Prior to the tailings being covered, it appeared that wind blew quantities of tailings off the impoundment, primarily to the north and northwest. Off-site soils have not been investigated to determine the extent and degree of effects from these windblown tailings. Such data are needed to complete the risk assessment and planned sampling is described in this SAP.

2.2.4.4 Exposure Points

Exposure points are those specific physical locations where receptors may be exposed to site-related materials. The locations are selected for evaluation of risk based on existence of elevated chemical concentrations, present or future existence of receptors of concern, sensitivity of receptors, or similar considerations. Measured or modeled concentrations of site chemicals in media of concern at the exposure points are usually needed to complete the risk assessment, unless cost outweighs the benefits of reducing uncertainty about risk at those points. At present, identified potential exposure points include:

- tailings or soils mixed with tailings outside of the impoundment;
- areas on the impoundment where there may be exposure to tailings, or tailings mixed with cover soils, on the impoundment;
- water and sediments in the diversion ditch;
- water and sediments in Silver Creek;
- off-site areas that received historic wind-blown tailings; and

 areas that may have received water runoff, seeped water, or sediments containing site-related metals, including the wetland area to the northwest of the site at the base of the embankment.

2.2.4.5 Potential Receptors and Routes of Exposure

Potential human receptors are discussed here as "site visitors" and "off-site receptors" until they are defined further. Site visitors now include:

- people working on or studying the site, and
- recreational users of the nearby bike path who may wander or trespass on to the site.

In the future human receptors may include on-site recreational users. Off-site receptors are those residents, workers, or recreational users such as hikers or hunters who may encounter water, sediments, or windblown tailings that have migrated off the site. At present, no one is known to live within about 3 km or more from the site; the closest residences are about 3 km upstream near Prospector Square in Park City, and at least 5 km downstream, and the closest off-site workers are at the facility about 0.5 km to the north across Highway 248.

Humans may be exposed to metals via incidental ingestion of soils and sediments, via dermal absorption of metals that may adhere to skin, and via inhalation of dust derived from the soils and suspended in the air. The most significant of these routes relative to the others is incidental ingestion. Dermal absorption of metals is usually assumed by EPA to be very low, and it can readily be shown that inhalation of suspended dusts derived from soils presents an insignificant risk compared to directly ingesting those same soils. For example, in the risk assessment for the Murray Smelter, EPA indicated that the dermal and inhalation exposure routes "may be complete; however, risk is low" and the exposure routes would receive "qualitative evaluation only." (See Figure 3-1, Weston, Inc., Baseline Risk Assessment, Murray Smelter, 1996)

To conduct the ecological risk assessment, the ecological receptors of concern must be specifically selected. Since that has not yet been done, we have used two general habitat categories (wetland and upland) and four general biota categories (avian, invertebrates), other non-avian vertebrates (including fish, frogs, mammals), and plants (vascular and non-vascular) to start thinking about ecological risk and to identify general data gaps. At this point, the potential for uptake through a food chain or web is noted with the route "food chain exposure," which is intended to indicate consumption of plants, benthic or

soil invertebrates, or other food items by herbivores or carnivores. Obviously this exposure route is not possible to the plants themselves. For many invertebrates the food chain pathway is also unlikely or insignificant, although some herbivorous or carnivorous invertebrates such as beetles may take up metals from food items. The ecological risk assessor will develop significant details of food web issues later, if necessary.

Ecological receptors live in intimate contact with environmental media and may receive exposures via a number of exposure routes, depending on their life habits. To accommodate the variety of potential routes, at this point we have simply divided the exposure routes into "direct exposure" (which could include ingestion of soil or sediment while feeding or grooming, dermal absorption while burrowing or dirt bathing, and inhalation or water breathing, just for example) and "food chain exposure" which would cover anything else. Again, details will be developed later by the ecological risk assessor.

2.2.4.6 Discussion of Potential Exposure Pathways

In this section, each identified potential exposure pathway from source to receptor will be discussed relative to existing data, probable completeness or significance, and data needs to reduce uncertainty. If a pathway is designated as "potentially significant" in this CSM description, that does not mean that there is a risk; it merely indicates that further evaluation by the risk assessors is appropriate based either on existing data or on current uncertainties about potential risk. Data needs are briefly summarized in Table 1; specific sampling plans are detailed later in this SAP.

2.2.4.7 Background Metals

All potential exposure pathways related to this site must be considered relative to naturally occurring background levels of metals in soils, sediments, and water. Many site-related metals occur naturally in soils and food items and some, including zinc and copper, are essential for life. The area surrounding the site is highly mineralized, so typical natural background levels of these metals may be higher than those found in other areas. At present, some background information is available for nearby areas, but none is available for soils typical of the site itself. This information would provide important context for the risk assessment. This SAP describes the planned collection of background soil samples in Section 3.1.3.

2.2.4.8 Direct Contact with Tailings

Direct contact with tailings in the impoundment can only occur if human visitors, animals or plants penetrate the existing soil cover. This cover varies in thickness from perhaps 5 feet down to less than a foot in limited areas. For humans, such contact is considered unlikely, and even if it occurred, it would be insignificant. Humans are unlikely to eat any food products from plants growing anywhere on the site, so any sort of food chain exposure pathway for humans due to plants contacting tailings is also considered insignificant. This is indicated in Figure 8a. The soil used to cover the tailings was obtained locally from sources that were free of mine waste, such as excavation waste from residential construction. For ecological receptors, direct contact could only occur in areas where the soil cover is thin and where animals burrow or plants grow roots that are deep enough to penetrate the cover. This sampling plan includes an investigation of whether the soil cap is thin or absent in some areas, and whether soil/tailings mixing has occurred.

Off site, contact with tailings could occur for both human and ecological receptors. This pathway is therefore probably complete and potentially significant for ecological receptors. Additional sampling of the tailings is planned. The areal extent of these tailings appears to be limited but more data on their nature and extent is needed and will be collected as part of this sampling program.

2.2.4.9 Contact with Tailings Mixed with Soil

Because the cap is thin in some areas, it is possible that placement of the soil cover, animal burrowing, root transport, or other processes have mixed the cover soils with the tailings. No data currently exist to evaluate this possibility. If mixing has occurred, exposure is possible for both human site visitors and ecological receptors and the pathway should be considered potentially complete.

2.2.4.10 Contact with Ground Water

Direct exposure to the shallow ground water is not likely (except by very deep-rooted plants, not currently known to live at the site, or plants in the diversion ditch if indeed the ditch sometimes intercepts groundwater), but if groundwater is being pumped to drinking water wells, then people living or working off-site may be exposed if they ingest the

water. Contact with ground water on site is an incomplete pathway for humans; on-site well water is not used for human consumption in any way.

Human contact with ground water potentially affected off-site is not known to be a complete pathway but a well survey is needed to be more confident of this. As part of the Remedial Investigation (RI) for this site additional data will be collected on nearby wells. The data will include drilling depths, geological sequences encountered, and water quality data. The hydrogeological report by Weston Engineering presented in Appendix A of the workplan (RMC, 2000) will be updated with the additional data. The scope of the hydrogeological conceptual model will be enlarged to include these off-site wells.

Based on existing information, the possibility that metals might migrate more than a short distance in ground water before surfacing appears unlikely. The groundwater report currently in preparation by Montgomery Watson is expected to show that the very low permeability clay layer underneath of the tailings impoundment prevents significant downward migration of metals into the aquifer (see Figure 4.0 in the Workplan), preventing entry of metals into the aquifer and preventing groundwater transport off-site. Short-distance lateral movement in the shallow alluvial aquifer is still possible through the sides of the impoundment, potentially allowing metals to be released into the diversion ditch to the east, south, and west of the impoundment, and at the toe of the embankment nearest to Silver Creek on the west. However, available data show that very low-flow, seasonal seeps appear not to reach Silver Creek, over 500 feet away.

2.2.4.11 Contact with Surface Water

Surface water potentially affected by the site can be found in at least some years and seasons ponded on top of the impoundment cover itself; in the diversion ditch; in the wetland area to the northwest of the impoundment; and possibly in Silver Creek. Silver Creek itself may have naturally enhanced concentrations of metals and does have enhanced levels of metals due to historic mining activity as well as potential impacts from other recreational and residential uses upstream. Data from standing surface water on the impoundment collected in 1999 (see, focused RI/FS Workplan, RF-9) detected only barium and zinc, both of which were below aquatic wildlife criteria, so exposure is unlikely by this pathway. Metals concentrations measured in the south diversion ditch have declined particularly since the mid to late 1980's when the ditch was reconstructed and when the tailings were covered with clean soil. As stated earlier, data collected in

1999 indicate that zinc concentrations measured at the outfall of the ditch meet applicable water quality standards. Zinc concentrations exceed water quality criteria in the central portion of the diversion ditch and both upstream and downstream of the site in Silver Creek. The downstream Silver Creek zinc concentrations are less than the upstream concentrations indicating that flow from the diversion ditch may be diluting the upstream zinc concentrations. Existing data appear adequate to evaluate human health risk. While the human pathway appears complete, existing data indicate it is unlikely to be a significant pathway.

The surface water pathway therefore appears complete and potentially significant for ecological receptors in the diversion ditch, but not on top of the impoundment itself. Additional data have been requested to support the ecological risk assessment and to better understand the migration of metals from the site sources, and this sampling plan describes activities to help further evaluate upstream and downstream water quality in the Silver Creek alluvial aquifer to assess impacts from the tailings impoundment. Data provided by Weston show that the creek typically loses water into the alluvial aquifer in the stream reach near the impoundment. Depending on the season, the water may flow both ways depending on ground water levels and stream water elevation. This SAP describes plans to collect ground water and surface water elevation data to quantify this.

Human site visitors may be exposed to this water occasionally if they try to explore or play in the area or fish in the creek. There are no water recreation areas for swimming or boating on or near the site. At certain times of year the water flow in the ditch is quite high and contact is unlikely. Indirect exposure to metals in surface water (as well as sediments, depending on the circumstances) is possible through ingestion of fish exposed at the site; although this pathway might be complete at some point, it is unlikely to be a significant route of exposure due to low uptake of the metals by fish and low consumption rates due to the small size of the stream. Human exposure to site-related metals in surface water is possible but is expected to be fairly low overall.

Exposure to surface water by ecological receptors is potentially significant and further evaluation will probably be required. Surface water data for each of the exposure points do exist. If additional data are collected for the purpose of ecological evaluation, they should be collected as part of a specific plan that will need to be developed with the ecological risk assessor. The planned collection of ground water elevation data will be critical to determining whether the actual pathway from Richardson Flat tailings to

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ecological receptors is complete and a significant contributor to any exposure in Silver Creek.

Tailings outside the impoundment, may be contributing to elevated levels of metals in surface water at some exposure points such as the diversion ditch. A focus of the current data collection program is to clarify which source is the significant contributor. This information will help to clarify effective remediation strategies.

2.2.4.12 Contact with Sediments

Similar to the situation with surface water, human contact with sediments is a possible, even probable complete exposure pathway, but it is unlikely to be significant. To be significant, there would need to be high concentrations of bioavailable metals in the sediments and children would have to routinely and frequently play in the stream and wetland areas, dirty their hands, and ingest the sediments through hand-to-mouth behaviors. Existing and planned sediment data are adequate for evaluating this pathway to humans. The EPA report in the work plan presents data from the wetland below the main embankment. The source of these sediments (Silver Creek or Richardson Flat impoundment tailings) is not certain.

Again similar to the situation with surface water, ecological receptor contact with site-related metals in sediments may be a complete exposure pathway, but risk cannot be evaluated at this time. Sediment sampling is currently planned for 6 locations in the diversion ditch. Understanding the ecological risk from these metals may require additional data collection but the additional data needs cannot be determined at this point As with all potential data collection, data needs will be balanced with considerations of their value versus the cost of remedial alternatives that could be conducted despite uncertainty about risk. Data needs may include a consideration of bioavailability since if the sediment metals were derived by precipitation from the water, they may have very low solubility and bioavailability.

2.2.4.13 Contact with Wind Deposited Tailings Off Site

Humans off-site and upland ecological receptors may be exposed to off-site winddeposited tailings, if the tailings have caused elevated levels of metals. No data currently exist to evaluate this. There currently are no residential areas in the suspected area of potential impact. This sampling plan describes a tiered, cost effective approach to sampling soils in the potentially affected areas; if elevated metals are detected above the levels designated in the sampling plan, then additional samples will be collected in areas further away until the affected area can be delineated.

2.2.4.14 Summary of Potentially Complete Exposure Pathways

Human receptors

- incidental ingestion of tailings outside the impoundment
- incidental ingestion of affected surface water
- incidental ingestion of affected sediment
- incidental ingestion of wind-deposited tailings off-site
- offsite use of groundwater
- direct contact with tailings mixed with soil on-site

Ecological receptors

- contact with tailings in the impoundment;
- contact with tailings outside the impoundment;
- contact with affected surface water:
- contact with affected sediment;
- contact with wind-deposited tailings off-site (off-site or highly mobile species only)
- food chain exposure to metals from food items exposed to the above exposure media

2.3 PROJECT/TASK DESCRIPTION (A6)

As summarized in the focused RI/FS Work Plan, extensive investigation work has already been completed at the Site. Moreover, over the years, United Park and others have taken actions to support final closure of the Site, including the installation of a soil cover over the tailings, drainage ditches, and a security fence. In order to evaluate the need for further remedial measures needed to support final Site closure and to assure that the existing remedies in place are adequate and have longevity, United Park will conduct the additional remedial investigation work described in Sections 5.0 to 5.7 of the focused RI/FS Work Plan.

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The following objectives of the remedial investigation are:

Soils and Tailings

- 1. Determine metal concentrations in imported cover soils, provide data of sufficient quality and quantity for analyzing risks to human health and for comparison with ecological screening levels and background. Verify extent (vertical and lateral) of imported cover soils on tailings.
- 2. Screen for impacts to off-impoundment soils and delineate the site boundary.

 Delineate all areas of potential impacts through the use of human and ecological screening levels.
- 3. Collect geochemical and physical data on tailings to evaluate long-term fate and chemical stability.
- 4. Delineate extent of tailings outside of impoundment and determine impacts to surface and ground water.
- <u>5.</u> Determine background soil metal concentrations.

Surface Water and Sediments

- 1. Collect sediment data in the south diversion ditch to aid in identification of the location of metal loading in the ditch. Use the data to determine long-term chemical fate and stability of metals in the sediments and in ecological and human health risk assessment.
- Collect data in Silver Creek and in drainages associated with the site to aid in determination of the background water quality relative to the site and the site's impact on water quality in Silver Creek, including seasonal variation. Provide additional surface water data for comparison with human health and ecological screening levels.

Groundwater

- 1. Determine metal concentrations in the Silver Creek shallow alluvial ground water both up and down stream of the impoundment.
- Collect surface and ground water elevation data to quantify the interaction between shallow ground water and Silver Creek, including seasonal variations.
- 3. Conduct a well survey to determine use of potentially affected groundwater.

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2.4 DATA QUALITY OBJECTIVES for MEASUREMENT DATA (A7)

Data Quality Objectives (DQOs) are quantitative and qualitative statements specifying the quality of the data required to support decisions during site investigations and removal actions. DQOs and associated data quality levels are based on the end uses of the data and are determined by the methods of analysis and the level of QC and documentation that are used to produce the data. Tables 1 and 4 summarize the DQOs, data uses, analytical methods, and QC level required for the sampling. All data collected during the focused RI/FS, except for decontamination water samples collected for pH testing in the field, will be considered "definitive" consistent with EPA Superfund Data Categories (EPA, 1993) and (Table 1). pH testing does not fall within the Superfund Data Categories (EPA, 1993), and for the purpose of this plan will be considered Screening. Screening and Definitive data are defined as follows:

- Screening data will be defined as data collected by Standard Operating Procedures
 (SOP's), such as water-level measurements and pH measurements, using field
 instruments calibrated according to manufacturer's specifications and SOP's. The
 data deliverables produced and QC documentation are not as rigorous as requirements
 for Definitive data. The data may be used for site monitoring and characterization.
- Definitive data will be defined as data produced using EPA-approved methods. The
 data deliverables may or may not be equivalent to Contract Laboratory Procedures
 (CLP) Routine Analytical Services (RAS). However, the data deliverables produced
 and the QC documentation may not be as rigorous as the CLP requirements for
 documentation. The data may be used for risk assessment, site characterization, and
 site monitoring.

DQOs are expressed in terms of precision, accuracy, representativeness, comparability, and completeness (PARCC). Table 3 summarizes the quality assurance goals in terms of the five PARCC criteria.

2.5 SPECIAL TRAINING REQUIREMENTS/CERTIFICATION (A8)

Besides OSHA training and certification for hazardous waste site workers (29 CFR 1910.120), no special training or certification requirements have been identified for this project.

2.6 DOCUMENTATION AND RECORDS (A9)

The following documentation and records will be maintained for the project:

Field Operation Records:

- Sample collection records (i.e., bound field notebooks)
- Chain-of-custody records
- QC sample records (information recorded in field notebook)
- Corrective action reports

Laboratory Records:

- Sample data
- Sample management records
- Test methods
- QA/QC reports.

Records will be maintained at UPCM offices for ten years after data collection.

3.0 MEASUREMENT / DATA ACQUISITION (B)

This SAP is intended to be a guide for United Park's Project Manager, field personnel and EPA's oversight contractor in implementing the remedial investigation at the Site. This section addresses all aspects of the collection and measurement systems design and implementation ensuring that appropriate methods for sampling, analysis, data handling, and QC are employed and documented.

A 2' topographic map is available. This digital map was prepared in the summer of 1998. All surveying to be completed for the focused RI/FS will be tied to the coordinate system for this map. Points will be established on the ground for referencing any hand-held GPS equipment used in the study.

3.1 SAMPLING PROCESS DESIGN (B1)

All sampling measurements described below are required to achieve the project objectives. The focus of sample collection activities proposed in this SAP is evaluation of the following environmental media:

- Surface water sampling of Silver Creek, the Diversion Ditch, and its tributaries, and the ponded water within the impoundment.
- Groundwater sampling for water quality and elevation in the existing wells and wells associated with the evaluation of the shallow aquifer in Silver Creek.
- Soils sampling off site to evaluate the migration of contaminants and determine background metal concentrations in soil.
- Sediment sampling in the Diversion Ditch to support the risk analysis, the study of the long-term viability of the wetlands in the ditch and to determine the contribution, if any, to the metal loading in surface waters flowing in the ditch.
- Samples of the soil cover over the tailings to determine depth of cover and any surface contamination.
- Tailings sampling to support the determination of the long-term chemical stability of the tailings.

A summary of sample media, locations, analytes and sampling objectives is provided in Table 5.

3.1.1 Surface Water

Surface water samples will be collected at a minimum of five (5) locations on or near the Site as depicted on Figure 4. The sample locations were selected based on data collected in 1999 and 2000. Data from these sample locations will be used to characterize seasonal water quality and quantity in the main Ditch, as well as the unnamed drainages flowing into the south diversion ditch and Silver Creek. Data from the unnamed drainages will provide limited background water chemistry data, the unnamed drainages only flow in response to snowmelt or significant storm events. Furthermore, the data will be used to determine the effect(s) of the Site on Silver Creek water chemistry and provide information in support of a study to determine the source of elevated zinc concentrations found in the middle reach of the diversion ditch. Samples will be collected monthly at each location through at least one complete seasonal time period. United Park intends to use data collected at these stations in 1999 to complete the annual cycle of data collection. The 1999 and 2000 data will be evaluated for QA/QC requirements found in

this document. If the 1999 and 2000 data do not pass QA/QC requirements, then additional data will be collected to fill in the data gaps.

The samples will be collected according to the RMC Standard Operating Procedures (SOP) for Surface Water Sampling SOP 1 presented in Appendix C. Field and laboratory analytical parameters are shown on Table 2 of this SAP. Sample collection will be conducted according to procedures in Section 3.2. Analytical and laboratory procedures that will be followed are detailed in Section 3.4 of this SAP. Actual sample locations and surface water elevation data will be logged with a Global Positioning Survey (GPS) unit or located with conventional survey methods.

3.1.2 Ground Water

United Park will install two monitoring wells in the Silver Creek shallow alluvial aquifer. The wells will be installed upgradient and downgradient of the Site as shown on Figure 4. Monitoring Well RT-11 shall be installed in the vicinity of Silver Creek to the west of the impoundment area. Monitoring Well RT-12 shall be placed in the vicinity of Sliver Creek downgradient from the impoundment area. Existing wells may be used to the extent possible. Groundwater and surface water levels will be measured at each location to determine flow characteristics between the surface and groundwater system. Samples will be collected monthly starting after installation and continuing for twelve (12) months. Water level and quality data from the wells will be used to verify the Site conceptual hydrogeologic model and determine what, if any, impacts to the shallow alluvial aquifer adjacent to Silver Creek are occurring from the Site. The hydrogeological model presented by Weston (Appendix A, focused RI/FS Workplan, 2000) shows that the shallow aquifers present near the Site are separated from the tailings aquifer by a 10-15 foot clay soil layer. Groundwater quality data to date only suggest that the monitoring wells on the downstream face of the main embankment contain elevated levels of metals and only in wells completed within the top six feet of the ground surface. Beyond seepage across the tailings embankment, there is no apparent hydraulic connection between groundwater stored in the tailings and underlying and adjacent shallow alluvial aquifer(s). (See Weston, 1999)

The monitoring wells will be installed according to the RMC SOP 3a (Ground Water Monitoring Well Installation) presented in Appendix C. Field and laboratory analytical

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between groundwater stored in the tailings and underlying and adjacent shallow alluvial aquifer(s). (See Weston, 1999)

The monitoring wells will be installed according to the RMC SOP 3a (Ground Water Monitoring Well Installation) presented in Appendix C. Field and laboratory analytical parameters are shown on Table 2. Groundwater samples collected from monitoring wells will be analyzed for total and dissolved metals as specified in Table 2. The proposed monitoring well locations are shown on Figure 4. Sample collection will be conducted according to procedures in Section 3.2; analytical and laboratory procedures that will be followed are detailed in Section 3.4 of this SAP. Section 3.2, Sampling Methods, describes the procedures that will direct sample collection. The well locations will be logged with a Global Positioning Survey (GPS) unit or located with conventional survey methods.

3.1.3 Onsite Soils Cover Sampling

At approximately forty-two (42) locations, soil samples will be collected on the tailings impoundment to determine: 1) the extent and thickness of the soil cover and 2) chemical characteristics of the surface soils. EPA will use this data, in the risk assessment process to evaluate the potential for impacts to human health and the environment from the Site soils. Sample locations are shown on Figure 5. Samples will be collected at the surface (0-2") at each location and initially analyzed for lead and arsenic to characterize the cover materials for potential human and ecological risks from exposure to the cover soils. All samples will be analyzed for lead and arsenic and 20% of all surface samples collected will be analyzed for RCRA metals including copper and zinc. All samples will be archived in the event that additional analyses are required. The thickness of the soil cover will be determined by excavating either by hand, core sampler or backhoe down to the soil/tailings interface. The interface will be visually verified at each location; the tailings are a characteristic grey color, sandy texture, while the soil cover is red-brown color and has a clayey texture. Approximately ten (10) samples will be collected just above the cover/tailings interface and submitted for laboratory analyses to verify the visual method. Table 2 lists the target metals for the depth of cover samples, target metals were selected based on results of the E&E air monitoring activities conducted in 1984. The surface sample (0-2") data will be used by EPA determine if the cover material presents a threat to human health or the environment. Based on the outcome of this sample event

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additional analyses may need to be performed to determine risk to human health or the environment.

On-site soil sample locations are shown on Figure 5. Sample collection will be conducted according to procedures in Section 3.2. Analytical and laboratory procedures are detailed in Section 3.4 of this SAP. Actual sample locations will be logged with a Global Positioning Survey (GPS) unit or located with conventional survey methods.

3.1.3.1 Off-Site Soils Cover Sampling

At approximately thirty-two (32) locations, soil samples will be collected along three transects, oriented perpendicular to the prevailing wind direction, to assess the extent and potential human health and/or environmental impacts from wind blown tailings. The prevailing wind direction is from the southeast as determined by EPA's contractor in the 1986 Air Sampling at Richardson Flat. A wind rose from the EPA Air Sampling Report is presented as Figure 7. Sample locations are shown on Figure 6. Samples will be collected at 0-2" and 1-6" depth intervals along the transects indicated on Figure 6. Data from this sampling effort will be used in the risk assessment process to evaluate if there is a threat to human health or the environment from exposure to off-site soils. If impacts to off-site soils are found additional sampling may need to be performed to further define the extent and nature of the impacts and to evaluate risk to human health or the environment.

Off-site soil sample locations are shown on Figure 6. Sample collection will be conducted according to procedures in Section 3.2. Analytical and laboratory procedures are detailed in Section 3.4 of this SAP. Actual sample locations will be logged with a Global Positioning Survey (GPS) unit or located with conventional survey methods.

3.1.4 Sediment

At approximately six (6) locations, sediment samples will be collected in the south diversion ditch. These samples will aid in the determination of the source of elevated zinc concentrations in water samples collected in 1999 and 2000. Sufficient sample will be taken for additional testing if desired. Long term fate and transport modeling of metals in the sediments (See, focused RI/FS Workplan, Table 3.4) will be performed to evaluate risk to the environment from metals bound in the organic substrates within the diversion ditch sediments. The samples will be collected at locations shown on Figure 5. At each location, samples will be collected at the surface and down to a depth of six (6)

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All sediment sample locations are shown on Figure 5. Sample collection will be conducted according to procedures in Section 3.2. Analytical and laboratory procedures are detailed in Section 3.4 of this SAP. Actual sample locations will be logged with a Global Positioning Survey (GPS) unit or located with conventional survey methods.

3.1.5 Tailings

At three (3) locations, samples of tailings will be collected within the impoundment. The purpose of this sample collection effort will be to collect data to evaluate the long-term fate of the metal in tailings and the chemical stability of the tailings. Presently, environmental data suggest that very little if any leaching of metals is occurring. The samples will be collected at locations shown on Figure 5. At each location, five (5) discrete samples will be collected at one (1) foot vertical increments, starting from the bottom of the cover over the tailings down to a depth of five (5) feet below the ground surface. Sample and analytical procedures will be consistent with sections 3.2 and 3.4 of this SAP. The discrete samples will be analyzed for metals in soils presented in Table 2. In addition, a composite sample comprising a split of each increment will be prepared and analyzed for acid/base potential in order to allow prediction of long-term geochemical characteristics of the tailings materials. The samples will either be collected by excavating a test pit with a backhoe or with direct push methods. The Project Manager, following discussions with the EPA Project Coordinator, will make this determination in the field.

To maximize visual observations of tailings, soils and the tailings/soils interface as well as maximize sample quantities, a backhoe will be used to dig test pits in selected locations. The test pit will enable site personnel to view the soils/tailings interface in a three-dimensional view. This will provide an understanding of the physical characteristics of the interface as well as provide information about the spatial configuration of the interface. Test pits will be excavated with minimal disturbance and shall not be excavated below the current water table. Excavated soils will be sorted and stockpiled adjacent to the test pit. Upon completion of sampling activities the test pit will be backfilled. To prevent soil mixing, each soil horizon will be backfilled with soils removed from that horizon. Soils will be compacted with the bucket of the backhoe during backfilling.

3.1.5.1 Delineation of Tailings South of the Diversion Ditch

The tailings outside of the impoundment have been covered with at least one to up to five feet of clean soil (See Section 4.2, focused RI/FS Workplan). The actual limit and extent of the tailings south of the diversion ditch will be identified using a combination of aerial photography review and investigative field methods. The approximate limits of these tailings are marked with a dashed green line as the "tailings outside of the impoundment" on Figure 5.0. The results of this investigation will aid in providing a definitive model of the extent of the tailings located south of the diversion ditch and to define study boundaries. Subsurface samples will be collected to determine: 1) the extent of tailings south of the south diversion ditch, 2) the thickness of soil cover on these tailings, and 3) whether these tailings are contributing to elevated zinc levels in the diversion ditch.

As described in Section 5.2 of the focused RI/FS Workplan (RMC, 2000), the purpose of this sampling effort is to evaluate the potential for tailings in this location to impact groundwater and surface water in the south diversion ditch. Data collected will be used in conjunction with the sediment samples collected from the Diversion Ditch. Subsurface samples will be collected using a combination of shallow hand tool excavation, backhoe test pits, boreholes or direct push methods. These four methods will involve the visual inspection of subsurface soils. To confirm the results of visual inspection, analytical samples will be collected at 10% of the locations visually inspected. The analytical soil samples will be collected above and below any color or texture changes. The soil samples will be analyzed for metals in soils presented in Table 2.

Two piezometers, designated RT-13 and RT-14, will be installed in the tailings outside of the impoundment. Figure 4.0 shows the locations, the piezometers will be completed in the tailings. To prevent the migration of water between the tailings and underlying native soils, borehole intervals below the tailings/native soil interface will be plugged with bentonite pellets. The borehole will be plugged to approximately one foot (1') above the tailings/native soils interface. The piezometers will be installed and monitored even if groundwater is not observed during drilling/sampling activities. The piezometers will be installed in specific areas to further define the hydraulic gradient that exists in any water that may be within the tailings outside the diversion ditch. The presence of water in these piezometers and the relationship that its' gradient has to the water in the diversion ditch will help identify areas of possible metal loading in the diversion ditch. Water quality data from these piezometers will not be collected. Water quality data is not as critical as the relative elevation data is in

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determining the potential metal loading to the diversion ditch that any tailings south of the diversion ditch may be causing. Based on the outcome of the groundwater evaluation, United Park may collect additional subsurface data to determine the volume, extent and environmental impacts from tailings in this area.

A review of historical aerial photographs will be conducted to assess the outermost limits of the tailings south of the diversion ditch. The approximate location of tailings will be determined from reviewing a series of historical aerial photographs. Where possible, the location of the tailings will be determined by examining the photographs for discontinuities that may be indicative of the boundaries of the tailings and native ground. These discontinuities may include changes in plant cover, drainage patterns and general geomorphology. The locations of the tailings/native ground boundary will be compared to the locations of known points such as fencing and roads. The boundary will then be staked on the ground using the known points as reference locations. The staked boundary locations will act as a starting point for the field delineation of the tailings/native ground boundary.

The extent of the tailings will be staked during the field investigation. At the completion of field activities, the boundary will be surveyed using conventional survey techniques. The survey data will be used to update the current boundary on Figure 5.

Sample location, collection and laboratory procedures that will be followed are detailed in Section 3.2 of this SAP.

Analytical soil sample collection will be conducted according to procedures in Section 3.2, analytical and laboratory procedures that will be followed are detailed in Section 3.4 of this SAP. Actual sample locations will be logged with a Global Positioning Survey (GPS) unit or located with conventional survey methods.

3.1.6 Background Soils Samples

At approximately twenty-five (25) locations, background soil samples will be collected in areas that have not been affected by tailings deposition. The samples will be used: 1) to determine the concentrations of metals in areas not affected by tailings deposition and 2) to obtain baseline, background values for soils metals concentrations in the vicinity of the tailings impoundment. Sample locations are shown on Figure 2.

3.1.6 Background Soils Samples

At approximately eleven (11) locations, background soil samples will be collected in areas that have not been affected by tailings deposition. The samples will be used: 1) to determine the concentrations of metals in areas not affected by tailings deposition and 2) to obtain baseline, background values for soils metals concentrations in the vicinity of the tailings impoundment. Sample locations are shown on Figure 2.

Discrete samples will be collected at the surface (0-2") at each location and initially analyzed for lead and arsenic to characterize the background concentrations of metals in the area surrounding the tailings impoundment. All samples will be analyzed for lead and arsenic and 20% of all surface samples collected will be analyzed for RCRA metals including copper and zinc.

Background soil sample locations are shown on Figure 2. Sample collection will be conducted according to procedures in Section 3.2 and RMC SOP 2a. Analytical and laboratory procedures are detailed in Section 3.4 of this SAP. Actual sample locations will be logged with a Global Positioning Survey (GPS) unit or located with conventional survey methods.

3.2 SAMPLING METHOD REQUIREMENTS (B2)

Sampling method requirements for each sample type are described below. Table 4 summarizes the sample containers, preservation requirements, and analytical holding times for each analytical method.

3.2.1 Surface Water Samples

Surface water samples will be collected following RMC's SOP 1, included in this SAP as Appendix C. These procedures will be followed for any water samples collected in this focused RI/FS. Water samples will be collected at five (5) locations on or near the Site, as shown on Figure 4. Analytical parameters vary depending on the data objective at each location. For the most part, complete cation/anion and metals (dissolved and total) samples will be collected. Table 1 presents the data quality objectives. Table 2 presents the parameters, analytical methods, laboratory methods, container types, preservation requirements and holding times for each type of analyte. Table 5 presents the analytes to

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3.2.2 Ground Water Samples

Groundwater quality samples will be collected at two (2) new monitoring wells installed in the Silver Creek alluvial aquifer and two monitoring wells installed in the tailings south of the diversion ditch. Additional data collected will include field parameters (as with surface water samples) as well as water elevation in the well. Elevations will be referenced to the ground surface adjacent to the well. Figure 4 shows the approximate locations of the monitoring wells. In addition, sampling of existing wells will occur in accordance with this RMC SOP 3c and this SAP. The wells will be installed according to SOP 3a. Prior to sampling, the wells will be developed according to SOP 3b. The wells will be sampled according to RMC SOP 3c. Table 1 presents the data quality objectives. Table 2 presents the parameters, analytical methods, laboratory methods, container types, preservation requirements and holding times for each type of analyte. Table 5 presents the analytes to be collected at each location. At each location, field data, including pH, temperature and specific conductivity will be collected after the laboratory sample has been collected, preserved and stored.

After well installation, a licensed land surveyor registered in the State of Utah will survey the locations of the new wells. Surveying will include horizontal coordinates (Northing and Easting) for the well location, and vertical elevation datum for the top of casing. Vertical accuracy will be within 0.01 feet and horizontal accuracy will be within one (1) foot.

One existing groundwater well, No. RF-1, will be sampled and then closed in accordance with standard procedures for abandoning and closing wells as set forth by the State Engineers Office.

3.2.3 Soils

All soil samples collected shall be analyzed on a dry weight basis. Surface soil samples (samples collected at a depth of 0-2") collected for human health and ecological risk assessment purposes will be submitted as bulk samples to the laboratory. The laboratory will split the sample into two equal portions. One sample will be archived by the laboratory, and the second half of the bulk sample will be analyzed. Since human health risk is expected to be low or absent, the samples will not be sieved to <250 microns (per discussions with Susan Griffin, EPA, January 8, 2001.)

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will split the sample into two equal portions. One sample will be archived by the laboratory, and the second half of the bulk sample will be analyzed. Since there is no residential use or other significant human exposure to site soils, human health risk is now (and will remain) low, and therefore the samples will not be sieved to <250 microns (per discussions with Susan Griffin, EPA, January 8, 2001.)

3.2.3.1 Surface Soil Samples

Surface soils (0-2") will be collected following RMC's SOP 2a, included in this SAP in Appendix C. The surface of the soil will be scraped free of vegetation from the sample location with a shovel, stainless steel spoon, or disposable sampling instrument. The underlying soil sample will then be collected with a stainless steel spoon or gloved hand and placed into a plastic bag, labeled and sealed with a chain-of-custody seal. Coarse-grained soils such as gravel and rock fragments will be discarded. The sample location will be surveyed with either a GPS unit or a licensed land surveyor.

3.2.3.2 Soil Samples at Depth

Samples collected at depth will follow SOP 2b and/or SOP 2c, included in this SAP in Appendix C. Surface vegetation will be scraped away from the sample location with a shovel, stainless steel spoon, or disposable sampling instrument. The target depth increment sample will be collected by one of the following methods: hand-powered auger, soil probe, shovel, or stainless steel trowel. For tailings samples, equipment such as a backhoe or geoprobe will be used as appropriate. This equipment will be operated by a professional operator and arranged for by the Project Manager. All appropriate safety precautions will be taken when working around this equipment. At each increment, the sample will be placed into a glass jar or plastic bag, labeled and sealed with a chain-of-custody seal. The sampling equipment will be decontaminated between each depth increment. If the sampling is to evaluate the cover material over the existing tailings, then the character (color and texture) of the sediment encountered will be noted. The samples will be analyzed in accordance with section 3.4 of this SAP.

This section also applies to sampling of tailings below ground surface. When tailings are encountered for sampling, the resulting exposed tailings will be covered with soil, not tailings. The objective is to not have tailings exposed after sampling.

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will be collected using a hand auger, shovel, or geoprobe methods. All samples will be analyzed as bulk samples.

3.2.5 Background Soil Samples

Background soil samples will be collected in accordance with RMC SOP 2a, included in Appendix C. At each of the locations shown on Figure 2, a discrete sample will be collected from a depth of zero to two inches (0-2") below the ground surface. Any surficial organic matter and debris will be removed prior to sample collection.

3.3 SAMPLE LABELING, HANDLING, CUSTODY, & DOCUMENTATION (B3)

3.3.1 Labeling

Each sample container will be immediately labeled with the following information:

- Project name
- Project number
- Sample identification and location
- Date and time collected
- Analyses requested
- Sampler's initials

Samples will also be labeled as to whether or not they contain any preservatives. Also, their origin will be noted. For example, the sample will be labeled as to whether or not it is tailings, sediment, soils surface or ground water.

Duplicate samples will be given a sample identification number in the same fashion as normal samples such that the laboratory cannot distinguish them as duplicates.

3.3.2 Sample Handling

All samples will be collected in appropriate containers supplied by the laboratory as specified in Table 4, and promptly placed in an iced cooler to maintain a temperature of 4 °C. Typically, samples selected for chemical analysis are delivered at the end of each day to the analytical laboratory. If they are not submitted to the laboratory on the same day collected, they will be refrigerated in a locked sample storage room at RMC's office until delivery to the laboratory.

3.3.3 Chain-of-Custody

Chain-of-custody procedures will be observed and documented. Chain-of-custody documentation will begin in the field for each sample submitted to the laboratory and will also be maintained by laboratory personnel. A chain-of-custody form (Appendix D) will be completed and will accompany each sample cooler to the analytical laboratory.

Sample custody means that all samples will remain in the possession or observation of the sampler at all times, or in a locked facility until delivery to the analytical laboratory.

3.3.4 Documentation

The field personnel will maintain a weather resistant, hardbound sample logbook. The logbook will include the sample identification, sample date, type of sample, and analyses requested. Information specific to the type of sample will also be included, as follows:

- Confirmation Soil Samples Include the sample number and location to confirm the surveyed location.
- Decontamination Fluid Include the quantity of water in the sample batch and the type of sampling equipment decontaminated.

3.4 ANALYTICAL METHOD REQUIREMENTS (B4)

Analytical methods, with corresponding laboratory reporting limits (LRLs) are specified on Tables 2 and 2b. Samples will be submitted to American Environmental Consultants Laboratory (AEC) in Salt Lake City, Utah. AEC is certified with the State of Utah. Appendix E contains AEC's QA/QC manual, and certification letters from the Utah Department of Health and Division Bureau of Laboratory Improvement. If another lab performs analyses, it must meet the following criteria and submit all QA documentation to the EPA for approval as described above:

- Demonstrated ability to achieve the required detection limits,
- Certified by the State of Utah, and
- Follows an internal QA/QC program.

If contradictions between the laboratory QA/QC manuals or other documents are identified, information in this SAP supersedes all other documents.

3.5 QUALITY CONTROL REQUIREMENTS (B5)

Quality control will include collecting field duplicates at a rate of 10 percent of the sample load for each sample type, and ensuring that the laboratory runs matrix

spike/matrix spike duplicates at a rate of five percent of the sample load for each sample type. The field duplicates will be submitted "blind" to the sample laboratory, i.e., they will be given a separate sample identification number from the environmental sample, unidentifiable to the laboratory, as described above. Field duplicates will be run for the same analytical suite as the environmental samples.

Matrix spikes and matrix spike duplicates will be selected at random by the laboratory. Separate samples do not need to be collected in the field. The laboratory will perform and report all analyses under QA/QC procedures that include the results of method blanks, laboratory control samples, matrix spikes, matrix spike duplicates, and the relative percent difference between the matrix spike sample and duplicate.

Equipment rinsate blanks will be collected during the decontamination process. Equipment rinsate blanks will consist of distilled water rinsate blanks collected from a post-decontamination rinse of soil sampling tools (spoons, shovels, split-spoons, mixing bowls, etc.). At least one equipment rinsate sample will be collected per every twenty field sampling locations or 1 per day, whichever is greater.

Due to the nature of the contaminants at this site, ambient, equipment and trip blanks will not be collected.

3.6 INSTRUMENT/EQUIPMENT TESTING, INSPECTION, AND MAINTENANCE REQUIREMENTS (B6)

All instruments and equipment will be regularly tested, inspected, and maintained according to manufacturers' instructions. Field equipment will be tested and inspected daily before use. Any equipment found to be not functioning properly will be repaired or replaced. Laboratory equipment will be tested, inspected and maintained in accordance with the laboratory QA/QC manual and manufacturers' recommendations.

3.7 INSTRUMENT CALIBRATION & FREQUENCY (B7)

3.7.1 Field Instruments

RMC will follow the manufacturer's specifications to calibrate any field equipment prior to each use. These manufacturers specifications are included in RMC's SOP's. A record of the calibration will be kept in the field logbook.

3.7.2 Laboratory Equipment

Procedures and schedules for the calibration of laboratory equipment are described in the appropriate SW-846 and EPA methods, and in the laboratory's Quality Assurance Plan. These procedures and schedules will be followed for all laboratory work.

3.8 DATA MANAGEMENT (B10)

Data from AEC Laboratory and Frontier Geosciences will be submitted to UPCM in both hard copy and electronic form. To avoid transcription errors, report tables will be prepared directly from the electronic submittals.

4.0 ASSESSMENT / OVERSIGHT (C)

4.1 Assessments and Response Actions (C1)

This section describes the number, frequency, and type of assessment activities needed for this project. Assessments coordinated by the Project QA Officer will include: (1) a readiness review prior to initiating each major phase of field work; (2) surveillance during representative phases of the project; (3) a technical systems audit (TSA) conducted toward the end of the first week of field work; and (4) a data quality assessment (DQA).

The readiness review will be conducted with both the field staff and analytical laboratories as a technical check to determine if the staff, subcontractors, equipment, and record keeping systems are in place to start work in accordance with this QAPP. At the review, the QA Officer will review the project objectives, methodologies, record keeping requirements, and schedule with the field team and laboratories to make sure they are

familiar and prepared to meet project requirements. The QA Officer will make sure all systems are ready before field work is initiated.

Surveillance will include weekly reviews of project progress and compliance with QAPP requirements. The project QA Officer will visit the field teams at the Site and observe their work habits and review project records. Based on the surveillance results, the QA Officer may propose corrective actions or changes to the field methods to the Project Manager.

A TSA will be conducted about halfway through the field portion of the project. The TSA is a thorough and systematic on-site qualitative audit where facilities, equipment, personnel, training, procedures, and record keeping and is conducted to determine conformance to the QAPP.

The DQA will be conducted to determine whether the data meet the assumptions that the DQOs and data collection design were developed under and whether the total error in the data are tolerable. This assessment activity will include complete data verification and validation as described in Section 5.0. Guidance for the Data Quality Assessment Process (EPA QA/G-9) will be consulted.

The QA Officer will report results of the assessment activities directly to the Project Manager who, with the assistance of the QA Officer, will be responsible for implementing any necessary corrective actions. The occurrence and resolution of major quality issues identified during assessment activities will be documented in memorandum to UPCM, the EPA Project Manager Jim Christiansen, and the UDEQ Project Manager Muhammad Slam.

5.0 DATA VALIDATION AND USABILITY (D)

5.1 Data Review, Validation & Verification Requirements (D1)

The data validation process evaluates whether the specific requirements for an intended use have been fulfilled and ensures that the results conform to the users needs. The data validation process develops the QC acceptance criteria or performance criteria.

Data verification confirms that the requirements of the specified sampling and analytical methods were followed. This process involves reviewing the results of sampling and

analysis to determine conformance with the QC requirements described for the project. The data verification process ensures the accuracy of data by using validated methods and protocols, and is often based on comparison with reference standards.

Requirements and methods for data validation and verification are listed in Tables 3, 4, 4b and 4c.

5.2 Validation & Verification Methods (D2)

Data will be reviewed to ensure that the requirements stated in Tables 3 and 4 were met. Data validation and verification will be conducted using the methods described in Table 4. Superfund's working definitions for data verification and validation are as follows:

<u>Data Validation</u>: A consistent, systematic process that determines whether the data have been collected in accordance to the specification as listed in the approved QAPP. The process is independent of data validation and is conducted at various levels both internal and external to the data generator (laboratory).

<u>Data Validation</u>: An evaluation of the technical usability of the verified data with respect to planned objectives. Data validation is performed external to the data generator (laboratory), using a defined set of performance criteria to a body of data in the evaluation process. This may include checks on some or all of the calculations in the data set and reconstruction of some or all final reported data from initial laboratory data (e.g., chromatograms, instrument printouts). It is in the data validation process that data qualifiers for each verified data are evaluated. It extends beyond the analytical method to protocols or QAPPs to address the overall technical usability of the generated data.

One hundred percent (100%) of the data will be validated according to Table 4 requirements by the Project QA Officer or a subcontractor experienced in conducting this type of data verification. Data will be reviewed as it is received, continuously throughout the project. If problems are uncovered as a result of the validation effort, the QA Officer and Project Manager will be immediately notified. The QA Officer or Project Manager will discuss possible corrective actions with the laboratory prior to implementation. The

Project Manager will immediately notify EPA and UDEQ of any data verification or validation issues that may affect the success of the project.

Any deviations from the analytical control limits specified in Table 3 and 4 will be evaluated in terms of their effect on the data usability. Data usability will be assessed using the National Functional Guidelines for Data Review (Inorganic & Organic, February 1994). The completeness goal for the project is 90 percent valid data.

The results of the data validation and verification will be summarized in a Data Review Report, to be prepared after the completion of sampling and analysis activities at the site.

5.3 RECONCILIATION WITH DATA QUALITY OBJECTIVES (D3)

The data validation and verification results will be compared to the DQOs stated in Table 1 and with the PARCC parameters described in Table 3. This evaluation will summarize the QA/QC performance by PARCC criteria including completeness calculations expressing the percent complete of valid data compared to the total number of samples collected. The result of the data validation and verification will be summarized in the Data Review Report described above.

References

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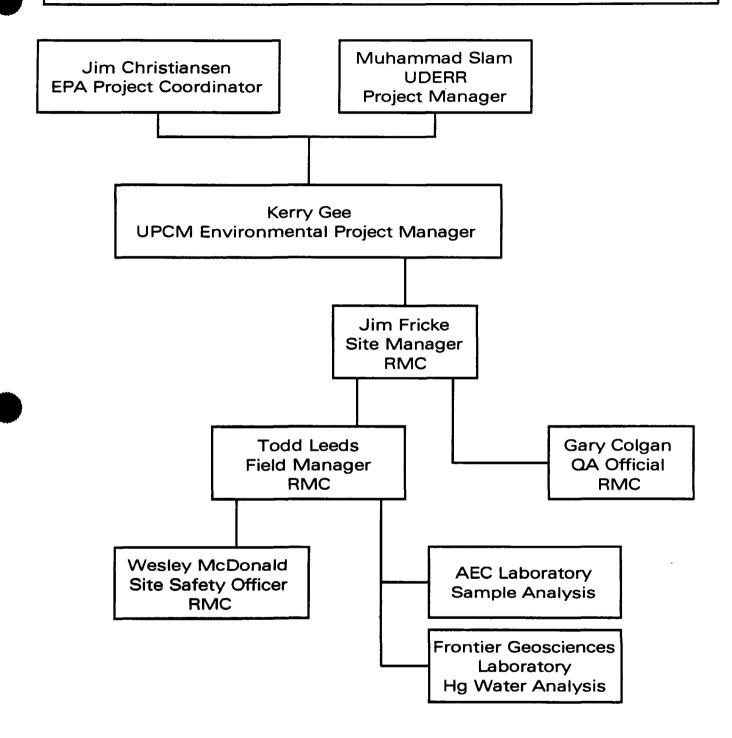
Weston, Inc., 1996, Baseline Risk Assessment, Murray Smelter.

Tabbed Page: Figures

FIGURES

Organizational Chart
Geographic Location of Site
Site Map
Surface Water and Ground Water Sample Locations
Soil/Sediment/Tailings Sample Locations
Offsite Soil Sample Locations
Wind Rose
Conceptual Site Model

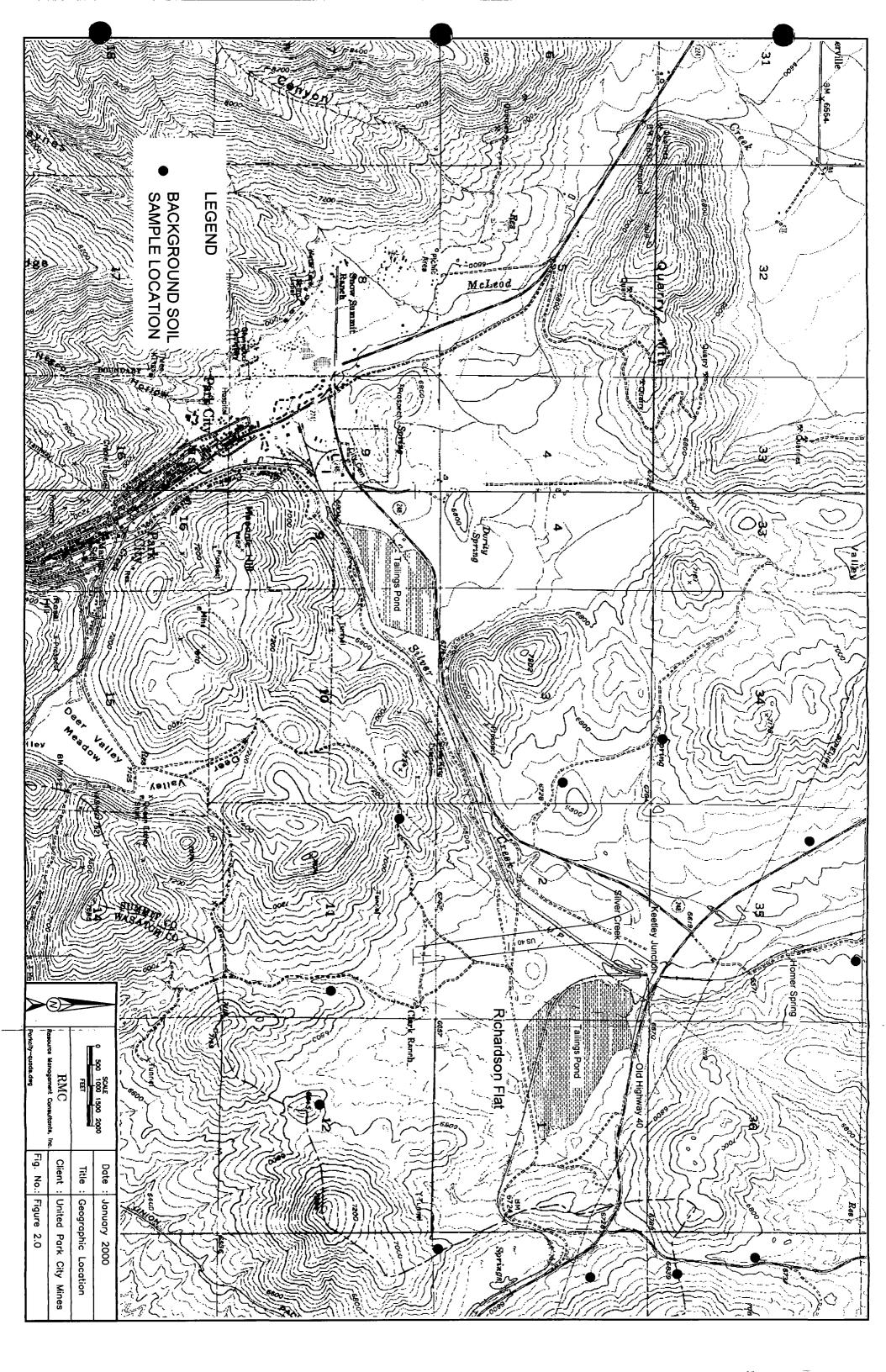
FIGURE 1 - Richardson Flat RI/FS Organizational Chart

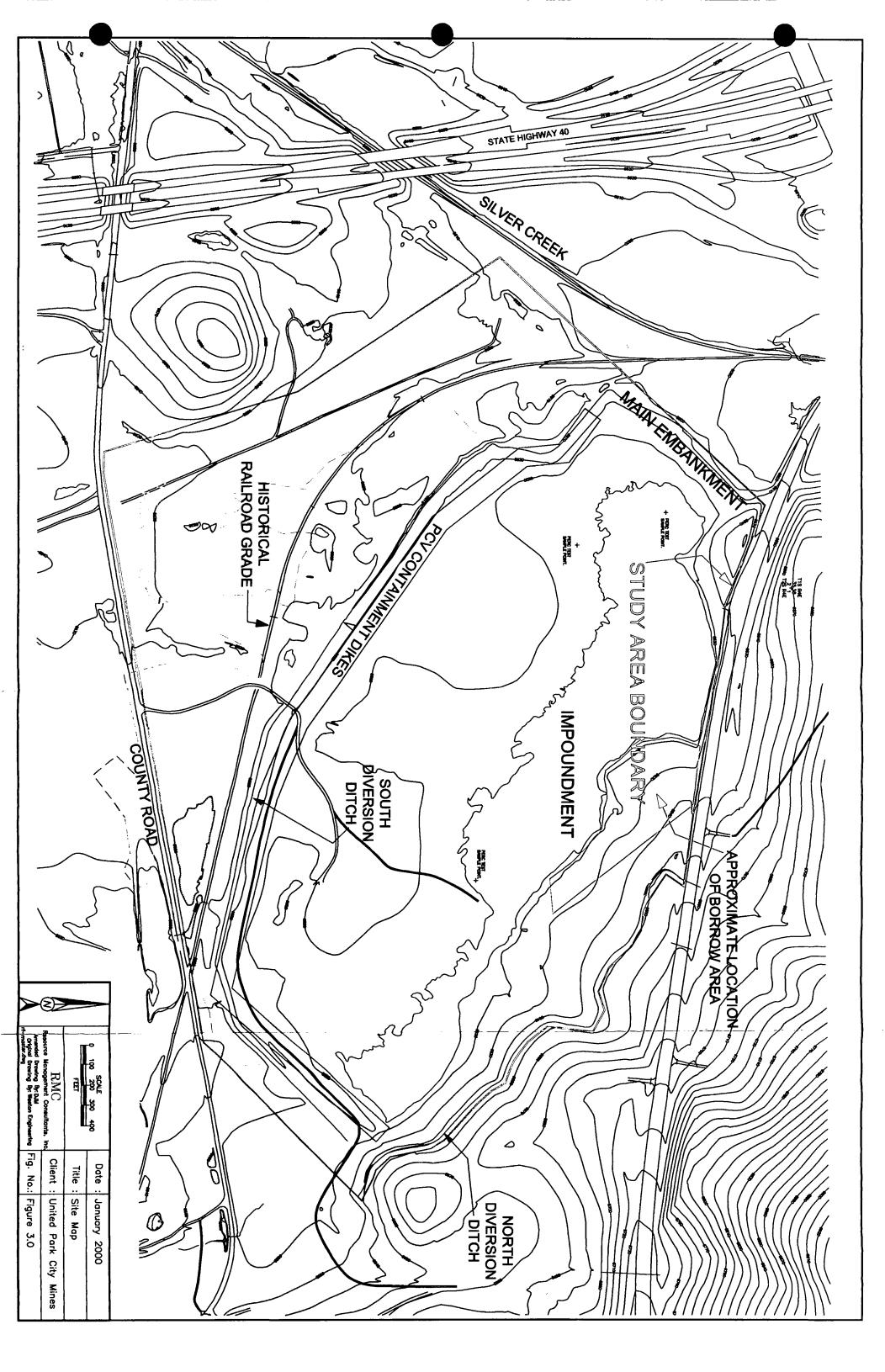


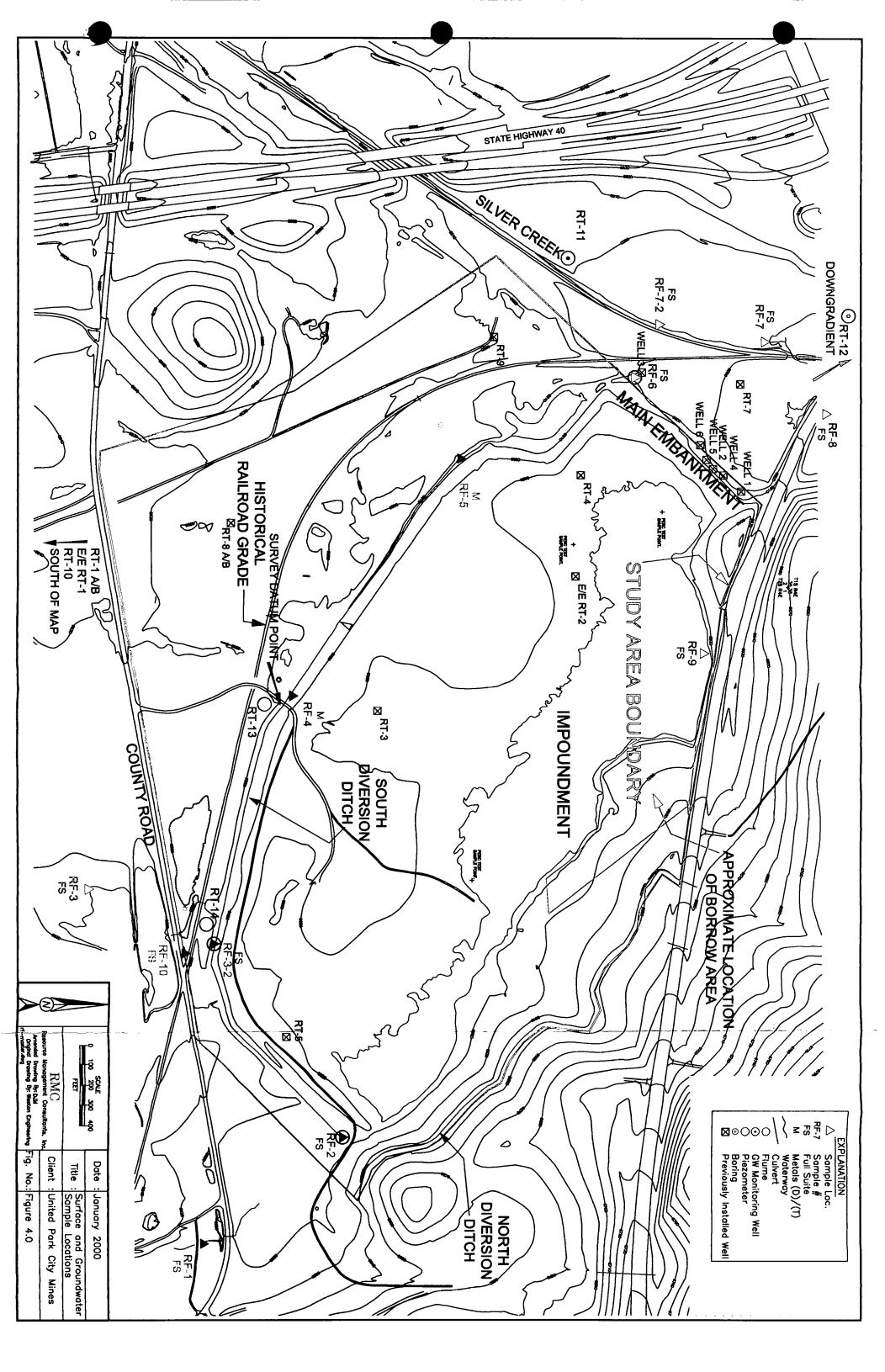
Color Map(s)

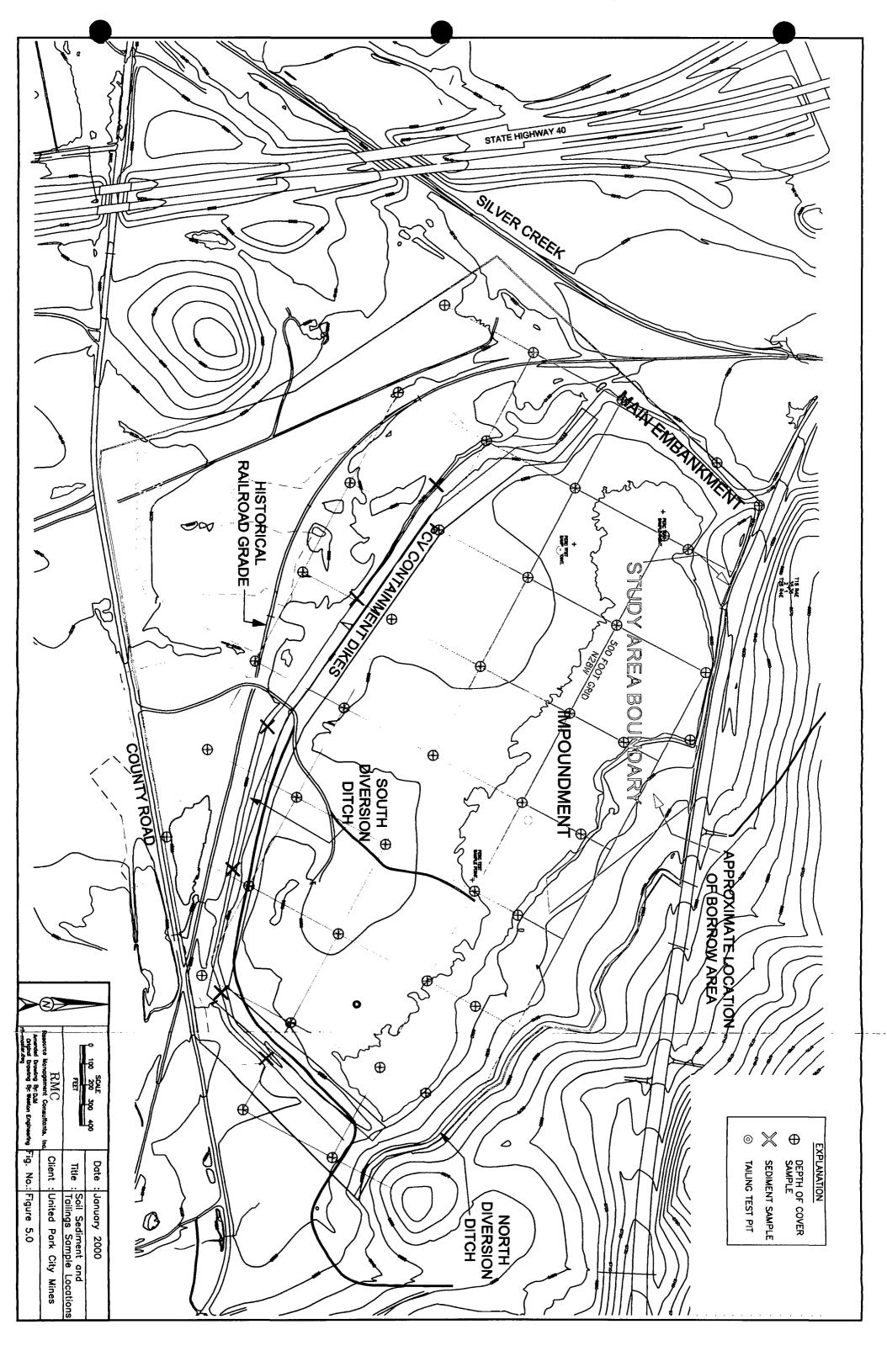
The following pages contain color that does not appear in the scanned images.

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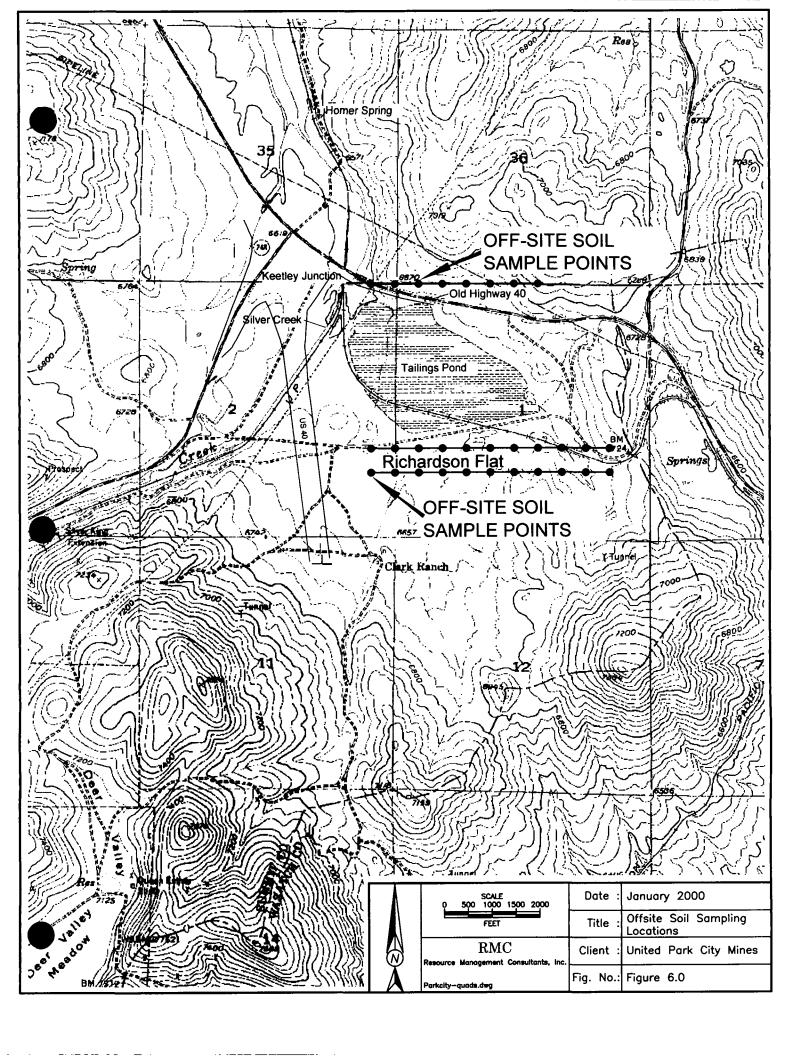
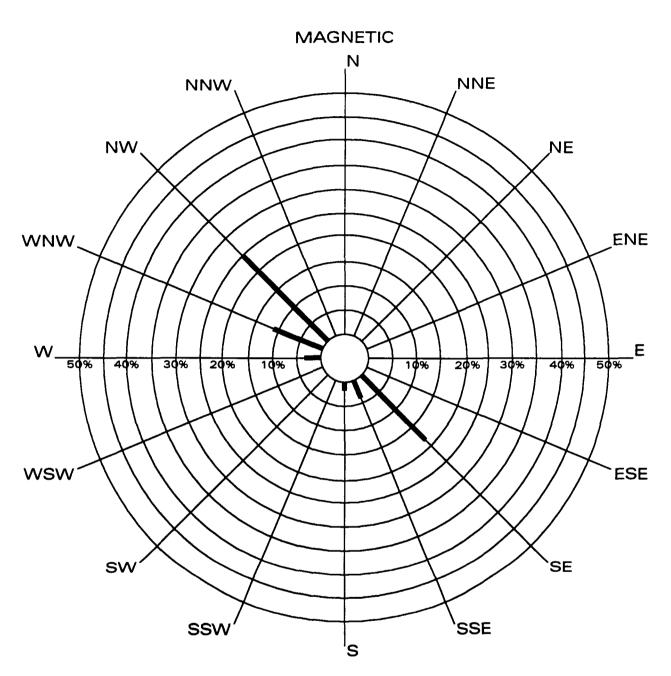


FIGURE 7 - Richardson Flat Wind Rose



DATA COLLECTION DATE: JULY 12, 1986 21.4% CALM FIGURE 8A RICHARDSON FLAT CONCEPTUAL SITE MODEL

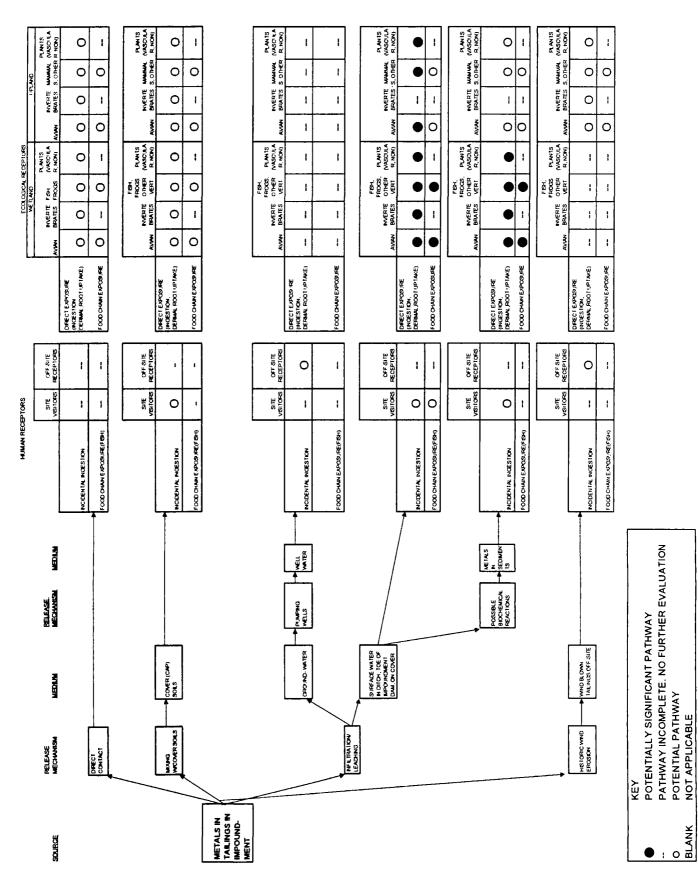
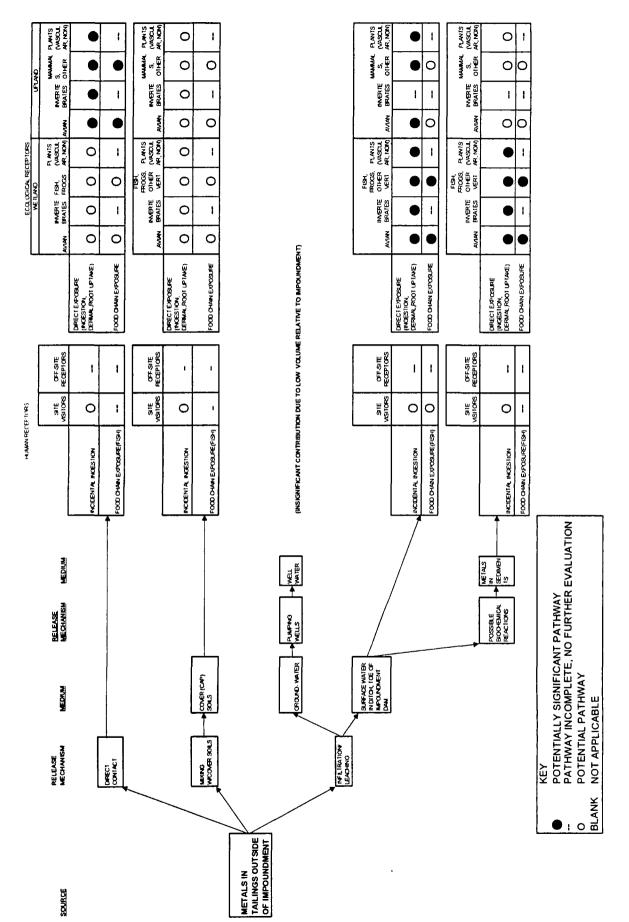


FIGURE 8B RICHARDSON FLAT CONCEPTUAL SITE MODEL



Tabbed Page: Tables

TABLES

Data Quality Objectives, Uses, Types and QC Levels

Sample Collection Guide – Target Analytes and Collection Requirements

Rationale for Proposed Laboratory Reporting Limits

PARCC

Data Validation and Verification Requirements

Data Flagging Convention for Metals Analyses

Data Flagging Convention for Ion Chromatography and Wet Chemistry

Sample Locations, Media and Target Analytes

Tane 1
Data Quality Objectives, Data Uses, Data Type, and QC levels
Richardson Flat
Sampling and Analysis Plan

Data Quality Objectives	Conceptual Site Model Exposure Pathway or Other Evaluation	Existing Data Summary	Data Use	Data Needs	Analysis Type	QC Level
Background Soil Sampling	Comparison of pattways exposures to natural background	Limited soil data from off-site, upland areas; water data from nearby areas	off-site, Evaluate whether potential risk tata from exceeds existing, naturally occurring background exposures	Sampling of natural surface soils in area surrounding site	Soil analysis	Definitive
Biological assessment of tallings and vicinity	Identification of ecological receptors of concern	Highway species survey, county surveys	Problem formulation phase of ecological risk assessment	Survey of typical, endangered, Site-specific field survey and threatened species on site conducted by trained and in affected off-site areas biologists	Site-specific field survey conducted by trained biologists	Definitive
Sample tailings on and near Direct contact with tailings impoundment outside of the impoundmen	Direct contact with tailings outside of the impoundment	Lacking	Evaluate human health, upland ecological receptor risk	Metals concentrations, aerial extent of covered and uncovered tailings	Soil (tailings) samples	Definitive
Surface water sampling	Outside-impoundment tailings contribution to surface water	Preliminary	Evaluate source contribution for remediation purposes. Augment previous data.	Water quality analysis, pH Measurement	Water quality analysis pH Measurement	Definitive Screening
Tailings soil cover sampling	Mixing of cover soils with tailings	Lacking although cover soils were obtained from clean sources	Evaluate potential risk, particularly ecological risk. Evaluate existing soil cover.	Inspect cover for potential mixing: Sample cover soils in thin areas	Soil analysis	Definitive
Groundwater sampling and assessment	Contact with groundwater by frumans or through seeps or surfacing of groundwater	Groundwater report is forthcoming, existing data do not indicate potential for vertical migration of metals or impacts on off-site water wells	Evaluate potential human exposure; Confirm lack of elevated metals concentrations in seeps and ' surface water	Off-site well surveys	Investigation	Definitive
Surface water sampling	Contact with surface water	Water quality data for diversion ditch and Silver Creek	Evaluate wetland processes that remove metals from water	Water quality analysis, pH Measurement	Water quality analysis pH measurement	Definitive Screening
Sample sediment in south diversion ditch	Contact with sediments	Metals data for sediments in wetland	If sediment concentrations exceed ecological risk screening levels additional data may be needed	Obtain sediment samples from diversion ditch	Soil metals analysis (dry weight) of surficial sediments and sediments in biotic zone	Definitive
Sample offsite tailings	Contact with wind-blown tailings	Limited to knowledge that dust storms occurred	Determine extent of windblown tailings and resulting soil metals concentrations	Tiered sampling plan to delineate extent and estimate concentrations	Soil analysis of surface soil samples	Definitive

Sample Collection Guide - Target Analytes and Collection Requirements Sample and Analysis Plan Richardson Flat

Hold Days 8 8 82 Preservative **\$**\$\$\$ इ इ इ 4 02. 4 oz. 4 oz. 4 oz. Glass Jar or Plastic Bag Container 0.1 ppm 0.5 ppm 5 ppm 0.10% SW-846 6010B SW-846 6010B SW-846 7471 Ag.Al,As,Cr,Cu,Fe,Pb,Sb,Se,Zn Moisture Content Mercury \mathfrak{Z}

Hold Days 쭁 8 8 8 8 8 2 ml HNO₃ (ph<2) None None H₂SO₄ None H₂SO₄ 2 ml HNO₃ (ph<2) Preservative H₂SO₄ None None Temperature Ş ဂ္ဂ 5 Ş န 5 ဦ Volume Bottle 1,2 Bortle 1,2 Bottle 1,2 Bottle 1,2 Bottle 1,2 Bottle 1,2 Bottle 3 Bottle 3 Bottle 4 Bottle 1,2 Bottle 1,2 Bottle 3 Bottle 1,2 Bottle 4 Bottle 3 Bottle 5 Bottle 5 Polyethylene Polyethylene Polyethylene Polyethylene **Polyethylene** Polyethylene **Polyethylene** Container 0.0002 0.00000 0.005 0.004 ĒŽ 0.00 0.05 0.0 0.0 9 0. 6020 SW-846 6010B or SW-846 6010B or 6020 SW-846 6010B or 6020 SW-846 60108 or 6020 SW-846 6010B or 6020 SW-846 6010B or SW-846 6010B or SW-846 9036 2340B⁵ Calculation EPA 160.2 EPA 160.1 SW-846 6010 EPA 325.2 EPA 353.2 Digital Meter 6020 EPA 245.1 **EPA 310.1** EPA 1631B **EPA 365.4 EPA 310.1** Method **EPA** 350.1 EPA 150. Ag, As, Cu, Mn, Pb, Sb, (Total & Hardness Cation/Anlon Balance TSS TDS Cd (Total & Dissolved) Se (Total & Dissolved) Cr (Total & Dissolved) Zn (Total & Dissolved) Fe (Total & Dissolved) Hg (Total & Dissolved) Hg (Total & Dissolved) Al (Total & Dissolved) Ca, K. Mg, Na Parameters NO₃, NO₂ Conductivity Dissolved) S F S NH₃ Total P Alkalinity SO.

LRL - Laboratory Reporting Umit

Flekt Data Collected for each sample statlon/event includes:

pH, Conductivity, Temperature, Flow
² All units are Parts Per Million (ppm) based upon dry weight unless otherwise noted.

³ All unhs in mg/l except as noted. *Laboratory analysis for the above parameters will require collection of the following sample volumes/preservation at each sample station:

Bottle 1 - 500 mt bottle filtered to 0.45 µm and preserved with 2 mi HNO3 Bottle 2 - 500 ml bottle unfiltered and preserved with 2 ml HNO₃

Bottle 3 - 1000 mi bottle unfiltered and unpreserved Bottle 4 - 500 ml bottle unfiltered and preserved with 2 ml H₂SO₄ Bottle 5 - 500 ml bottle unfiltered and unpreserved for field parameters.

Standard Methods, 20th edition

⁶Laboratory will measure the temperature of each cooler upon receipt to ensure proper temperature was maintained (4°C +/- 2°C)

Sample Collection Guide - Target Analytes and Collection Requirements Sample and Analysis Plan Richardson Flat

Parameters	Method	LRL ²	Container	, eum	Temperature	Preservative	Hold Days
Ag,AJ,As,Cr,Cu,Fe,Pb,Sb,Se,Zn	SW-846 6010B	2 ppm	Glass Jar or Plastic Bag	4 02.	2. †	ΝΑ	180
පි	SW-846 6010B	0.5 ppm	Glass Jar or Plastic Bag	4 02.	4.0	NA	£ 08;
Mercury	SW-846 7471	0.1 ppm	Glass Jar or Plastic Bag	40.	4. C	N/A	28
Molsture Content		0.10%	Glass Jar or Plastic Bag	4 02.	J. *	¥	

Parameters 1	Method	LRL	Container	Volume*	Temperature	Preservative	Hold Days
pH (Field)	EPA 150.1	¥	Polyethylene	Bottle 5	5. †	None	-
Ag.As,Cd.Cu,Mn,Pb,Sb,Se (Total & Dissolved)	SW-846 6010B or 6020	0.005	Polyethylene	Bottle 1,2	4. C	2 ml HNO ₃ (ph<2)	190
Cd (Total & Dissolved)	SW-846 6010B or 6020	0.001	Polyethylene	Bottle 1,2	7.	2 ml HNO ₃ (ph<2)	180
Se (Total & Dissolved)	SW-846 6010B or 6020	0.004	Polyethylene	Bottle 1,2	9.7	2 mt HNO ₃ (ph<2)	180
Cr (Total & Dissolved)	SW-846 6010B or 6020	0.01	Polyethylene	Bottle 1,2	0. ₽	2 ml HNO ₃ (ph<2)	92
A (Total & Dissolved)	SW-846 6010B or 6020	0.05	Polyethylene	Bottle 1,2	9. ₹	2 mt HNO ₃ (ph<2)	96
Zn (Total & Dissolved)	SW-846 6010B or 6020	0.01	Polyethylene	Bottle 1,2	9.	2 ml HNO ₃ (ph<2)	180
Fe (Total & Dissolved)	SW-846 6010B or 6020	0.1	Polyethylene	Bottle 1,2	5	2 ml HNO ₃ (ph<2)	9 <u>5</u>
Hg (Total & Dissolved)	EPA 245.1	0.0002	Polyethylene	Bottle 1,2	2.4	2 ml HNO ₃ (ph<2)	28
Hg (Total & Dissolved)	EPA 1631B	0.000001	Polyethytene	Bottle 1,2	J. *	2 mt HNO ₃ (ph<2)	82
Ca, K, Mg, Na	SW-846 6010B	2	Polyethylene	Bottle 2	4. C	None	180
٥	EPA 325.2	1	Polyethylene	Bottle 3	0. *	None	28
NO ₃ , NO ₂	EPA 353.2	0.1	Potyethylene	Bottle 4	0. ₩	H ₂ SO ₄	58
CO, HCO,	EPA 310.1	-	Potyethylene	Bottle 3	0.	None	7
NH,	EPA 350.1	0.1	Polyethylene	Bottle 4	0. †	H ² SO ⁴	28
Total P	EPA 365.4	0.1	Polyethylene	Bottle 4	0.	H ₂ SO,	28
*os	9£06 978-MS	2	Polyethylene	Bottle 3	∵ •	None	28
Alkalinity	EPA 310.1	1	Polyethylene	Bottle 3	7.7	None	14
Conductivity	Digital Meter	10 uS	Polyethylene	Bottle 5	4.C	None	28
Hardness	2340B ⁵	N/A	Polyethylene	Bottle 3	0.	None	180
Cation/Anion Balance	Calculation	N/A	Polyethylene	N/A	J.+	None	¥
TSS	EPA 160.2	1	Polyethylene	Воще з	4.C	None	7
TDS	EPA 160.1	10	Polyethylene	Bottle 3	4.C	None	7

Water

N/A - Not Applicable LRL - Laboratory Reporting Limit ^I Field Data Collected for each sample station/event includes: pH, Conductivity, Temperature, Flow

² All units are Parts Per Million (ppm) based upon dry weight unless otherwise noted.

All units in mg/l except as noted.

Standard Methods, 20th edition

⁶Laboratory will measure the temperature of each cooler upon receipt to ensure proper temperature was maintained (4°C +/- 2°C)

Laboratory analysis for the above parameters will require collection of the following sample volumes/preservation at each sample station:

Bottle 1 - 500 mi bottle filtered to 0.45 mm and preserved with 2 mi HNO₂.

Bottle 2 - 500 mi bottle unfiltered and preserved with 2 mi HNO₃.

Bottle 3 - 1000 mi bottle unfiltered and unpreserved.

Bottle 4 - 500 mi bottle unfiltered and preserved with 2 mi H₂SO₄.

Bottle 6 - 500 mi bottle unfiltered and unpreserved for field parameters.



A SW-8 A SW-8 A SW-8 A SW-8 C SW-8	Method SW-846 6010B SW-846 6010B	Consensus-Based Threshold Effect	Consensus-Based			
	46 6010B 46 6010B	(TEC)* (mg/kg)	Probable Effect Concentrations (PEC)¹ (mg/kg)	Other Criteria	Explanation for "Other Criteria"	LRL (ppm)
	46 60108					
	46 60108			4.5	Toxicity Threshold	2
				2.55%	Lowest threshold effect lever	ç
	SW-846 6010B	67.6	33.0			5
	SW-846 6010B	060	86.4			0.5
	SW-846 6010B	43.4	111			5
	SW-846 6010B	31.6	149			9
	SW-846 6010B					2
	SW-846 7471	0.18	1.06			0.1
	SW-846 6010B	35.8	128			9
	SW-846 6010B			8	Upper effects threshold*	
	SW-846 6010B			4	Toxicity Threshold ⁵	3
	SW-846 6010B	121	459			5
Water						
Parameters	Method	Aquatic Wildlife Criteria (chronic)² (mg/l)	Aquatic Wildilfa Criteria (acute) ² (mg/l)	Utah WQS for Stream Class. IC ² (mg/l)	Drinking Water MCL ³ (mg/l)	LRL (mg/l)
	SW-846 6010B or 6020	NA	0.072	0.05	90.0	0.005
	010B or 6020	0.19	90.36	0.05	50.0	0.005
	3010B or 6020	0.003	0.016	0.01	0.005	0.001
	3010B or 6020	0.572	4.8	0.05	0.1	10.0
	3010B or 6020	0.033	0.055	1	1.3 (TTR)	0.005
	3010B or 6020				0.3 (Secondary)	0.1
	EPA 245.1	0.000012	0.0024	0.002	0.002	0.0002
	31B (RF-4,6,8)	0.000012	0.0024	0.002	0.002	0.000001
	010B or 6020				0.05 (Secondary)	0.005
Pb SW-846 6	SW-846 6010B or 6020	0.015	0.375	0.05	0.015 (TTR)	0.005
	010B or 6020				900'0	0.005
	010B or 6020	0.005	0.02	0.01	0.05	0.004
	010B or 6020	0.292	0.322	none	none	0.01

Consensus-based sediment quality quidelines for freshwater ecosystems (MacDonald et al., 2000, Arch. Contam. Toxicology)

² Lowest values (based on hardness) listed in Table 3.4 of Focused RUFS Wortplan for Richardson Flat Tailings Site (RMC, 1999)

³ Safe Drinking Water Act Primary Maximum Contaminant Levels (MCLs), Secondary Drinking Water Standard, or Treatment Technique Requirement (TTR) ⁴ Screening Quick Reference Table for Inorganics in Freshwater Sediments (NOAA, 1999)
⁵ DOI Guidelines for Interpretation of the Biological Effects of Selected Constituents in Biota, Water, and Sediment (1998)

Precison, Accuracy, Representativeness, Comparability and Completeness (PARCC) Sample and Analysis Plan Richardson Flat TABLE 3

Parameter	QC Program	Evaluation Criteria	Acceptance Criteria	Recommended Corrective Actions
Precision	Field Duplicate	Relative Percent Offference (RPD)	RPDs: soil, sediment and water samples +/- 35 percent if > 5 times LRL, or, +/- LRL if < 5 times LRL	Verify the RPD calcusation. If connect, determine if matrix interference or heterogeneous samples are factors in poor RPD. If matrix effects or heterogeneous samples are not observed, resmalyze the associated investigator samples and MSMSD. If appropriate, neutract or redigest and nearal/yze the associated investigators and MSMSD.
	Matrix Spike Matrix Spike Dupitcate (MSMSD)	Relative Percent Difference (RPD)	See method-specific control limits ¹	Verify the RPD catculation. If this is correct, determine if maint invariant invariants or heterogeneous samples are factors in poor RPD. If maint effects or heterogeneous samples are not observed, resnalyze the method duplicate and associated investigative samples.
Acuracy	Matrix Spike (MS)	Percent Recovery	See method-specific control limits*	Verify the matrix splice percent recovery calculations and evaluate the LCS percent recovers. If the calculations are corner and the LCS recovertes are acceptable, determine if matrix interhence is a factor in the poor recoveries. If matrix effects not observed, reamilyze the MS and associated samples. If appropriate, resettant or redigest and reanalyze the MS and associated investigative samples.
	Matrix Spike Duplicate (MSD)	Percent Recovery	See method-specific control limits	Same as above.
	Laboratory Control Samples (LCS)	Percent Recovery	See method-specific control limits ¹	Verify the percent recovery calculations. Evaluate the standard to determine if it is faulty. If it is, prepare a new standard and reanalyze the LCS and associated investigative samples. If necessary, recalibrate the instrument. Do not continue analysis until problem solved.
Representativeness	Holding Times	Representative of Environmental Conditions	Holding Times Met 100 Percent	Evaluate whether data is critical to decision making. If so, resample and reanalyze for parameter exceeding holding time.
	Method Blanks	Qualitative Degree of Confidence	See method specific requirements ¹	Evaluate instrument, locate source of contamination, perform system blanks meet performance criteria. Rearwayse method blank and associated samples, if method blank still method blank still above acceptance criteria, resortand or redigest the method blank and all associated samples.
	Equipment/Rinsate Blanks	Qualitative Degree of Confidence	Target analytes <1 X LRL; 5-10 X LRL for laboratory-inducted contaminants.	
	· Field Dupitcates	Qualifative Degree of Confidence	90 Percent of Field Duplicates Meet RPD Goals	If acceptance other a not met, evaluate reasons for not meeting oriteria (i.e., matrix interferences or heterogeneous samples) and make recommendations on whether resampling and/or reanalysis is necessary to improve degree of confidence.
Comparability	Standard Units of Measure	Qualitative Degree of Confidence	Laboratory Methods Followed	Revise analytical reports with correct units.
	Standard Analytical Methods		SOPs Followed	If SOPs not followed, evaluate whether reanalysis is necessary to obtain reliable data.
Competeness	Complete Sampling	100 Percent Valid ² Samples	90 Percent Valid² Data	If not enough samples were collected for project needs, collect and analyze additional samples for parameters needed for key decisions.

^{*}Laboratory Control limits are specific to individual analytical/digestion methods and any deviation outside control limits are reported (see method-specific SOPs in Appendix F).

*Valid means that samples meet all evaluation criteria (I.e., are not rejected for any reason).

Precision is a measure of how repeatable data are and is othen measured by sample displicate.

Accuracy is a measure of how chose the data are to the actual, or real value, measured by certified reference materials and matrix spikes.

Representativeness is a measure of how representative a sample from the sample population and is achieved by accurate sampling procedures and appropriate sample homogenization.

Comparability looks at ongoing projects and how variable one set of data is relative to another. Comparability helps the measure of how many data points collected are usable; 90% usable data is considered to be an acceptable value for completeness.

TABLE 4

Data Validation and Verificaiton Requirements Richardson Flat Sample and Analysis Plan

Data Validation and Verification Steps		Data Validation and Verification Methods
Samples were collected according to established locations and frequencies.	↑	Comparison with Sampling Plan
Sample collection and handling followed established procedures.	↑	Review of field notes, field procedures and COCs
Appropriate analytical methods were used; internal laboratory calibration checks were performed according to the method-specified protocol.	↑	Review of analytical methods and case narratives provided with laboratory reports. Documentation of any communications with laboratory concerning problems or corrective actions.
Required holding times and laboratory reporting limits were met.	↑	Comparison with established holding times and LRLs.
Field Duplicates for QA/AC	↑	Field duplicates met acceptance criteria tabulation of RPDs and comparison with PARCC parameters
Acceptance criteria (see Table 3) for field and laboratory QC samples (field blanks, field dups, equipment/rinsate blanks, method blanks, LCS) were met.	†	Tabulation of RPDs and spike recoveries, and direct comparison with method-specific acceptance criteria (see SOPs in Appendix F). Comparison with PARCC parameters.
Appropriate steps were taken to ensure the accuracy of data reduction, including reducing data transfer errors in the preparation of summary data tables and maps.	↑	Maintain permanent file for laboratory hardcopies of analysis reports. Minimize retyping of data and error check data entered into database, tables, maps, etc.

RPD = Relative Percent Difference LRL = Laboratory Reporting Limit

TABLE 4b
Data Flagging Convention for Metals Analyses

		DAT	ΓA QUAL	IFIER FLAG	
		L	etects	Nondetects	SAMPLE(S) QUALIFIED
QUALITY CONTROL ITEM	EVALUATION	Non Biased	Biased		
HOLDING TIMES	1) Holding time exceeded by 2 times or less	J	J-	UJ	Sample
	2) Holding time exceeded by greater than 2 times	J	J-	R	
METHOD BLANK CONTAMINATION	Sample results less than or equal to 5 times the blank contamination	U	U	No qual.	All samples in the same Preparation Batch
MATRIX SPIKE RECOVERY	1) % Recovery < CL but ≥ 30%	J	J-	UJ	All samples in the same Method
	2) % Recovery <30%	J	J-	R	Batch
	3) % Recovery > CL 4) RPD > CL	J	J+	No qual.	
	4) N D > CL	J	J	נט	
LABORATORY CONTROL	1) % Recovery < CL but ≥ 50%	J	J-	UJ	All samples in the same Preparation
SAMPLE RECOVERY	2) % Recovery <50%	J	J-	R	Batch
	3) % Recovery > CL	J	J+	No qual.	
	4) RPD > CL	J	J	ប្រ	
REPORTING LIMITS	Reporting limits not matching the project specified limits	No qual.	No qual.	No qual.	Sample (noted in outlier report)
	2) Reported result less than the project reporting detection limit.	J	J	No qual.	Sample
FIELD DUPLICATES	RPD > CL	No qual.	No qual.	No quai.	Non-compliant results listed in the ADR outlier report
FIELD BLANKS EQUIPMENT BLANKS	Sample results within 5 times blank contamination	U	U	No qual.	All samples in the same sampling event

TABLE 4c
Data Flagging Convention for Ion Chromatography and Wet Chemistry

		DATA	QUALIFI	ER FLAG	
QUALITY		Dete	ects]
CONTROL ITEM	EVALUATION	Non Biased	Biased	Nondetects	SAMPLE(S) QUALIFIED
HOLDING TIMES	1) Holding time exceeded by 2 times or less	J	J-	UJ	Sample
	2) Holding time exceeded by greater than 2 times	J	J-	R	
METHOD BLANK CONTAMINATION	Sample results less than or equal to 5 times the blank contamination	U	U	No qual.	All samples in the same Preparation Batch
MATRIX SPIKE RECOVERY	1) % Recovery < CL but ≥ 30%	J	J-	UJ	All samples in the same Method Batch
	2) % Recovery <30%	J	J-	R	
	3) % Recovery > CL	J	J+	No qual.	
	4) RPD > CL	J	J	UJ	
LABORATORY CONTROL SAMPLE	1) % Recovery < CL but ≥ 50%	J	J-	UJ	All samples in the same Preparation Batch
RECOVERY	2) % Recovery <50%	J	J-	R	
	3) % Recovery > CL	J	J+	No qual.	
	4) RPD > CL	J	J	UJ	
REPORTING LIMITS	Reporting limits not matching the project specified limits	No qual.	No quai.	No qual.	Sample (noted in outlier report)
	2) Reported result less than the project reporting detection limit.	J	J		Sample
FIELD DUPLICATES	1) RPD > CL	No qual.	No qual.	No qual.	Non-compliant results listed in the ADR outlier report
FIELD BLANKS EQUIPMENT BLANKS	Sample results within 5 times blank contamination	U	ប	No qual.	All samples in the same sampling event

TABLE 5

Sample Locations, Media and Target Analytes Richardson Flat Sample and Analysis Plan

Surface water samples at RF-1,RF-2, frequency, see Fig.4 for locations. Fo will be conducted by Method 1631E conducted by Method 245.1. Groundwater List 3 and upstream of Site. Mercury Analy Method 245.1. Groundwater List 3 and upstream of Site. Mercury Analy Method 245.1. Inpoundment Soils List 1 Tailing cover, visual with laboratory conductions, see Fig 5.	Surface water samples at RF-1,RF-2,RF-3-2,RF-4,RF-5,RF-6-2,RF-7-2,RF-8, monthly frequency, see Fig.4 for locations. For locations RF-1,RF-6-2 and RF-8 Mercury Analyses in	
Groundwater List 3 Soils List 1		Collect data in Silver Creek and in drainages associated with the site to aid in determination of the background water quality relative to the site and the sites' impact on water quality in Silver Creek, including seasonal variation. Provide additional surface water data for comparison with human health and ecological screening levels.
Soils List 1	wells completed in Silver Creek altuvium, downstream	Determine metal concentrations in the Silver Creek shallow altuvial ground water both up and down stream of the impoundment. Collect surface and ground water elevation data to quantify the interaction between shallow ground water and Silver Creek; including seasonal variations.
	with laboratory confirmation at 10% of locations. Approximately 42	Determine metal concentrations in imported cover soils, provide data of sufficient quality and quantity for analyzing risks to human health and for comparison with ecological screening levels and background. Verify extent of imported soil cover in tailings. Determine suitability of the soil cover to minimize surface water infiltration into the tailings.
Diversion Soils - List 2 Sediment Samples in the south	C Sediment Samples in the south diversion ditch. Approximately 6 locations, see Fig. 5. te	Collect sediment data in the south diversion to aid in the identification of the location of metal loading in the ditch. Use the data to determine long-term chemical fate and stability of metals in the sediments and in ecological risk assessment.
Impoundment Tailings List 2 Tailings below cover, three test pits, d	w cover, three test pits, discrete samples at 1' increments down to 5' bgs, see Fig. et st	iscrete samples at 1' increments down to 5' bgs, see Fig. evaluate the long-term fate of the metal in tailings and the chemical stability of the tailings.
Site Soils List 2 Evaluate tailings south of diversio	n ditch, determine extent of tailings and impacts to diversion	Provide a definitive model of the extent of the tailings located south of the diversion ditch and to define study boundaries.

List 1 Analytes: pH, As, Pb,

List 2 Analytes: pH, Ag, As,Cd,Cu, Fe, Hg, Pb, Sb, Zn,CEC List 3 Analytes: TSS, TDS, pH, hardness, Alk, Ag,Ag-d,As,As-d,Cd,Cd-d,Cr, Cr-d,Cu,Cu-d,Fe,Fe-d,Hg,Hg-d,Pb,Pb-d,Sb,Se-d,Sn,Zn,Zn-d,Ca,K,Mg,Mn,Na,Cl,SO4,CO3,HCO3,NO2,NO3, CA/AN Bai.

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Appendix ASite Heath and Safety Policy

United Park Health and Safety Policy Richardson Flat Site Park City, Utah

Site ID Number: UT980952840

Prepared for:

United Park City Mines Company Park City, Utah

Prepared by:

Resource Management Consultants, Inc Midvale, Utah

January 2001

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10 INTRODUCTION

This Health and Safety Policy (HASP) is intended to protect all employees, general contractors, subcontractors, and/or visitors conducting or observing any activities under the direction of United Park City Mines Company (United Park). This HASP is intended to apply to activities taking place at the Richardson Flat Tailings Site (hereafter referred to as the Site), and covers both investigation and construction. The policy is intended to minimize potential exposures and/or accidents that may occur, and details the actions to be taken during an emergency. The HASP will establish required procedures intended to minimize exposures of United Park personnel, contractors, visitors and the surrounding community. Guidelines contained herein that are appropriate to the activities taking place at the Site will be observed at all times.

All personnel will be required to understand and observe the provisions of this plan. Any tasks associated with investigation or remediation activities on the Site must be performed in accordance with this policy, designed to ensure that employees are adequately protected from any potential chemical and/or physical hazards present at the Site. To help ensure safety compliance, all field participants and observers must read this plan and sign a certification stating that they agree to comply with the conditions of the policy. All activities conducted will be in accordance with 29 CFR part 1910, OSHA standards for general industry.

1.1 Site Description

The Site covers approximately 700 acres in a small valley in Summit County, Utah, located one and one-half miles northeast of Park City, Utah. The Site includes a tailings impoundment covering approximately 160 acres in the northwest corner of the Property and lies within the NW quarter of Section 1 and NE quarter of Section 2, Township 2 South, Range 4 East, Summit County, Utah.

United Park personnel will be investigating the soil and water in and around the tailings impoundment. During the course of this investigation, there exists a potential for personnel to have limited contact with tailings contained on the Site. The mill tailings present at the Site consist mostly of sand-sized particles of carbonate rock with some minerals containing silver, lead, zinc and other metals. Currently, tailings at the Site are completely covered with a layer of clean fill.

1.2 Site Activities

This HASP is intended to address the risks associated with sampling and construction activities, which will take place at the Site. During the course of investigation by United Park, personnel will be required to visit the Site in order to collect soil and water samples for chemical analysis. Personnel will also visit the Site to survey and perform other miscellaneous tasks. The procedures contained in this HASP are intended to protect

those personnel from potential hazards while carrying out their duties, and provide them with information necessary in the event of an emergency.

It is anticipated that investigations by United Park may also involve limited construction activities, including excavation and/or removal of soils. The HASP has therefore included procedures for equipment and personnel involved in construction activities at the Site.

2.0 PROJECT MANAGEMENT

Efficient implementation of this policy requires that the roles, responsibilities and scope of authority for key personnel be identified. United Park shall identify individuals responsible the following positions:

2.1 Project Manager

The Project Manager is responsible for implementation of the work plan and compliance with the HASP.

2.2 Health and Safety Manager

The Health and Safety Manager will have a thorough working knowledge of state and federal occupational safety and health regulations in addition to thorough knowledge and understanding of this policy. The Health and Safety Manager will have the authority to temporarily suspend site operations in order to ensure site safety and resume normal operations once the appropriate measures have been taken. The Health and Safety Manager will report directly to the Project Manager.

2.3 Site Manager

The Site Manager will be present during the majority of site activities and will be responsible for general site activities, supervision and enforcement of this HASP. The Site Manager will report directly to the Health and Safety Manager.

2.4 Supervisor

The Supervisor(s) will be present during all on-site activities and will report directly to the Site Manager.

Note: The aforementioned personnel may be increased, or personnel may share responsibilities dependent upon specific site conditions.

3.0 TRAINING

3.1 Off-Site Training

All full-time, part-time and short-duration workers must hold current certification of the Occupational Safety and Health Administration (OSHA) Hazardous Waste Operations and Emergency Response (HAZWOPER) 40-hour training. Visitors must hold current certification of OSHA/HAZWOPER 40-hour training and shall be escorted at all times by an experienced and trained Site Manager.

3.2 On-Site Training

An informational training program implemented by United Park will cover on-site training.

3.3 Weekly Health and Safety Meetings - Construction

During any construction or excavation activities, the site Health and Safety Manager will conduct mandatory weekly safety meetings for all site personnel. The meetings will provide time for refresher courses, and new site conditions will be examined as they are encountered.

3.4 CPR and First Aid Training Requirements - Construction

During any construction or excavation activities, a minimum of one worker per work crew or shift shall have a current certificate of training in first aid and CPR. These workers must have appropriate training and medical surveillance to enter the Site.

4.0 MEDICAL SURVEILLANCE

4.1 Medical Surveillance - General

Medical surveillance will be obtained if personnel:

- Receive, or may have received, a possible overexposure to on-site contaminants;
- · Received an injury requiring hospital or medical attention;
- Experience an unexplained or serious illness.

4.2 Medical Surveillance - Construction

A yearly physical examination shall be provided for field personnel involved with excavation of any tailings material in excess of 500 yd³. The examination shall emphasize skin, renal, hepatic, immunological, neurological, and hematological systems, and shall include tests for liver and kidney function. If construction personnel are exposed to tailings materials on-site for thirty (30) days or more, they will participate in a medical examination program according to OSHA's lead (29 CFR 1926.65) standard.

5.0 HEALTH AND SAFETY PROTECTION

5.1 Substance Hazards

Lead, arsenic and cadmium are known to exist on the Site, and personnel should be briefed on exposure and health hazards. It is not anticipated that exposures to these substances will exceed OSHA's Personal Exposure Limit (PEL). The following table lists the primary hazards associated with significant exposure to each substance.

Lead

Toxic on inhalation and ingestion.

Arsenic

Toxic on inhalation and ingestion; skin irritant; known human

carcinogen.

Cadmium

Toxic on inhalation and ingestion; suspected human carcinogen

through inhalation only.

5.2 Safety Hazards

Investigation activities may expose field personnel to potential physical hazards including, but not limited to:

- Holes and ditches
- Uneven terrain
- Slippery surfaces
- Electrical equipment
- Mobile equipment
- Overhead hazards
- Underground hazards

5.3 Personal Protection Equipment - Construction

The minimum level of protection used during any construction activities is level D, requiring the following items:

- Hardhat;
- Steel-toed boots;
- Safety glasses;
- Cotton coveralls;
- Work gloves;
- Sampling gloves;
- Hearing protection, when needed.

5.4 Personal Air Monitoring – Construction

During construction activities involving contact of tailings material, personal air monitoring will be conducted to verify and document exposures to lead, arsenic, and cadmium on this project do not exceed the OSHA PEL's. Personal air monitoring will only occur when tailings are contacted in excess of 500 yd³. If monitoring reveals exposures above an OSHA PEL, then field personnel will be upgraded to level C protection.

5.4.1 Work Practices to Reduce Employee Exposure - Construction

While performing any construction/excavation activities, work practices shall be instituted to ensure worker exposure remains below the applicable PEL. Work practices will include wetting down excavation-sites as needed throughout any excavation operation. The site safety officer will be responsible to monitor the dust control operations when needed.

5.5 Exposure to Elements

5.5.1 Heat Stress

The potential for heat stress depends on the type of protective gear being worn, the ambient temperature, and the amount of activity. Personnel will report any cases of dizziness, excessive sweating, increased respiratory rate, or pulse and are to leave the work area immediately if these conditions are noted. Work cycle lengths will be based initially on subjective input from personnel, and will be reduced and a monitoring program will be initiated if the above are noted. Work cycles will also be reduced if a pulse rate of greater than 110 is noticed during rest. Personnel with elevated rates will not return to work until the pulse has lowered to their resting rate.

Workers exhibiting signs of heat stress will have their oral temperature measured at the beginning of a rest period before liquid intake. If oral temperature exceeds 99.6° F, the next work cycle will be shortened by one-third without changing the rest period. If the oral temperature still exceeds 99.6° F at the beginning of the next rest period, the next work cycle will be shortened by another one-third. If the oral temperature exceeds 100.6° F, the worker will not be allowed to wear semi-permeable or impermeable clothing. If an employee is overcome with heatstroke or becomes unconscious, the 9-1-1 service will be called. First-aid procedures will be used for heat related conditions, as necessary. Some of the signs and symptoms of heat stress are as follows:

5.5.1.1 Heat Rash

Symptoms of Heat Rash include:

- Profuse tiny raised vesicles on the skin
- Pricking sensations during heat exposure

5.5.1.2 Heat Cramps

Symptoms of Heat Cramps include:

- Painful spasms of muscles used during work
- Onset during or after work hours

5.5.1.3 Heat Exhaustion

Symptoms of Heat Exhaustion include:

- Fatigue
- Nausea
- Headache
- Giddiness
- Clammy and moist skin
- Pale complexion
- Upon standing, fainting possible, with rapid, thready pulse and low blood pressure

5.5.1.4 Heatstroke

Symptoms of Heatstroke include:

- Hot dry skin usually red, mottled or cyanotic
- Confusion, loss of consciousness, and convulsions

Note: Heat stroke may be fatal if treatment is delayed

5.5.2 Cold Stress

During on-site activities, workers may be exposed to cold temperatures. Exposure to cold temperatures increases the likelihood and potential for disorders or conditions that could result in injury or illness. Factors leading to hypothermia and frostbite include ambient temperature, wind velocity, exposure time and insufficient cold-weather protective gear. Signs of excess cold exposure include uncontrollable fits of shivering, slurred speech, memory lapses, immobile hands, stumbling, drowsiness, and exhaustion. Treatment for these symptoms are to get the victim out of the wind and cold, remove wet clothing, supply a warm drink, and keep victim warm with blankets or clothing.

5.5.2.1 Hypothermia

The first symptoms of this condition are uncontrollable shivering and the sensation of cold, irregular heart beat, weakened pulse, and change in blood pressure. Severe shaking

of rigid muscles may be caused by a burst of body energy and changes in the body's chemistry. Vague or slow slurred speech, memory lapses, incoherence, and drowsiness are some of the additional symptoms. Symptoms noticed before complete collapse are cool skin, slow and irregular breathing, low blood pressure, apparent exhaustion, and fatigue even after rest. As the core body temperature drops, the victim may become listless and confused, and may make little or no attempt to keep warm. Pain in the extremities can be the first warning of dangerous exposure to cold. If the body core temperature drops to about 85° F, a significant and dangerous drop in the blood pressure, pulse rate, and respiration can occur. In extreme cases, death will occur.

5.5.2.2 Frostbite

Frostbite can occur, in absence of hypothermia, when the extremities do not receive sufficient heat from central body stores. This can occur because of inadequate circulation and/or insulation. Frostbite occurs when there is freezing of fluids around the cells of the body tissues due to extremely low temperatures. Damage may result, including loss of tissue around the areas of the nose, cheeks, ears, fingers, and toes. This damage can be serious enough to require amputation or result in permanent loss of movement. The potential for both heat and cold related disorders or conditions can occur in many common situations. Cold early morning temperatures can give way to warm daily temperatures, resulting in heavy perspiration within protective clothing. As temperatures cool again in the evening, the potential for cold related disorders or conditions can occur. Managers should be aware of the potential for this occurrence and should monitor workers accordingly.

5.5.3 Wind Exposure

Extreme low temperatures may not be the only element necessary to create the potential for cold exposure disorders or conditions; strong wind accompanied by cold temperatures can lead to these types of disorders or conditions. The windchill factor is the cooling effect of any combination of temperature and wind velocity or air movement. The windchill factor should be considered when planning for exposure to low temperatures and wind.

5.5.4 Logs and Reports

United Park will maintain all records required by OSHA, Worker's Compensation Insurance and similar regulations. This will include the maintenance of accident logs, the OSHA annual summary report and the posting of all prescribed notices.

6.0 SITE CONTROL

Site control will be implemented for both investigation and construction activities as needed.

6.1 Investigation

6.1.1 Work Zone

All investigatory activities, including but not limited to surveying and sampling, at the Site will take place within the work zone; this includes the tailings impoundment and the area immediately surrounding it, currently demarcated with fencing. This area will be restricted to appropriately trained personnel, and any non-approved personnel will immediately be escorted off-site.

6.1.2 Cleaning/Maintenance Area

At the entrance(s) of the work zone, an area will be provided for removal of gross contamination from both hand tools and personnel. United Park personnel and/or representatives will remove gross contamination from their boots and coveralls. Facilities will be provided for personnel to wash their hands and face as needed. At a minimum, facilities will include fresh water, soap, towels and waste receptacle.

6.2 Construction

6.2.1 Work Zone

All construction activities carried out at the Site will occur within the work zone, currently demarcated by fencing. This area poses a potential hazard and will therefore be restricted to trained workers with the appropriate personal protective equipment. Any excavation-sites will be demarcated by yellow barrier tape, if not backfilled prior to the end of each workday. An area that has been backfilled will be considered as lacking hazards, unless exposed utilities, etc. create a hazard. Such hazards will be demarcated with barrier tape.

6.2.2 Cleaning/Maintenance Area

At the entrance(s) of the work zone, an area will be provided for removal of gross contamination from both equipment and personnel. United Park personnel and/or representatives having contact with any tailings material will be required to remove gross contamination from their vehicles, equipment, boots and coveralls prior to leaving the Site. At a minimum, facilities will be provided including pressurized water, scrub tools for vehicles and equipment, and fresh water, soap, towels and waste receptacle.

6.3 General Maintenance

General cleaning maintenance is key in helping to maintain acceptable exposure levels for lead, arsenic, and cadmium. General cleaning/maintenance will be required for all equipment and facilities used by on-site as well as off-site personnel. This will include, but is not limited to a change and/or shower facility, office areas, and lunch facilities.

6.4 Equipment Safety

All mobile equipment with limited visibility to the rear shall be equipped with audible back-up alarms. If mobile equipment is operated at night, it shall be equipped with head lights and taillights. All equipment will be maintained in good condition. When the operator leaves the cab of mobile equipment, emergency brakes shall be set and any hydraulics released. If a truck is parked on an incline, it shall have the tires chocked.

When refueling, engines on all equipment shall be shut off. All mobile equipment will be supplied with a fire extinguisher with a rating of not less than 5-B rating, and the service truck will be supplied with a fire extinguisher with a rating of not less than 20-B rating.

6.5 Electrical Safety

Electrical power tools will continuously be inspected for damage. Electric tools with frayed cords or broken housings will be tagged and taken out of service.

If tools are used in wet conditions, they must be listed or labeled as double insulated. All extension cords will be of the three-wire ground type and be connected to a ground fault circuit interrupter (GFCI). If extension cords are not plugged into a permanently mounted GFCI, then the extension cord must be supplied with a waterproof GFCI. Extension cords that are spliced, worn, or frayed are not to be used. Extension cords must have the manufacturers rating on the cord and it must be legible; if it is not legible the cord must be taken out of service.

6.6 Miscellaneous Site Safety Rules

Miscellaneous Site Safety Rules include the following:

- Smoking, eating, chewing, applying cosmetics, etc. is not allowed on-site.
- A minimum of two personnel shall be on-site at all times.
- No horseplay is permitted at any time
- Vehicles used to transport personnel shall have seats firmly secured and adequate for the number of persons to be carried.
- Seat belts and anchors meeting the requirements of 49 CFR part 571 (department of transportation, federal motor vehicle safety standards) shall be installed in all motor vehicles.

7.0 DECONTAMINATION

7.1 Field Personnel

Decontamination procedures for field personnel shall be:

- Gross contamination removal from clothing and boots prior to leaving the Site.
- Wash hands and face at facility provided
- Containment of dirty coveralls.
- Launder coveralls at commercial laundry.

7.2 Equipment

The decontamination procedures for equipment contacting tailings shall be:

- Clean vehicles (inside and out) as needed prior to leaving the Site.
- Construction equipment, backhoes, loaders, dump trucks, hand tools, trailers
 hoses, etc contacting any tailings material will be cleaned of gross excavated soil
 material before leaving the Site and pressure washed upon culmination of
 scheduled work.
- Sampling equipment and hand tools not contacting tailings material will be cleaned of gross contamination prior to leaving the Site.

8.0 EMERGENCY RESPONSE

Accidents or potentially hazardous conditions will be handled in a manner to minimize the health risk to personnel. Accidents and hazardous conditions will be reported to the site safety officer. Prior to the start-up of this project, methods of communication will be established in order to summon emergency services in a timely manner. Supervisory personnel and the Site Safety Officer will be trained in first aid/CPR.

8.1 Emergency Route to Hospital

The emergency route to local medical facilities is shown in Figure 1 and emergency contacts with phone numbers are listed in Appendix A

8.2 Incident Command System

The Incident Command System used on this project will utilize different senior response officials depending on the nature of the incident. Front line supervisors are the initial "Senior Official" until the Project Manager or the Health and Safety Manager arrives. When emergency officials arrive, they shall become the "Senior Official".

8.3 Response Procedures

All United Park personnel will be trained in general procedures in the event of an emergency. Prior to beginning any work, personnel will be required to review the emergency procedures of this plan and ensure that all necessary equipment is ready for use in the event of an emergency. Visitors to the Site should also be briefed on these procedures.

Common forms of emergency include, but are not limited to fires, explosions, spills, sudden changes in weather, and personal illness or injury. The following emergency response procedures have been developed to help ensure a timely and efficient response to emergency situations that may arise.

8.3.1 Major and Minor Personal Injury

If field personnel are injured, the incident scene will be evaluated for immediate hazards and actions taken to eliminate those hazards. Once the incident scene is safe, the "Senior Official" will make an evaluation of the injured person. Seriously injured personnel should not be moved unless their life is in immediate danger and until a person trained in first-aid and CPR has made an assessment.

If the victim is conscious, first-aid may only be administered with the injured person's permission. If the victim is unconscious or unable to respond, then no permission is required to provide standard first aid. If no outside emergency services are needed, the

"Senior Official" will arrange for the injured person to be transported to the predetermined medical facility.

If it is determined that emergency medical services are needed, the emergency services listed in Appendix A will be contacted as soon as possible. Calling for help is often the most important action to be taken. If you are the only person with the injured employee and urgent care is needed, provide initial critical care and then contact the outside emergency services. Return to care for the victim as soon as possible.

First-aid or other appropriate actions can be administered by the initial "Senior Official" or by the victim. For injuries requiring medical treatment such as a laceration requiring stitches or a sprained ankle, the "Senior Official" shall arrange transportation to the emergency facility as noted in Figure 1. For major injuries, the "Senior Official" may administer first-aid. The "Senior Official" rendering assistance will not place themselves in a situation of unacceptable risk.

8.3.2 Fire or Explosion

In the event of a fire or explosion, the local fire department will be notified immediately. The "Senior Official" will notify the emergency services and inform them of the location, nature and identification of any hazardous materials on-site.

During the beginning stages, the closest person to the incident will take measures to extinguish the fire using a fire extinguisher or water hose. If the fire progresses beyond the beginning stages, the "Senior Official" will evacuate workers and any other occupants on the property from the immediate area and allow local fire officials to attend to the situation.

8.4 Notification and Documentation Procedures

As soon as practical following an accident/incident, the accident/incident will be documented using the appropriate report forms and the site safety officer will be notified.

8.5 On-Site Emergency Equipment

The following emergency equipment will be maintained at all work sites.

- Cellular Telephone;
- First-aid kit;
- Fire extinguisher; and
- Emergency eye wash solution.

APPENDICES

Appendix A – Emergency Contact Phone Numbers

Organization	Telephone
Any Emergency	911
Ambulance:	911
Local Police:	435-645-5500
Fire:	911
State Police:	801-576-8606
Hospital (Primary)	435-649-7640
Hospital (Secondary)	435-655-0055
Poison Control Center:	801-581-2151
Regional EPA:	800-227-8917
EPA Emergency Response Team:	800-227-8914
National Response Center:	800-424-8802
Center for Disease Control:	404-639-3311
Chemtrec:	800-262-8200
Spill Center:	978-897-6461
Site Emergency Operations Center:	801-355-2350
DOE Emergency Operations Center (National Center):	202-586-5000

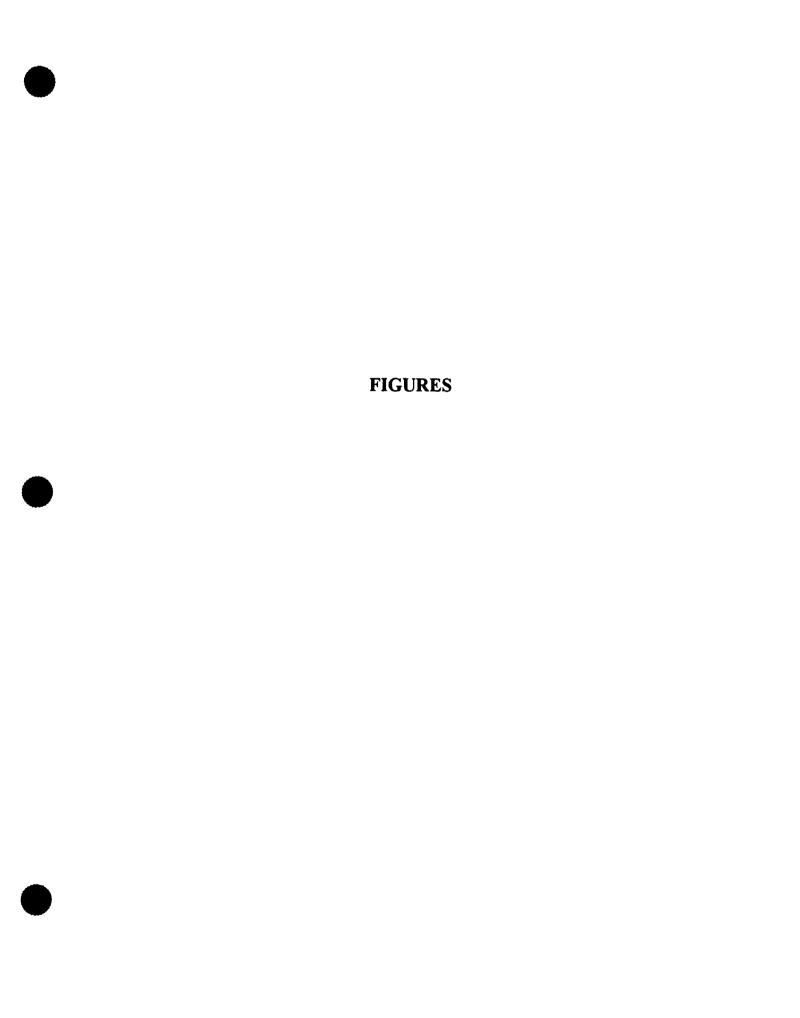
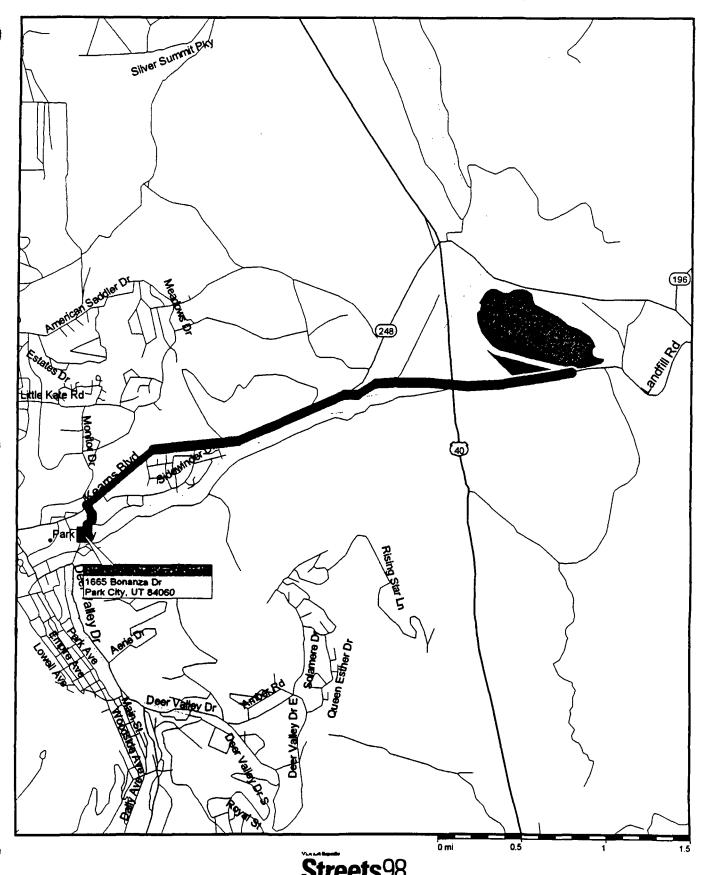


Figure 1 - UPCM Richardson Flat Emergency Route



Tabbed Page: Appendix B

Appendix B Standard Operating Procedures

RMC SOP 1 STANDARD PROCEDURES FOR COLLECTION OF SURFACE WATER SAMPLES AND GENERAL WATER SAMPLE HANDLING

1.0 Purpose

This SOP describes the procedures that will be used for collection of surface water samples. The procedures will ensure that samples are collected and handled properly and that appropriate documentation is completed. The procedures outlined in this SOP detail the procedures used for the treatment/handling of water samples collected from other sources (e.g. monitoring wells).

2.0 Sampling Equipment

- Field data sheets / Field notebook / Chain of Custody Forms (COC) Documentation of sample activities, field notes and sample custody.
- Sample containers Containers provided by laboratory for the collection, storage and transportation of samples.
- Direct reading instruments field instruments to measure pH, conductivity and temperature.
- Disposable sampling gloves to prevent exposure to water and the prevention of cross-contamination.
- Custody seals seals to be placed on sample containers to maintain sample integrity.
- 0.45 um filter apparatus with inert filters for filtering samples in preparation for the analysis of dissolved metals.
- Nitric acid (HNO₃, supplied by the analytical laboratory) for sample preservation.
- Water velocity meter and tape measure to measure stream flow (where applicable).
- Distilled water for rinsing direct reading instruments.
- Custody seals seals to be placed on sample containers to maintain sample integrity.

3.0 Procedure

Sample bottles will remain sealed until the water sample is collected. At that time, the bottle lid will be removed and placed, top down, in an appropriate place. The sample bottle will be placed under the flow of water. The container will be rinsed three times prior to sample collection. If wading is required for sample collection, the sample must be collected upstream of wading personnel to avoid the sampling of suspended sediments. After rinsing, the sample container will be completely filled; any overflow of the sample container will be kept to a minimum. Sediment disturbance shall be kept to an absolute minimum. The sample cap will then be replaced on the sample bottle. All surface water samples will be collected in accordance with containers, volumes, preservatives, temperatures and holding times as outlined in Table 2 of the Sampling and Analysis Plan.

4.0 Dissolved Metals Analysis

Surface water samples collected for analysis of Dissolved (D) Metals will be a minimum volume of 500 ml, collected in a poly or glass container. The samples will be field filtered. The field filtering methodology will include the following steps:

- 1: Sample shall be collected in a 1000 ml bottle.
- 2: Sample is poured into the top of the disposable plastic filter.
- 3: Vacuum pump is attached to the filter and pumped.
- 4: When the bottom compartment of the filter is full, the water is to be transferred into a 500 ml sample container which shall be rinsed three times, the sample will be preserved with 2 ml of nitric acid (HNO₃), sufficient to bring the sample to pH <2.

5: The pH level in samples will be verified using pH paper before bottles are sealed.

5.0 Total Metals Analysis

Surface water samples collected for analysis of Total (T) Metals will be a minimum volume of 500 ml, collected in a poly or glass container, and preserved with 2 ml of nitric acid (HNO₃), sufficient to bring the sample to pH <2. The pH level in samples will be verified using pH paper before bottles are sealed.

6.0 Cations/Anions and Total Suspended Solids

Cations/Anions and Total Suspended Solids samples shall be collected in accordance with the methodologies outlined in the Procedure section of this SOP. Samples will not be preserved.

7.0 Field Parameter Collection - pH, Temperature and Conductivity

Temperature, pH and Conductivity measurements will be collected in the field at the time of sampling. All Direct reading field instruments will be calibrated prior to daily use according to the manufacturers specification manual/instructions provided with each instrument. Field data will be collected according to the manufacturers specification manual/instructions for each instrument. Water samples shall be placed into a clean container such as a sample bottle or 1 gallon bucket. The instrument probes will be placed into the water sample. Field data collected in this procedure will be recorded in the field notebook and Field data sheet for each sample.

7.0 Stream Flow Measurement

Stream flow volumes shall be measured during surface water sampling activities. To minimize sediment disturbance during sampling, the stream flow measurements should be conducted either downstream from the sampling point or after the completion of sample collection. RMC uses an electronic flow meter. The procedure for measuring stream flows is as follows:

- 1: Measure the width of the stream and divide the width into 0.5 foot increments.
- 2: At the midpoint of each 0.5 foot increment, record the total depth of the stream. The water velocity shall be measured at 0.6 of the total height of the water (e.g. if the water is one foot deep the velocity is measured at a depth of 0.4 foot from the surface or 0.6 feet from the streambed).
- 3: Turn the electronic stream meter gauge on. Set the meter to record the average velocity. Insert the stream flow gauge into the water at the midpoint of each segment with the arrow pointing in the direction of flow. Measure the velocity for approximately one minute and record the average.
- 4: Calculate the stream flow by calculating the area of each 0.5 foot wide segment by multiplying the width times depth. To obtain the flow volume for each 0.5 wide segment multiply the area of the segment by the average flow velocity for the segment. To obtain the total stream flow, add the total stream flow for each segment. An Excel spreadsheet is typically used for the calculations.

Calculations:

Segment flow volume = depth of 0.5 foot segment x width x flow velocity (feet/sec.) = cubic feet/ second Total flow volume = sum of segment flow volumes.

8.0 Labeling

Each sample will be labeled with the following information:

Sample identification;

- Project number/name;
- Analyses requested;
- Preservatives (if required);
- Date/time collected; and
- Samplers initials.

9.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheet. Field notes shall include all pertinent information including but not limited to:

- Date and time samples were collected;
- Physical description of sample area;
- Identification of samples collected;
- Total number of samples collected;
- Total number of samples collected from each sample location;
- Physical description of samples;
- · Preservatives used for samples;
- Sample container types;
- Filtered vs. Unfiltered samples;
- Analysis to be performed;
- Weather conditions;
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

Sample handling and Chain of Custody documentation shall be in accordance with RMC SOP 5 found in this document.

10.0 Demobilization

After Decontamination, sample equipment will be stored in the appropriate, clean containers. Any equipment that suffers damage or excessive wear while conducting sampling will be labeled and reported to the equipment manager for the necessary maintenance, repair and/or replacement.

SOP 2a STANDARD PROCEDURES FOR COLLECTION OF SURFACE SOIL SAMPLES

1.0 Purpose

This SOP describes the procedures that will be used for sampling surface soils from ground surface to a maximum depth of 18 inches below surface. Samples will be collected with a Decontamination shovel or hand auger/probe. Specific soil sampling locations will be determined from the project work plan.

2.0 Sampling Equipment

- Hand Auger/Probe and/or Shovels For the collection of soil samples below the ground surface.
- Field data sheets / Field notebook / Chain of Custody (COC) Documentation of sample activities, field notes and sample custody.
- Sample containers Containers provided by laboratory for the collection, storage and transportation of samples.
- Stainless steel sample spoons For the collection of surface soil samples and composite sample mixing.
- Stainless steel bowl or sealable plastic bags for mixing composite samples.
- Sample location staking For the marking and identification of sample locations. Staking should be easily visible for surveying.
- Disposable sampling gloves to prevent exposure to soils and the prevention of cross-contamination.
- Custody seals seals to be placed on sample containers to maintain sample integrity.

3.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

4.0 Procedure

All samples shall be collected using Decontaminated equipment. Decontamination procedures are detailed in RMC SOP 6.

4.1 Discrete Samples

If significant vegetation, rocks, or debris prevent collecting the surface samples then the upper 2-3 inches of soil will be scraped away from the sample location with a shovel or stainless steel spoon. The underlying soil will then be collected and placed into sample containers with a stainless steel spoon or gloved hand. Composite samples will be homogenized as described below. Coarse grained soils, gravel and rock fragments will be removed wherever possible.

4.2 Composite Samples

Composite samples will be collected (as described above) by placing sub samples into a stainless steel mixing bowl or a clean plastic bag, or by hand with new, clean sampling gloves. The sample will be homogenized with a stainless steel spoon or gloved hand. The homogenized soil will be packaged in a laboratory-supplied sample container, labeled and placed in a cooler to maintain temperature.

4.3 Sample Preparation

Soil Samples collected for human health risk assessment shall be sieved to <250 microns. The <250 micron fraction is then analyzed for metals. For ecological screening/risk assessment purposes, sieving should not occur. Sieving shall be performed by the laboratory.

5.0 Labeling

Each soil sample will be labeled with the following information:

- Sample identification;
- Project number/name;
- Analyses requested;
- Date/time collected; and
- Samplers initials.

6.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheet. Field notes shall include all pertinent information including but not limited to:

- Date and time samples were collected;
- Physical description of sample area;
- Identification of samples collected;
- Total number of samples collected per sampling event;
- Total number of samples collected from each sample location;
- · Physical description of samples;
- Preservatives used for samples;
- Sample container types;
- Filtered vs. Unfiltered samples (water);
- Analysis to be performed;
- Weather conditions;
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

Sample handling and Chain of Custody documentation shall be in accordance with RMC SOP 5 found in this document.

7.0 Demobilization

After Decontamination, sample equipment will be stored in the appropriate, clean containers. Any equipment that suffers damage or excessive wear while conducting sampling will be labeled and reported to the equipment manager for the necessary maintenance, repair and/or replacement.

SOP 2b

HAND AUGER SOIL SAMPLING

1.0 Purpose

Hand auger equipment will be used for collecting shallow soil samples to approximately 5 feet below ground surface. This SOP describes the procedures for collecting soil samples using hand auger equipment.

2.0 Equipment

- Hand augers
 - Clay auger barrel for the collection of clay rich soils.
 - Sand auger barrel for the collection of sandy soils.
 - Extension rods For connecting the sample barrel to the handle
 - T handle- for turning the auger assembly.
- Two crescent wrenches For attaching/breaking down the hand auger.
- Tape measure for the measurement of sample depths/intervals.
- Field data sheets / Field notebook / Chain of Custody (COC) Documentation of sample activities, field notes and sample custody.
- Sample containers for sample storage and transportation.
- Disposable sampling gloves to prevent exposure to soils and the prevention of cross-contamination.
- Surface patching supplies, if necessary (asphalt patch/post mix)
- Stainless steel bowl or sealable plastic bags for mixing composite samples.
- Custody seals seals to be placed on sample containers to maintain sample integrity.

3.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

4.0 Preliminaries

All boring locations will be determined using the project specific Work Plan. Arrangements will be made for the location of underground utilities using Blue Stakes. A private locating service will be used for utilities that are not covered by Blue Stakes.

5.0 Procedures

The borehole will be advanced using the clay bucket for fine-grained soils and the sand bucket for coarse-grained soils. Each auger bucket of soil will be described and recorded on the soil boring log. Soil samples selected for laboratory analysis will be placed in an appropriate container.

6.0 Labeling

Each soil sample will be labeled with the following information:

- Sample identification;
- Project number/name;
- Analyses requested;

- · Date/time collected; and
- · Samplers initials.

7.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheet. Field notes shall include all pertinent information including but not limited to:

- Date and time samples were collected;
- Physical description of sample area;
- Identification of samples collected;
- Total number of samples collected per sampling event;
- Total number of samples collected from each sample location;
- Physical description of samples;
- Preservatives used for samples;
- Sample container types;
- Filtered vs. Unfiltered samples (water);
- Analysis to be performed;
- Weather conditions;
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

Sample handling and Chain of Custody documentation shall be in accordance with RMC SOP 5 found in this document.

8.0 Decontamination

All samples shall be collected using Decontaminated equipment. Decontamination procedures are detailed in RMC SOP 6.

9.0 Demobilization

After Decontamination, sample equipment will be stored in the appropriate storage containers. If any equipment is damaged while conducting soil sampling, the damaged equipment will be labeled and reported to the equipment manager for maintenance or replacement.

SOP 2c

GEOPROBE SAMPLING

1.0 Purpose

GeoprobeTM sampling equipment will be used to advance shallow soil borings (30 feet or less) to collect soil and groundwater samples and for sites where access restrictions prevent mobilization of a drill rig. Standard operating procedures for geoprobe soil and groundwater sampling are described below.

2.0 Preliminaries

Geoprobe sample locations will be marked or staked in the field and coordinated with the RMC project manager and, if necessary, the client project manager. Blue Stakes utility clearance will be requested for each boring location prior to geoprobe sampling. Borings will be located at least two feet from marked underground utilities.

All sampling equipment will be decontaminated prior to mobilizing to the site. This equipment includes all geoprobe rods, geoprobe samplers, and stainless steel bowls and spoons.

3.0 Geoprobe Equipment and Procedures

Soil borings will be advanced and sampled using a geoprobe hydraulic hammer mounted to a truck, van, four-wheeler, or small tractor. Each borehole will be started by hydraulically hammering a 3 foot length of 1 inch outside diameter steel drill rod with a stainless steel sample collection tube into the ground. Each sample tube shall be Decontaminated prior to use. The borehole will be advanced in 3 foot increments by adding 3 foot sections of flush threaded drill rod to the drill stem. No lubricants or additives will be used while advancing geoprobe borings.

4.0 Soil Sampling Equipment

The following equipment will used to conduct soil sampling:

- Field data sheets / Field notebook / Chain of Custody (COC) Documentation of sample activities, field notes and sample custody.
- Geoprobe core sampler (supplied by the geoprobe contractor).
- New sample liners (supplied by the geoprobe contractor).
- New sample liner end caps (supplied by the geoprobe contractor).
- Disposable sampling gloves to prevent exposure to soil and water as well as the prevention of crosscontamination.
- Sealable plastic bags for sample storage.
- Laboratory supplied glass soil sample jars and labels (optional).
- Razor blade knife for splitting open sample tubes.
- Stainless steel bowl and spoon for mixing composite samples.
- Custody seals seals to be placed on sample containers to maintain sample integrity.

5.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

6.0 Decontamination

All samples shall be collected using Decontaminated equipment. Decontamination procedures are detailed in RMC SOP 6.

7.0 Soil Sampling Procedures

Samples will be collected as specified in the site specific sampling plan. At a minimum, soil samples will be collected at 5 foot intervals if lithologic information is needed. Each soil sample will be collected in a 2 foot long lined core sampler. The sampler will be attached to the drill rod, lowered to the sample interval and then hydraulically hammered two feet into the subsurface.

8.0 Groundwater Sampling Procedures

To facilitate the collection of groundwater samples at sites where the water table is penetrated, a temporary well point will be installed in the geoprobe borehole. After the water table has been encountered, the borehole will be advanced at least three more feet to ensure adequate sample volume. The well point may consist of either a three foot long stainless steel screen, attached to polyethylene tubing, or a length of 3/8inch polyethylene tubing with perforations in the bottom 3 feet. New tubing and well screens will be used for each well point. After approximately 15 minutes, a peristaltic pump will be attached to the tubing to obtain groundwater.

Groundwater samples shall be handled in accordance to the methods detailed for the handling/treatment of surface waters samples in RMC SOP1.

9.0 Sample Labeling

Each sample will be labeled with the following information:

- Sample identification;
- Project number/name;
- Analyses requested;
- Preservatives (water samples);
- Date/time collected; and
- Samplers initials.

10.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheets. Field notes shall include all pertinent information including but not limited to

- Date and time samples were collected;
- Physical description of sample area;
- Identification of samples collected;
- Total number of samples collected per sampling event;
- Total number of samples collected from each sample location;
- Physical description of samples;
- Preservatives used for samples;
- Sample container types;
- Filtered vs. Unfiltered samples (water);
- Analysis to be performed;
- Weather conditions;
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

Sample handling and Chain of Custody documentation shall be in accordance with RMC SOP 5 found in this document.

11.0 Boring Abandonment

After all soil and groundwater samples have been collected, each soil boring will be backfilled with granular bentonite. Borings that were drilled through asphalt or concrete will be backfilled with granular bentonite to within six inches of the ground surface and the asphalt and concrete cores will be restored.

12.0 Demobilization

After the equipment has been rigged down and loaded, the site will be cleaned and restored as close to its original condition as possible. All sampling equipment will be Decontaminated prior to mobilizing to the next geoprobe sample location.

HOLLOWSTEM AUGER DRILLING, SOIL SAMPLING AND MONITORING WELL INSTALLATION.

1.0 Purpose

Hollowstem auger drilling techniques will be used to advance intermediate depth borings of 100 feet or less. Standard operating procedures for hollowstem auger drilling and soil sampling are described below. Hollowstem auger drilling techniques will be used to install groundwater monitoring wells.

2.0 Preliminaries

Final soil boring locations will be marked or staked in the field and coordinated with the RMC project manager and, if necessary, the client project manager. Blue Stakes utility clearance will be requested for each drilling location to identify any subsurface utilities prior to drilling and sampling. If required, drilling and/or monitoring well permits will be requested by supplying the appropriate forms to the corresponding regulatory agency.

Boring locations will be located the following distances from overhead power lines:

Power Lines Nominal System (kV)	Minimum Required Clearance (ft)	
0-50	10	
51-100	12	
101-200	15	
201-300	20	
301-500	25	
501-750	35	
751-1000	45	

All drilling and sampling equipment will be Decontaminated with a steam cleaner prior to drilling. This equipment includes all drill pipe, auger flights, split-spoon samplers, brass sleeves, stainless steel bowls and spoons, tools, and non-packaged well screen and casing. Steam cleaning will be conducted after placing equipment, tools, and non-packaged screen and casing on racks or sawhorses to keep them off the ground. After steam cleaning is completed, the equipment will remain off the ground until it is used. Borings will be located according to the site specific work plan. No borings will be drilled within 5 feet of marked underground utility lines or within 10 feet of active overhead power lines. Boring locations will be adjusted, as necessary.

3.0 Drilling Equipment and Procedures

A truck mounted hollow stem auger drill rig will be used to drill borings of 100 feet or less. Augers will be sized to accommodate the well casing diameter, if a well is to be installed in the borehole. If flowing sands are encountered a center plug will be used to prevent liquefied sands from entering the inside of the auger string during monitoring well installation. No lubricants, circulating fluid, drilling muds, or other additives will be used during drilling. All drilling equipment will be supplied by the drilling contractor. The drilling contractor will supply a steam cleaner for drilling equipment decontamination.

4.0 Soil Sampling Equipment

The following equipment will be used to conduct soil sampling:

- Log forms / Field data sheets / Field notebook / Chain of Custody Forms (COC) Documentation of sample activities, field notes and sample custody.
- Split-spoon samplers and sand catcher (supplied by the driller)
- New sample liners (supplied by the drilling contractor).

- New sample liner end caps (supplied by the drilling contractor).
- Disposable sampling gloves to prevent exposure to soil and water as well as the prevention of crosscontamination.
- Sealable plastic bags for sample storage.
- Laboratory supplied glass soil sample jars and labels (optional).
- Razor blade knife for splitting open sample tubes.
- Stainless steel bowl and spoon for mixing composite samples.
- Custody seals seals to be placed on sample containers to maintain sample integrity.

5.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

6.0 Monitoring Well Equipment

Monitoring well equipment shall be supplied by the drilling contractor.

- Well screen materials and intervals to be based on site conditions or specified in Workplans and/or Sample analysis plans. Screen size is to be determined based on specific site conditions.
- Well casing materials and intervals to be specified in Workplans and/or Sample analysis plans.
- Sand and/or gravel pack gradation to be determined based on site conditions.
- Betonite well seal to provide annular well seal.
- Concrete for well surface seal.
- Locking standpipe to protect well assembly.
- Water proof locking well cap to seal well and tamper prevention.
- Total depth probe to measure the total depth of the open borehole and/or monitoring well annular pack.
- File to cut a datum notch in the top of the well assembly.

7.0 Decontamination

All samples shall be collected using Decontaminated equipment. Decontamination procedures are detailed in RMC SOP 6.

8.0 Soil Sampling Procedures

Samples will be driven at intervals specified in the work plan. At a minimum, samples will be driven at 5 foot intervals, if lithologic data is needed. If loose, unconsolidated soils are encountered, a sand catcher will be placed at the end of the sampler so that unconsolidated soils are not lost as the sampler is retrieved from the borehole. The sampler will be advanced by blows from a 140-pound downhole hammer. The number of blows required to drive the sampler 6 inches will be recorded on the Soil Boring Log Form. The amount of recovery for each sample driven will be recorded on the soil boring log form.

Each site-specific sampling plan will identify the appropriate sample containers used to collect soil samples. If sample analytes do not include volatile or semi-volatile organic compounds, laboratory supplied glass jars or plastic bags may be used. Otherwise, samples should be submitted in brass or plastic (for inorganic analyses) sleeves.

Sleeves in the sampler will be separated using a stainless steel putty knife and the soil between the sleeves will be carefully cut so that the soil within the sleeve is flush at each end. Each sleeve will be sealed with an end cap. Each sleeve will be labeled with the sample identification and immediately placed in an iced cooler to maintain a temperature of 4°C. The remaining sample(s) will be used for soil classification. Samples may be removed from the sleeves for the mixing of composite samples.

9.0 Soil Boring Abandonment Procedures

Soil borings not used for well installations will be backfilled. If water is not encountered in the boring, the boring will be backfilled with drill cuttings. If water is encountered, the saturated portion of the boring will be backfilled with granular bentonite. Cuttings will be used to backfill the remainder of the boring. Borings that were drilled through asphalt or concrete will be patched to match existing conditions.

10.0 Storage and Disposal of Drill Cuttings

Drill cuttings and unused soil samples will disposed of on-site within the tailings impoundment.

11.0 Monitoring Well Installation

Monitoring well installation will occur in completed soil borings according to the procedure detailed below:

- 1: A soil boring shall be drilled to the anticipated total depth of the monitoring well.
- 2: The center tube and bit shall be removed from the auger assembly.
- 3: If flowing and/or heaving sands are encountered a center plug shall be used. If a center plug is required the auger assembly shall be removed from the hole and a new wood or plastic center plug will be placed at the base of the bottom section of auger. The auger will then be redrilled to the total depth of the borehole.
- 4: The monitoring well will assembly will be assembled and lowered into the center of the auger until the well is resting on the bottom of the borehole. The well casing will installed so that the top of the well assembly is approximately two to three feet above the ground surface. The well assembly will be handled using clean disposable gloves. If a center plug is used the well shall be lowered until the well assembly is resting on the center plug. The well will then be lifted slightly and dropped to release the center plug.
- 5: The sand/gravel pack will be poured into the annular space between the well assembly and the inner wall of the auger assembly. The sand/gravel pack shall be poured in three foot intervals. A Decontaminated total depth probe shall be used to measure the depth of the sand/gravel pack. Upon the completion of a three foot section of sand/gravel pack the auger shall be lifted two feet. This will allow the sand pack to fill the annular space between the walls of the borehole and the well assembly while keeping a portion of the sand/gravel pack inside of the auger assembly. This will prevent the collapse of the borehole and assuring the complete filling of the annular space between the borehole and monitoring well assembly. The sand/gravel pack installation shall continue until the sand/gravel pack is two feet above the top of the well screen.
- 6: Upon the completion of the sand/gravel pack an annular bentonite well seal shall be installed. The annular well seal will consist of bentonite pellets or chips. The bentonite seal shall be installed using the same procedure as outlined above for the sand/gravel pack. The bentonite well seal shall be installed to a depth of two feet below ground surface.
- 7: Upon the completion of the bentonite well seal, a cement surface seal and stand-pipe shall be installed. A steel stand-pipe shall be inserted into the bore hole to a depth of two feet. The stand-pipe shall contain a locking cover. The standpipe and cover assembly will be used to prevent unauthorized access to the well. The cement well seal shall be installed to ground surface in the annular space between the well casing and the inner wall of the stand-pipe. Cement will also be placed in the annular space between the outer wall of

the stand-pipe and the wall of the borehole. The outer cement seal shall be configured to form an apron that slopes away from the well and hence aids in the prevention of surface water runoff flowing into the well.

- 8: Upon the completion of well construction a V-shaped notch shall be cut into the top of the well casing. This notch shall act as a permanent datum point for surveying. The stand-pipe shall be locked upon the completion of well construction activities.
- 9: The well shall be surveyed according to the datum requirements specified in individual Workplans and/or Sample Analysis Plans.

12.0 Labeling

Each sample will be labeled with the following information:

- Sample identification;
- Project number/name;
- Analyses requested;
- Date/time collected; and
- Samplers initials.

13.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheets. Field notes shall include all pertinent information including but not limited to:

- Date and time samples were collected;
- Physical description of sample area;
- Lithologic descriptions of soils encountered;
- Identification of samples collected;
- Total number of samples collected per sampling event;
- Total number of samples collected from each sample location;
- · Physical description of samples;
- Preservatives used for samples;
- Sample container types;
- Analysis to be performed;
- Well construction details;
- Weather conditions:
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

Sample handling and Chain of Custody documentation shall be in accordance with RMC SOP 5 found in this document.

14.0 Demobilization

After the site has been cleaned and restored as close to its original condition as possible. All drilling and sampling equipment will be Decontaminated with a steam cleaner prior to drilling and sampling the next soil boring.

SOP 3B STANDARD PROCEDURES FOR MONITORING WELL DEVELOPMENT

1.0 Purpose

This SOP describes the procedures that will be used for developing monitoring wells after installation activities have been completed. Monitoring well installation procedures are detailed in RMC SOP 3a. Well development ensures that drilling fluids and/or sand pack materials are removed from the well prior to sampling and that water from the aquifer enters the well as designed.

2.0 Equipment

- Decontaminated pump/bailer or surge block.
- Direct reading instruments field instruments to measure pH, conductivity and temperature.
- Water level probe to measure water level.
- Total depth probe to measure total depth of well.
- Disposable sampling gloves to prevent exposure to water and the prevention of cross-contamination.
- Field notebook for recording field data.
- Clean new twine for lifting bailer and/or surge block.

3.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

4.0 Procedure

After the monitor well has been installed the well will require development to ensure that all materials introduced during installation are removed and that water entering the well is representative of the aquifer.

Measure the total depth of well with sounding device, measure standing water level and determine well bore volume (V):

V in gallons = $\pi r^2 h \times 7.48$

Where $\pi = 3.14$

r = radius of well casing converted to feet

and h = Water level - total depth of well (determined from drillers log or previous well sounding)

Purge three (3) well volumes of water from the well and measure pH, conductivity and temperature from the 3rd well volume. Continue to purge the well until there are three consecutive readings from the field measurements that have similar values and the water is clear and the turbidity is low. The pH, conductivity and temperature should stabilize when the well is properly developed.

5.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheets. Field notes shall include all pertinent information including but not limited to:

- Water level at start and end of development activities;
- Calculated well volume:
- Log of field pH, temperature and conductivity readings;
- Physical characteristics of water (color and turbidity) during development process;

6.0 Fluid Disposal

All well development purge water shall be disposed of on-site within the tailings impoundment.

7.0 Decontamination

Clean well development equipment according to procedures outlined in RMC SOP 6.

8.0 Demobilization

After Decontamination, sample equipment will be stored in the appropriate, clean containers. Any equipment that suffers damage or excessive wear while conducting sampling will be labeled and reported to the equipment manager for the necessary maintenance, repair and/or replacement.

SOP 3C STANDARD PROCEDURES FOR GROUNDWATER SAMPLING

1.0 Purpose

This SOP describes the procedures that will be used for collecting groundwater samples. Samples will be collected with a new disposable bailer and/or a Decontaminated downhole pump. Specific monitoring well locations will be determined from the project work plan.

2.0 Sampling Equipment

- Field data sheets / Field notebook / Chain of Custody Forms Documentation of sample activities, field notes and sample custody.
- Sample containers Containers provided by laboratory for the collection, storage and transportation of samples.
- Direct reading instruments field instruments to measure pH, conductivity and temperature.
- Disposable sampling gloves to prevent exposure to water and the prevention of cross-contamination.
- Custody seals seals to be placed on sample containers to maintain sample integrity.
- 0.45 um filter apparatus with inert filters for filtering samples in preparation for the analysis of dissolved metals.
- Nitric acid (HNO₃, supplied by the analytical laboratory) for sample preservation.
- Distilled water for rinsing direct reading instruments.
- Water level probe to measure water level
- Disposable bailers to extract water from monitoring wells
- Clean new twine to lift bailers out of wells.
- Downhole pump if required for deep wells.
- Water level probe to measure water level.
- Field notebook for recording field data.

3.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

4.0 Procedure

Unlock and open the well, obtain a water level by inserting a Decontaminated water level probe into the well and measuring the standing water surface to the established datum point on the top of the well head. The established datum point can be installed by using a file to insert a notch in the PVC casing.

Purge the well with appropriate water removal device (Decontaminated bailer/pump or disposable bailer). A total of three well bore volumes of water are normally removed.

Determine the well volume (V) by the following formula:

V in gallons = π^2 h x 7.48

Where $\pi = 3.14$

r = radius of well casing converted to feet

and h = Water level - total depth of well (determined from drillers log or previous well sounding)

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Pump or bailer discharge during purging is directed to a bucket or container to determine purge rate.

Samples are collected after a sufficient purge volume is withdrawn. Bottles are filled directly from discharge from the well or from another clean container.

After the bottles are filled, the appropriate preservatives are added, if required. The pH level in samples will be verified using pH paper before bottles are sealed.

If dissolved metals analysis is required, filtration is required and the samples will be field filtered. The field filtering methodology will include the following steps:

- 1: Sample shall be collected in a 1000 ml bottle.
- 2: Sample is poured into the top of the disposable plastic filter.
- 3: Vacuum pump is attached to the filter and pumped.
- 4: When the bottom compartment of the filter is full, the water is to be transferred into a 500 ml sample container which shall be rinsed three times, the sample will be preserved with 2 ml of nitric acid (HNO_3), sufficient to bring the sample to pH < 2.
- 5: The pH level in samples will be verified using pH paper before bottles are sealed.

5.0 Labeling

Each soil sample will be labeled with the following information:

- Sample identification;
- Project number/name;
- Analyses requested;
- Preservatives;
- Date/time collected; and
- Samplers initials

6.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheets. Field notes shall include all pertinent information including but not limited to:

- Date and time samples were collected;
- Physical description of sample area;
- Lithologic descriptions of soils encountered;
- Identification of samples collected;
- Total number of samples collected per sampling event;
- Total number of samples collected from each sample location;
- Physical description of samples;
- Preservatives used for samples;
- Sample container types;
- Analysis to be performed;
- Well construction details;
- Weather conditions;
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

7.0 Fluid Disposal

All well development purge water shall be disposed of on-site within the tailings impoundment.

8.0 Decontamination

If cross contamination of sampled wells is a potential problem, the following procedure should be followed:

- 1. Decontaminate equipment according to RMC SOP 6.
- 2. Design sampling to proceed from best quality water to the poorest quality water.
- 3. If a pump is used rinse the pumping apparatus if well yields are too low to allow sufficient water to purge the pump.
- 4. Use one disposable bailer for both purging and sampling per well.

9.0 Demobilization

After Decontamination, sample equipment will be stored in the appropriate, clean containers. Any equipment that suffers damage or excessive wear while conducting sampling will be labeled and reported to the equipment manager for the necessary maintenance, repair and/or replacement.

SOP 4 STANDARD PROCEDURES FOR COLLECTION OF STREAM SEDIMENT SAMPLES

1.0 Purpose

This SOP describes the procedures that will be used for sampling stream sediment to a maximum of 18 inches below surface. Samples will be collected with a decontaminated shovel, stainless steel spoon or hand auger/probe. Specific soil sampling locations will be determined from the project work plan.

2.0 Sampling Equipment

- Field data sheets / Field notebook / Chain of Custody Forms (COC) Documentation of sample activities, field notes and sample custody.
- Hand Auger/Probe (if necessary) for the collection of at-depth samples.
- Shovels for the collection of near-surface samples.
- Log forms / Field notebook / COC for field documentation.
- Sample containers for sample storage and transportation.
- Stainless steel mixing bowl for mixing composite samples.
- Stainless steel sample spoons for the collection of surface samples and mixing composite samples.
- Disposable sampling gloves to prevent exposure to soils and water and the prevention of crosscontamination.
- Custody seals seals to be placed on sample containers to maintain sample integrity.

3.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

4.0 Procedure

Sediment samples shall be collected as discrete grab samples or composite samples. Sample types and frequency shall be specified in the Sample and Analysis Plan.

4.1 Discrete Samples

If water samples are being concurrently sampled with stream sediment samples the water samples will be collected prior to the collection of the sediment samples. Sediment samples will be collected from streambeds with standing water or slow flow rates such that there will be no significant impact while sampling. Vegetation, rocks, and/or debris will be scraped away from the sample location with a shovel or stainless steel spoon. The underlying sediment will then be collected and placed into sample containers with a stainless steel spoon or gloved hand. Composite samples will be homogenized as described below. Coarse grained soils, gravel and rock fragments will be removed wherever possible.

4.2 Composite Samples

Composite samples will be collected (as described above) by placing sub samples into a stainless steel mixing bowl or a clean plastic bag, or by hand with new, clean sampling gloves. The sample will be homogenized with a stainless steel spoon or gloved hand. The homogenized soil will be packaged in a laboratory-supplied sample container, labeled and placed in a cooler to maintain temperature.

5.0 Labeling

Each soil sample will be labeled with the following information:

- Sample identification;
- Project number/name;
- Analyses requested;
- Date/time collected; and
- Samplers initials.

6.0 Documentation

Field activities shall be recorded in a hard bound field notebook and field data sheets. Field notes shall include all pertinent information including but not limited to:

- Date and time samples were collected;
- Physical description of sample area;
- Lithologic descriptions of soils encountered;
- Identification of samples collected;
- Total number of samples collected per sampling event;
- Total number of samples collected from each sample location;
- Physical description of samples;
- Preservatives used for samples;
- Sample container types;
- Analysis to be performed;
- Well construction details;
- Weather conditions:
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

7.0 Demobilization

After Decontamination, sample equipment will be stored in the appropriate, clean containers. Any equipment that suffers damage or excessive wear while conducting sampling will be labeled and reported to the equipment manager for the necessary maintenance, repair and/or replacement.

SOP 6 STANDARD PROCEDURES FOR DECONTAMINATION

1.0 Purpose

This SOP details the Decontamination protocols for sampling equipment. In order to reduce the risk of transferring materials from one sample site to another, and to assure that there is no cross-contamination of samples, the following procedures will be used.

2.0 Decontamination Equipment

- 5 gallon buckets For washing and the collection of rinsate.
- Alconox Soap
- Scrub brushes For cleaning sampling equipment.
- Distilled water For final equipment rinse.
- Culinary tap water for equipment rinse.
- Garbage bags for clean equipment storage.

3.0 Decontamination Procedures

RMC uses the following decontamination procedure for equipment:

1. Gross contaminant removal

This step involves scrubbing the equipment using an Alconox and water solution and a stiff scrub brush. The scrubbing will continue until all visible contaminants are removed from the equipment. This water will be changed as necessary. The Alconox and water solution is typically prepared and stored in a clean 5-gallon bucket.

2. Clean detergent wash

This step involves using a clean volume of Alconox and water solution. Equipment will be washed in this solution once all gross contaminants have been removed during Step 1. This solution will also be changed as necessary. The Alconox and water solution is typically prepared and stored in a clean 5-gallon bucket.

3. Clear water rinse

This step involves rinsing the equipment in clear, culinary tap water. This water will be changed as necessary to maintain its purity. The water solution is typically collected and stored in a clean 5-gallon bucket.

4. Distilled water rinse

Distilled water will be used as a final rinse for all Decontamination procedures. The water will be poured from a new container, or sprayed from a suitable container or the equipment will be submerged in a suitable container. Decontamination (equipment) blanks will be collected as required in the Sampling and Analysis Plan. The water solution is typically collected and stored in a clean 5-gallon bucket.

4.0 Rinsate Blank Sample Collection

The collection and analysis of rinsate (equipment) blanks is intended to provide information on the cross-contamination and contamination potential introduced by sampling equipment and methods. Rinsate blank sample collection will be performed after decontamination procedures. Rinsate blank sample frequency is

specified in the Sampling and Analysis Plan. The rinsate blank sample is collected by running distilled over or through a piece of decontaminated equipment.

The rinsate blank sample collection procedure is:

- 1. Place the equipment in a catch basin. The catch basin should be made by cutting the top of a 1 liter sample bottle with a decontaminated knife.
- 2. Pour distilled water over or through the piece of decontaminated samples. To avoid dilution, pour and collect only enough rinsate required for analysis.
- 3. Transfer the water sample into a laboratory supplied sample container. Refer to RMC SOP 1 for the handling and preparation of the sample.
- 4. Label the sample with a sample identification that does not identify the sample as a blank.
- 4.0 Decontamination fluid disposal

Decontamination fluids shall be disposed of on-site in the tailings impoundment area.

SOP 5 STANDARD PROCEDURES FOR SAMPLE HANDLING AND DOCUMENTATION

1.0 Purpose

This section describes the handling and documentation procedures that will be used once soil and water samples are collected. The procedures will ensure that samples are handled properly and that appropriate documentation is completed.

2.0 Sample Handling

All samples will be promptly placed into a cooler to maintain a temperature of 4°C. Typically, samples selected for chemical analysis will be delivered at the end of each day to the analytical laboratory. If they are not submitted to the laboratory on the same day, they will be stored in a refrigerator in a locked storage room until they can be delivered to the laboratory.

3.0 Sample Identification and Labeling

Soil samples will be labeled in such a way as to identify the area and depth from which they were taken. Water samples will be labeled as to identify when and where they were collected from. Duplicate samples will always be labeled in the same manner such that the laboratory cannot tell they are duplicate (i.e., as a "blind duplicate"). Each sample container will be immediately labeled with the following information:

- Project name
- Project number
- Sample identification
- Date and time collected
- Analysis requested
- Filtered or unfiltered (water)
- Samplers initials
- Preservative used (water)

This information will also be recorded in the field logbook and on a Field data sheet.

4.0 Custody Seals

Custody seals shall be used to prevent tampering and to maintain sample integrity. A seal shall be placed across the top of sample jars or across the seals of plastic sample bags. The seal shall placed on the sample container directly preceding sampling and will be signed and dated by the sampler who collected the sampler.

5.0 Chain-of-Custody (COC)

COC documentation will begin in the field for each sample submitted to the laboratory and will also be maintained by laboratory personnel. Samples that are submitted to AEC will use the COC provided by AEC. COC forms will be requested from labs other than AEC prior to sample collection. A COC for each sampling event will be completed and will accompany each sample batch to the analytical laboratory. Sample custody means that all samples will remain in the possession or observation of the sampler at all times, or in a locked facility until delivery to the analytical laboratory. A sample COC form is provided in Appendix D. Copies of the COC forms shall be stored in a three ring binder for sample tracking.

6.0 Field Book

RMC field personnel will maintain a field logbook to record all field activities. The field logbook will be a weather-resistant bound field book. All data generated during the project and any accompanying comments will be entered directly into the logbook in indelible ink; any corrections will be made with single line-out deletions. At no time will any pages be removed from the field logbook.

Each day's field activities will be documented, including the following minimum information:

- Date of field activity:
- Time of field activity;
- RMC field personnel's initials;
- Project name;
- Project number;
- Date and time samples were collected;
- Physical description of sample area;
- Identification of samples collected;
- Total number of samples collected per sampling event;
- Total number of samples collected from each sample location;
- Physical description of samples;
- Preservatives used for samples;
- Sample container types;
- Filtered vs. Unfiltered samples (water);
- Analysis to be performed;
- Weather conditions:
- Hand sketches of subject area(s); and
- Description and date of any photograph(s) taken.

7.0 Field Data Sheets and Borehole/Monitoring Well Logs

Field data sheets will be used to collect and organize data in the field. Field data sheets will be completed for each sample location. Completed originals of the field data sheets shall be placed in a three ring binder and numbered chronologically in a three ring binder to be archived in RMC's office. A set of copies will be stored in a three ring binder for on-site reference.

Borehole/monitoring well logs will be completed for each borehole and monitoring well completed.

Borehole/monitoring well logs and Field data sheets for soil and groundwater sampling are included in Appendix A of this SOP.

8.0 RMC Sample Logbook

RMC will maintain a sample logbook, which will track all samples collected and/or accepted by RMC. The logbook will provide a unique, six digit alphanumeric identifier that will be assigned to each sample collected. All samples collected will be assigned an identifier number, regardless of that samples' submission to a laboratory. The next available chronological number in the sample logbook will determine the identifier, and this number will be cross-referenced with a sample description number, assigned in the field.

The RMC Sample logbook will be a covered, bound journal with non-removable pages. At no time will any pages be removed from the sample logbook.

All entries into the sample logbook will be made in indelible ink; and all corrections shall consist of initialed, line-out deletion. Data contained therein will include:

- Unique identifier number;
- Date;

- Project number;
- Sample description number; Sampler initials; and Lab acceptance initials.

Appendix A

Borehole/Monitoring Well Log Soil Sample Field Data Sheet Groundwater Filed Data Sheet



A CONTRACTOR OF THE CONTRACTOR	· · · · · · · · · · · · · · · · · · ·
CLIENT	BORING ID.
PROJECT.	START DATE:
LOGGED BY:	END DATE

CONSULTANTS					
Graphic Log Geotechnical		Samples	LITHOLOGIC DESCRIPTION/ COMMENTS		
Depth LITHOLOGY					
O	BLOW COUNTS PER USCS 6" CLASS	INTERVAL RECOVERY 33			
5 —					
10					
			• .		
- 75 			· · · · · · · · · · · · · · · · · · ·		
-					
20 -					
25_					
70					
30 <u>Borin</u> Total Depth.	-	NOTES:			
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The second secon			
Client:	Boring/Monitor Well ID:		
Project Name.	Start Date. / /		
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Loaged By	Location:		
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CONSULTANTS				Static Depth to Water:			
G	raphic Log	Geotechnical	Samples	LITHOLOGIC DESCRIPTION/ COMMENTS			
FEET	WATER LEVEL AND	BLOW COUNTS PER USCS 6" CLASS	INTERVAL RECOVERY S and				
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5 -							
10 -							
20 -							
25							
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Depth to	o 1st Water:			Annular Seal #1:			
Abandor	nment.			Annular Seal #2:			

RESOURCE MANAGEMENT CONSULTANTS, INC. SOIL SAMPLING DATA SHEET

PROJECT:			DATE:			
PROJECT:SAMPLED BY:			SAMPLE ID:			
TIME: SAMPLE TYPE:			DUPLICATE ID:			
			RMC SAM	PLE ID:		
WEATHER (CONDITIONS: _	<u>. </u>				
SAMPLE MI	ETHOD <u>:</u>					
SITE DESCR	LIPTION:					
FIELD DAT						
GROUND CO COLOR:	OVER:	4	· · · · · · · · · · · · · · · · · · ·			
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				· · · · · · · · · · · · · · · · · · ·		
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LABORATO	RY INSTRUCTI	ONS:				
SPLIT AND	ARCHIVE SAM	PLES:				
	S SAMPLED:					
INTERVAL	RMC SAMPLE ID	COLOR	SOIL TYPE	PARAMETERS		
			**	<u> </u>	<u> </u>	
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RESOURCE MANAGEMENT CONSULTANTS, INC. WATER SAMPLING DATA SHEET

PROJECT	·		_ D.	ATE:	
SAMPLEI) BY:		SA	AMPLE ID:	
TIME:			_ D	UPLICATE ID:	
SAMPLE '	TYPE:		_ RI	MC SAMPLE ID:	
WEATHE	R CONDIT	IONS:			
SAMPLE	METHOD <u>:</u>		741		
SITE DES	CRIPTION				<u>—</u>
	RAMETE		C(OND:	_
COLOR:_		TURB	IDITY:		
FLOW DA	ATA:				
					_
WELL DA	ATA:				
TD.		CWT ·	DI IR GE N	ÆTHOD:	
VOLUME		5 W L	_ FORGE N	ETHOD.	
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and n = w	ater level – to	nai deput of well	(determined from ari	lers log or previous well sounding)	
=	3.14 X		² X	X 7.48	
V (GAL)	= π	RADIUS (FT) ²	WATER LEV	X 7.48 EL-WELL DEPTH (FT)	
	COLLEC			In parettee	
QUANTITY	SIZE (MI)	FILTRATION	PRESERVATIVE	PARAMETERS	
			<u> </u>		
				7	

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Tabbed Page: Appendix C

Appendix C Sample Chain of Custody

AEC LABORATORIES

Laboratory Services Request Form

CLIENT I	NFORMATION				SEND REQUESTS	S TO:
Client Nan	ne:				AEC LABORAT	TORIES
	ress:				3422 South 700) West
					Salt Lake City,	UT
Client Pho	ne:	_			84119	
Client Fax					Your Customer S	ervice
	IT INFORMATION				Representative is	::
Account N	ame:				Maureen Ottley	
	ddress:				Phone # (801) 261-1426	
ŀ					Fax # (801) 264-98	B38
P.O. No: _		<u> </u>				
	T INSTRUCTIONS			·		
Report Re	sults To:	···.				
1	dress:					
Please Fo	rward Results By: US Mail ()	Fed Ex () Fa	x () O	her	
Services F	Requested below are required no later than				(date)	
IV. TYPE O	F SERVICE REQUESTED					
lease anal	yze the enclosed environmental samples for:					
Lab Use		Sampling	No.			
Only	Field Sample	Date & Time	of		Analysis	
Lab No.	No./Description		Cont.		Requested	
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Relinquished	by:	Date	Time	<u> </u>	Airbill #	
Received by:		Date	Time		Custody Seal Intact?	
Received for	lab by:	Date	Time		Yes	No

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Appendix D Glossary

GLOSSARY

Accuracy: Ensures how close the results are to a true or expected value and can be determined by comparing the analysis of a standard or reference sample to its actual value.

Aliquot: A measured portion of a sample taken for analysis. One or more aliquots make up a sample.

Bioavailability: A characteristic of a chemical that describes its ability to be absorbed into the tissues or circulatory system of a human or ecological receptor. Bioavailability of metals may be affected by the specific mineral or chemical form and its solubility, environmental pH, the rock matrix, other metals present, the presence of organic material in soil or the receptor's gut, and other factors.

Biota: Any living organisms, including plants, mammals, birds, invertebrates, bacteria, and so forth.

Blind Sample: A type of sample used for quality control purposes, a blind sample is a sample submitted to an analyst without their knowledge of its identity or composition. Blind samples are used to test the laboratory's expertise in performing the sample analysis. CLP: The EPA's Contract Laboratory Program. The CLP provides analytical services to the 10 EPA Regions through contracted commercial laboratories.

Comparability: The extent to which data can be compared between sample locations or periods of time within a project, or between projects.

Completeness: The comparison between the amount of valid data originally planned to be collected, versus how much was collected.

Concentration: Defined as high, medium, or low, and used to determine how much volume is collected or the analytical protocol to be followed.

Data quality objectives (DQOs): Quantitative and qualitative statements describing the degree of the data's acceptability to the data user(s). They include indicators such as accuracy, precision, representativeness, comparability, and completeness. DQO's specify the quality of the data needed in order to meet the project's goals. The planning process for ensuring environmental data are of the type, quality, and quantity needed for decision making is called the DQO process.

Data turnaround time: The maximum length of time allowed for laboratories to submit analytical data to EPA in order to avoid liquidated damages. Data turnaround time begins at the validated time of sample receipt (VTSR) at the laboratory.

Detection limit: Applied to both methods and equipment, the lowest concentration of a

target analyte that a given method or piece of equipment can reliably ascertain and report as greater than zero.

Duplicate sample: Used for quality control, two samples taken at the same time from, and representative of, the same site that are carried through all assessment and analytical procedures in an identical manner. Duplicate samples are used to measure natural variability as well as the

precision of a method, monitor, and/or analyst. More than two duplicate samples are referred to as replicate samples.

Equipment or rinsate blank: Used for quality control, types of field blanks used to check specifically for carryover contamination from reuse of the same sampling equipment (see *field blank*).

Exposure pathway: The route by which a chemical moves from a chemical source to a living receptor. If the chemical cannot reach the receptor, the pathway is incomplete and poses no risk. If the chemical reaches the receptor but in low enough concentrations, risk may be insignificant.

Field blank: Used for quality control, a field blank is a "clean" sample (e.g., distilled water) that is otherwise treated the same as other samples taken from the field. Field blanks are submitted to the analyst along with all other samples and are used to detect any contaminants that may be introduced during sample collection, storage, analysis, and transport.

Fraction: A specific subunit of an analytical protocol. For example, for low/medium organics, the fractions are volatiles, semivolatiles, and pesticides/Aroclors.

Instrument detection limit: The lowest concentration of a given substance or analyte that can be reliably detected by analytical equipment or instruments (see also detection limit).

Matrix: A matrix is a specific type of medium, such as water, soil, or sediment, in which the analyte of interest may be contained.

Method detection limit (MDL): The MDL is the lowest concentration of a given substance or analyte that can be reliably detected by an analytical procedure (see detection *limit*).

Precision: The degree of agreement among repeated measurements of the same characteristic. It may be determined by calculating the standard deviation, or relative percent difference, among samples taken from the same place at the same time.

Preservative: A chemical added to inorganic and volatile water samples to maintain the integrity of the sample. Some common preservatives include nitric acid, hydrochloric acid,

and sodium hydroxide.

Quality Assurance (QA): Refers to the overall management system which includes the organization, planning, data collection, quality control, documentation, evaluation, and reporting activities of a particular project. QA is designed to ensure that a product or service meets defined standards of quality with a stated level of confidence.

Quality Assurance Project Plan (QAPP): A formal written document describing the detailed *quality control* procedures that will be used to achieve a specific project's data quality requirements.

Quality Control (QC): Refers to the routine technical activities designed to measure quality and limit error in a product or service. Since errors can occur in either the field, the laboratory, or the office, QC must be part of each of these functions.

Quality Control (QC) Samples: Samples used to estimate the precision and accuracy of analytical results in the field and in the laboratory.

Receptor: A living being (or sometimes a significant ecological habitat) that may receive an exposure to a chemical

Representativeness: The extent to which measurements actually represent the true environmental condition or population at the time a sample was collected.

Sample: A single, discrete portion of the environment collected from a specified physical location at a specific time. The single sample may be placed in multiple vessels.

Sample container: The individual bottle that contains the sample or an aliquot of the sample. The type of sample container varies for different sample fractions and concentrations.

Sample custody: Legal possession of and responsibility for a sample. Documentation of sample custody is maintained on the chain-of-custody part of the traffic report or packing list. The sample is in your custody if any of the following criteria are met: 1) the sample is in your possession or is in your view after being in your possession, 2) the sample was in your possession and then locked up or sealed to prevent tampering, or 3) you have placed the sample in a secured area.

Sample label: Taped or adhesive labels that provide the sample numbers to be assigned to the samples.

Sample number: The sample number from the sample label that identifies the sample or an aliquot of the sample.

Spiked samples: Used for quality control, a sample to which a known concentration of

the target analyte has been added. When analyzed, the difference between an environmental sample and the analyte's concentration in a spiked sample should be equivalent to the amount added to the spiked sample.

Split sample: Used for quality control, a split sample is one that has been equally divided into two or more subsamples. Splits are submitted to different analysts or laboratories and are used to measure the precision of the analytical methods.

Standard deviation(s): Used in the determination of *precision*, the most common calculation used to measure the range of variation among repeated measurements. The standard deviation of a set of measurements is expressed by the positive square root of the variance of the measurements.

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Appendix E
Previous Site Studies

Previous Site Studies

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Appendix F Laboratory QA/QC Documents

Section 1.1 Frontier Geosciences' Quality Assurance

1.1.1 Frontier Geosciences' QA Policy Statement

Quality Assurance (QA) is a system for ensuring that all information, data, and interpretation resulting from an analytical procedure are technically sound, statistically valid and appropriately documented. Quality Control (QC) is the mechanism used to achieve quality assurance.

Frontier Geosciences Inc. (Frontier) has a strong commitment to quality assurance, both at the bench and the management level. Frontier realizes that without quality control, data may become suspect and of less value to our client. Frontier is therefore dedicated to producing data of highest quality, usability, and coherence.

Data quality is achieved through Frontier's Data Quality Objectives (DQO's). Our DQO's consist of five components: precision, accuracy, representativeness, comparability and completeness (PARCC).

• Precision is a measure of how repeatable data is and is often measured by sample replicates.

• Accuracy is a measure of how close the data is to the actual, or real value,

measured by certified reference materials and matrix spikes.

 Representativeness is a measure of how representative a sample is to the sample population and is achieved by accurate, artifact-free sampling procedures and appropriate sample homogenization.

• Comparability looks at ongoing projects and how variable one set of data is to another. Comparability helps to measure the scientific coherence of the

system to past work.

Completeness is a measure of how many data points collected are usable;
 Frontier considers 95% usable data to be an acceptable value for completeness.

Frontier routinely provides data packages in one of three QA formats. The first, called "Screening Level", is equal to US EPA Level 1. "Research Level" is between US EPA Levels 2 and 3, while the third, called "Litigation Level" is approximately equivalent to the US EPA Level 4. In addition, Frontier will provide custom QA/QC packages to meet the individual needs of the client. The various QA/QC levels above "screening level" do not represent differences in analytical data quality, but rather, the degree of documentation provided, and therefore the ability to defend the data in legal proceedings. The quality of the data produced under QA reporting schemes above the screening level, as measured by quantitative indicators such as precision, accuracy, and detection limits, are equivalent.

1.1.2 Frontier Geosciences' Quality Assurance Policy

Frontier recognizes that accurate and precise data depends upon an effective and consistent QC program. Frontier's program is implemented collaboratively by the entire laboratory group and subscribes to the following basic tenets:

<u>Sample integrity must be preserved.</u> Integrity is preserved by following documented sample handling procedures for the preservation, custody, storage, labeling and record keeping of samples received by the laboratory.

Trace metal-free ("ultra-clean") sample handling must be employed. All samples to be analyzed for low level or ambient metals concentrations are handled according to protocols: including the use of our class-100 clean room, wearing of clean room gloves, and using only pre-tested and approved reagents, water, and equipment. High-level (contaminated) samples are kept segregated from ultra-clean samples during storage and sample preparation.

Approved analytical methods must be followed. The analyst's fundamental understanding of analytical methods is paramount for effective, first-defense QC. Emphasis on understanding and following the correct methods is part of every analyst's training. QC results from each method are evaluated to identify and correct method weaknesses, and to detect any need for further training.

Analytical instrumentation must be in proper working order. Optimum instrument performance is assured by the use of daily calibration and performance evaluation samples. Rigorous preventative maintenance is performed on a regular basis and is well documented.

Raw data must be properly reduced and accurately transcribed into the correct reporting format. Various levels of data review, from acquisition to the final report, are performed to minimize error.

The laboratory-specific precision and accuracy of analytical methods must be documented and monitored continuously. Accuracy and precision are monitored and compared to historical data from Standard Reference Materials (SRM's). All data is scrutinized according to our scientific understanding of the biogeochemistry of the particular situation. The scientific coherence of the data set is considered to be as important a QA parameter as precision or accuracy.

Section 1.2 Corporate Ethics Policy on Fraud, Waste, and Abuse

1.2.1 Environmental Responsibility

As an environmental science research company and specialty laboratory, Frontier aspires to be a model of environmental responsibility. Thus, employees should be ever-vigilant to avoid waste, conserve resources, reuse, recycle, and minimize the production of hazardous wastes. The procedures used in a modern scientific laboratory often employ toxic materials and disposable items. Therefore, employees are encouraged to develop innovative ways to minimize the production of these materials, while finding safe ways to reuse disposed of items (plastic bags, bottles, unused chemicals, etc.), once their laboratory usefulness is finished.

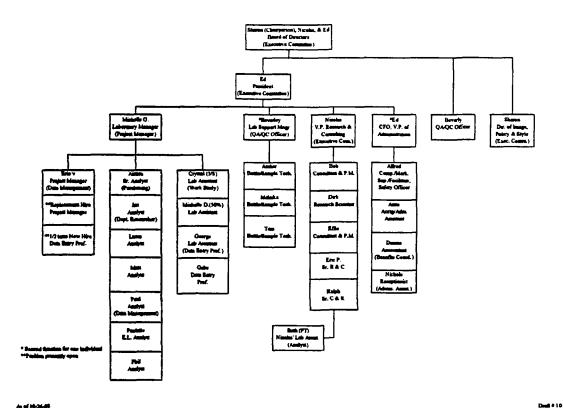
1.2.2 Intellectual Honesty

Frontier performs environmental research for government, industry, academic institutions, and environmental activist groups. Hence, it is important for the company's credibility that all reported results and interpretations be objective and honest. Although individuals within the company may differ on the political implications of various results, Frontier must remain above this in its data and research. All obtained results must be reported with complete honesty, with no regard to the expected, or preferred (by client or researcher) outcome. In the case of researcher-selection of "best" or "most accurate" data from an analytical set, all assumptions used in choosing data or rejecting data, must remain readily available to the sponsor and/or outside review. Fabrication of data or its deliberate misinterpretation is considered grounds for immediate employment termination.

All employees are required to adhere to Frontier's Intellectual Honesty policy. Any employee witnessing an act that may be considered to go against the Intellectual Honesty policy is required to report the instance to management. The reporting employee may maintain anonymity. Several avenues are available to employees wishing to report suspect behavior: they may directly confront the offending employee, they may report the instance to the employee's direct supervisor, the lab manager, the president or the QA officer. There may be times when the decision to reject data is not absolutely clear. Any time the decision is ambiguous, the decision must be made with management consent. Unambiguous decisions need not be overseen, or explained, but all data must be presented and noted on the original data set. The important factor is that management become aware of the situation and that all employees feel comfortable reporting suspect behavior.

Section 2.1 Frontier Geosciences' Organizational Chart

Frontier Geosciences Inc.



Frontier subscribes to a relaxed hierarchy, where teamwork is more important than job status. The lab manager oversees the entire lab on a day to day basis. All technical employees report to the QA officer and lab manager about QA issues. The lab manager is able to answer questions regarding quality assurance issues if the QA officer is out. The QA officer works with the lab manager to implement and assure QA policies.

Laboratory personnel are responsible for quality control at the bench level. Frontier considers the analyst as the first line of defense. Quality control is then looked at by the lab manager. Finally, each dataset's quality control measures are reviewed by the QA officer, who is the last line of defense for quality control before the data is sent to the client. The client performs a final review of the data when he/she receives it. For projects less than level 4 QA, the samples may be discarded after the project manager and QA officer have reviewed the data, but before the client has done so (unless special provisions for sample archiving were made in advance). For litigation level QA work, any requests to re-run the sample must come within one month of report submission. If re-run results are equivalent to initial results, the client assumes responsibility for paying for the

extra work. Frontier will answer QC related questions after one year of report submission if the client is willing to pay for the time.

In addition, if any employee feels that the appropriate person is not fixing a given problem, the employee may speak to the lab manager or QA officer, enlisting their aid in getting the issue resolved.

Section 2.2 Frontier Geosciences' Personnel Qualifications

Scientific Staff	Highest Academic Degree	Responsibilities
Nicolas S Bloom, Sr. Research Scientist	MS Oceanography 1987	mentoring business dev., supervision, research
Raph R. Tumer, St. Research Scientist	PhD Oceanography 1975	research, consulting field work
Eric M. Prestbo, Recentch Scientist	PhD Chemistry 1992	research, consulting, field work
Efrosini Tsalkitzis, Research Scientist	MS Blology 1995	research, consulting, staff biologist
Dirk Waltschläger, Post-Doc Researcher	PhD Natural Sciences 1996	research, methods development
Robert C. Brunette, Research Scientist	BS Chemistry 1993	research, consulting, field operation
Michelle L. Geuftrier, Lab Manager	BS Chemistry/Physics 1992	laboratory management, project management
Eric J. von der Geeet, Project Menager	BS Chemistry & Math 1993	mercury project management, computer resource
Amara M. Vandervort, Sr. Analyst	BS Biology & Psychology 1994	sample prep and analysis supervision, field work
Jacqueline G. London, Analyst, Research Specialist	BS Chemical Engineering 1998	sample prep and analysis, research
Paul Laskowski, Anahyst	BS Chemistry 1990	sample prep and analysis, computer resource
Mett Horrobin, Analyst	BS Honors Biology 1989, PGCE	sample prep and analysis
Phil Kiner, Analyst	BS Environmental Science 1997	sample prep and analysis
Lucas Hawkins, Analyst	BS Chemistry 1993	sample prep and analysis
Paulette Jones, Entry-Level Analyst	MS Environmental Science 1997	deud ejdues
George Scriba, Lab Assistant, Data Entry Prof.	BS Environmental Science 1998	sample prep, misc. laboratory duties, data entry
Crystal Howard, Lab Assistant (student)	BS Geology 1999 (to be awarded)	sample prep, misc. laboratory duties
Gabriel Choy, Electronic Data Deliverables Specialist	AA Art 1996	data entry, peer-review of datasets
Michelle Dedman, Lab Assistant	BS Zoology 1982	sample prep, misc. laboratory duties
A. Malaika Lafferty, Technical Specialist & Sample Custodian	High School Diploma	laboratory equipment processing, sample receiving, shipping
Amber Steward, Technical Specialist & Sample Custodian	BS Geological Oceanography 1998	laboratory equipment processing, sample receiving, shipping
Thomas E. Smith, Technical Specialist & Sample Custodian	High School Diploma	laboratory equipment processing, sample receiving, shipping

Administrative Staff	Academic Degree	Responsibilities
Sharon K. Goldblatt, CEO	BS Human Development 1987	P.R. administration, marketing, editing
Ed G. Geiger, President & CFO	MBA (CPA, Retired) 1983	financial mgt., administration, marketing
Beverly A. Heaphey, Quality Assurance Officer	BA Philosophy 1990	quality assurance and control
R. Dianne Shepard, Accountant & Benefits Admin.	BA Education 1974	accounts receivable, payroll, general ledger
Anna M. Cortez, Administrative Assistant	AA General Studies 1992	administrative duties, accounts payable
Alfred Rordame, MIS, Safety Officer	BUS Choreography and Composition 1983	safety coordination, facilities mangrint, computers, graphics

Section 2.3 Facility Description and Capital Equipment

2.3.1 Facilities

Frontier's laboratory and office facilities are in downtown Seattle, Washington. The location is close to Sea-Tac Airport, and the University of Washington.

The space contains a small (100 ft²), clean room, four mercury analysis laboratories (c.a. 500 ft² each), four sample preparation rooms (c.a. 200 ft² each), two graphite furnace AA laboratories, an ICP-MS laboratory, an atmospheric

reactions laboratory, a conference room/library, and ten staff offices.

The laboratories are served by a custom-designed HVAC system with ESP pre-cleaners, providing an atmosphere that is clean and well isolated from outside dust and dirt. Each laboratory atmosphere is routinely monitored for Hg in the gas phase, and appropriate action is taken if it exceeds 25 ng/m³ in any location, or 10 ng/m³ in the clean room. Frontier uses an acid neutralization discharge systems for liquid acid-waste disposal. Disposal of toxic materials is carried out under contract to a certified disposal company.

The offices are equipped with document production equipment: laser printers, document and image processing software, color printer, large capacity collating copier, and a binding machine. A network connects personnel computers and printers for local access, as well as allowing for external email and faxes. Frontier has a Fed-Ex Powership shipping computer and access to Fedex pick-up

as late as 5:00 PM Pacific Coast time.

The entire Frontier space has been inspected and passes all city and state code requirements for fire, emissions, and low level radioactive samples.

Frontier owns all necessary equipment for ultra low level trace metal research.

2.3.2 Security

Access to Frontier offices and laboratories is regulated and limited to authorized personnel. During normal work hours, when the front office is staffed, visitors are required to check in and sign the guest list if they have not previously done so. All visitors, including delivery personnel, must be accompanied by an employee in the laboratory. The outside front door and the two back doors are kept locked at all times. Visitors must first press the Frontier button on the building's security door buzzer system outside the main entrance, identify themselves, and then be admitted by an employee-actuated electronic door lock release.

All computers are backed up five days per week onto the main system, which is stored in a fire-proof safe. Backup tapes are rotated on a five day schedule, so that at all times backups are insured for the last five business days. Once a month, a tape is taken to a safety deposit box and stored there.

Employee safety is an important concern to Frontier. In addition to the overall facility security system, Frontier also has two remote alarms. The alarms

are worn around the neck, and alert the police when activated.

2.3.3 Frontier Geosciences' Capital Equipment

Qty.	Description	Manufacturer
4	Cold Vapor Atomic Fluorescence Hg Detector	Oriel/Frontier-built
3	Cold Vapor Atomic Fluorescence Hg Detector	Tekran
4	Isothermal GC for Hg Speciation	lab-built
2	0.1 mg Analytical Balance	VWR
2	0.01 g Lab Balance	VWR
2	Class 100 Clean Air Hood (4' and 2')	Labconco
4	Class 100 Clean Bench (6' with gold Hg scrubber)	Labconco
1	Milli-Q 50 Reagent Water System	Millipore
2	Ultra-Pure Reagent Water System	US Filter
1	Large Volume Water Deionization System	US Filter
10	Macintosh Computer (networked + e-mail)	Apple
23	IBM Compatible Computer (486 processor)	Toshiba
1	Zeeman 5000 GFAAS + Hydride System	Perkin-Elmer
1	4' Hg-Free Nitrogen-Purge Glove Box	Labconco
1	Large Volume Centrifuge (250 mL bottle)	Centra
1	Tungsten Carbide Ball Mill	Spex
6	Dual Pen Chart Recorder	Kipp & Zonen
2	Chromatographic Integrator	Hewlett-Packard
7	Constant Temperature Lab Ovens	Precision
6	5 Tube Methyl Hg Distillation Units	lab built
2	Muffle Furnace	Precision
6	Refrigerator/Freezer (sample storage)	Whirlpool
5	Complete sets of fluegas sampling equipment	Frontier-built
24	Teflon [®] bulk deposition collectors	Frontier-Built
>500	Ultra-Clean Teflon® Bottles, various sizes	Nalgene
>1000	Ultra-Clean Teflon [®] Vials, various sizes	Savillex
1	Dissection Microscope	Leica
1	Specific Ion/Conductance/pH Meter	Orion
1	UV-VIS Spectrophotometer	Spectronic
1	Low Level Ozone Analyzer/Calibrator	Dasibi
10	Digital Mass Flowmeter	Sierra
1	4110 ZL Atomic Absorption Spectrometer	Perkin-Elmer
1	ICP-MS ELAN-6000	Perkin-Elmer
1	PSA "Excalibur" HGAFS system (As, Se, Sb)	PSA

Section 2.4 Frontier Geosciences' Preventative Maintenance

The responsibility for preventative maintenance belongs to the laboratory group leader. Verification that preventative maintenance is being performed is the responsibility of the QA Officer. Logbooks for each instrument are maintained, and a central file is located in the QA officer's office to document major actions. Preventative maintenance procedures are laid out according to the type of instrument.

Mercury Analyzers:

Daily - Make sure work area is clean

Check all traps and bubblers for particulate contamination

Clean bubblers at the end of the analytical day
Monitor gold trap performance for blanks,
reproducibility, recoveries

Test and/or replace degraded traps as needed

Every 3 months - Thoroughly clean analyzer

AA:

Daily - Clean contacts
Clean analyzer windows

Weekly - Compare slopes

As needed - Routine service contract for cleaning, testing and/or service

Balances:

Daily - Check with ASTM Class 1 weights Every 6 months - Certified calibration performed

Pipettes:

Weekly - Calibration checked
As needed - Pipettes are taken apart and cleaned

pH meters:

Before each use - Calibration checked Every 6 months - Certified calibration performed

Clean Hoods:

Weekly - Outside air filters cleaned Monthly - Pre-Hepa filters checked As needed - Pre-Hepa filters changed

Section 3.1 Frontier Geosciences' Corrective Action

3.1.1 Qualified Data Corrective Actions

In the event that a breech of security or other out-of-control event occurs before sample receipt at the laboratory (e.g., custody seal is broken, samples received do not match the COC, cooler temperature or preservation status not as specified in SOW or COC, etc.), the discrepancy will be noted in the receiving logbook, on the COC form, and in the project file, by the sample custodian. The project manager or laboratory manager will then be immediately notified. The manager will contact the client immediately, via fax or telephone, to decide upon a plan of action. In the intervening time until the client can respond, the samples will be held in secure storage under conditions appropriate for the sample type (i.g., in refrigerator for water samples, in freezer for frozen samples, etc.). If the client cannot be reached within reasonable sample storage time, the analyst and lab manager will use their best judgment on deciding what course of action to take.

If there is any suspicion that a sample set may contain radioactive, dangerous or toxic materials, high levels of gaseous mercury, unsterilized human or primate blood, tissues, excreta, or any other threat to the health or cleanliness of the laboratory or its staff, the sample container will be immediately closed up, placed in a secure storage area outside the building, and the safety manager immediately notified. The client will be immediately contacted via fax or telephone, to decide upon a plan of action. Under no circumstances will processing or analysis of the samples begin until the client can certify the safety of the materials. If the client cannot do so, the samples will be returned exactly as received.

3.1.2 Analytical Issues

Any data that is analytically suspect due to laboratory problems, biogeochemical improbability, poor precision or accuracy on QA samples will be qualified (flagged) and an explanation will be included in the case narrative. If the sample is re-analyzed, all results will be presented on the original data set with an explanation as to why the laboratory selected specific certain results over others. It is initially the task of the analyst to identify any out-of-control occurrences and immediately notify her/his supervisor and/or the lab manager, to obtain further instructions. If the incident appears to be a unique random occurrence (e.g., a low LCS recovery) the sample will be re-run as soon as possible. If upon re-running the sample an acceptable result is obtained, and no other out-of-control events occur in the data set, the data set will be considered valid and in control. If multiple out-of-control QC results, and poor system calibration, or any other suspected systematic analytical problems are observed, the analyst will immediately suspend further analysis, and contact the project or laboratory manager. The analyst, under the manager's direction, will then investigate the probable causes of the system failure, correct them, and then successfully re-calibrate and QC-test the system before continuing to analyze samples. Occurrences of this type will be noted in the case narrative, and all data collected while the system was out-of-control will be either discarded or reported and flagged and the instance noted in the corrective action file.

Once the data has been checked and found to be in control by the analyst, it is submitted to the project manager, who makes the formal calculations of results, notes any QC sample discrepancies (low SRM's, high blanks, etc.), and any problems generated from field QC samples (field duplicate precision), or problems related to scientific coherence (i.e., out-of-place results in a depth vs. concentration plot, methyl Hg higher than total Hg, etc.). For suspect samples, the raw data and analyst notes are first consulted to help resolve the discrepancy (i.g., is it plausible that the error is due to a sample labeling mix-up, transcription error, etc.?). If the project manager feels that one or more samples should be re-analyzed based upon his/her evaluation, a written note is placed in the project file, and a re-analysis request is made to the analyst. When the re-run results are evaluated by the project manager, she/he will make a decision as to which data to retain or flag, and that will be noted in the case narrative.

3.1.3 Data Package Verification

On litigation level QA projects, before submission of final results to the client, the raw data package and calculated results are submitted to the QA officer. The QA officer or his/her delegate checks over the package to be sure all QC samples have been run and are within internal and/or client specifications. An exception to this would be for fast turn-around-time projects, when an "un-revised" report is initially sent and then later backed up with a reviewed, revised version. At this time, approximately 5% of the calculations are rechecked for errors. If none are found, the package is cleared for final submission. If errors are found, the entire package is rechecked, and a memo of discrepancies is sent to the project manager for rectification. The project manager then writes the final report, noting any unresolvable QC issues, and qualifying any data, if necessary, in the case narrative.

On some litigation level QA data packages, the final step in the QA/QC process occurs with the independent validation of the data package by the client. If discrepancies or concerns are noted by the independent validator, they are communicated in a formal letter to the project manager, who then investigates and explains each result in a formal response. If any of the results are changed (e.g., due to calculation or transcription errors), the affected pages of the final report are modified, annotated with the revision date, and re-submitted. It is Frontier's convention to report all data to one more significant figure than is warranted by the precision and accuracy of the methods employed (typically three figures). It is further the position of Frontier that discrepancies of up to 5 units in the last significant figure, typically attributable to numerical rounding differences between Frontier and independent QC validators, need not be rectified, as they are meaningless to the interpretation of the data. This position results in significant savings in personnel resources for both Frontier and the client.

3.1.4 Sample Re-Analysis

Frontier will normally, if sufficient sample volume remains and if the degree to which the samples are out-of-control warrants, re-analyze, at our cost, all suspect results on our own initiative. Upon agreement of the sponsor, we will also re-run analyses at sponsor request, but if the re-run data supports the notion that poor results were due to natural or client causes rather than laboratory causes, the client must pay for those additional requested analyses. Further, it is our position that if data is slightly outside QC bounds, but that lower quality does not materially affect the overall data interpretation, the results should not be re-analyzed at Frontier cost, but rather flagged as approximate.

3.1.5 Corrective Action Reports

In all cases where investigation by the project manager or the QA officer results in data being changed, qualified, or in samples being re-analyzed, a written note will be placed in the client file, and a copy placed into a chronological "Corrective Action File" maintained by the QA officer. These reports are the basis for corrective actions by the project manager and her/his staff, and for notations in the project case narrative. Once the QA issue has been investigated, resolved and/or corrected, a response describing the course of action and final result is appended to the original corrective action report.

Infrequent, random, and singular out-of-control events (e.g., a single low spike recovery), which are not part of a trend, are not considered evidence that the system is out-of-control. However, the QA officer must keep a record of such events (in control charts, for example), to assure that their frequency is maintained at less than 5% (running average n>100 occurrences). Occurrences of greater than 5% random deviations from acceptable control limits in any monitored parameter are considered an "out-of-control" condition, and analysis is stopped until the cause is identified and rectified.

3.1.6 Control Charts

In addition to project-specific QC corrective actions, the project manager and QA officer will maintain records of key analytical parameters (i.e., SRM's, spike recoveries, RPD results, blanks) as control chart files. The control chart files are updated weekly, and control chart analyses conducted on a monthly basis. The QA officer or the project manager may elect to perform specific control chart analyses more frequently to ascertain if any trends in data quality are developing. Examples of data not included in control charts include: data from highly contaminated soils or other very inhomogenous materials, values close to the MDL, spikes too close to the sample concentrations, specialty (rarely used) and/or research-only methods, samples, etc.

Section 3.2 Frontier Geosciences' Laboratory Audits

3.2.1 Internal Audits

On a quarterly basis, the QA officer will conduct an internal laboratory audit, (FGS-041), via review of all control chart information, logbook entries, and client specific QA issues and corrective actions. In addition, he/she will take a detailed unannounced walk-through inspection of the laboratories, noting the QA and safety practices of the staff. Randomly requested data packages will be reviewed with the project manager to verify the accuracy and retrievability of particular data points, starting with raw data records (only litigation level data can currently guarantee this). The internal audit may also include submission of blind performance evaluation samples. These findings will then be evaluated by the QA Officer, reviewed by the laboratory manager, and discussed at a general staff meeting, with commendations and recommendations for areas of improvement. If any serious breaches of safety or QA practice (i.e., fire hazard, toxic fumes, poor calibration results, high water or air Hg levels, etc.) are noted during an internal audit or at any other time, all affected laboratory work will immediately be ceased, the health and safety officer notified and a meeting will be called to resolve the matter. Once the system has been brought back into control, routine work is allowed to continue and the incidence is noted as soon as possible in the corrective action file.

3.2.2 External Audits

Frontier views external audits as a form of free consultation and welcomes the opportunity to improve the quality of the lab. External audits are conducted at the discretion of the client, either prior to award of a contract, or as part of an ongoing laboratory monitoring process. Such audits may include submission of blind Performance Evaluation results, control chart information, data packages for complete independent validation, or a complete personal walk-through interview by the client QA representative. The laboratory may also be audited pending application for government certification, safety, or environmental regulation. Records of all such audits, their findings, and the corrective actions taken will be maintained in chronological order by the QA officer. In addition, health and safety related documentation will be retained by the health and safety officer. A copy of each project-specific audit will also be maintained by the project manager in the appropriate project file.

Performance Evaluations are performed twice a year. Samples from the Washington State PE program and from the APG program are analyzed and compared to other labs. In addition, other round-robin performance evaluations and regular performance evaluations will be conducted at the client's request or

as deemed necessary by the lab manager.

Section 3.3 Quality Assurance Reports to Management

The QA officer is responsible for preparing a quarterly report. The purpose of the report is to present QA issues to the scientific staff in a condensed, graphical representation. The report should include at a minimum: changes in quality assurance, quality control changes, audit report findings, and control charts detailing current issues of interest or problem areas. The main objective of the monthly report should always be a tool to the scientific staff, helping them recognize trends and/or trouble spots. These quarterly reports should summarize and draw attention to quality control issues, both good and bad.

Section 3.4 Lab Documentation and Forms

Lab documentation and forms are created on an as-needed basis. The creator of the form is responsible for checking to see if the new form is replacing an old form. The new form is placed into the central Form Library and if it is replacing an old form, the old form is removed and discarded. The creator of the new form is also responsible for putting the new form into the computer system's file server. If it is replacing an old form, the old form is deleted (old forms are retained in back-up files). The creator of the new form then e-mails all staff informing them of the change.

All newly created forms are given a version number and date. This will assist staff and clients in recognizing current versus old forms. All newly created forms have Frontier's name on them, as well as the revision date and form number. If it is recognized that an old form is being used, then the person who noticed it is responsible for replacing the old form with the new version. It is the responsibility of all laboratory personnel to ensure that clients are receiving the new versions.

The QA officer maintains a document management file, wherein all major forms (i.g. SOP's, QAP's and QAM's) are logged out of the office so that when new versions are created, the correct persons may receive the new copies.

Section 3.5 Sub-Contracting of Services

On occasion, Frontier finds it necessary to sub-contract services for specialty analyses that we do not perform. We do not send overflow of samples within our expertise to outside labs. When this happens, Frontier will first contact the client and inform them of the situation and obtain their permission to sub-contract the work. Frontier will then contact laboratories that it usually uses and request their services. If they are not able to meet our needs, we will look for other laboratories. In either case, Frontier will remain ultimately responsible for the samples and results. Laboratory integrity and competence are of utmost importance and our paramount concern. When using another lab's services, Frontier routinely sends, along with the samples, blind spiked samples and

blanks to double-check and enforce the quality and integrity of the other laboratory's technical work. In addition, Frontier maintains a copy of the subcontracted laboratory's quality assurance plan, performance evaluation results, and some of their benchsheets, to further ensure the quality of the other laboratory.

Section 3.6 Staff Training and Documentation

3.6.1 Hiring Process

People often begin their careers at Frontier as temporary employees provided by employment agencies. Under these circumstances, screening and reference confirmation is undertaken by the agency, and copies of that information are retained in the person's newly created employee folder. If/when the employee becomes permanently hired by Frontier, she/he signs the agency information, and those papers are returned to the personnel file. Even in situations where Frontier finds the employee directly, an employment agency may be used as a screening and payroll service for a temporary period.

New employees hired directly by Frontier provide documentation of skills they already possess via publications, detailed resumes, letters of recommendation, and self-certification. Subsequent staff training is documented via training checklists and written management and/or peer reviews, which are maintained in chronological order in the personnel files. Every staff member is formally evaluated quarterly using a combination of self and supervisor evaluation. Job descriptions are reviewed and may be updated at that time.

3.6.2 Training

Employees trained in a new skill learn by a mentorship process. The employee is assigned to a staff member, who teaches each method as follows:

- a) First familiarization with the equipment/process by observation
- b) Detailed reading of the SOP and attendant literature references
- c) Second familiarization with the equipment/process by observation
- d) Supervised practice of the method using standards and SRM's
- e) Unsupervised practice of the method on standards/SRM's
- f) Blind re-analysis of actual project samples, which have also been independently analyzed/processed by the mentor
- g) Evaluation of the blind intercomparison by the new staff member, mentor, and laboratory manager.
- h) Supervised analysis of low QC level samples or samples generated by internal research projects.

Completion of these steps is documented using a training checklist, signed by the trainee and by her/his supervisor. Once the new employee has successfully inter-compared on at least ten samples (i.e.; all results are found to replicate the mentor's results within ± 20%) he/she is considered trained in that task, and a note is placed in her/his personnel file.

For newly trained employees, pertinent QC data is reviewed on a daily basis by the laboratory manager or project supervisor for a period of at least one week, until the supervisor and/or laboratory manager is satisfied that the employee is competent in the procedure.

As the employee sets up her/his equipment and begins to obtain actual data, ongoing performance evaluations occur via control chart information, interlaboratory intercomparison exercises, and blind Performance Evaluation (PE) samples documenting precision and accuracy. Records of performance are submitted by the staff member and/or supervisor (quarterly basis) to the lab manager, for inclusion in the staff member's personnel file.

Training documentation differs according to employment category. Principle Investigator's training is primarily documented by the quantity and quality of peer-reviewed literature and complete client reports. For technical staff, Frontier relies on documented analytical performance criteria (accuracy and precision on blind samples). Copies of all documentation are maintained by the QA officer.

3.6.3 Seminars and Meetings

Records are also maintained on all seminars, classes, and training sessions attended by each employee. On a yearly basis, all staff attend at least one safety seminar and one QA/QC review seminar, where current practices and new procedures are discussed and overall laboratory compliance is reviewed. In addition, Frontier holds monthly public seminars. The topic of the seminars varies from month to month and speakers may be invited from outside sources or may be internal staff. Finally, occasional, internal seminars are presented with speakers from outside the company or from within, on specific applied topics of interest.

In the event that any serious safety or QA/QC deficiencies are discovered by the lab manager, or if a new employee is hired, or if new project-specific requirements are mandated, a staff meeting will immediately be called to provide the appropriate information and establish any required protocols. The agendas from all Frontier staff meetings which include training issues, health and safety or discussion of QA/QC concerns will be signed by the attendees, and copies will be retained in personnel files.

Section 4.1 Frontier Geosciences' Sample Handling Procedures

4.1.1 Sample Receipt and Holding Time

Samples are delivered to the sample receiving area and may be accepted by any staff employee. Because the samples are sent by overnight carrier, signing the receipt and the COC form upon delivery fulfills the Chain-of-Custody requirement. The sample custodian opens the cooler, which may or may not be sealed with a client-supplied custody seal, checks the condition of the samples (intact, temperature, broken, leaking, etc.), and notes observations on the COC form and in the sample receipt log.

If the samples are from a contaminated or potentially radioactive site, the contents will be pre-monitored for the appropriate contaminants prior to unpacking by the health and safety officer. If the contents contain substances in concentrations which might contaminate the laboratory, endanger personnel or the environment, the container will be re-sealed, placed in a secure outside

storage area, and the client notified.

The sample custodian verifies that each container is properly labeled and sealed, and compares the sample identification with the COC form. If Frontier bottle numbers are not utilized by the sampling crew, the sample custodian correlates these numbers with the client numbers directly on the COC form. If the sample identifications and the COC do not match, or if the seals on any of the containers are broken, the sample custodian notifies the laboratory supervisor.

4.1.2 Sample Tracking Procedures

Once the samples have been examined and their labels compared to the COC form, their information is entered into a master receipt logbook along with the client or project name, date and time received, matrix type, and any special client notes or anomalous observations. Each sample is assigned a unique laboratory sample identification number. In general, the sample tracking number is the client ID number, or, if the sample is in a Frontier provided Teflon bottle, the engraved bottle number is the sample tracking number. Damaged samples are disposed of in an appropriate manner, and the laboratory supervisor and the project manager are notified (please see Frontier Geosciences' Hazardous Materials Management Plan for more information on sample disposal). The system for tracking samples through preparation and analysis consists of COC records, sample receipt logbook, project logbook, laboratory worksheets, laboratory notebooks, instrument operation logbooks, instrument printouts (raw data), and final analytical reports. Because the laboratory is small, secured, and the analysts/sample receipt personnel are in constant communication, no additional internal chain-of-custody documentation is maintained.

4.1.3 Sample Transport, Preservation and Storage

Special consideration is given to the procurement, storage, and transportation of samples to be analyzed. These procedures enhance the probability that any analyte originally present in the sample matrix will not degrade or alter in concentration, and that contaminants that might interfere with the analysis have not been added. For low level (ambient) mercury work, only rigorously acid-cleaned Teflon® containers (or Borosilicate glass or quartz containers with Teflon® lids) may be used for water, as outlined in Table 4.1.

Tissues, sediments, and contaminated water samples should be stored in acid-cleaned glass containers with Teflon® lids. The client is responsible for potential sample contamination resulting from the use of polyethylene, polypropylene, or other plastics not approved for mercury work. Aqueous samples are sent unpreserved by overnight courier, while solid samples are preserved by freezing in the field unless specifically requested otherwise. Each sample container is sealed inside a zip-loc bag which is labeled with a unique sample number and geochemically relevant information (location, depth, date, etc.). After the samples are logged in, rinsed and dried, the sample custodian transports them to the refrigerator, freezer, or shelf space designed and allocated for sample storage. All company employees have access to the sample storage area, which is within the (locked) analytical laboratory.

Summary of Containers and Preservatives Table 4.1

(G=Glass with Teflon[®] Lids, T=Teflon[®], P = Polyethylene, GT=Gold Trap, CT=Carbotrap, IC=Iodated Carbon, KCL/Lime=KCL/soda lime, IX=Ion Exchange Membrane, QFF=Quartz Fiber Filter).

Parameter	Min. Vol.		Preservative	
Tissues, Sediments				
All Metals	1 g	G,T	Freeze	1 year
Water				
*Total Hg	100 mL	T	0.5% HCl	28 days
Se IV and VI	100 mL	G	0.5% HCl	3 months
Non-Hg Metals	200 mL	T,G,P	0.1% HNO ₃	6 months
Methyl Hg	50 mL	T	0.5% HCl	6 months
Dimethyl Hg	500 mL	G	1°C, Dark	48 hours
Elemental Hg	500 mL	G	1°C, Dark	48 hours
As(III)/As(V)	200 mL	P	-196°C Freeze*	3 months
Dissolved/Particulate	250 mL	T	1°C, Dark	48 hours
Air				
Total Hg	10 L	G,T	none	6 months
Methyl Hg	10 L	CT	1°C, Dark	7 days
Dimethyl Hg	10 L	CT	1°C, Dark	48 hours
Gaseous Hg (II)	10 m ³	IX	20°C, Dark	7 days
Particle Hg	10 m ³		20°C, Dark	7 days
Flue Gas				
Total Hg/Hg°	15 L	IC	none	30 days
Methyl Hg	60	KCL/Lime	none	30 days
Ionic Hg	15 L	KCL/Lime	none	30 days

^{**}Holding times are as specified unless previous approval is given by the client. This approval may be in written or verbal form.

NOTE: Samples may not be packed in vermiculite, as the dust from this material represents a contamination risk. The client should use bubble wrap or foam as packing materials. All samples known or suspected to contain high mercury levels, or any other hazardous

^aAlso good for all other trace metals.

^{*}If freezing is not possible, then 0.5% HCl, 1°C, dark.

constituents, must be so identified. Failure to do so may result in additional costs and the return of the samples, as well as liability for damages or injuries which result.

4.1.4 Chain of Custody for Samples in the Laboratory

A sample is considered to be "in custody" if it meets the following criteria:

a) It is actually in the analyst's possession.

- b) It remains in the analyst's visual range once possession of the sample has been assumed.
- c) The analyst has locked or sealed the sample to prevent tampering.

d) The sample has been stored in a secure area.

To satisfy these custody provisions, the laboratory implements the following procedures:

1. Samples are stored in a secure area.

2. Outside laboratory doors are locked at all times.

3. Visitors are accompanied by a laboratory staff member.

4. Samples remain in the secure area until acceptance of the final report by the client.

Aside from signed and dated records of activities in the lab note books, bench sheets, and sample prep logs, no additional internal chain of custody documentation is maintained.

4.1.5 Sample and Waste Disposal

Samples must eventually be disposed of to preserve laboratory storage space. Proper disposal is emphasized for the sake of efficiency, and, in the case of hazardous substances, safety. On litigation level QA projects, samples are stored for 30 days following submission of the final report unless the client requests otherwise. For levels 2 QA and below, samples may be disposed of immediately following data review, with permission of the group leader or lab manager. Disposal is notated on the sample chain-of-custody form. (Please see Frontier's Hazardous Materials Management Plan for detailed information on hazardous waste disposal.) Frontier will not accept hazardous samples without prior agreement that the client is responsible for sample handling and disposal after the analytical report has been provided. Exceptions may be made if prior arrangements are approved by the lab manager and the health and safety officer. Frontier reserves the right to reject any samples that may pose a reasonable threat to the health or safety of personnel (for example, unsterilized human biological tissue, radioactive materials, unknown industrial wastes, etc.)

Section 4.2 Frontier Geosciences' Analytical Procedures

The analytical procedures used at Frontier (Table 4.2) are derived from peerreviewed literature representing state-of-the-art methods, and are thoroughly tested prior to adoption as Frontier SOP'Ss. The analytical methods used are generally not EPA approved techniques, as Frontier methods operate with lower detection limits, more wide-ranging chemical speciation, and/or greater

precision than currently approved EPA techniques.

Recently, however, the method used for total Hg in water (FGS-011) has been given interim approval by the US EPA as Method 1631, and the method used for ultra-clean sampling (FGS-007) as Method 1669. Frontier has been commissioned by the EPA to write more methods in the 1600 series, relating to ultra-clean techniques. When conducting direct injection graphite furnace AAS, Frontier closely follows EPA protocols (Method 200.1). Our method for total arsenic by hydride generation (FGS-022) is functionally equivalent to EPA Method 1632.

Frontier's methods are periodically reviewed and updated to represent the latest thinking of the research community and/or to improve the economics of performing the analyses. Before new procedures are implemented, the analyst conducts the analysis at least three times using standards, spikes, and duplicates in order to establish reproducibility. Once the procedure is properly understood by the analyst, high quality data has been achieved, and it is approved by the lab manager, a standard operating procedure (SOP) is written for the method.

SOP's are given a numerical number that is sequential to the other SOP's. All SOP's are given an SOP number and revision or creation date. The QA officer is responsible for assigning the SOP number and then making sure that the new SOP is followed by laboratory staff. The QA officer is also responsible for checking the document management file and sending out the new version to all

applicable persons.

On a yearly basis, new SOP's and versions of SOP's will be compiled and made into an SOP appendix or new SOP manual, depending on how many revisions and newly created SOP's there are.

Literature and SOP References to Analytical Procedures Table 4.2

Parameter	Frontier SOP	Literature Reference
Total Hg	FGS-009 FGS-011 FGS-012	Bloom, 1993 Bloom & Crecelius, 1987 Bloom & Crecelius, 1983 Fitzgerald & Gill, 1979 US EPA Method 1631
Methyl Hg	FGS-010 FGS-013 FGS-017 FGS-018	Liang, Horvat, and Bloom, 1994 Horvat, Bloom, and Liang, 1993 Bloom, 1989 Bloom and Fitzgerald, 1987
Labile Hg(II), MMHg	FGS-034	Bloom, 1994
Flue Gas Hg	FGS-023 FGS-024 FGS-031	Prestbo and Bloom, 1995 Bloom, 1993
Ultra-Clean Sampling	FGS-007 FGS-008	Bloom, 1995 Fitzgerald and Watras, 1989 Gill and Fitzgerald, 1987 Patterson and Settle, 1977 Bothner and Robertson, 1975 US EPA Method 1669
Ultra-Clean Filtration	FGS-029	Bloom, 1995 Bloom and Effler, 1990
Selenium Speciation	FGS-037	Cutter, et. al, 1986
Arsenic Speciation	FGS-022	Crecelius, et. al, 1986 US EPA Method 1632
Trace metals by GFAAS	FGS-020 FGS-021 FGS-032	Slavin, 1984, US EPA 200.1 Bloom and Crecelius, 1984

Section 4.3 Frontier Geosciences' Calibration Procedures

4.3.1 Standards and Instrument Calibration

Every instrument used to analyze samples at Frontier must pass the calibration criteria established in the appropriate operating procedure document. Initial calibration criteria for instrument reproducibility and sensitivity must be met before samples may be analyzed. Continuing calibration checks establish whether ongoing instrument calibration is acceptable or not. Detailed presentations of Frontier's analytical instrument calibration procedures are provided in SOP's FGS-001 and FGS-020.

All calibration standards are obtained from chemical suppliers and are of high purity and concentration. The standards are routinely checked by the laboratory for traceability to National Research Council of Canada (NRCC) or National Institute of Standards and Technology (NIST) Standard Reference Materials (SRM's). Solutions are normally labeled as follows: name of solution, concentration of solution, date prepared, analyst's initials, and expiration date (if needed). A certificate attesting to the concentration ranges of the covered analytes is maintained in a central Frontier file, entitled "Standard Certificates."

4.3.2 Periodic Calibration Procedures for Other Laboratory Equipment

Periodic calibrations are performed for associated equipment that are required in analytical methods but are not routinely calibrated as part of the analytical procedure. Such equipment includes balances (daily, FGS-002), pipettes (prior to first daily use, FGS-003), ovens (when in use), refrigerators and freezers (daily, FGS-004), and the water purification system (daily). All calibration measurements are recorded in laboratory logbooks.

4.3.3 Supplies, Reagents, Water

All supplies (e.g., glassware, chemicals, reagents, etc.) are of the best possible quality to ensure proper instrument calibration and to avoid contamination. All reagents used are prepared from analytical reagent grade chemicals or higher purity grades, unless such purity is not available. Reagent (18 meg W) water is prepared at Frontier by double deionization of filtered tap water. Each reagent is clearly labeled with the composition, concentration, date prepared, initials of preparer, and expiration date, if necessary. Reagents that have a long, unquantified shelf life, and standards, which are re-calibrated monthly, are not given expiration dates.

Reagent solutions are stored in appropriate glass or plastic containers, under conditions designed to maintain their integrity (refrigerated, dark, etc.). Reagent solutions are checked for contamination before use in any analysis. If known to be finite, shelf life is listed on the label. Expired reagents are appropriately discarded. To avoid chemical waste and excess pollution, chemicals and standards with no known storage shelf life are kept indefinitely, except at specific contractor request. All acids for trace metal analysis other than mercury are either glass-distilled or of special grade for trace metals analysis.

Laboratory water is double deionized and checked at least weekly for mercury concentration. Our goal is to maintain [Hg] at less than 0.2 ng/L and MMHg at less than 0.01 ng/L. In the event that concentrations are observed above 1.0 ng/L for total Hg and 0.025 ng/L for MMHg, all low level analysis is discontinued until the problem is identified and remediated.

The laboratory air is monitored for mercury on a monthly basis, both in the clean room and in general laboratory space. The quality maintenance standard levels for the clean room and laboratory air are 5 and 10 ng/m³ respectively,

with corresponding action levels of 15 and 30 ng/m.

Acid-cleaned sample bottles are stored with 0.5% HCl for Hg, As, and Se, and 0.5% HNO3 for other trace metals. At least six bottles per week are randomly selected and the acidified water contents checked for mercury (EPA 1631) or trace metals scan by ICP-MS. Control charts will be kept of these data points for the following metals: Hg, As, Se, Zn, Cu, Pb, Cd, Ni, Fe. The occurrence of contamination (Hg > 1 ng/L, all others > 0.5 μ g/L) must be maintained with a running frequency of less than 3% of bottles tested. Each new lot of EPA-clean ("I-CHEM") bottles of each size will be tested for Hg by filling with 0.5% BrCl in distilled water. These values will be logged. Any lot found to contain blanks of > 1 ng/L will not be used for low-level work.

Section 4.4 Data Precision and Accuracy

4.4.1 Calibration Checks

Calibration check samples are used to verify the standard calibration curve. At least one check sample is analyzed with each batch of samples at the beginning of each calibration period. Calibration check samples can either be prepared in the laboratory (additional standards) or are available as Standard Reference Materials (SRM's) from the NRCC or NIST. Performance Evaluation samples serve as an accuracy check of laboratory operations and measurement systems by comparing results with those of other laboratories. The experimental results of the check sample are compared with the true values, and the percent recovery of the check standard is calculated. If the percent recovery falls within the acceptance range, then sample analysis proceeds. However, if the percent recovery does not fall within the acceptance range, the sample is run again, and if it still does not fall within the acceptance range, the problem is identified, corrected, and the instrument calibration sequence begins again.

SRM data is maintained chronologically in a computer database, and control charts are examined on a weekly basis to assess long-term laboratory performance trends. The control chart is used to assess trends, and therefore a single deviant point will not be considered evidence that an analytical procedure is out-of-control. If, however, two or more points in any five day period are found to be more than three sigma away from the long-term mean for that parameter, the analytical procedure will be considered out-of-control, and all analyses halted until the problem is identified, corrected, and a triplicate analysis of that parameter is found to be in control. In cases where two or more QC points are found to deviate by between two and three sigma over a five day period, the system will be considered under probation, and additional QC

checks will be performed. If, over another five day period, performance continues to give results greater than two sigma from the mean, the system will be considered out-of-control. In all cases where >95% of all QC measurements over a given period for a method are within two sigma of the long-term mean (or certified range, whichever is smaller), then the system shall be considered in control.

Data collected from inter-laboratory intercomparisons and performance evaluation (PE) samples are maintained by the QA officer. In the event that a result from such a sample deviates from the acceptance or group mean interval reported, the source of the discrepancy will be located, and corrective action will be taken. Aliquots of all PE and intercomparison samples will be retained until the group results are reported so that the samples may be re-run if necessary.

4.4.2 Matrix Spike/Duplicate Spike Analysis

For litigation level QA work, and any new sample matrix/analyte combination, one in every twenty samples will be analyzed with a matrix spike and matrix spike duplicate. In this type of analysis, predetermined quantities of stock analyte are added to a sample matrix prior to sample extraction or digestion and subsequent analysis. Because the nature of such digest is well understood in the case of simple total metals digestions, the matrix spike can be added to the digest after digestion, but prior to dilution, unless specifically disallowed by contract language. This procedure allows initial analysis of samples to calculate appropriate spiking levels.

Percent recovery is calculated for the amount of added analyte detected. Spike recoveries of 75-125% (70-130% for MMHg) are considered acceptable. The relative percent difference between the samples is calculated and used to assess analytical precision (see Duplicate Samples, Section 4.4.4). The concentration of the spike should be 2-5 times the expected concentration of the analyte in the unspiked media. In general, this means that MS/MSD analyses will be performed after the unspiked samples are analyzed, and the concentrations calculated.

For levels of QA other than litigation, the client or project manager will specify spiking frequency.

4.4.3 Duplicate Samples

Duplicate samples provide information about sampling plus analysis precision and accuracy. Duplicate samples may be either true duplicates or split samples. True duplicates are two samples collected from a common sampling location in two independent sampling events. Ideally, these samples should have identical compositions, although in fact, a degree of field variability always contributes to the observed difference between duplicates. Split samples are one or more sub-samples of a homogeneous sample. These samples should be identical in concentration, and are a direct indicator of analytical-only precision. However, in the cases of unhomogenizable samples, such as Hg⁰ contaminated soils, splits will be considered as functionally equivalent to field duplicates. The frequency of duplicate sample collection and analysis is specified in the contract with the client. Typically, one duplicate or split sample is performed per twenty samples collected, or one per sampling event, whichever is more frequent. The

relative percent difference (RPD) between the two values is calculated as 2*(R1-R2)/(R1+R2). The RPD for sample splits must be <25% at concentrations greater than 10 times the MDL to be considered in control. The laboratory has no control over field and sampling induced variability, and so the relative precision of field duplicates may be viewed as serving informational purposes only.

Section 4.5 Frontier Geosciences' Quality Control Checks

4.5.1 QC Samples

The laboratory uses QC samples to assess validity of the analytical results of QC samples include method blanks, calibration checks, performance evaluations, duplicates, and spiked samples. QC samples are analyzed in the same manner as field samples, at a frequency described either in the individual procedures, or in the contract with the client. If the OC sample results fall within the acceptance criteria (also detailed in the method or prescribed by the client), then the field sample data is considered to be valid or acceptable as-is. Unless specified otherwise by the client, the acceptance criterion for data based on QC samples is specified in the SOP on Calibration Procedures (FGS-001), and in Table 4.3. These criteria are followed per set for litigation level QA and per day for research level QA. Of particular importance to the client is Frontier's position that a single non-compliant result on a QC sample does not automatically invalidate a data set. If the other QA data in the same data set is of acceptable quality, and a re-run of the out-of-control sample is also of acceptable quality, then the entire data set is considered to be in control and acceptable.

4.5.2 Method Blanks

A method blank is a sample of reagent water or analytical reagents that undergoes the same analytical process as the corresponding field samples. Method blanks are used to monitor laboratory performance as well as to detect contamination that could have been introduced during the analytical procedure. For litigation level work, a minimum of three method blanks are required per batch, or one per ten samples, whichever is higher. For research level QA, the number of blanks required is determined by the project manager based upon historical information on blank values for the same analytical method, and the expected concentrations in the samples. All Frontier results will be reported as corrected for the mean of the method blanks analyzed with the samples, unless previously specified by the contract language.

4.5.3 Certified Reference Materials (CRM's)

Frontier maintains the position that matrix equivalent CRM's are the best measure of precision and accuracy, as problems associated with homogeneity and spikes not matching the true analyte forms are avoided. Unfortunately, CRM's do not exist for all matrices. Frontier will utilize CRM's at a rate of 1 per

20 samples or 1 per set, whichever is greater, whenever the appropriate matrix is available (sediments, tissues, sewage sludge, sea water).

4.5.4 Interlaboratory Intercomparisons

For matrices where no CRM's are available at ambient levels, Frontier will endeavor to participate in at least one interlaboratory intercomparison or round robin per year. These include low level speciation of Hg, As, and Se in water. In addition, Frontier routinely participates in several interlaboratory intercomparisons per year, including the EPA's WP series, APG, and USGS.

Litigation Level QC Requirements for Data Validation Table 4.3

QC Requirement	Minimum Frequency	Limits
all standards, force through zero	min. 4 points/day	r > 0.995
continuing calibration	1 per 10 samples	80-120% of initial slope
laboratory duplicate	1 per 20 samples	± 25% RPD @ > 10xMDL a
method blanks ^a	3 per batch or 1 per 10 samples, whichever is greater	a
Standard Reference Material	1 per dataset	a
matrix digestion MS/MSD	1 per 20 samples	75-125% Rec. a
ongoing precision and recovery	as stipulated by contract language	
low-level spike recovery	as stipulated by contract language	
filtration blanks	as stipulated by project manager	

^aNumeric values in corresponding SOPs

Section 4.6 Data Reduction, Review, and Verification

After the data has been acquired, and any necessary calculations performed, the initial review is performed by the analyst. Items included in the review include: sample identity, peak height verification, instrument calibration, QC samples, detection limits, numerical computations, accuracy of transcriptions, and compliance with the individual method. For research level QA projects, this is the full extent of the formal data review. Screening level data verification is dependent upon the project, but is generally less stringent than research level data verification.

On litigation level QA projects, following the analyst's review, the QA officer or his/her designate reviews 100% of the raw data, the analyst's chemical interpretation and any out-of-control conditions that may be identified by the analyst (FGS-038). Additionally, the QA officer examines the QC sample data and ensures that the analytical results are within Frontier-prescribed criteria for accuracy and precision. Finally, as specified by the client, the data may be further reviewed by the client, or by an independent data validator.

Data verification is part of the review process whereby data is inspected and either accepted or rejected based on a set of criteria. Evaluation parameters that can be used for validation include, but are not limited to the following:

- a) Performance on SRM's (precision and accuracy)
- b) Calibration data
- c) Specific checks unique to each measurement

Section 4.7 Data Reporting

Data is reported using a format specified in the client's contract. Data is generally reported via US Mail or 2nd day Fed-Ex in tabular form with a case narrative and/or cover letter attached. All of the data, including standard spike recoveries, control samples, duplicate analyses, and results from blank analyses, is reported along with the sample results in the calculation spreadsheets. Samples re-run for reasons of analytical error (i.g. spiked incorrectly, pipetting error, aliquot too large or too small, gas flow problem, etc.) will not be addressed in the sample report. Samples with results that are unaccountable to physical, laboratory problems, will be addressed in the sample report. requests for sample re-runs for litigation level work must come within 30 days of report submittal. The detection limits of an analytical procedure are reported if the analyzed value is less than the detection limit. Footnotes are referenced to specific data if an explanation of reported values is required. All the reports are signed and transmitted by the project manager or designee. If previously negotiated, data may be transmitted via electronic media (diskette, modem, email, fax), and/or in specific formats (i.e., Excel file, ASCII, etc.). Negotiable aspects may include the additional cost of specific computer programs, equipment, or on-line services. Our guaranteed turn-around-time for a summary data report is four weeks from receipt of the last sample in a set, with the full QA report coming no more than two weeks later. Generally a faster response time is achieved (1-2 weeks), especially for research level projects.

QC Documentation Provided with Frontier Data Packages Table 4.4

Document	Abbreviated Package	Full Package
Chain-of-Custody Form	archived	provided
Analytical Bench Sheets	archived	provided
Sample Prep Logbook	archived	at client request
Raw Data	archived	at client request
Uncorrected results	not provided	provided
Corrected results	provided	provided (unless specified)
QC data	summary	complete
Case narrative	abbreviated	complete
Instrument Logs	not provided	at client request
Reagent Prep Logs	not provided	at client request
Control Chart data	not provided	at client request
Phone/Fax Records	not provided	at client request
Data Interpretation	at client request	at client request

Thank you for your time and effort in reading this document. We at Frontier Geosciences Inc. realize that technical documents are not always the most entertaining papers to read and we are constantly striving to be as succinct and readable as possible. Your comments are most welcome and encouraged. Please feel free to call, fax or email Beverly A. Heaphey, Quality Assurance Officer, with any questions, comments or suggestions.

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ASARCO/AEC

QUALITY ASSURANCE MANUAL

LABORATORY QUALITY ASSURANCE PROGRAM

Reviewed

Laboratory Coordinator

Approved

Laboratory Manager

Laboratory Quality Assurance Program

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Attachments

- Organization Chart and Employee Responsibilities 1.
- Submittal/Chain of Custody Form Log-in/Chain of Custody Form 2.
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INTRODUCTION

Quality assurance has always been an important part of laboratory operations at ASARCO Inc./AEC. The following program describes the quality control procedures designed specifically to meet or exceed EPA, OSHA and various states requirements for an environmental chemistry laboratory. The program addresses the following areas: I) Goals and program addresses the following areas: laboratory quality control policy; II) Personnel and training; III) Sample custody, sampling techniques and preservatives; IV) Sample receipt and log-in procedures; V) Flow of samples lab; through the VI) Analytical methodology instrumentation; Quality VII) control and performance requirements VIII) Data reduction and validation; IX) Analytical reports and other deliverables, and; X) Standa Standard operating procedures. The quality assurance manual available to all personnel and is located in the supervisor's office.

I. GOALS AND LABORATORY POLICY

The purpose of the ASARCO/AEC Quality Assurance Program is to insure that all data generated and processed is scientifically valid and of known precision and accuracy. It is the policy of the laboratory to meet, as a minimum, all of the criteria set forth in this program or the method specific QC criteria and to insure that approved procedures and QA data are satisfactorily documented. The Quality Assurance manual is reviewed and approved on an annual basis by laboratory management. The effectiveness of the program is measured by means of a systems audit conducted on an annual basis by the QA coordinators. A written report of the audit findings is provided to the Analytical Services Manager. In addition, quarterly quality assurance update reports are prepared by the QA coordinators and submitted to the Analytical Services Manager.

II. PERSONNEL AND TRAINING

ASARCO field service staff, technicians and outside consulting field service personnel operate independently of the laboratory. These individuals consult with laboratory personnel and staff environmental scientists on issues such as sample volume requirements, containers and methods of preservation.

All laboratory personnel participate in an extensive on the job training program tailored specifically to the individual chemical analysis. This in-house training consists of acquiring an in-depth understanding of the appropriate methodology (eg: EPA, NIOSH and OSHA procedures) and applying these procedures to synthetic standards, standard reference materials and various other check samples prior to the analysis of unknowns. Each analyst is also required to attend laboratory safety training seminars on a quarterly basis. In addition, analysts are encouraged to attend training seminars conducted by analytical instrument manufacturers. Training records are kept on file in the manager's office. An organizational chart and laboratory personnel responsibilities forms can be found in attachment #1.

III. SAMPLING TECHNIQUES, PRESERVATION AND SAMPLE CUSTODY

Samples are taken, labeled, preserved and prepared for shipment by qualified personnel usually under the direct supervision of the on-site environmental scientist using the EPA approved procedures listed in 600/4-79 Revised 1983 and SW 846 Third Edition; NIOSH Manual of Analytical Methods, Third Edition; and the OSHA Manual of Analytical Methods. Specific chain of custody procedures are followed utilizing the form included as attachment #2.

IV. SAMPLE RECEIPT AND LOG-IN PROCEDURES

The samples are shipped by common carrier, next day service in appropriate shipping containers such as insulated ice chests. The samples are sent to one of the following individuals:

Ms. Maureen Ottley Customer/Systems Support Rep. c/o ASARCO Incorporated 3422 South 700 West Salt Lake City, Utah 84119

Ms. Jodie Haynes Customer/Systems Support Rep. c/o ASARCO Incorporated 3422 South 700 West Salt Lake City, Utah 84119

Upon arrival, the shipment is inspected for completeness by the sample custodian. Chain-of-custody forms are signed as evidence of receipt and accompany samples and login/chain-of-custody work sheets while in the laboratory. The samples are then individually logged into the ASARCO Laboratory Information Management System (LIMS).

The log-in program assigns a unique lab number to each sample and prints login/chain-of-custody work sheets for each batch of samples with all pertinent information (see attachment #3). The samples are then logged into a hardbound logbook listing the computer assigned batch number, client, date of receipt, the number of samples in the batch and the matrix (see attachment #4). The condition of the shipping and sample containers are noted on the log-in checklist along with preservation parameters (see attachment #5). Field personnel are notified immediately of any inconsistencies with the chain-of-custody forms or problems regarding the condition of the sample containers or sample preservation.

V. FLOW OF SAMPLES THROUGH THE LAB

The login/chain-of-custody work sheets are given to the analyst(s) assigned to the project along with the corresponding samples. Samples are analyzed in accordance with the methods listed in Section VI of this manual. samples are kept either in a locked refrigerator appropriate) or in the custody of the analysts while they are being worked on. All necessary sample and analytical information is logged in ink into a hardbound laboratory workbook (see attachment #6). This may include analytical data, graphs, calibration curves, quality control information, sample preparation and/or digestion procedures and any comments. All calculations such as adjustments for dilutions or concentrations, unit changes, etc. are made in this book. Any corrections are made by drawing a single line through the entry, and placing the revised entry to the side. The correction is initialed and dated. Unused portions at the bottom of a page are to have a single line extending from the last entry to the bottom of the page. Upon completion of analyses the sample and quality control data are logged into the LIMS by the analyst and a final report is generated. samples and digestates are stored for approximately two after the final report has been Miscellaneous and solid samples (eg. soils, solid waste, filters, etc.) are stored indefinitely. Samples for disposal are disposed of in accordance with all local, state and federal regulations.

VI. ANALYTICAL METHODS AND INSTRUMENTATION

The procedural and instrumentation manuals are available to the analyst in the supervisor's office at any time. For projects requiring contract laboratory protocol, the CLP ILM03.0 is followed. Other frequently used references are the EPA's "Methods for Chemical Analysis of Water and Wastes", 6009/4-79 Revised 1983, "Test Methods for Evaluation of Solid Wastes", SW846; "Standard Methods for

Water and Wastewater" 19th Edition and the NIOSH Manual of Analytical Methods. Methods are thoroughly tested using matrix matched standard reference materials or other materials of known composition to insure accurate and precise results before testing of unknowns is conducted.

The analytical instrumentation is operated in accordance with the manufacturer's guidelines to obtain the desired sensitivity. One example of this would be the use of the Perkin-Elmer "cookbook" for samples analyzed by atomic absorption spectrophotometry. Method detection limits are compiled annually and an example is included as attachment #7. Instrument standards are used on a daily basis to determine response and calibration of the instruments. These standards are traceable to high purity stock material utilizing a unique numbering system and documented in a hardbound logbook (refer to section VII). All calibration information is documented in the analyst's workbook and available for review.

Instrumentation maintenance records and service agreements are kept on file are included as attachment #8. Operating and routine maintenance SOP's are are in the ASARCO Laboratory Operations manual and on the company's intranet. Calibration and maintenance procedures are documented for support equipment such as balances and micro pipettes. Defective equipment is removed from service until repaired.

Instrument detection limits and linear dynamic ranges are determined in accordance with method specific requirements.

VII. QUALITY CONTROL AND PERFORMANCE REQUIREMENTS

Following is a summary of the ASARCO laboratory QC program:

1. Instrument Calibration and Acceptance Criteria

For atomic absorption spectrophotometry, a blank and three standards in graduated amounts must be used. Standards absorbance values will be recorded and a correlation coefficient of .995 or better must be achieved. At least two standards must be used for ICP calibration. One of the standards must be a blank. Interference check samples (ICS's) are performed daily and must be within + or - 20%. Calibrations will be conducted daily and each time the instrument is set up. Initial calibration verification (ICV) is conducted with an independent reference standard immediately after instrument calibration and must be within + or - 10%.

Continuing calibration verifications are conducted at a frequency of 1 in 10 and also must fall within + or - 10%. Initial calibration blanks (ICB's) and continuing calibration blanks (CCB's) are conducted at the frequency as the ICV's and CCV's, and are to fall within + or - the reported limit of detection.

Quality Control Samples

The table on the following page summarizes the type and frequency of QC samples normally used to validate. Also included are control limits and corrective action procedures.

The analysts are largely responsible for monitoring trends and trouble shooting problems with laboratory supervision in the role of reviewing the program and providing assistance in trouble shooting more difficult problems. Quality control charts are included as Attachment #9. As part of the laboratory alert system, quality control data, which are out of acceptable limits, are automatically flagged via the statistical quality control program and action taken in accordance with the control loop (see attachment #10).

On rare occasions when samples are sent to other laboratories, synthetic standards, NIST type certified materials, duplicates, and blanks are sent to monitor the quality of the data returned. In addition, only properly accredited and/or certified laboratories are used.

3. Proficiency Testing

The laboratory participates in many inter-laboratory proficiency testing programs. Proficiency test samples are analyzed in a manner similar to routine samples. The following is a list of current programs:

AIHA Proficiency Analytical Testing Program (PAT)

ELPAT Program for soil, paint and wipes

ERA WP Program

Ambient Source Sampling Audits (by EPA) for lead

The laboratory adheres to the following guidelines regarding proficiency testing:

• Conduct proficiency testing for each analyte or

where proficiency testing is not available, maintain appropriate validating documentation.

- Conduct proficiency testing at the certifying agencies required frequency.
- Utilize an approved proficiency testing service.
- Follow proficiency testing provider's analyzing and reporting instructions.
- Notify certifying agency of a change in provider.
- Maintain a copy of all proficiency testing records.
- Submit corrective action to certifying agency for values outside acceptable limits.

4. Laboratory Reagents and Standards Log and Traceability

All laboratory reagents are to meet or exceed the quality specified by the method. Generally, unless a higher grade is recommended, reagents are ACS grade or better. For all standards the following information will be recorded in the appropriate standards logbook and on the standards bottles:

- 1) Date of preparation
- 2) Acid matrix of standards
- 3) Concentration of standards
- 4) Initials of the analyst preparing the standard
- 5) Source of standards

Laboratory reagents are inspected, dated and initialed upon receipt. Reagents are not used beyond the the the thickness of the internal QC suggests a problem. Intermediate reagent containers are labeled with the following information:

- 1) Date of preparation
- 2) Matrix
- 3) Concentration
- 4) Initials of the analyst

Item number 5, <u>Source of Standards</u>, requires some specific guidelines to ensure a consistent, useable method of traceability. All standards are made from high purity metals and confirmed by third party reference materials or previous standards. Purchased standards are NIST traceable. Standards are not used beyond expiration dates or if the internal QC's (eg. ICV's or LCS's) suggest a problem.

Source of Standards for Stock Solutions will be conducted as follows:

Laboratory SOP's		Location	ו
Sample Preservation	Quality Section	Assurance III	Program,
Sample Custody	11	lt	
Holding Times	11	11	
*Sample Receipt	Quality Section	Assurance IV	Program,
Sample Handling & Distribution	Quality Section	Assurance V	Program,
Sample Preparation and Analyses	Quality Section	Assurance VI	Program,
*Standard Traceability & Source	н	11	
Laboratory QC Samples	Quality Section	Assurance VII	Program,
Data Assessment	Quality Section	Assurance VIII	Program,
Data Validation	11	II	
Data Useability	11	ţ1	
Data Generation	Quality Section	Assurance IX	Program,
Sample Storage Sample Disposal Glassware Cleaning Sample Digestion	Posted of Necessar	or Distribu ry	ited as

^{*}Also posted or distributed.

FIELD SOP'S

The following field SOP's are addressed by the sampling contractor and are to be included in the project specific sampling and laboratory QAPP:

- -Sample Collection
- -Field Log Book
- -Sample Packaging
- -Field Measurements
- -Consumable Procurement
- -Documentation
- -Sample Shipments

Standard reference materials will be disposed of when they met the expiration date. Standards will always be validated with a current standard reference material and will be disposed of when they can no longer meet the + or - 10% ICV guidelines set for laboratory quality assurance.

6. Examples

Refer to attachment #11 for an example of the standards log and its analytical documentation.

VIII. CERTIFICATIONS AND ACCREDITATIONS

The laboratory maintains the following certifications and accreditations:

- American Industrial Hygiene Association
- State of Utah
- State of Arizona

Refer to attachment #12 for copies of the certificates and list of licensed parameters.

IX. DATA REDUCTION, VALIDATION AND REPORTING

All standard curves, quality control information, raw data, and mathematical manipulations of data are recorded in ink in the hard bound laboratory workbook. Any corrections are made with a single strike out line dated and the analyst's initials. Each analyst maintains his or her workbook and files completed workbooks for future reference.

The sample and quality control data is logged into the LIMS data base where final reports are generated. After the reports are generated sample and quality control data are independently reviewed by a second analyst. Below is an outline of the checking procedures used to verify the reportable data. At least 20% of all data entries are checked in this manner. If during the checking process, errors are determined, 100% of the data set will be checked.

 The analyst performing the data processing shall give the data package to an analyst independent of the work for checking. The package shall

X. ANALYTICAL REPORTS AND OTHER DELIVERABLES

The following information is included with each analytical report (see attachment #12).

- 1. A cover letter referencing the project, pertinent sample information and any sample or analytical abnormalities.
- 2. Batch and sample number.
- 3. Date collected.
- Sample description.
- Parameter.
- 6. Value and Units of Concentration.
- 7. Analyst.
- 8. Date Analyzed.
- 9. Holding Times.
- 10. Method of Analyses.
- 11. Quality Control Data (provided if requested).

Each report is reviewed for accuracy of calculations, transcription errors, etc. by a second analyst and approved by the laboratory manager prior to distribution.

It is the policy of ASARCO Inc. to keep all ASARCO Laboratory records and reports on computer file and in hard copy indefinitely. Records for commercial clients, both hard copy and computer files are kept a minimum of ten years.

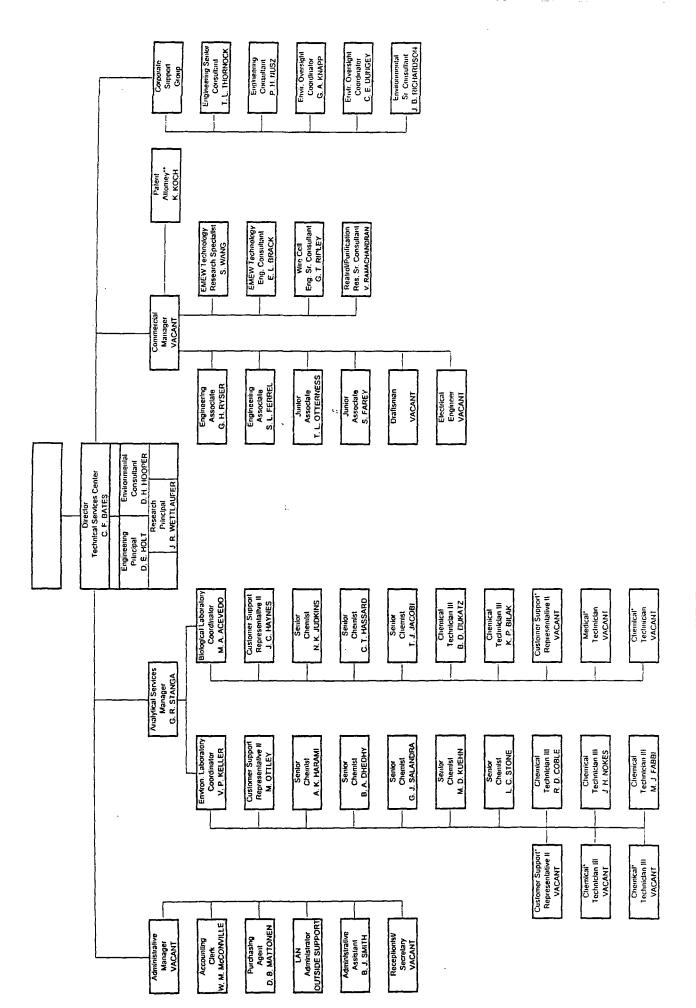
XI. STANDARD OPERATING PROCEDURE (SOP's)

The utilization of SOP's in both the field and laboratory is considered to be a critical component in maintaining a high level of quality and consistency. All field and laboratory personnel are to have ready access to any of the applicable SOP's. SOP's are to be reviewed and dated by the laboratory coordinator and approved and dated by the laboratory manager on an annual basis. The adoption of new SOP's or revision of existing SOP's are performed by the lab coordinator and lab manager. The following table is a list of specific laboratory functions for which a written SOP is available.

XII. Physical Facility

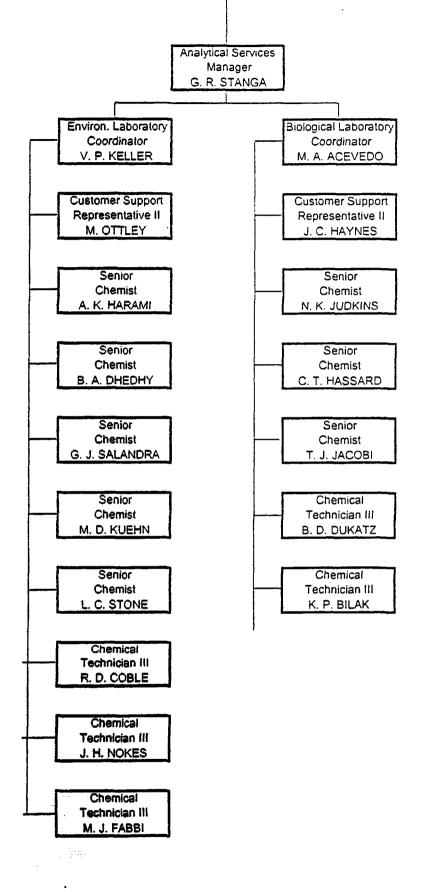
The physical facility is designed to insure that the laboratory staff has suitable space, utilities and equipment to enable them to conduct their tests in a safe and healthful environment and produce quality data. Separate areas have been designated for sample receipt, storage, chemical storage, waste storage, data handling, and incompatible tests. The facility is inspected on a regular basis for health and safety purposes. Both the facility and records are available for inspection by the certifying agency.

ORGANIZATION CHART AND EMPLOYEE RESPONSIBILITY



Needed to go from 160,000 analyses/yr to 240,000 analyses/yr.
 To be used as needed

TECHNICAL SERVICES CENTER ORGANIZATIONAL CHART January 11, 2000



CHEMICAL TECHNICIAN JOB FAMILY

POSITION	DESCRIPTION	GRADE
Chemical Technician I	Entry level position. Under direct supervision performs a variety of simple and routine chemical tests and analyses.	ဘ
Chemical Technician II	Under general supervision, performs a variety of chemical tests and analyses. Typically, requires a minimum of I year of Chemical Analyst I experience. Performs method development for laboratory processes.	9
Chemical Technician III	Under limited supervision, performs more complex chemical test and analyses on major projects. Is competent to work in most ureas of the laboratory. Uses judgment in the independent evaluation, selection and adaptation of standard methods. Performs method development and evaluation for laboratory processes. Participates in the technical training of new employees. Typically, position requires a minimum of 6 years experience.	17

CHEMICAL TECHNICIAN JOB FAMILY (cont.)

GRADE conducts work requiring mastery of specialized analytical techniques. Supervises This position represents the highest technical skill. Independently performs most Irains and supervises less experienced analysts. Position requires a bachelors degree in Chemistry or related ullied field with minimum of 8 years of diverse and performs method development and evaluation for laboratory processes. assignments with instruction as to the general results expected. Plans and DESCRIPTION laboratory work. POSITION Senior Chemist

Laboratory Coordinator

requires a bachelors degree in Chemistry or allied related field with a minimum of Supervises the daily activities of the Senior Chemists and Chemical Technicians. laboratory programs. Insures implementation of quality assurance program and programs on the basis of specialized knowledge. Organizes and implements oversees proficiency testing programs and conective action plans. Position commercial clients on a wide variety of issues. Participates in planning lab Cnordinates and distributes sample workload. Tracks progress of assigned projects to ensure timely completion. Interacts with plant personnel and 10 years of diverse analytical laboratory operations.

CUSTOMER SERVICE REPRESENTATIVE JOB FAMILY

POSITION	DESCRIPTION	GRADE
Customer Service Representative f	Unity level - Acts as the secondary customer contact for intercompany and commercial accounts regarding pricing, scheduling and shipping inquiries. Ensures that customers receive the best service possible through processing orders and preparing general correspondence. Strong communication skills is required.	∞
Customer Service Representative II	Acts as the customer contact for intercompany and commercial accounts regarding pricing, scheduling and shipping inquiries. Ensures that customers receive the best service possible through processing orders, preparing general correspondence, and coordinating with other functions as required. Strong communication skills and knowledge of laboratory environment required.	
Senior Customer Service Representative	Primary customer contact for intercompany and commercial accounts regarding pricing, scheduling and shipping inquiries. Ensures that customers exertive the best service possible through processing orders, preparing general correspondence, and coordinating with other functions as required. Assists management in planning and coordinating the marketing and advertising of the laboratory. Supervises other Customer Service Representatives.	13

Minimum of five years of customer service work experience in Laboratory

environment.

MEDICAL TECHNOLOGIST JOB FAMILY

POSITION	DESCRIPTION	GRADE
Medical T echnolog ist l	Entry level position. Under direct supervision, performs simple and routine analysis of biological samples with same day turnaround time. Position requires a bachelors degree in Medical Technology or allied reluted field.	91
Medical Technologist il	Under minimal supervision, performs more complex analysis and tests of biological samples. Has a working knowledge of instrument operation, maintenance and troubleshooting. Maintains internal quality assurance program. Reviews and enters data into the laboratory information system. Participates in method development and laboratory certification procedures. Position requires a bachelors degree in Medical Technology or allied related field with a minimum of 2 years experience in a clinical laboratory including chemistry, hematology and urinalysis.	2
Senior Medical Technologist	Independently performs most assignments with instruction as to the general	4

results expected. Oversecs all activities in the clinical section and is responsible

requires a bachelors degree in Medical Technology or allied related field with a

minimum of 8 years experience in a clinical laboratory including circuistry,

hematology and urinalysis.

for timely reporting and quality of analyses. Selects, modifies and develops methodologies to enhance analytical capabilities. Trains and supervises less

experienced employees. Maintains quality assurance programs. Position

SUBMITTAL/CHAIN-OF-CUSTODY FORMS

Your Customer Service TSC LABORATORY 3422 South 700 We Sall Lake City, Clay SEND REQUESTS TO Representative is: ASARCO USMON() Redux () Par() Other() vices Request Form II. REPORT INSTRUCTIONS Additional Copy Of Report To: are Requested No Later Than: Please Forward Results by: Services Requested Below **Enviro**rimental Orginal Report To: Laboratory . erson Requesting Service: iant Requesting Service: PLANT INFORMATION amples Collected By: roject Description: our Project No.:

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AEC LABORATORIES

Laboratory Services Request Form

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Account Add	iress:							Phone # (801)261-142
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LOGIN/CHAIN -OF-CUSTODY FORM

LOGIN CHAIN OF CUSTODY REPORT (1n01) Dec 18 1998, 09:52 am

Login Number: L982521 Account: 7818 East Helena Project: 3119 Water and Solid Waste

Laborator Sample Nu		ent Collec ole Number Date	t Receive Due Date PR Date
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CENTRAL LOGBOOK RECORD

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LOGIN CHECKLIST FORM

LOG-IN CHECKLIST

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CN-	(NaOH)				
Phenol & P	(H ₂ SO,)				
Oil&Grease	(H,50,)				
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COMMENTS:					

Signature

LAB WORKBOOK RECORD

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METHOD DETECTION LIMITS

PARAMETER: AL METHODOLOGY: ICP							
INSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP							
METHOD of PREPARATION: EPA 3010							
				EPA 6010/200.7			
MATRIX:	WATER						
PREPARER:	AKH		DAT	- E OF PREPARATION:	06-Dec-00		
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00		
Ana	alyte Conc.: _	0.0868		Analyte Units:	PPM	(e.g. ppm)	
		Students	T Value:	3.143			
	Replicate #		Result	Units			
	1		0.0906	PPM	=		
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	4		0.1089	PPM	•		
	5		0.0871	PPM	-		
	6		0.0767	PPM	_		
	7		0.0815	PPM	-		
Mean Analyte	Recovery: _	0.0868	PPM	2xMDL:	0.0676		
Standard	Deviation:	0.01075	PPM	Analyte Conc.:	0.0868		
	MDL:	0.03379	РРМ	10xMDL:	0.3379		

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

SPIKE SOLUTION EEICQ7

PARAMETER:	R: AS METHODOLOGY: ICP							
INSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP								
METHOD of PREPARATION: EPA 3010								
METHOD of A	ANALYSIS: _		EPA 6010/200.7					
MATRIX:	WATER							
PREPARER:	AKH	DATE	OF PREPARATION:	06-Dec-00				
ANALYST:	MDK	D	ATE OF ANALYSIS:	06-Dec-00				
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	6	0.0519	PPM					
	7	0.0559	PPM					
		0.0333	3 1 141					

Mean Analyte Recovery: 0.0536 PPM 2xMDL: 0.0115

Standard Deviation: 0.00183 PPM Analyte Conc.: 0.0500

MDL: 0.00577 PPM 10xMDL: 0.0577

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

SPIKE SOLUTION EEEICQ19

PARAMETER:	AG		METHO	DDOLOGY: <u>I</u>	СР		
INSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP							
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METHOD of A	ANALYSIS:	EPA 6010/200.7					
MATRIX:	WATER			_			
PREPARER:	AKH		DAT	E OF PREPA	RATION:	06-Dec-00	
ANALYST:	MDK			DATE OF AN	IALYSIS:	08-Dec-00	
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		Students	T Value:	3.143			
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	3		0.0093		PPM	_	
	4		0.0091		PPM	_	
	5		0.0101		PPM	_	
	6		0.0108		PPM_	_	
	7		0.0090		PPM	•	
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PPM

PPM

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

Analyte Conc.:

10xMDL:

0.0100

0.0203

SPIKE SOLUTION EEICQ7

Standard Deviation: 0.00064

MDL: 0.00203

PARAMETER:	PARAMETER: BE METHODOLOGY: ICP						
INSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP							
METHOD of PREPARATION: EPA 3010							
METHOD of A	ANALYSIS: EPA 6010/200.7						
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PREPARER:	AKH	_	DAT	E OF PREPARATION:	29-Nov-00		
ANALYST:	MDK	_		DATE OF ANALYSIS:	30-Nov-00		
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	1		0.0024	PPM	_		
	2		0.0024	PPM	_		
	3		0.0025	PPM	_		
	4		0.0024	PPM	_		
	5		0.0025	PPM	<u>-</u>		
	6		0.0025	PPM	_		
	7		0.0025	PPM	_ _		
Mean Analyte	Recovery:	0.0025	PPM	2xMDL:	0.0004		
Standard	Deviation:	0.00006	PPM	Analyte Conc.:	0.0020		
	MDL:	0.0002	PPM	10xMDL:	0.0020		
Acceptable Analyte Concentration YES							

SPIKE SOLUTION EEEICQ19

PARAMETER:	R		METHO	DDOLOGY: ICP		
CARCITETY. D			IVILITI	7D01001: 101	- \	
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	<u> </u>	
METHOD of PREPARATION: EPA 3010						
METHOD of ANALYSIS:			· -	EPA 6010/200.7		
MATRIX:	WATER	 		_		
PREPARER:	AKH		DAT	E OF PREPARATION:	29-Nov-00	
ANALYST:	MDK			DATE OF ANALYSIS:	30-Nov-00	
Ana	alyte Conc.: _	0.1000		Analyte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units	=	
	11	·	0.1052	PPM	_	
	2		0.1120	PPM	_	
•	3		0.1066	PPM PPM	-	
			0.1096 0.1016	PPM	-	
	6		0.1018	PPM	_	
	$\frac{7}{7}$		0.1283	PPM	-	
Mean Analyte	Recovery:	0.1107	PPM	2xMDL:	0.0540	
Standard	Deviation: _	0.00859	PPM	Analyte Conc.:	0.1000	
	MDL: _	0.02701	PPM	10xMDL:	0.2701	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

SPIKE SOLUTION EEICQ7

PARAMETER:	Ва		METHO	DDOLOGY: ICP	
INSTRUMENT IDENT	TIFICATION	AND DESCR	RIPTION:	OPTIMA 3000 DV ICE)
METHOD of PREP	ARATION:	EPA 3010			
METHOD of A	NALYSIS:			EPA 6010/200.7	
MATRIX:	WATER				
PREPARER:	AKH		DATI	- E OF PREPARATION:	29-Nov-00
ANALYST:	MDK			DATE OF ANALYSIS:	30-Nov-00
Ana	lyte Conc.:	0.0020		Analyte Units:	PPM (e.g. ppm)
=					
		Students	T Value:	3.143	
	Replicate #		Result	Units	_
	1		0.0054	PPM	=
	2		0.0054	PPM	- -
	3		0.0054	PPM	_
	4		0.0053	PPM	-
	5		0.0058	PPM	_
	6		0.0057	PPM	_
	7		0.0060	PPM	-
Mean Analyte	Recovery:	0.0056	PPM	2xMDL:	0.0017
Standard	Deviation:	0.00026	РРМ	Analyte Conc.:	0.0020
	MDL:	÷0.00083	**PPM	10xMDL:	0.0083

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	Са		METHO	DOLOGY: 1	СР				
INSTRUMENT IDENT	TIFICATION 1	AND DESCF	RIPTION:	OPTIMA 300	00 DV ICE) ————————————————————————————————————			
METHOD of PREP	ARATION:	EPA 3010				·			
METHOD of A	METHOD of ANALYSIS:			EPA 6010/200.7					
MATRIX:	WATER			_					
PREPARER:	AKH		DAT	E OF PREPA	RATION:	29-Nov-00			
ANALYST:	MDK			DATE OF AN	IALYSIS:	30-Nov-00			
Ana	lyte Conc.:	0.1000		Anal	yte Units:	PPM	(e.g. ppm)		
		Students	T Value:	3.143			;=====		
	Replicate #	!	Result		Units				
	1		0.1209		PPM	=			
	2	·····	0.1320		PPM	-			
	3		0.1239		PPM	<u>-</u>			
	4		0.1256		PPM	- -			
	5		0.1205		PPM	_			
	6		0.1132		PPM	-			
	7		0.1147		PPM	-			
Mean Analyte	Recovery:	0.1215	PPM	-	2xMDL:	0.0405			

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

Analyte Conc.:

10xMDL:

0.1000

0.2026

SPIKE SOLUTIONEEICQ19

Standard Deviation: 0.00645 PPM

MDL: MUNICO NEW PARTY

PARAMETER:	Cd METHODOLOGY: ICP					
INSTRUMENT IDENT	IFICATION .	AND DESCF	RIPTION:	OPTIMA 3000 DV ICE) 	
METHOD of PREP	ARATION: 1	EPA 3010				
METHOD of A	NALYSIS:			EPA 6010/200.7		
MATRIX:	WATER					
PREPARER:	AKH		DATE	OF PREPARATION:	29-Nov-00	
ANALYST:	MDK		ı	DATE OF ANALYSIS:	30-Nov-00	
Ana	lyte Conc.: _	0.0020		Analyte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.0020	PPM	=	
	2		0.0018	PPM	-	
	3		0.0018	PPM	-	
	4		0.0017	PPM	_	
	5		0.0018	PPM	_	
	6		0.0019	PPM	-	
	7		0.0017	PPM	_	
Mean Analyte	Recovery: _	0.0018	PPM	2xMDL:	0 0007	
Standard	Deviation: _	0.00011	PPM	Analyte Conc.:	0.0020	
	MDL: 🖁	0.00034	PPM	10xMDL:	0.0034	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

	=======================================						
PARAMETER:	Со		METHO	DOLOGY: ICP	·		
INICTOLIMENT IDENT	TIEICATION	AND DECC		ODTIMA 3000 DV ICE)		
INSTITUTE INTERVI	HEICATION	AND DESCR	RIPTION.	OPTIMA 3000 DV ICE			
METHOD of PREPARATION: EPA 3010							
METHOD of ANALYSIS: EPA 6010/200.7							
MATRIX:	WATER			_			
PREPARER:	AKH		DAT	E OF PREPARATION:	29-Nov-00		
ANALYST:	MDK			DATE OF ANALYSIS:	30-Nov-00		
Ana	alyte Conc.:	0.0020		Analyte Units:	РРМ	(e.g. ppm)	
		Students	T Value:	3.143			
	Replicate #		Result	Units			
	1		0.0010	PPM	=		
		·	0.0010	PPM	-		
	3		0.0011	PPM	-		
	4		0.0010	PPM	-		
	5		0.0008	PPM	-		
	6		0.0016	PPM	-		
	7		0.0013	PPM	- -		
Mean Analyte	Recovery:	0.0011	PPM	2xMDL:	0.0016		
Standard	Deviation:	0.00026	PPM	Analyte Conc.:	0.0020		
	MDL:		THE STATE	10xMDL:	0.0081		

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	PARAMETER: Cu METHODOLOGY: ICP							
INSTRUMENT IDENT	TIFICATION	AND DESCR	RIPTION:	OPTIMA 3000 DV ICE	<u> </u>	·····		
METHOD of PREPARATION: EPA 3010								
METHOD of A	ETHOD of ANALYSIS: EPA 6010/200.7							
MATRIX:	WATER							
PREPARER:	AKH		DATE	OF PREPARATION:	29-Nov-00			
ANALYST:	MDK		!	DATE OF ANALYSIS:	30-Nov-00			
Ana	lyte Conc.:	0.0200		Analyte Units:	PPM	(e.g. ppm)		
		Students	: T Value:	3.143				
	Donlingto #			Units				
	Replicate #		Result		=			
	1		0.0204	PPM	- -			
	2		0.0194	PPM	_			
	3		0.0200	PPM	_			
	4	· · · · · · · · · · · · · · · · · · ·	0.0211	PPM	-			
	5 6		0.0190	PPM PPM	-			
	7		0.0200 0.0198	PPM	-			
			0.0196	Frivi	-			
Mean Analyte	Recovery:	0.0200	PPM	2xMDL:	0.0043			
Standard	Deviation:	0.00068	РРМ	Analyte Conc.:	0.0200			
	MDL:	0.00213	EPM.	10xMDL:	0.0213			

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

			<u> </u>			
PARAMETER:	Cr	•	METHO	DOLOGY. ICP		
INSTRUMENT IDENT	TIFICATION A	AND DESCR	RIPTION:	OPTIMA 3000 DV ICP) 	
METHOD of PREP	ARATION: I	EPA 3010				
METHOD of A	ANALYSIS: _			EPA 6010/200.7		
MATRIX:	WATER					
PREPARER:	AKH		DATE	OF PREPARATION:	29-Nov-00	
ANALYST:	MDK		1	DATE OF ANALYSIS:	30-Nov-00	
Ana	lyte Conc.: _	0.0020		Analyte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.0022	PPM	3	
	2		0.0027	PPM	-	
	3		0.0026	PPM	-	
	4		0.0023	PPM	-	
	5		0.0025	PPM	-	
	6	 	0.0029	PPM	-	
	7		0.0020	PPM	-	
Mean Analyte	Recovery:	0.0025	PPM	2xMDL:	0.0019	
Standard	Deviation: _	0.00031	PPM	Analyte Conc.;	0.0020	
	MDL:	2000975	PPML	10xMDL:	0.0097	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	FE		METHO	DDOLOGY: ICP				
INSTRUMENT IDEN	TIFICATION A	AND DESC	RIPTION:	OPTIMA 3000 DV ICE)			
METHOD of PREP	METHOD of PREPARATION: EPA 3010							
METHOD of A	ANALYSIS: _	·····		EPA 6010/200.7				
MATRIX:	WATER			_				
PREPARER:	AKH		DAT	E OF PREPARATION:	06-Dec-00			
ANALYST:	MDK			DATE OF ANALYSIS:	06-Dec-00			
Ana	llyte Conc.: _	0.0500		Analyte Units:	PPM	(e.g. ppm)		
		Students	T Value:	3.143				
	Replicate #		Result	Units				
	1		0.0545	PPM	=			
		·	0.0570	PPM	-			
	3		0.0680	PPM	-			
	4		0.0516	PPM	-			
	5		0.0552	PPM	-			
	6		0.0491	PPM	-			
	7		0.0528	PPM	•			
Mean Analyte	Recovery: _	0.0554	PPM	2xMDL:	0.0383			
Standard	Deviation: _	0.00610	PPM	Analyte Conc.:	0.0500			
	MDL:	0.01916	РРМ	10xMDL:	0.1916			

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	MN		METHO	DDOLOGY: ICP	
INSTRUMENT IDEN	TIFICATION A	AND DESC	RIPTION:	OPTIMA 3000 DV IC	P
METHOD of PREP	ARATION: E	PA 3010			
METHOD of A	ANALYSIS: _			EPA 6010/200.7	
MATRIX:	WATER			_	
PREPARER:	AKH		DAT	E OF PREPARATION	: <u>06-Dec-00</u>
ANALYST:	MDK			DATE OF ANALYSIS	: <u>06-Dec-00</u>
Ana	llyte Conc.:	0.5000		Analyte Units	: PPB (e.g. ppm)
		Students	T Value:	3.143	
	Replicate #		Result	Units	
	1		0.3100	PPB	==
	2		0.3200	PPB	-
	3	·	0.3200	PPB	
	4		0.2900	PPB	_
	5		0.3700	PPB	
	6		0.2800	PPB	_
	7		0.3100	PPB	
Mean Analyte	Recovery: _	0.3143	PPB	_ 2xMDL	:0.1809
Standard	Deviation: _	0.02878	PPB	_ Analyte Conc.	: 0.5000

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

MDL: 0.09047 PPB 10xMDL: 0.9047

PARAMETER:	NI		METHO	OOLOGY: ICP				
INSTRUMENT IDEN	TIFICATION .	AND DESC	RIPTION:	OPTIMA 3000 DV ICE)			
METHOD of PREP	ARATION:	EPA 3010						
METHOD of A	ANALYSIS:			EPA 6010/200.7				
MATRIX:	WATER							
PREPARER:	AKH		DATE	OF PREPARATION:	06-Dec-00			
ANALYST:	MDK		C	ATE OF ANALYSIS:	08-Dec-00			
Ana	llyte Conc.:	0.0147		Analyte Units:	PPM	(e.g. ppm)		
	Students T Value: 3.143							
	Replicate #		Result	Units				
	1		0.0152	PPM	=			
	2		0.0146	PPM	-			
	3		0.0152	PPM	-			
	4		0.0159	PPM	•			
	5		0.0157	PPM	-			
	6		0.0161	PPM	-			
	7	 	0.0101	PPM	•			
Mean Anaiyte	Recovery:	0.0147	PPM	2xMDL:	0.0131			
Standard	Deviation:	0.00208	PPM	Analyte Conc.:	0.0147			
	MDL:	0.00655	PPM 🦈	10xMDL:	0.0655			

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	К		METHO	DOLOGY: ICP		
INSTRUMENT IDENT	ΓΙ F ICATION	AND DESCI	RIPTION:	OPTIMA 3000 DV ICF) —————	
METHOD of PREP	ARATION:	EPA 3010	~= ~			
METHOD of A	ANALYSIS:		<u> </u>	EPA 6010/200.7		
MATRIX:	WATER					
PREPARER:	AKH	,	DATE	OF PREPARATION:	29-Nov-00	
ANALYST:	MDK			DATE OF ANALYSIS:	30-Nov-00	
Ana	lyte Conc.:	1.0000		Analyte Units:	РРМ	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.9694	PPM	=	
	2		1.0720	PPM	-	
	3		1.0650	PPM	-	
	4	-	1.1000	PPM	-	
	5		0.8997	PPM	-	
	6		1.1000	PPM	-	
	7		1.1780	PPM	-	
Mean Analyte	Recovery:	1.0549	PPM	2xMDL:	0.5796	
Standard	Deviation:	0.09221	PPM	Analyte Conc.:	1.0000	
	MDL:	TUZESIA	REPN)	10xMDL:	2.8981	

<u>Acceptable Analyte Concentration</u> **YES - 2xMDL < Analyte Conc. < 10xMDL**

PARAMETER:	Mg METHODOLOGY: ICP					
INSTRUMENT IDENT	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE) 	
METHOD of PREP	ARATION:	EPA 3010				
METHOD of A	NALYSIS:			EPA 6010/200.7		
MATRIX;	WATER					
PREPARER:	AKH		DATE	OF PREPARATION:	29-Nov-00	
ANALYST:	MDK		Ε	DATE OF ANALYSIS:	30-Nov-00	
Ana	lyte Conc.:	0.1000		Analyte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.0985	PPM	=	
	2		0.0934	PPM	-	
	 3		0.0910	PPM	-	
	4		0.0998	PPM	-	
	5		0.1007	PPM	-	
	6		0.0979	PPM	-	
	7	···	0.0977	PPM	-	
Mean Analyte	Recovery:	0.0970	PPM	2xMDL:	0.0221	
Standard	Deviation:	0.00351	PPM	Analyte Conc.:	0.1000	
÷	MDL: 🚪	9 01 (94)	ZPRV N	10xMDL:	0.1104	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

						
PARAMETER:	Мо		METHO	DOLOGY: ICP		
INSTRUMENT IDENT	TIFICATION	AND DESCR	RIPTION:	OPTIMA 3000 DV ICE) 	,
METHOD of PREP	ARATION:	EPA 3010	 			
METHOD of A	ANALYSIS:	IALYSIS: EPA 6010/200.7				
MATRIX:	WATER					
PREPARER:	AKH	_	DATE	OF PREPARATION:	29-Nov-00	
ANALYST:	MDK	_		DATE OF ANALYSIS:	30-Nov-00	
Ana	alyte Conc.:	0.0200		Analyte Units:	РРМ	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #	1	Result	Units	_	
	1		0.0187	PPM	-	
	2		0.0177	PPM	-	
	3		0.0203	PPM	-	
	4		0.0186	PPM	-	
	5		0.0189	PPM	_	
	6		0.0190	PPM	-	
	7		0.0191	PPM	_	
Mean Analyte	Recovery:	0.0189	PPM	2xMDL:	0.0049	
Standard	Deviation:	0.00077	PPM	Analyte Conc.:	0.0200	
	MDL:	0.002450	YEP MEN	10xMDL:	0.0243	

<u>Acceptable Analyte Concentration</u> **YES - 2xMDL < Analyte Conc. < 10xMDL**

PARAMETER:	Na		МЕТНО	DOLOGY: ICP		
INSTRUMENT IDENT	TIFICATION A	ND DESCF	RIPTION:	OPTIMA 3000 DV ICE)	
METHOD of PREP	ARATION: E	PA 3010				
METHOD of A	NALYSIS: _			EPA 6010/200.7		
MATRIX:	WATER					
PREPARER:	AKH		DATE	OF PREPARATION:	29-Nov-00	
ANALYST:	MDK			DATE OF ANALYSIS:	30-Nov-00	
Ana	llyte Conc.: _	0.1000		Analyte Units:	PPM (e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.1325	PPM	=	
	2		0.1242	PPM	-	
	3		0.1153	PPM	-	
	4		0.1308	PPM	-	
	5		0.1143	PPM	-	
	6		0.1428	PPM	-	
	7		0.1199	PPM	_	
Mean Analyte	Recovery:	0.1257	PPM	2xMDL:	0.0649	
Standard	Deviation: _	0.01032	РРМ	Analyte Conc.:	0.1000	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

10xMDL; 0.3243

SPIKE SOLUTIONEEICQ7

MDL: 0.03243- PPM

PARAMETER:	Pb		METHO	DOLOGY: IC	P		
INSTRUMENT IDENT	TIFICATION .	AND DESC	RIPTION:	OPTIMA 300	0 DV ICF)	
METHOD of PREP	ARATION:	EPA 3010					
METHOD of A	ANALYSIS:	EPA 6010/200.7					
MATRIX:	WATER			-			
PREPARER:	AKH		DATE	OF PREPAR	RATION:	29-Nov-00	
ANALYST:	MDK			DATE OF ANA	ALYSIS:	30-Nov-00	
Ana	lyte Conc.:	0.0200		Analy	te Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143			
	Replicate #		Result		Units	_	
	1		0.0190		PPM	-	
	2		0.0211		PPM	_	
	3		0.0210		PPM	_	
	4		0.0187		PPM	_ _	
	5		0.0196		PPM	_	
	6		0.0163		PPM	_	
	7		0.0230		PPM	_	
Mean Analyte	Recovery:	0.0198	РРМ		2xMDL:	0.0135	
Standard	Deviation:	0.00214	РРМ	- Analyt	e Conc.:	0.0200	

<u>Acceptable Analyte Concentration</u>
YES - 2xMDL < Analyte Conc. < 10xMDL

10xMDL: 0.0673

SPIKE SOLUTIONEEICQ19

MDL: 2000678 3 2 2 2 1 M

PARAMETER:	Se		метно	DOLOGY: ICP		
INSTRUMENT IDENT	ΓΙΓΙCATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	<u> </u>	
METHOD of PREP	ARATION:	EPA 3010				
METHOD of A	ANALYSIS:			EPA 6010/200.7		
MATRIX:	WATER					
PREPARER:	AKH		DATE	OF PREPARATION:	29-Nov-00	
ANALYST:	MDK		[DATE OF ANALYSIS:	30-Nov-00	
Ana	lyte Conc.:	0.1000		Analyte Units:	PPM (e.g. ppm)
	** **********************************	Students	s T Value:	3.143	· · · · · · · · · · · · · · · · · · ·	
			•			
	Replicate #	<u> </u>	Result	Units		
	1		0.0873	PPM	-	
	2		0.0695	PPM	_	
	3		0.0736	PPM	-	
	4	····	0.0728	PPM	-	
	5		0.0908	PPM	-	
	6		0.0838	PPM	-	
	7		0.0801	PPM	_	
Mean Analyte	Recovery:	0.0797	PPM	2xMDL:	0.0505	
Standard	Deviation:	0.00803	PPM	Analyte Conc.:	0.1000	
	MDL:	0.02524	PPM N	10xMDL:	0.2524	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	PARAMETER: Si			DDOLOGY: I	CP				
INSTRUMENT IDENT	TIFICATION	AND DESCR	RIPTION:	OPTIMA 300	0 DV ICF) ————————————————————————————————————			
METHOD of PREP	ARATION:	EPA 3010					·		
METHOD of A	METHOD of ANALYSIS:			EPA 6010/200.7					
MATRIX:	WATER			_					
PREPARER:	AKH_	_	DAT	E OF PREPAI	RATION:	29-Nov-00			
ANALYST:	MDK			DATE OF AN	ALYSIS:	30-Nov-00			
Analyte Conc.:		0.1000		Analy	te Units:	РРМ	(e.g. ppm)		
		Students	T Value:	3.143					
	Replicate #	<u> </u>	Result		Units	_			
	1		0.1266		PPM	=			
	2		0.1276		PPM	•			
	3		0.1252		PPM	-			
	4		0.1251		PPM	-			
	5		0.1233		PPM				
	6		0.1248		PPM	-			
	7		0.1353		PPM	-			
Mean Analyte	Recovery:	0.1268	РРМ		2xMDL:	0.0250			

Standard Deviation: 0.00397

PPM

Analyte Conc.:

0.1000

MDL: MOLE PROPERTY

10xMDL: 0.1248

Acceptable Analyte Concentration YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	V		METHO	DOLOGY: ICP				
INSTRUMENT IDENT	TIFICATION A	AND DESCR	RIPTION:	OPTIMA 3000 DV ICE)			
METHOD of PREP	ARATION: E	PA 3010				· 		
METHOD of A	ANALYSIS: _			EPA 6010/200.7	····			
MATRIX:	WATER							
PREPARER:	AKH		DATE	OF PREPARATION:	29-Nov-00			
ANALYST:	MDK		[DATE OF ANALYSIS:	30-Nov-00			
Analyte Conc.: 0.0200 Analyte Units: PPM (e.g. pp						(e.g. ppm)		
	Students T Value: 3.143							
	Replicate #	_	Result	Units	_			
	1		0.0188	PPM	=			
	2		0.0189	PPM	-			
	3		0.0201	PPM	_			
	4		0.0196	PPM	-			
	5		0.0183	PPM	-			
	6		0.0178	PPM	-			
	7		0.0188	PPM	•			
Mean Analyte	Recovery: _	0.0189	PPM	2xMDL:	0.0048			
Standard	Deviation: _	0.00077	PPM	Analyte Conc.:	0.0200			
	MDL:	0.00241	PPM	10xMDL:	0.0241			

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	SB		METHO	DDOLOGY: ICP		
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE)	
METHOD of PREP	ARATION: E	EPA 3010				
METHOD of A	ANALYSIS: _			EPA 6010/200.7	· · · · · · · · · · · · · · · · · · ·	
MATRIX:	WATER			_		
PREPARER:	AKH		DAT	E OF PREPARATION:	29-Nov-00	
ANALYST:	MDK			DATE OF ANALYSIS:	30-Nov-00	
Ana	alyte Conc.:	0.1000		Analyte Units:	РРМ	(e.g. ppm)
Students T Value:3.143_						
	Replicate #		Result	Units		
	1		0.1076	PPM	=	
	2		0.1040	PPM	-	
	3		0.1055	PPM	-	
	4		0.1139	PPM	-	
	5		0.1070	PPM	-	
	6		0.1132	PPM	_	
	7		0.1156	PPM	_	
Mean A na lyte	Recovery: _	0.1095	PPM	2xMDL:	0.0288	
Standard	Deviation: _	0.00459	PPM	Analyte Conc.:	0.1000	
	MDL: _	0.01442	РРМ	10xMDL:	0.1442	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	TL		METHO	DOLOGY: ICP		
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	<u> </u>	
METHOD of PREP	ARATION:	EPA 3010				
METHOD of A	ANALYSIS:			EPA 6010/200.7		
MATRIX:	WATER					
PREPARER:	AKH		DATE	OF PREPARATION:	06-Dec-00	
ANALYST:	MDK			DATE OF ANALYSIS:	06-Dec-00	
Analyte Conc.: 0.0500 Analyte Units: PPM (e.g. pp						(e.g. ppm)
		Students	s T Value:	3.143		
	Replicate #		Result	Units		
	1		0.0671	PPM	=	
			0.0590	PPM	-	
	3		0.0590	PPM	-	
	4		0.0612	PPM	-	
	5		0.0612	PPM	_	
	6		0.0627	PPM	-	
	7		0.0663	PPM	_	
Mean Analyte	Recovery:	0.0634	PPM	2xMDL:	0.0199	
Standard	Deviation:	0.00316	PPM_	Analyte Conc.:	0.0500	
	MDL:	0.00994	PPM	10xMDL:	0.0994	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	Ag		МЕТН	ODOLOGY: ICP				
INSTRUMENT IDENT	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE)			
METHOD of PREP	ARATION:	EPA 3050	· · · · · · · · · · · · · · · · · · ·					
METHOD of A	ANALYSIS: _			EPA6010/3050				
MATRIX:	SOIL			_				
PREPARER:	VPK		DAT	E OF PREPARATION:	07-Dec-00			
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00			
Ana	lyte Conc.:	0.0200		Analyte Units:	PPM	(e.g. ppm)		
	Replicate #	Students	Result	3.143 Units	=			
	1		0.0133	PPM	_			
	2		0.0172	PPM	_			
	3		0.0160	PPM	_			
	4		0.0139	PPM	_			
	5		0.0192	PPM	_			
	6		0.0151	PPM	_			
	7		0.0130	PPM	_			
Mean Analyte	Recovery: _	0.0154	PPM	2xMDL:	0.0143			
Standard	Deviation: _	0.00227	РРМ	_ Analyte Conc.:	0.0200			
**	MDL:	0.00713	PPM	10xMDL:	0.0713			
Soil MDL Base	Soil MDL Based On One Gram To 50 ml 0.356 ug/g							
		_		Analyte Concentration				
		162	- ZXIVIUL	< Analyte Conc. < 10x	MIDF			

PARAMETER:	PARAMETER: As METHODOLOGY: ICP								
INSTRUMENT IDENT	TIFICATION A	AND DESC	RIPTION:	OPTIMA 3000	DV ICE	>			
METHOD of PREP	ARATION: <u>E</u>	PA 3050							
METHOD of A	NALYSIS: _			EPA6010/	3050				
MATRIX:	SOIL			_					
PREPARER:	VPK		DAT	E OF PREPAR	ATION:	07-Dec-00			
ANALYST:	MDK			DATE OF ANA	ALYSIS:	08-Dec-00			
Analyte Conc.: 0.1000 Analyte Units: PPM (e.g. p						(e.g. ppm)			
	Students T Value: 3.143								
	Replicate #		Result		Units	_			
	1		0.0910		PPM	-			
	2		0.0887		PPM	_			
	3		0.0889		PPM				
	4		0.0938		PPM	_			
	5		0.0826		PPM	_			
	6		0.0708		PPM	-			
	7		0.0830		PPM	_			
Mean Analyte	Recovery: _	0.0855	PPM	_ ;	2xMDL:	0.0482			
Standard	Deviation: _	0.00767	PPM	_ Analyte	Conc.:	0.1000			
	MDL: _	0.02410	PPM	10	0xMDL:	0.2410			
Soil MDL Based On One Gram To 50 ml Acceptable Analyte Concentration YES - 2xMDL < Analyte Conc. < 10xMDL									

PARAMETER:	BE		METHO	DOLOGY: IC	<u> </u>				
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 300	0 DV ICE)			
METHOD of PREP	ARATION:	E PA 3050							
METHOD of A	ANALYSIS:			EPA6010	/3050				
MATRIX:	SOIL			_					
PREPARER:	VPK	DATE OF PREPARATION: 07-Dec-00							
ANALYST:	MDK	DATE OF ANALYSIS: 08-Dec-00							
Ana	lyte Conc.:	0.0040 Analyte Units: PPM (e.g. ppm							
Students T Value: 3.143									
	Replicate #		Result		Units	=			
	1		0.0041 PPM			_			
	2		0.0038		PPM	-			
	3		0.0036		PPM PPM	-			
	5		0.0035		PPM	-			
	6		0.0036		PPM				
	7		0.0036		PPM	•			
Mean Analyte	Recovery:	0.0037	PPM	-	2xMDL:	0.0011			
Standard	Deviation:	0.00018	PPM	Analyte	e Conc.:	0.0040			
	MDL:	0.00057	PPM	1	0xMDL:	0.0057			
Soil MDL Based On One Gram To 50 ml 0.029 ug/g Acceptable Analyte Concentration YES - 2xMDL < Analyte Conc. < 10xMDL									

PARAMETER:	CD		METHO	DDOLOGY: 10	CP		
INSTRUMENT IDENT	TIFICATION	AND DESC	RIPTION:	OPTIMA 300	00 DV ICF) 	
METHOD of PREP	ARATION:	EPA 3050					
METHOD of A	NALYSIS:			EPA6010)/3050		
MATRIX:	SOIL			_			
PREPARER:	VPK		DAT	E OF PREPAI	RATION:	07-Dec-00	
ANALYST:	MDK			DATE OF AN	IALYSIS:	08-Dec-00	
Ana	lyte Conc.:	0.0040		Analy	/te Units:	PPM	(e.g. ppm)
		Students	s T Value:	3.143	·=		
	Replicate #		Resúlt		Units	_	
	1		0.0034		PPM	=	
	2		0.0030		PPM	-	
	3		0.0035		PPM	-	
	4		0.0034		PPM	-	
	5		0.0032		PPM	-	
	6		0.0035		PPM	-	
	7		0.0034	· 	PPM	-	
Mean Anaiyte	Recovery:	0.0033	РРМ	_	2xMDL:	0.0011	
Standard	Deviation: _	0.00017	PPM	Analyt	te Conc.:	0.0040	
	MDL:	0.00053	PPM	.i 	10xMDL:	0.0053	
Soil MDL Based On One Gram To 50 ml 0.026 ug/g Acceptable Analyte Concentration							
		YES	2xMDL	< Analyte Cor	1c. < 10x	MDL	

						
PARAMETER:	Cu		METHO	DDOLOGY: ICP		
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV IC	P	
METHOD of PREP	ARATION:	EPA 3050				
METHOD of A	ANALYSIS:			EPA6010/3050		
MATRIX:	SOIL			_		
PREPARER:	VPK		DAT	E OF PREPARATION	I: <u>07-Dec-00</u>	
ANALYST:	MDK			DATE OF ANALYSIS	8: <u>08-Dec-00</u>	
Ana	llyte Conc.:	0.0200		Analyte Units	s: PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.0234	PPM		
	2		0.0238	PPM		
	3		0.0256	PPM	_	
	4		0.0237	PPM		
	5		0.0245	PPM		
	6		0.0247	PPM		
	7		0.0232	PPM PPM		
Mean Analyte	Recovery: _	0.0241	PPM	_ 2xMDL	.: 0.0054	
Standard	Deviation: _	0.00086	PPM	Analyte Conc	0.0200	
	MDL:	0.00271	PPM	10xMDL	.: 0.0271	
Soil MDL Based On One Gram To 50 ml 0.135 ug/g Acceptable Analyte Concentration						
				< Analyte Conc. < 10		

						
PARAMETER:	Со		METHO	DDOLOGY: ICP		
INSTRUMENT IDENT	TFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	<u> </u>	
METHOD of PREP	ARATION:	EPA 3050				
METHOD of A	NALYSIS:			EPA6010/3050	 	
MATRIX:	SOIL			-		
PREPARER:	VPK		DAT	E OF PREPARATION:	07-Dec-00	
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00	
Ana	lyte Conc.:	0.0200		Analyte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.0212	PPM	=	
			0.0193	PPM	_	
	$\frac{-\frac{2}{3}}{3}$		0.0195	PPM	-	
	4		0.0208	PPM	-	
•	5		0.0191	PPM	-	
	6		0.0197	PPM	-	
	7		0.0199	PPM	_	
Mean Analyte	Recovery:	0.0199	PPM	2xMDL:	0.0049	
Standard	Deviation:	0.00078	РРМ	Analyte Conc.:	0.0200	
	MDL:	0.00247	PPM -	10xMDL:	0.0247	
Soil MDL Base	ed On One (A	cceptable	0.123 ug/g Analyte Concentration Analyte Conc. < 10x	-	
		5		· , · · · · · · · · · · · · · · ·	=	

PARAMETER:	Cr		METHO	ODOLOGY: ICP					
INSTRUMENT IDENT	NSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP								
METHOD of PREP	ARATION:	EPA 3050							
METHOD of A	ANALYSIS:	· · · · · · · · · · · · · · · · · · ·		EPA6010/3050					
MATRIX:	SOIL			_					
PREPARER:	VPK		DAT	E OF PREPARATION	: <u>07-Dec-00</u>				
ANALYST:	MDK			DATE OF ANALYSIS	: <u>08-Dec-00</u>				
Ana	lyte Conc.:	0.0200		Analyte Units	PPM (e.g. ppm)				
		Students	T Value:	3.143					
	Replicate #		Result	Units					
	1		0.0258	PPM	==				
	2		0.0264	PPM	-				
	3		0.0276	PPM					
	4	 -	0.0275	PPM	_				
	5		0.0275	PPM	_				
	6		0.0269	PPM	_				
	7		0.0265	PPM	_				
Mean An a lyte	Recovery:	0.0268	PPM	2xMDL	0.0041				
Standard	Deviation: _	0.00065	PPM	Analyte Conc.:	0.0200				
	MDL: 4	0.00203	PPM	10xMDL	0.0203				
Soil MDL Base	ed On One G	<u>A</u>	cceptable	0.101 ug/g Analyte Concentration Analyte Conc. < 10:					

PARAMETER:	Ni		METHO	DOLOGY: ICP	·				
INSTRUMENT IDENT	NSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP								
METHOD of PREP	ARATION: <u>E</u>	PA 3050							
METHOD of A	NALYSIS: _			EPA6010/305	50				
MATRIX:	SOIL			_					
PREPARER:	VPK		DATI	E OF PREPARAT	ION:	07-Dec-00			
ANALYST:	MDK			DATE OF ANALY	'SIS:	08-Dec-00			
Ana	lyte Conc.:	0.0200		Analyte U	Inits:	PPM	(e.g. ppm)		
		Students	T Value:	3.143					
	Replicate #		Result	Un	its	_			
	1		0.0211	PF	M	- -			
	2		0.0200	PF	M_	-			
	3		0.0212	PF	M	_			
	4		0.0216	PP		<u>-</u>			
,	5		0.0199	PP		•			
	6		0.0202	PP		_			
	7		0.0200	PP	M	_			
Mean Analyte	Recovery: _	0.0206	РРМ	. 2xN	/IDL:	0.0043			
Standard	Deviation:	0.00068	PPM	Analyte Co	onc.:	0.0200			
	MDL:	0.00214	РРМ	10xN	IDL:	0.0214			
Soil MDL Base	ed On One Gr	A	cceptable	0.107 ug/g Analyte Concentr Analyte Conc. <					

PARAMETER:	МО		METHO	DDOLOGY: IC	Р		·
INSTRUMENT IDENT	IFICATION .	AND DESC	RIPTION:	OPTIMA 3000	DV ICF)	
METHOD of PREP	ARATION: [EPA 3050					
METHOD of A	NALYSIS:			EPA6010/	3050		
MATRIX:	SOIL			_			
PREPARER:	VPK		DAT	E OF PREPAR	ATION:	07-Dec-00	
ANALYST:	MDK			DATE OF ANA	ALYSIS:	08-Dec-00	
Ana	lyte Conc.: _	0.0040		Analyt	e Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143	<u>;=====</u> :	 	
	Replicate #		Result		Units	_	
	1		0.0039		PPM	=	
	2		0.0043		PPM	_	
	3		0.0041		PPM	_	
	4		0.0035		PPM	_	
	5		0.0035		PPM	_	
	6		0.0038		PPM		
	7		0.0048	· · · · · · · · · · · · · · · · · · ·	PPM		
Mean Analyte	Recovery: _	0.0040	PPM	_	2xMDL:	0.0029	
Standard	Deviation: _	0.00046	PPM	Analyte	Conc.;	0.0040	
	MDL:	0.00146	PPM	_ 10	0xMDL:	0.0146	
Soil MDL Base	d On One G	A	cceptable	0.073 ug. Analyte Conce Analyte Conc	ntration		

PARAMETER: Pb METHODOLOGY: ICP								
INSTRUMENT IDENT	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	D			
METHOD of PREPARATION: EPA 3050								
METHOD of A	NALYSIS:			EPA6010/3050				
MATRIX:	SOIL	-		_				
PREPARER:	VPK		DATI	E OF PREPARATION:	07-Dec-00			
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00			
Ana	lyte Conc.:	0.0200		Analyte Units:	PPM (e.g. ppm)			
		Students	T Value:	3.143				
	Replicate #		Result	Units				
	1		0.0274	PPM	=			
	2		0.0233	PPM	-			
	3	·	0.0231	PPM	-			
	4		0.0271	PPM	_			
	 5		0.0224	PPM	-			
	6		0.0210	PPM	-			
	$\frac{-}{7}$		0.0213	PPM	-			
Mean Analyte	Recovery:	0.0237	PPM	2xMDL:	0.0163			
Standard	Deviation:	0.00259	PPM	Analyte Conc.:	0.0200			
	MDL:	0.00814	⊊PPM ∩	10xMDL:	0.0814			
Soil MDL Base	ed On One C	A	cceptable	0.407 ug/g Analyte Concentration Analyte Conc. < 10x				
		0						

							
PARAMETER:	SB		METH	ODOLOGY: ICP			
INSTRUMENT IDEN	FIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	<u> </u>		
METHOD of PREPARATION: EPA 3050							
METHOD of A	ANALYSIS:			EPA6010/3050			
MATRIX:	SOIL			_			
PREPARER:	<u>VPK</u>		DAT	E OF PREPARATION:	07-Dec-00		
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00		
Ana	lyte Conc.:	0.1000		Analyte Units:	PPM	(e.g. ppm)	
		Students	T Value:	3.143			
	Replicate #		Result	Units	_		
	1		0.0924	PPM	=		
			0.0980	PPM	-		
	3		0.0861	PPM	-		
	4		0.0935	PPM	-		
			0.0953	PPM	-		
	6		0.0950	PPM	-		
	7		0.1045	PPM	-		
Mean Analyte	Recovery:	0.0950	PPM	2xMDL:	0.0351		
Standard	Deviation: _	0.00559	PPM	Analyte Conc.:	0.1000		
	MDL:	0.01756	PPM	10xMDL:	0.1756		
Soil MDL Base	ed On One G	A	cceptable	0.878 ug/g Analyte Concentration			
		YES	- 2xMDL	< Analyte Conc. < 10x	MDL		

PARAMETER: Se METHODOLOGY: ICP							
INSTRUMENT IDENT	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	o 		
METHOD of PREP	ARATION:	EPA 3050					
METHOD of A	ANALYSIS:	- 		EPA6010/3050			
MATRIX:	SOIL	 -	<u>.</u>	_			
PREPARER:	VPK		DAT	E OF PREPARATION:	07-Dec-00		
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00		
Ana	lyte Conc.:	0.1000		Analyte Units:	PPM	(e.g. ppm)	
		Students	s T Value:	3.143			
	Replicate #		Result	Units	_		
	1		0.0630	PPM	_		
	2		0.0698	PPM	_		
	3		0.0501	PPM	_		
	4		0.0753	PPM	_		
	5		0.0723	PPM	_		
	6		0.0731	PPMPPM_	_		
	7		0.0707	PPM	_		
Mean Analyte	Recovery:	0.0677	PPM	2×MDL:	0.0546		
Standard	Deviation:	0.00868	PPM	Analyte Conc.:	0.1000		
	MDL:	0.02729	PPM	: 10xMDL:	0.2729		
Soil MDL Base	ed On One C	A	cceptable	1.365 ug/g Analyte Concentration Analyte Conc. < 10x			

		·····			_		=	
PARAMETER:	PARAMETER: TI METHODOLOGY: ICP							
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 D	V ICF	<u> </u>		
METHOD of PREPARATION: EPA 3050								
METHOD of A	ANALYSIS:		·	EPA6010/305	50			
MATRIX:	SOIL			-				
PREPARER:	VPK		DAT	E OF PREPARAT	ION:	07-Dec-00		
ANALYST:	MDK			DATE OF ANALY	'SIS:	08-Dec-00		
Ana	lyte Conc.:	0.1000		Analyte U	Inits:	PPM (e.g. ppn	n)	
		Students	T Value:	3.143			=	
	Replicate #		Result	Un	its			
	1	-	0.0731	PF	M			
	2		0.0892	PF		-		
	3		0.0749	PF		-		
	4		0.0913	PF		-		
	5		0.0898	PF		-		
	6		0.0722	PF		-		
	7	······································	0.1106	PF	M	-		
Mean Analyte	Recovery:	0.0859	PPM	2×N	IDL:	0.0866		
Standard	Deviation: _	0.01377	PPM	Analyte Co	onc.:	0.1000		
	MDL: _	0.04329	PPM	10xN	IDL:	0.4329		
Soil MDL Base	ed On One G	Δ	cceptable	2.164 ug/g Analyte Concentr Analyte Conc. <				

PARAMETER: V METHODOLOGY: ICP									
INSTRUMENT IDENT	NSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP								
METHOD of PREP	ARATION:	EPA 3050							
METHOD of A	NALYSIS:			EPA6010/3050					
MATRIX:	SOIL			-					
PREPARER:	VPK	_	DAT	E OF PREPARATION:	07-Dec-00				
ANALYST:	MDK	-		DATE OF ANALYSIS:	08-Dec-00				
Ana	lyte Conc.:	0.0200		Analyte Units:	PPM (e.g. ppm				
		Students	T Value:	3.143					
	Replicate #	<u> </u>	Result	Units	_				
•	1	·	0.0344	PPM	-				
•	2		0.0362	PPM	-				
•	3		0.0325	PPM	-				
,	4		0.0357	PPM					
•	5		0.0342	PPM	-				
•	6		0.0341	PPM	-				
	7		0.0347	PPM	- -				
Mean Analyte	Recovery:	0.0345	PPM	2xMDL:	0.0074				
Standard	Deviation:	0.00118	PPM	Analyte Conc.:	0.0200				
	MDL:	0.00371	PPM	10xMDL:	0.0371				
Soil MDL Base	d On One	<u>A</u>	cceptable	0.186 ug/g Analyte Concentration Analyte Conc. < 10x	•				
		153-	TVILIDE,	- Analyte Conc. < 10x	1110 L				

						
PARAMETER:	Zn		METHO	DDOLOGY: ICP		
INSTRUMENT IDENT	TIFICATION .	AND DESC	RIPTION:	OPTIMA 3000 DV	СР	
METHOD of PREPARATION: EPA 3050						
METHOD of A	NALYSIS: _			EPA6010/3050		
MATRIX:	SOIL		- · · · · · · · · · · · · · · · · · · ·	<u>-</u>		
PREPARER:	VPK		DAT	E OF PREPARATIO	N: <u>07-Dec-00</u>	
ANALYST:	MDK			DATE OF ANALYSI	S: <u>08-Dec-00</u>	
Ana	lyte Conc.: _	0.0200		Analyte Uni	ts: PPM (e.g. ppm)	
		Students	T Value:	3.143		
	Replicate #		Result	Units		
,	1		0.0596	PPM		
	2		0.0577	PPM	·	
	3		0.0627	PPM		
	4		0.0645	PPM		
,	5	 	0.0552	PPM		
	6		0.0613	PPM		
			0.0614	PPM		
Mean Analyte	Recovery: _	0.0603	PPM	_ 2xMD	L: 0.0196	
Standard	Deviation: _	0.00312	PPM	Analyte Con	0.0200	
	MDL: _	0.00980	РРМ	10xMD	L: <u>0.0980</u>	
Soil MDL Based On One Gram To 50 ml Acceptable Analyte Concentration YES - 2xMDL < Analyte Conc. < 10xMDL						

Method Detection Limit Study

40 CFR Part 136, Appendix B

Anal. Method: EPA 6010

Instrument: OPTIMA

Prep Method: EPA 7470

Analyte Conc:

0.15ug/l

Analyte: HG

Analyst: ESH

Matrix: TCLP Sol.#1

Study Date: 12/5/00

Replica	tes
#1	0.12
#2	0.16
#3	0.17
#4	0.17
#5	0.14
#6	0.16
#7	0.16

Mean 0.1543

Standard Dev. 0.0181

Student's T Value

3.143

* MDL 0.0570 ug/l

RL 0.1(0ug/l

	011			225	^	
PARAMETER:	CN-		METH	DDOLOGY: 335		
INSTRUMENT IDENT	TIFICATION	AND DESCR	RIPTION:			
METHOD of PREP	ARATION:	Distillation				
METHOD of A	NALYSIS:	335.2				·
MATRIX:	WATER			_		
PREPARER:	RDC		DAT	E OF PREPARATION	: 29-Nov-00	
ANALYST:	RDC			DATE OF ANALYSIS	: 29-Nov-00	
Ana	lyte Conc.:	0.0040		Analyte Units	E PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1	·	0.0038	PPM	==	
	2		0.0032	PPM		
	3		0.0038	PPM		
	4		0.0036	PPM		
	5		0.0038	PPM	_	
	6		0.0038	PPM	_	
	7	·····	0.0040	PPM		
Mean Analyte	Recovery:	0.0037	PPM	_ 2xMDl	.: 0.0016	
Standard	Deviation:	0.00025	PPM	_ Analyte Conc	0.0040	
	MDL:	A SHIP US	WEEN'S	3 10×MDl	.: 0.0080	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

Spike Solution: dilution of stock standard

·					
PARAMETER:	Total A	Italinity	METHO	DDOLOGY:	
INSTRUMENT IDENT	TIFICATION A	AND DESCR	RIPTION:		
METHOD of PREP	ARATION: _				
METHOD of A	NALYSIS: _	Total Alkalin	ity EPA 3	10.1	
MATRIX:	_Wates	-			
PREPARER:	RDC		DAT	= E OF PREPARATION:	11/1
ANALYST:	RDC			DATE OF ANALYSIS:	04-Dec-00
Ana	lyte Conc.: _	1.4400		Analyte Units.	PPM (e.g. ppm)
		Students	T Value:	3.143	
	Replicate #		Result	Units	
	1		1.1000	PPM	=
	2		1.6000	PPM	-
	3		1.4000	PPM	-
	4		1.6000	PPM	-
	5	-	1.4000	PPM	-
	6		1.1000	PPM	-
	7		1.4000	PPM	-
Mean Analyte	Recovery: _	1.3714	PPM	2xMDL:	1.2941
Standard	Deviation: _	0.20587	PPM	Analyte Conc.:	1.4400
	MDL: 🂆	£0.64704	PPM 1	10xMDL:	6.4704

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

Spike Solution:

ERA 99101

							
PARAMETER:	TDS		METHO	DOLOGY: _			<u>-</u>
INSTRUMENT IDENT	TIFICATION A	AND DESC	RIPTION:		<u></u> ,,,		
METHOD of PREP	ARATION: _				<u>:</u>		
METHOD of A	ANALYSIS: _	160.1					
MATRIX:	WATER			_			
PREPARER:			DAT	E OF PREPAR	RATION:		
ANALYST:	RDC			DATE OF AN	ALYSIS:	29-Nov-00	
Ana	llyte Conc.: _	6.7400		Analy	rte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143			
	Replicate #		Result		Units	_	
	1		8.4000		PPM	-	
	2		9.6000		PPM	•	
	3		8.2000		PPM	-	
	4		10.8000		PPM	-	
	5		11.4000		PPM	-	
	6		8.6000		PPM	_	
	7		7.6000		PPM	- -	
Mean Analyte	Recovery: _	9.2286	PPM	-	2xMDL:	8.9320	
Standard	Deviation:	1.42093	РРМ	Analyt	e Conc.:	6.7400	
	MDL:		《新西斯斯		10xMDL:	44.6599	
Acceptable Analyte Concentration NO - Analyte Conc. < 2xMDL							

Spike Solution:

PARAMETER: TSS							
METHOD of PREPARATION: METHOD of ANALYSIS: 160.2 MATRIX: WATER PREPARER: RDC DATE OF PREPARATION: ANALYST: RDC DATE OF ANALYSIS: 29-Nov-00 Analyte Units: PPM (e.g. p.	PARAMETER:	TSS		METHO	DDOLOGY:		
METHOD of ANALYSIS: 160.2 MATRIX: WATER PREPARER: RDC DATE OF PREPARATION: ANALYST: RDC DATE OF ANALYSIS: 29-Nov-00 Analyte Conc.: 7.3800 Analyte Units: PPM (e.g. p. pr.) Students T Value: 3.143 Replicate # Result Units 1 8.8400 PPM 2 9.4400 PPM 3 9.2000 PPM 4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066	INSTRUMENT IDENT	TIFICATION .	AND DESCF	RIPTION:			
MATRIX: WATER PREPARER: RDC	METHOD of PREP	ARATION: _	- <u>.</u>				
PREPARER: RDC DATE OF PREPARATION: ANALYST: RDC DATE OF ANALYSIS: 29-Nov-00 Analyte Conc.: 7.3800 Analyte Units: PPM (e.g. p. Students T Value: 3.143 Replicate # Result Units 1 8.8400 PPM 2 9.4400 PPM 3 9.2000 PPM 4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066	METHOD of A	NALYSIS: _	160.2				
ANALYST: RDC DATE OF ANALYSIS: 29-Nov-00 Analyte Conc.: 7.3800 Analyte Units: PPM (e.g. posterior of the property of the prop	MATRIX:	WATER		·····	_		
Students T Value: 3.143	PREPARER:	RDC		DAT	E OF PREPARATION:		
Students T Value: 3.143	ANALYST:	RDC			DATE OF ANALYSIS:	29-Nov-00	
Replicate # Result Units 1 8.8400 PPM 2 9.4400 PPM 3 9.2000 PPM 4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066	Ana	lyte Conc.: _	7.3800		Analyte Units:	PPM	(e.g. ppm)
1 8.8400 PPM 2 9.4400 PPM 3 9.2000 PPM 4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066			Students	T Value:	3.143		, , , , ,
1 8.8400 PPM 2 9.4400 PPM 3 9.2000 PPM 4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066		Replicate #		Result	Units		
2 9.4400 PPM 3 9.2000 PPM 4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066			- : 		PPM	=	
3 9.2000 PPM 4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066						-	
4 8.7200 PPM 5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066						-	
5 9.0800 PPM 6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066						-	
6 8.9200 PPM 7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066			-			-	
7 9.0000 PPM Mean Analyte Recovery: 9.0286 PPM 2xMDL: 1.5066		6			PPM	_	
		7			PPM	-	
Standard Deviation: 0.23968 PPM Analyte Conc.: 7.3800	Mean Analyte	Recovery:	9.0286	PPM	2xMDL:	1.5066	
	Standard	Deviation: _	0.23968	PPM	Analyte Conc.:	7.3800	
MDL: 9.75332 PPM 10xMDL: 7.5332		MDL:	0.75332	PPM	10xMDL:	7.5332	

<u>Acceptable Analyte Concentration</u> **YES - 2xMDL < Analyte Conc. < 10xMDL**

Spike Solution: 8076 20/250 DILUTION

PARAMETER:	F-		METH	DDOLOGY: EPA 340.2		
INSTRUMENT IDENT	TIFICATION	AND DESCR	RIPTION:	ORION EA940		
METHOD of PREP	ARATION:			·····		· · · · · · ·
METHOD of A	NALYSIS:	340.2				
MATRIX:	WATER	····		_		
PREPARER:	EH		DAT	E OF PREPARATION:		
ANALYST:	EH			DATE OF ANALYSIS:	11-Dec-00	
Ana	lyte Conc.:	0.0500		Analyte Units:	PPM (e.g. ppm)
		Students	T Value	3.143		
		Otadents	i i value.	3.143		
	Replicate #	!	Result	Units	=	
	1		0.0522	PPM	-	
	2		0.0630	PPM	-	
	3		0.0661	PPM	-	
	4		0.0645	PPM	_	
	5		0.0655	PPM	_	
	6		0.0588	PPM	_	
	7		0.0612	PPM	-	
Mean Analyte	Recovery:	0.0616	PPM	_ 2xMDL:	0.0306	
Standard	Deviation:	0.00487	РРМ	Analyte Conc.:	0.0500	
	MDL:	0.01530	PPM	10xMDL:	0.1530	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

Spike Solution: STOCK STANDARD

40 CFR Part 136, Appendix B

Anal. Method: 350.1

Instrument: alpchem M6000

Prep Method:

Analyte Conc: 0.2

Analyte: NH3/N

Analyst: JHN

Matrix: WATER

Study Date: 3/3/00

UNITS: PPM

Replicates				
#1	0.1849			
#2	. 0.1936			
#3	0.1904			
#4	0.1851			
#5	0.1995			
#6	0.1954			
#7	0.2022			

Mean 0.1930

Standard Dev. 0.0067

Student's T Value 3.143

MDL 0.0210 P.P.M

RL 0.05

PARAMETER:	TKN		METHO	DDOLOGY: EPA 351.2	2	
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	ALPCHEM ANION	ANALYZER	<u> </u>
METHOD of PREF	PARATION: _					
METHOD of	ANALYSIS: <u>E</u>	PA 351.2				
MATRIX:	WATER			_		
PREPARER:	JHN		DAT	E OF PREPARATION:	27-Nov-00	
ANALYST:	JHN			DATE OF ANALYSIS:	29-Nov-00	
Ana	alyte Conc.: _	0.1000		Analyte Units:	PPM	(e.g. ppm)
		Students	s T Value:	3.143		
	Replicate #		Result	Units	=	
	1		0.1096	PPM	- -	
	2		0.1212	PPM	-	
	3	· -	0.1150	PPM	_	
	4		0.1152	PPM	_	
	5		0.1063	PPM	-	
	<u>6</u> 7		0.1020	PPM	-	
			0.1112	PPM	-	
Mean Analyte	Recovery: _	0.1115	PPM	2xMDL:	0.0398	
Standard	Deviation:	0.00634	PPM	Analyte Conc.:	0.1000	
	MDL:			10xMDL:	0.1992	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

Spike Solution: STOCK STANDARD

PARAMETER:	Total Phospi	norous	METHO	DOLOGY:	365.1		
INSTRUMENT IDENT	TIFICATION A	AND DESCR	IPTION:	Alpchem	Anion	Analyzer	
METHOD of PREP	ARATION: 1	ecator Bloci	k Digestio	on			
METHOD of A	NALYSIS: _	365.1					
MATRIX:	WATER			_			
PREPARER:	JHN		DAT	E OF PREP	ARATION:	06-Nov-00	
ANALYST:	JHN			DATE OF A	ANALYSIS:	06-Nov-00	
Ana	llyte Conc.: _	0.0500		An	alyte Units:	РРМ	(e.g. ppm)
		Students	T Value:	3.143	3		
	Replicate #		Result		Units		
	1		0.0620		PPM	=	
	2		0.0560		PPM	_	
	3		0.0610		PPM	_	
	4		0.0680		PPM	_	
	5		0.0630		PPM	_	
	6		0.0650		PPM	_	
	7		0.0670		PPM	_	
Mean Analyte	Recovery: _	0.0631	РРМ	_	2×MDL:	0.0255	
Standard	I Deviation: _	0.00406	РРМ	_ Ana	alyte Conc.:	0.0500	
	MDL:	0.01276	PPM		10xMDL:	0.1276	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	Nitrate/Nitri	te as	N	METHO	DOLOGY:	Automated	l Colorimetric	<u> </u>	
INSTRUMENT IDENT	IFICATION	AND	DESCRI	PTION:	Alpchem	Anion	Analyzer		
METHOD of PREP	ARATION:	NA							
METHOD of A	NALYSIS:		353.2		·•		·· 		
MATRIX:	WATER				_				
PREPARER:	NA			DATI	E OF PREP	ARATION:	NA		
ANALYST:	JHN				DATE OF A	NALYSIS:	28-Nov-00		
Ana	lyte Conc.:	0	.1000		Ana	alyte Units:	PPM	(e.g.	ppm)
	· · · · · · · · · · · · · · · · · · ·	s	tudents 1	Γ Value:	3.143				

Replicate #	Result	Units
1	0.1050	PPM
2	0.1170	PPM
3	0.1060	PPM
4	0.1140	PPM
5	0.1180	PPM
6	0.1180	PPM
7	0.1180	PPM

Mean Analyte Recovery: 2xMDL: 0.1137 PPM 0.0364

0.1000 Standard Deviation: 0.00579 Analyte Conc.: PPM

> MDL: SEE BY LEVELY 10xMDL: 0.1821

> > Acceptable Analyte Concentration YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	ВА		METHO	DOLOGY: <u>IC</u>	Р		
INSTRUMENT IDENT	TIFICATION A	AND DESC	RIPTION: 1	OPTIMA 3000	DV ICE)	
METHOD of PREP	ARATION: <u>E</u>	PA 3010					
METHOD of A	NALYSIS: _		<u></u>	EPA 6010/	200.7		
MATRIX:	WATER (7	CLP S.	1. #1)				
PREPARER:	MJF		DATE	OF PREPAR	ATION:	12-Dec-00	
ANALYST:	MDK		[DATE OF ANA	ALYSIS:	12-Dec-00	
Ana	lyte Conc.: _	0.0500		Analyt	te Units:	PPM	(e.g. ppm)
		Students	s T Value: _	3.143			
	Replicate #		Result		Units		
	1		0.1051		PPM	=	
	2		0.1069		PPM	-	
	3		0.1020		PPM	-	
	4		0.1020		PPM	-	
	5		0.1035		PPM	-	
	6		0.1016		PPM	-	
	7		0.1011		PPM	= -	
Mean Analyte	Recovery: _	0.1032	PPM		2xMDL:	0.0134	
Standard	Deviation:	0.00213	PPM_	Analyte	Conc.:	0.0500	

MDL: 0.00670 PPM

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

10xMDL: 0.0670

PARAMETER: CU	METHO	DOLOGY: ICP				
INSTRUMENT IDENTIFICATION AND DESCRIPTION: OPTIMA 3000 DV ICP						
METHOD of PREPARATION	N: EPA 3010					
METHOD of ANALYSIS	S:	EPA 6010/200.7				
MATRIX: WATER	(TCLP Sol #1)					
PREPARER: MJF	DATE	OF PREPARATION:	12-Dec-00			
ANALYST: MDK	_	DATE OF ANALYSIS:	12-Dec-00			
Analyte Conc	.: 0.0500	Analyte Units:	PPM (e.g. ppm)			
	Students T Value:	3.143				
Replicate	# Result	Units				
1	0.0642	PPM				
2	0.0573	PPM				
3	0.0577	PPM				
4	0.0661	PPM				
5	0.0564	PPM				
6	0.0614	PPM				
	0.0631	<u>PPM</u>				

PPM

PPM

"ÞÞM

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

2xMDL: 0.0239

10xMDL: ___ 0.1193

0.0500

Analyte Conc.:

SPIKE SOLUTION EEICQ19

Mean Analyte Recovery: _____0.0609

Standard Deviation: 0.00380

MDL: 5.01193

PARAMETER:	AG		METHO	DDOLOGY: ICP		
INSTRUMENT IDENT	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 D\	/ ICP	
METHOD of PREP	ARATION:	EPA 3010				
METHOD of A	ANALYSIS:			EPA 6010/200	.7	
MATRIX:	TCLP SOL	. #1		_		
PREPARER:	MJF		DAT	E OF PREPARATI	ON:	05-Dec-00
ANALYST:	MDK			DATE OF ANALY	SIS:	08-Dec-00
Ana	lyte Conc.:	0.1000		Analyte U	nits:	PPM (e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Uni	its	
	1	<u>_</u>	0.1158	PP	М	
	2		0.1094	PP	M	•
	3		0.1439	PP	M	•
	4		0.1080	PP	M	•
	5		0.1175	PΡ	M	•
	6		0.1090	PP	М	•
	7		0.1188	PP	М	•
Mean Analyte	Recovery:	0.1175	PPM	2xM	IDL:	0.0782
Standard	Deviation:	0.01245	PPM	Analyte Co	nc.:	0.1000
	MDL:	0.03912	PPM	10xM	IDL:	0.3912

Acceptable Analyte Concentration YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	AS		METHO	DOLOGY: ICP		
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE)	
METHOD of PREP	ARATION:	EPA 3010				
METHOD of ANALYSIS: EPA 6010/200.7						
MATRIX:	TCLP SOL.					
PREPARER:	MJF		DATE	OF PREPARATION:	05-Dec-00	
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00	
Ana	llyte Conc.:	0.1000		Analyte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units_		
	1		0.1091	PPM	=	
	2		0.0961	PPM	-	
	3		0.0957	PPM	-	
	4		0.0969	PPM	-	
	5		0.0969 0.0936	PPM PPM	- -	
	5 6		0.0936 0.1083	PPM PPM	• • •	
	5		0.0936	PPM	- - - -	
Mean A nal yte	5 6 7	0.0993	0.0936 0.1083	PPM PPM	0.0407	
•	5 6 7		0.0936 0.1083 0.0957	PPM PPM PPM		

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	DDOLOGY: ICP				
INSTRUMENT IDENT	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE)
METHOD of PREP	ARATION:	EPA 3010			
METHOD of A	ANALYSIS:			EPA 6010/200.7	
MATRIX:	TCLP SOL.	#1		_	
PREPARER:	MJF		DAT	E OF PREPARATION:	05-Dec-00
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00
Ana	ilyte Conc.:]	0.0200		Analyte Units:	PPM (e.g. ppm)
		Students	T Value:	3.143	
	Replicate #		Result	Units	
	1		0.0196		=
	2		0.0222	PPM	-
	3		0.0192	PPM	-
	4		0.0187	PPM	•
	5		0.0182	PPM	
	6		0.0187	PPM	
	7	·	0.0186	PPM	-
Mean Analyte	Recovery:	0.0193	PPM	2xMDL:	0.0085
Standard	Deviation:	0.00135	PPM	Analyte Conc.:	0.0200
	MDL:	0.00424	PPM	10xMDL:	0.0424
		<u> </u>	cceptable	Analyte Concentration	

YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	CR		METHO	DOLOGY: ICP	
INSTRUMENT IDEN	TIFICATION	AND DESCR	RIPTION:	OPTIMA 3000 DV ICE	D
METHOD of PREF	PARATION: 1	EPA 3010			
METHOD of	ANALYSIS: _			EPA 6010/200.7	
MATRIX:	TCLP SOL.	#1			
PREPARER:	MJF		DATE	OF PREPARATION:	05-Dec-00
ANALYST:	MDK		ſ	DATE OF ANALYSIS:	08-Dec-00
Ana	alyte Conc.: _	0.0200		Analyte Units:	PPM (e.g. ppm)
		Students	T Value:	3.143	
	Replicate #		Result _	Units	
				0.1110	_
	1		0.0202	PPM	=
			0.0202 0.0207		-
	1 2 3		0.0207	PPM	= - -
	2 3		0.0207 0.0214	PPM PPM PPM	= - -
	2		0.0207 0.0214 0.0202	PPM PPM	= - - -
	3 4		0.0207 0.0214	PPM PPM PPM PPM	= - - -
	2 3 4 5		0.0207 0.0214 0.0202 0.0182	PPM PPM PPM PPM PPM	= - - -
Mean Analyte	2 3 4 5 6 7		0.0207 0.0214 0.0202 0.0182 0.0187	PPM PPM PPM PPM PPM PPM	0.0070
•	2 3 4 5 6 7	0.0199	0.0207 0.0214 0.0202 0.0182 0.0187 0.0198	PPM PPM PPM PPM PPM PPM PPM	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	CR		METHO	DDOLOGY: ICP		
INSTRUMENT IDEN	TIFICATION A	ND DESCI	RIPTION:	OPTIMA 3000 DV ICE)	
METHOD of PREP	PARATION: E	PA 3010				
METHOD of A	ANALYSIS:			EPA 6010/200.7		
MATRIX:	TCLP Sol. #1					
PREPARER:	MJF		DAT	- E OF PREPARATION:	05-Dec-00	
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00	
Ana	alyte Conc.:	0.0200		Analyte Units:	PPM	(e.g. ppm)
		Students	s T Value:	3.143		
	Replicate #		Result	Units	_	
	1		0.0202	PPM	-	
	2		0.0207	PPM	-	
	3		0.0214	PPM	_	
	4		0.0202	PPM	_	
	5		0.0190	PPM	-	
	<u>6</u> 7		0.0203	PPM PPM	-	
			0.0198		-	
Mean Analyte	Recovery:	0.0202	PPM	2xMDL:	0.0047	
Standard	Deviation:	0.00074	PPM	Analyte Conc.:	0.0200	
	MDL:	0.00233	PPM	10xMDL:	0.0233	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER: FE METHODOLOGY: ICP							
INSTRUMENT IDEN	TIFICATION A	ND DESC	RIPTION:	OPTIMA 30	00 DV ICF		
METHOD of PREP	ARATION: EF	PA 3010					
METHOD of A	ANALYSIS:	ALYSIS: EPA 6010/200.7					<u></u>
MATRIX:	TCLP Sol. #1			_			
PREPARER:	MJF		DAT	E OF PREPA	RATION:	05-Dec-00	
ANALYST:	MDK			DATE OF A	NALYSIS:	08-Dec-00	
Ana	lyte Conc.:	0.1000		Anai	yte Units:	PPM	(e.g. ppm)
		Students	s T Value:	3.143			
	Replicate #		Result		Units		
	1		0.1318		PPM	=	
	2		0.1421		PPM	- -	
	3		0.1368		PPM	- -	
	4		0.1666		PPM	-	
	5		0.1385		PPM	-	
	6		0.1536 0.1576		PPM PPM	-	
Mean Analyte	Recovery:	0.1467	PPM	_	2xMDL:	0.0800	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

Analyte Conc.: 0.1000

10xMDL: ___ 0.3998

SPIKE SOLUTION EEICQ19

Standard Deviation: 0.01272 PPM

MDL: 0.03998 PPM

PARAMETER:	NI		METHO	DOLOGY: ICP				
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICF) 			
METHOD of PREP	ARATION:	EPA 3010			<u> </u>			
METHOD of A	ANALYSIS:			EPA 6010/200.7				
MATRIX:	TCLP Sol.	#1						
PREPARER:	MJF		DATE	OF PREPARATION:	05-Dec-00			
ANALYST:	MDK		Ε	DATE OF ANALYSIS:	08-Dec-00			
Ana	llyte Conc.:	0.0200		Analyte Units:	PPM	(e.g. ppm)		
		Students	T Value:	3.143				
	Replicate #		Result	Units				
	1		0.0290	PPM	•			
	2		0.0281	PPM	-			
	3		0.0294	PPM	-			
	4	···	0.0294	PPM	-			
	5		0.0302	PPM	-			
	6		0.0308	PPM	•			
	$\frac{7}{7}$		0.0279	PPM	-			
Mean Analyte	Recovery:	0.0294	PPM	2xMDL:	0.0069			
Standard	Deviation:	0.00109	РРМ	Analyte Conc.:	0.0200			
,	MDL:	0.00344	PPM	10xMDL:	0.0344			

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	РВ		METHO	DOLOGY: ICP				
INSTRUMENT IDEN	TIFICATION .	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	D			
METHOD of PREP	ARATION: _	EPA 3010		<u> </u>				
METHOD of A	FANALYSIS: EPA 6010/200.7							
MATRIX:	TCLP Sol. #	1						
PREPARER:	MJF	<u> </u>	DAT	OF PREPARATION:	05-Dec-00			
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00			
Ana	llyte Conc.: _	0.1000		Analyte Units:	PPM	(e.g. ppm)		
		Students	T Value:	3.143				
		Otagerite	, value.	<u> </u>				
	Replicate#		Result	Units				
	1	`~''`	0.0915	PPM	=			
	2	· ·	0.0868	PPM	-			
	3		0.0858	PPM	-			
	4		0.0954	PPM	-			
	5		0.0936	PPM	-			
	6		0.0979	PPM	_			
	7		0.1072	PPM PPM	_			
Mean Analyte	Recovery: _	0.0940	PPM	2xMDL:	0.0457			
Standard	Deviation: _	0.00727	PPM	Analyte Conc.:	0.1000			
	MDL:	0.02284	PРM	10xMDL:	0.2284			

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	SB		METH	ODOLOGY: ICP		
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE		
METHOD of PREP	ARATION: E	PA 3010				
METHOD of A	ANALYSIS: _	-		EPA 6010/200.7		
MATRIX:	TCLP Sol. #	1		_		
PREPARER:	MJF		DAT	E OF PREPARATION:	05-Dec-00	
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00	
Ana	alyte Conc.: _	0.1000		Analyte Units:	PPM (e.g.	ppm)
		Students	T Value:	3.143		
	Replicate #	Students	T Value: Result	3.143 Units		
	Replicate #	Students			=	-
		Students	Result	Units	=	
	1	Students	Result 0.0568	Units PPM	=	
	1 2 3 4	Students	Result 0.0568 0.0456 0.0430 0.0545	Units PPM PPM PPM PPM	= - -	
	1 2 3 4 5	Students	Result 0.0568 0.0456 0.0430 0.0545 0.0485	Units PPM PPM PPM PPM PPM	= - -	
	1 2 3 4 5 6	Students	Result 0.0568 0.0456 0.0430 0.0545 0.0485 0.0492	Units PPM PPM PPM PPM PPM PPM	= - - -	
	1 2 3 4 5	Students	Result 0.0568 0.0456 0.0430 0.0545 0.0485	Units PPM PPM PPM PPM PPM	= - - -	
Mean Analyte	1 2 3 4 5 6 7		Result 0.0568 0.0456 0.0430 0.0545 0.0485 0.0492	Units PPM PPM PPM PPM PPM PPM	0.0326	
•	1 2 3 4 5 6 7	0.0488	Result 0.0568 0.0456 0.0430 0.0545 0.0485 0.0492 0.0442	Units PPM PPM PPM PPM PPM PPM PPM PPM		

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

						
PARAMETER:	SE		METH	DDOLOGY: ICP		
INSTRUMENT IDEN	TIFICATION A	ND DESC	RIPTION:	OPTIMA 3000 DV ICE)	
METHOD of PREF	ARATION: E	PA 3010				
METHOD of A	ANALYSIS:		······	EPA 6010/200.7		
MATRIX:	TCLP Sol. #1			_		
PREPARER:	MJF		DAT	E OF PREPARATION:	05-Dec-00	
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00	
Ana	alyte Conc.:	0.1000		Analyte Units:	<u>PPM</u> (e.	g. ppm)
		Student	s T Value:	3.143		
		0.200				
	Replicate #		Result	Units		
	1		0.1068	PPM	=	
	2	<u>, , , , , , , , , , , , , , , , , , , </u>	0.1135	PPM	-	
	3		0.1084	PPM	-	
	4		0.1051	PPM	-	
	5		0.1159	PPM	_	
	66		0.1048	PPM	_	
	7		0.1053	PPM	-	
Mean Analyte	Recovery:	0.1085	PPM	_ 2xMDL:	0.0279	
Standard	Deviation:	0.00444	PPM	Analyte Conc.:	0.1000	
	MDL:	0.01395	PPM	10xMDL:	0.1395	

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	TL	DOLOGY: ICP							
INSTRUMENT IDEN	TIFICATION	AND DESC	RIPTION:	OPTIMA 3000 DV ICE	<u> </u>				
METHOD of PREPARATION: EPA 3010									
METHOD of A	THOD of ANALYSIS: EPA 6010/200.7								
MATRIX:	TCLP Sol. :	#1		-					
PREPARER:	MJF		DAT	E OF PREPARATION:	05-Dec-00				
ANALYST:	MDK			DATE OF ANALYSIS:	08-Dec-00				
Ana	alyte Conc.:	0.1000		Analyte Units:	PPM	(e.g. ppm)			
		Students	T Value:	3.143					
		-1000							
	Replicate #		Result	Units					
	1		0.1162	PPM	=				
	2		0.1326	PPM	-				
	3		0.1218	PPM	-				
	4		0.1265	PPM	-				
	5		0.1209	PPM	-				
	6		0.1270	PPM	-				
	7		0.1415	PPM	- -				
Mean Analyte	Recovery:	0.1266	РРМ	2xMDL:	0.0527				
Standard	Deviation:	0.00839	PPM	Analyte Conc.:	0.1000				
	MDL:	0.02635	PPM	10xMDL:	0.2635				
		٨		Analyta Cancontration					

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

PARAMETER:	ZN		METH	ODOLOGY:	ICP		
INSTRUMENT IDEN	TIFICATION A	ND DESC	RIPTION:	OPTIMA 30	00 DV ICI	>	
METHOD of PREF	PARATION: E	PA 3010				·	
METHOD of A	ANALYSIS:			EPA 601	0/200.7		
MATRIX:	TCLP Sol. #1			_			
PREPARER:	MJF		DAT	E OF PREPA	RATION:	05-Dec-00	
ANALYST:	MDK			DATE OF A	NALYSIS:	08-Dec-00	
Ana	alyte Conc.:	0.1000		Ana	lyte Units:	РРМ	(e.g. ppm)
		Students	T Value:	3.143			
	Replicate #		Result		Units		
	1		0.3250		PPM	=	
	2		0.3206		PPM		
	3		0.3139		PPM	_	
	4		0.3255		PPM	_	
	5		0.3141		PPM	_	
	6		0.3352		PPM	_	
	7		0.3425		PPM	_	
Mean Analyte	Recovery:	0.3253	PPM		2xMDL:	0.0665	

PPM

MDL: 0.03325 PPM

<u>Acceptable Analyte Concentration</u> **YES - 2xMDL < Analyte Conc. < 10xMDL**

Analyte Conc.:

0.1000

10xMDL: 0.3325

SPIKE SOLUTION EEICQ-19

Standard Deviation: 0.01058

PARAMETER:						
INSTRUMENT IDEN	TIFICATION A	ND DESC	RIPTION:	OPTIMA 3000 DV ICE		
METHOD of PREP	ARATION: E	PA 3010				··
METHOD of ANALYSIS:				EPA 6010/200.7		
MATRIX:	TCLP Sol. #1	<u> </u>		_		
PREPARER:	MJF		DAT	E OF PREPARATION:	02-Jan-01	
ANALYST:	MDK			DATE OF ANALYSIS:	02-Jan-01	
Ana	llyte Conc.:	0.0020		Analyte Units:	PPM	(e.g. ppm)
		Students	T Value:	3.143		
	Replicate #		Result	Units		
	1		0.0020	PPM	=	
	2		0.0021	PPM	-	
	3	 	0.0020	PPM	-	
	4		0.0021	PPM	_	
	5		0.0023	PPM	_	
	6		0.0020	PPM	_	
	7		0.0021	PPM	-	
Mean Analyte	Recovery: _	0.0021	PPM	2xMDL:	0.0007	
Standard	Deviation:	0.00011	PPM	_ Analyte Conc.:	0.0020	
	MDL:	0.00034	РРМ	10xMDL:	0.0034	
				Analyte Concentration		

Acceptable Analyte Concentration
YES - 2xMDL < Analyte Conc. < 10xMDL

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Analyte Conc: 0.20 ppb

Prep Method: EPA 6020 CLP-M

Analyst:

KPB

Analyte: Ag Mass 107

Study Date:

10/1/99

Instrument: Perkin Elmer Elan 6000 ICP-MS

Matrix: .5% HCL 1% HCO3

0.13

Replicates		
0.13		
0.14		
0.13		
0.12		
0.13		
0.13		

#7

0.1300 Mean

0.005774 Standard Dev.

3.143 Student's T Value

MDL = 0.0181 PPB

1.0 RL

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 ppb

KPB

Analyte: Ag Mass 109

Analyst:

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replica	tes
#1	0.13
#2	0.13
#3	0.13
#4	0.12
#5	0.13
#6	0.13
#7	0.13

Mean 0.1286

Standard Dev.

0.00378

Student's T Value

3.143

* MDL - * 0.0149 PPB

1.0 RL

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 1.0 PPB

Analyte: Al Mass 27

Analyst:

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replica	tes
#1	1.82
#2	2.23
#3	2.95
#4	2.17
#5	1.1
#6	1.3
#7	1.28

1.8357 Mean

Standard Dev. 0.6641

Student's T Value 3.143

* MDE 2-0873 PPB

RL 5.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte: As Mass 75

Analyst:

KPB

(ArCl Corr. Se78)

Study Date:

10/1/99

Matrix:	.5%	HUL	1%	HCO	3

0.5

Replicates		
#1	0.49	
#2	0.45	
#3	0.39	
#4	0.47	
#5	0.4	
#6	0.47	

#7

Mean 0.4529

Standard Dev. 0.042706

Student's T Value 3.143

* MDL == 0.1342 PPB

RL 3.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte: As Mass 75

(ArCl Corr. Se82)

Analyst: **KPB**

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates		
#1	0.4	
#2	0.4	
#3	0.38	
#4	0.38	
#5	0.32	
#6	0.37	
#7	0.4	

0.3786 Mean

Standard Dev. 0.028536

Student's T Value 3.143

* MDL===0008974PPB=

RL 3.0

40 CFR Part 136, Appendix B

Anal. Method: EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte : Ba Mass 135

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replica	tes
#1	0.22
#2	0.22
#3	0.26
#4	0.22
#5	0.23
#6	0.22
#7	0.24

Mean 0.2300

Standard Dev. 0.015275

Student's T Value 3.143

* MDL - 0.0480 PPB

RL 5.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: Ba Mass 137

Analyst: **KPB**

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates		
#1	0.22	
#2	0.23	
#3	0.26	
#4	0.21	
#5	0.23	
#6	0.21	
#7	0.23	

0.2271 Mean

Standard Dev. 0.017043

3.143 Student's T Value

MDE 1 10:0536 PPB

5.0 RL

40 CFR Part 136, Appendix B

Anal. Method: EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte : Be Mass 9

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replicates		
#1	0.2	
#2	0.21	
#3	0.21	
#4	0.23	
#5	0.22	
#6	0.19	
#7	0.21	

Mean 0.2100

Standard Dev. 0.01291

Student's T Value 3.143

* MD1435004060PPB

RL 1.0

40 CFR Part 136, Appendix B

Anal. Method: EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte : Cd Mass 111

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replicates		
#1	0.21	
#2	0.2	
#3	0.21	
#4	0.21	
#5	0.21	
#6	0.2	
#7	0.21	

Mean 0.2071

Standard Dev. 0.00488

Student's T Value 3.143

* MDL& .0.0153/PPB

RL .10

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte : Cd Mass 114 (MoO corr.) Analyst:

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates		
#1	0.21	
#2	0.21	
#3	0.21	
#4	0.2	
#5	0.21	
#6	0.21	
#7	0.21	

Mean 0.2086

Standard Dev. 0.00378

Student's T Value 3.143

*MDL 00119 PPB

RL .10

40 CFR Part 138, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: Cd Mass 114

KPB Analyst:

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates		
#1	0.21	
#2	0.21	
#3	0.21	
#4	0.2	
#5	0.21	
#6	0.21	
#7 _	0.21	

0.2086 Mean

Standard Dev.

0.00378

Student's T Value

3.143

MDL= 0.0119 PPB

RL .10

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte: Co Mass 59

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

Analyst:

10/1/99

Replicates	
#1	0.4
#2	0.4
#3	0.4
#4	0.4
#5	0.39
#6	0.4
#7	0.4

0.3986 Mean

Standard Dev.

0.00378

Student's T Value

3.143

MDL 200119 PPB

RL 1.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte: Cr Mass 52

Analyst: **KPB**

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replica	Replicates	
#1	0.4	
#2	0.36	
#3	0.44	
#4	0.44	
#5	0.41	
#6	0.39	
#7	0.4	

Mean 0.4057

Standard Dev.

Student's T Value

3.143

0.0282

MDE 00886 P.P.B.

1.0 RL

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: Cu Mass 63

Analyst: **KPB**

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates	
#1	0.53
#2	0.5
#3	0.88
#4	0.55
#5	0.27
#6	0.45
#7	0.45

0.5186 Mean

Standard Dev.

0.184068

Student's T Value

3.143

* MDE 0.5785 PPB*

3.0 RL

40 CFR Part 136, Appendix B

Anal. Method: EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte : Cu Mass 65

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replica	Replicates	
#1	0.52	
#2	0.51	
#3	0.9	
#4	0.55	
#5	0.27	
#6	0.45	
#7	0.46	

Mean 0.5229

Standard Dev. 0.189887

Student's T Value 3.143

* MDL . 0 5968 PPB

RL 3.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 ppb

Analyte: Mn Mass 55

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replicates	
#1	0.23
#2	0.24
#3	0.24
#4	0.22
#5	0.26
#6	0.22
#7	0.26

Mean 0.2386

Standard Dev. 0.016762

Student's T Value 3.143

* MDL 35 0.0527 PPB

40 CFR Part 136, Appendix B

Anal, Method: EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 ppb

Analyte: Mo Mass 97

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replica	tes
#1	0.22
#2	0.21
#3	0.2
#4	0.2
#5	0.22
#6	0.2
#7	0.2

Mean 0.2071

Standard Dev. 0.009512

Student's T Value 3.143

* MDL 0.0299 PPB

40 CFR Part 136, Appendix B

Anal. Method: EPA

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 ppb

Analyte: Mo Mass 98

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replicates	
#1	0.21
#2	0.2
#3	0.2
#4	0.19
#5	0.21
#6	0.19
#7	0.19

Mean 0.1986

Standard Dev. 0.008997

Student's T Value 3.143

MDL - 0.0283 PPB

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: Ni Mass 60

Analyst:

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates	
#1	0.24
#2	0.22
#3	0.37
#4	0.24
#5	0.27
#6	0.21
#7	0.45

0.2857 Mean

Standard Dev.

0.089974

Student's T Value

3.143

MDE 0 2828 PPB

1.0 RL

40 CFR Part 136, Appendix B

Anal. Method: EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: Ni Mass 60 (CaO Corr.)

Analyst: k

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates	
#1	0.24
#2	0.22
#3	0.37
#4	0.24
#5	0.27
#6	0.21
#7	0.45

Mean 0.2857

Standard Dev. 0.089974

Student's T Value 3.143

* MDL 0:2828 PPB...

40 CFR Part 135, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte: Pb Mass 208

KPB Analyst:

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates	
#1	0.5
#2	0.55
#3	0.59
#4	0.5
#5	0.49
#6	0.48
#7	0.54

0.5214 Mean

Standard Dev. 0.039761

Student's T Value 3.143

* MDL 0.4250 PPB

RL 2.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte : Sb Mass 121

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replicates	
#1	0.54
#2	0.49
#3	0.55
#4	0.52
#5	0.44
#6	0.53
#7	0.47

Mean 0.5057

Standard Dev. 0.040356

Student's T Value 3.143

* MDL 1:01268:P.B.

RL 3:0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte: Sb Mass 123

Analyst: **KPB**

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates	
#1	0.55
#2	0.49
#3	0.55
#4	0.52
#5	0.45
#6	0.54
#7	0.47

Mean	0.5100

0.040415 Standard Dev.

Student's T Value 3.143

* MDL - 0.1270 PPB

RL 3.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte: Se Mass 78

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

Analyst:

10/1/99

Replicates						
#1	0.66					
#2	0.49					
#3	0.4					
#4	0.62					
#5	0.56					
#6	0.64					
#7	0.68					

0.5786 Mean

Standard Dev.

0.10205

Student's T Value

3.143

MDL: 03207 PPB

RL 5.0

40 CFR Part 136, Appendix B

Anal. Method: EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.40 PPB

Analyte : Se Mass 82

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replicates							
#1	. 0.37						
#2	0.32						
#3	0.36						
#4	0.32						
#5	0.28						
#6	0.31						
#7	0.36						

Mean 0.3314

Standard Dev. 0.032878

Student's T Value 3.143

* MDL = 0.1033 PPB

RL 5.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: Ti Mass 49

Analyst:

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replicates					
#1	0.23				
#2	0.22				
#3	0.25				
#4	0.21				
#5	0.2				
#6	0.19				
#7	0.2				

Mean 0.2143

0.020702 Standard Dev.

Student's T Value 3.143

MDL 3:00654 PPBH

RL 5.0

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: TI Mass 205

Analyst:

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

0.00488

Replica	tes
#1	0.21
#2	0.2
#3	0.2
#4	0.2
#5	0.2
#6	0.21
#7	0.2

0.2029 Mean

Standard Dev.

Student's T Value 3.143

* MDE7: 0.0153 RPB

1.Q RL

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 1.0 PPB

Analyte: V Mass 51

Analyst:

KPB

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

0.208121

Replicates					
#1	1.12				
#2	1.21				
#3	1.29				
#4	1.36				
#5	1.34				
#6	1.44				
#7	1.77				

1.3614 Mean

Standard Dev.

Student's T Value 3.143

* MDL . . 0 6541 PPB

1.0 RL

40 CFR Part 135, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 PPB

Analyte: Zn Mass 66

Analyst: **KPB**

Matrix: .5% HCL 1% HCO3

Study Date:

10/1/99

Replica	tes
#1	0.77
#2	1.02
#3	1.18
#4	0.85
#5	1.43
#6	0.91
#7	1.08

1.0343 Mean

Standard Dev.

0.223372

Student's T Value

3.143

5.0 RL

40 CFR Part 136, Appendix B

Anal. Method:

EPA 6020

Instrument: Perkin Elmer Elan 6000 ICP-MS

Prep Method: EPA 6020 CLP-M

Analyte Conc: 0.20 ppb

Analyte : Zn Mass 68

Analyst: KPB

Matrix: .5% HCL 1% HCO3

Study Date: 10/1/99

Replicates					
#1	0.8				
#2	1.05				
#3	1.21				
#4	0.91				
#5	1.46				
#6	0.95				
#7	1.13				

Mean 1.0729

Standard Dev. 0.219295

Student's T Value 3.143

* MDE - 0.6892 PRB-

RL 5.0

ATTACHMENT 8

INSTRUMENT MAINTENANCE AGREEMENTS

QUOTATION

Page 1 of 4

PERKIN ELMER

The Perkin-Elmer Corporation

76: Main Avenue Norwalk, CT 05859-0001 Priche (800) 762-8288. Fax: (203) 752-4300 www.perkin-eimer.com

MR GARY STANGA

ASARCO INC

TO

3422 SOUTH 700 WEST

SALT LAKE CITY UT 84119

Quotation No:

Q062701

Original Quotation Date:

10/03/1998

Quotation Validity Date:

09/24/1998 10 12/23/1998

Contract Coverage:

01/01/1999 to 12/31/1999

Contract Description:

STANDARD PROTECTION

TELEPHONE: 801 263 5251 FAX:

801 264 9838

YOUR REFERENCE:

Prior Customer P.O. No.:

NO	QTY	ITEM DESCRIPTION/ COVERAGE DATES	(SERIAL NUMBERI	COVERAGE		GROSS THE RICE/MTH		PRICE/MTH
210	1	AS60 FURNACE AUTOSAMPLER Begin Date: 01/01/1999 End Date: 12/31/1999	(6935)	P.L.T with OPM	\$	84.00	\$	71.40
20	1	AS60 FURNACE AUTOSAMPLER Begin Date: 01/01/1999 End Date: 12/31/1999	(8424)	P,L,T with OPM	\$	84,00	\$	71.40
030	1	EDLSYSTEM2 VOLTAGE MODULE Begin Date: 01/01/1999 End Date: 12/31/1999	(42078)	P.L.T with OPM	\$	38.00	\$	32.30
040	1	HGA600 FURNACE POWER SUPPLY Begin Date: 01/01/1999 End Date: 12/31/1999	(4599)	P.L.T with OPM	\$	98.00	\$	83.30
050	1	HGA600 FURNACE FOWER SUPPLY Begin Date: 01/01/1989 End Date: 12/31/1989	(5526)	P,L,T with OPM	\$	98.00	\$	83.30
060	1	ZEEMAN5100 BGC ACCY ** Begin Date: 01/01/1999 End Date: 12/31/1999	(6235A1)	P,L,T with OPM	6	103.00	\$	87.5
070	1	ZEEMAN5100 BGC ACCY ** Begin Date: 01/01/1999 End Date: 12/31/1999	(7056)	P.L.T with OPM	\$	103.00	ŧ	87.5

his quotation is subject to the terms and conditions attached.

NOTE: Customer is responsible for applicable taxes, including sales, use and/or excise tax.

PLEASE SIGN THIS MAINTENANCE AGREEMENT QUOTATION AND RETURN ORIGINAL COPY ALONG WITH YOUR PURCHASE ORDER.

QUOTATION

Page 2 of 4

PERKIN ELMER

The Perkin-Elmer Corporation

761 Main Avenue Norwaik, CT 06859-0001 Phone: (800) 762-8288, Fax: (203) 762-4300 www.perkin-eimer.com

Quotation No:

Q062701

Quotation Date:

10/03/1998

Quotation Validity Date:

09/24/1998 10 12/23/1998

TEM NO	.017	ITEM DESCRIPTION/ COVERAGE DATES	(SERIAL NUMBER)	COVER	AGE		GROSS PRICE/MTH		NET PRICE/MTH
080	1	5100PC AA INSTRUMENT ** Begin Date: 01/01/1999 End Date: 12/31/1999	(131163)	P,L,T with 0	PM	\$	272.00	\$	231.20
090	1	5100PC AA INSTRUMENT ** Begin Date: 01/01/1999 End Date: 12/31/1999	(145636)	P,L,T with 0)PM	\$	272.00	\$	231.20
100	1	703 AA INSTRUMENT Begin Date: 01/01/1999 End Date: 12/31/1999	(117194)	P.L.T with ()PM	\$	223.00	\$	189.55
120	1	OPTIMA3000DV ICP INSTRUMENT ** Begin Date: 01/01/1999 End Date: 12/31/1999	(069N7062302)	P.L.T with (DPM	6	868.00	\$	737.80
121	1	ICPSOFTWARE ICP SOFTWARE Begin Date: 01/01/1999 End Date: 12/31/1999	(069N7062302)	P,L.T with (ОРМ	\$	40.00	\$	34.00
130	1	AS91 FLAME AUTOSAMPLER** Begin Date: 01/01/1999 End Date: 12/31/1999	(3610)	P.L.T with (ОРМ	\$	84.00	\$	71.40
140	1	NESLABCHILLER COOLING SYSTEM Begin Date: 01/01/1999 End Date: 12/31/1999	(197127091)	P,L,T with	OPM	\$	44.00	3	37.40
150	1	PERISTALTICPUMP PUMP ASSEMBLY Begin Date: 01/01/1999 End Date: 12/31/1999	(7053007)	P.L.T with	ОРМ	\$	19.00	3	16.15
						Tota	al net price:	\$	2,065.50

Net Price includes a volume discount of 15.00% (on items eligible for volume discounts only) plus other discounts that may apply less applicable taxes.

NOTE: Item descriptions above marked with (**) see NOTE 1. Item descriptions above marked with (**) see NOTE 2.

Note 1: This equipment has not been tested to determine Year 2000 compliance or is not Year 2000 compliant. Problems due to Year 2000 non-compliance are excluded from this agreement. Please see the "Exclusions" section in the terms and conditions. To learn about new products that are Year 2000 compliant call us at 800-762-4000 or e-mail us at info @Perkin-Elmer.com.

QUOTATION.

Page 3 of 4

PERKIN ELMER

The Perkin-Elmer Corporation '61 Main Avenue Norwaik, CT 06859-0001 Prione: (800) 762-8288, Fax: (203) 752-4300 www.perkin-elmer.com

Quotation No:

Q062701

Quotation Date:

10/03/1998

Quotation Validity Date:

09/24/1998 to 12/23/1998

ITEM	OTV	ITEM DESCRIPTION/	ISERIAL	COVERAGE GROSS	NET I
	ω.,				*
I NO I		COVERAGE DATES	NUMBER)	PRICE/MTH	. PRICE/MTH '

Note 2: Please determine Year 2000 compliance using the attached documentation. If the equipment is not Year 2000 compliant, problems due to Year 2000 non-compliance are excluded from this agreement. Please see the "Exclusione" section in the terms and conditions and contact us at 800-762-4000 or e-mail us at info @Perkin-Elmer.com if you require more information. (www.perkin-elmer.com) and refer to the Terms and Conditions.

Zone: Zone 1

Region:

INORG RM REGION

Location:

USUTO1

Contract Notas:

1/99 MERGING OPTIMA3000DV/PUMP/CHILLER/AS91 FOR 12 MONTHS. 12/21 DELETED (1) 703 PER BRIAN.

QUOTATION

Page 4 of 4

PERKIN ELMER

The Perkin-Elmer Corporation

761 Main Avenue Norwalk, CT 06859-0001 Phone: (500) 762-6288, Fax: (203) 762-4300 www.perkin-elmer.com

Quotation No:

0062701

Quotation Date:

10/03/1998

Quotation Validity Date:

09/24/1998 to 12/23/1998

OTY ITEM DESCRI			GROSS PRICE/MTH	PRICE/MT
Monthly Billing Plan		Bill to/Payer:	<u> </u>	
Planned Invoice date	Monthly Amount			
01/01/1999 02/01/1999 03/01/1999	\$ 2.065.50 \$ 2.065.50 \$ 2.065.50	ASARCO INC 3422 SOUTH 700 S SALT LAKE CITY U		
04/01/1999 05/01/1999 06/01/1999	\$ 2,065.50 \$ 2,065.50 \$ 2,065.50	SALT DAKE CITT O	. 04113	
07/01/1999 08/01/1999 09/01/1999	\$ 2,065.50 \$ 2,065.60 \$ 2,065.60			
10/01/1999 11/01/19 99 12/01/1999	\$ 2,065.50 \$ 2,065.50 \$ 2,065.50			
Monthly Billing Plan Total:	\$24,786.00	Quoted By:	Vlcki F. Cook	
Pre-Payment Discount:	5.000 %	Telephone No:	203-762-6169	
Pre-Payment Plan Total:	\$23,546.40		1	
Note: Taxes will be applied to not have a copy of your Accepted By:	lied to your invoice. ur tax exemption certificate (ule: (Please select o	na only)
Signature of authorized pegood	n	Monthly Billing Pl	an	t 1
Please grint name and title	Manager 1/11/	99 Pre-Payment Plan		×

Legend

1PM Only (no P.L.T) = One Preventive Maintenance Visit only = Two Preventive Maintenance Visits Only = Two Preventive Maintenance Visits Only = Pow Only (no P.L.T) = Fixer Preventive Maintenance Visits Only = Fixer Preventive Maintenance Visits Only = Six Preventive Maintenance Visits Only

T with OPM = Emergency service Incl. normal Parts, Labor & Travel w/ No PM's - Emergency service Incl. normal Parts, Labor & Travel w/ One PM - Emergency service Incl. normal Parts, Labor & Travel w/ Two PM's - Emergency service Incl. normal Parts, Labor & Travel w/ Twee PM's - T with 6PM = Emergency service Incl. normal Parts, Labor & Travel w/ Five PM's - Emergency service Incl. normal Parts, Labor & Travel w/ Five PM's - Emergency service Incl. normal Parts, Labor & Travel w/ Six PM's

PERKIN ELMER

MR GARY STENGA

BIOTRACE LABS

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QUOTATION

Page 1 of 4

The Perkin-Elmer Corporation

761 Main Avenue Norwalk, CT 06859-0001 Prione: (800) 762-8268, Fax. (203) 762-4300 www.perkin-elmer.com

FROM-

Quotation No:

Q063363

Original Quotation Data:

10/04/1998

Quotation Validity Date:

11/30/1998 to 02/28/1999

Contract Coverage:

01/01/1999 10 12/31/1999

Contract Description:

STD PROT-ELAN 1 PM

3440 SOUTH 700 WEST SALT LAKE CITY UT 84119

TELEPHONE: 801 263 5251 FAX: 801 264 9838

YOUR REFERENCE:

Prior Customer P.O. No.:

This Agreement is entered into between PERKIN ELMER and the undersigned Customer in consideration of the payments provided for in this Agreement. Subject to the terms and conditions of this Agreement, PERKIN ELMER agrees to perform the services sat forth in the coverage of this Agreement on the agreement lested below for the period described.

TEM	QTY	ITEM DESCRIPTION/ COVERAGE DATES	(SERIAL NUMBER)	COVERAGE		GROSS PRICE/MTH	15.14	NET PRICE/MTH
020		AS60 FURNACE AUTOSAMPLER Begin Date: 01/01/1999 End Date: 12/31/1999	(8428)	P,L,T with OPM	9	84.00	\$	71.40
030	1	AS60 FURNACE AUTOSAMPLER Begin Date: 01/01/1999 End Date: 12/31/1999	(8630)	P,L,T with OPM	\$	84.00	\$	71.40
050	1	AS91 FLAME AUTOSAMPLER** Begin Date: 01/01/1999 End Date: 12/31/1999	(3263)	P,L,T with 1PM	\$	84.00	\$	71.40
070	1	EDLSYSTEM2 VOLTAGE MODULE Begin Date: 01/01/1999 End Date: 12/31/1999	{420295}	P,L,T with OPM	\$	38.00	\$	32.30
080	1	ELAN8000 MAS INSTRUMENT ** Begin Date: 01/01/1999 End Date: 12/31/1999	(91950860)	P,L,T with 1PM	\$	1,544.00	\$	1,312.40
110	1	HGASOO FURNACE POWER SUPPLY Begin Date: 01/01/1999 End Date: 12/31/1999	(5602)	P,L.T with OPM	\$	98.00	3	83.30
120	1	HGA600 FURNACE POWER SUPPLY Begin Date: 01/01/1999 End Date: 12/31/1989	(5622)	P.L.T with OPM	6	98.00	\$	83.30

PERKIN ELMER

The Perkin-Elmor Corporation

761 Main Avenue Norwalk, CT 06859-0001 Phone: (800) 752-8288, Fax: (203) 752-4300 www.perkin-eimer.com

FROM-

QUOTATION

Page 2 of 4

Quotation No:

Q063363

Quotation Date:

10/04/1998

Quotation Validity Date:

11/30/1998 10 02/28/1999

NO.	OTY	ITEM DESCRIPTION/ COVERAGE DATES	(SERIAL NUMBER)	COVERAGE		GROSS RICE/MTH		NET PRICE/MTH
30	1	NESLABCHILLER COOLING SYSTEM Begin Date: 01/01/1999 End Date: 12/31/1999	(695145140)	P.L.T with OPM	\$	44.00	\$	37.40
150	1	ZEEMAN5100 BGC ACCY ** Bagin Date: 01/01/1999 End Date: 12/31/1999	(706 7)	P,L,T with OPM	\$	103.00	\$	87.55
160	1	ZEEMAN5100 BGC ACCY ** Begin Date: 01/01/1999 End Date: 12/31/1999	(7068)	P,L,T with OPM	\$	103.00	3	87.55
170	1	3110 AA INSTRUMENT ** Begin Date: 01/01/1999 End Date: 12/31/1999	(311N3042304)	P.L.T with OPM	\$	140.00	\$	00.err
180	1	5000 AA INSTRUMENT Begin Date: 01/01/1999 End Date: 12/31/1999	(119979)	P,L,T with OPM	\$	333.00	\$	283.05
190	1	5100 AA INSTRUMENT ** Begin Date: 01/01/1999 End Date: 12/31/1999	(145322)	P,L,T with OPM	\$	272.00	3	231.20
200	1	5100PC AA INSTRUMENT ** Begin Date: 01/01/1999 End Date: 12/31/1999	(149621)	P,L,T with OPM	\$	272.00	9	231.20
					Total	net price:	6	2,802.45

Net Price includes a volume discount of 15.00% (on items eligible for volume discounts only) plus other discounts that may apply less applicable taxes.

NOTE: Item descriptions above marked with (*) see NOTE 1. Item descriptions above marked with (**) see NOTE 2.

Note 1: This equipment has not been tested to determine Year 2000 compliance or is not Year 2000 compliant. Problems due to Year 2000 non-compliance are excluded from this agreement. Please see the "Exclusions" section in the terms and conditions. To learn about new products that are Year 2000 compliant call us at 800-762-4000 or e-mail us at into @Perkin-Elmer.com.

Note 2: Please determine Year 2000 compliance using the attached documentation. If the equipment is not Year 2000 compliant, problems due to Year 2000 non-compliance are excluded from this agreement. Please see the "Exclusions" section in the terms and conditions and contact us at 800-762-4000 or e-mail us at into @Perkin-Elmer.com if you require more information. (www.perkin-elmer.com) and refer to the Terms and Conditions.

QUOTATION

Page 3 of 4

PERKIN ELMER

The Perkin-Elmer Corporation

761 Main Avenue Norwaik, CT 06859-0001 Phone: (800) 762-8288, Fax: (203) 762-4300 www.perkin-elmer.com

Quotation No:

Q063363

Quotation Date:

10/04/1998

Quotation Validity Date:

11/30/1998 to 02/28/1999

TEM	ΩTY	ITEM DESCRIPTION/ COVERAGE DATES		. (SERIAL : NUMBER)	COVERAGE	GROSS PRICE/MTH	NET PRICE/MTH
		Zone: Zone 1 Region: INORG RM REG	ON				· · · · · · · · · · · · · · · · · · ·

Contract Notes:

12/21 DELETED 6 ITEMS PER BRIAN: AS40,AS90,DEC,FIAS200,HGA500,ZEEMAN5000

QUOTATION

Page 4 of 4

PERKIN ELMER

The Perkin-Elmer Corporation

761 Main Avenue Norwalk, CT 06859-0001

Phone: (800) 762-8288, Fax: (203) 762-4300

www.parkin-elmar.com

Quotation No:

0063363

Quotation Date:

10/04/1998

Quotation Validity Date:

11/30/1998 10 02/28/1999

Monthly Billing Plan		Bill to/Payer:			
Planned Invoice date	Monthly Amount				
01/01/1999	\$ 2,802.45	BIOTRACE LABS			
02/01/1999	\$ 2,802.45	3440 SOUTH 700 WEST			
03/01/1999	\$ 2,802.45	SALT LAKE CITY UT 84119			
04/01/1999	\$ 2,802.45				
05/01/1999	s 2,802.46				
06/01/1999	\$ 2,802.45				
07/01/1999	\$ 2,802.45				
08/01/1999	\$ 2,802.45				
09/01/1999	\$ 2,802.45				
10/01/1999	3 2,802.45				
11/01/1999	\$ 2,802.45				
12/01/1999	\$ 2,802.45				
Monthly Billing Plan Tota	ai: \$33,629.40	Quoted By: Vicki F. Cook			
Pre-Payment Discount:	5.000 %	Telephone No: 203-762-6169	•		
Pre-Payment Plan Total:	631,947.72				
We do not have a copy Accepted By:	e applied to your invoice. of your tax exemption certificate	Payment Schedule: (Please select one only)			
Signature of authorized	pefsoy	Monthly Billing Plan	[]		
Gary Sta	title) Date	14/99 Pre-Payment Plan	×		

Legend



Service Contract Quotation

Quotation Contract RENEWAL FULL SERVICE For Type

Quote Valid Until:

Dates of Coverage for Service Contract

Period Covered By Agreement 1/01/99 to 12/31/99

RENEWAL OFFER EXPIRES: 4/01/99

Ship To Address

Bill To Address

ASARCO INC. ASARCO INC. TECHNICAL SERVICE CENTER TECHNICAL SERVICE CENTER

3422 SOUTH 700 WEST 3422 SOUTH 700 WEST SALT LAKE CITY UT 84119 SALT LAKE CITY UT 84119

8012622459 8012622459 GARY STANGA

Ouote Reference Number
Agreement #: 90000219 1 OF 1 PAGE Total Amount 8411.00 Coverage Includes all items listed below EQUIPMENT / SERIAL NUMBER INSTALLED / MODEL NUMBER **AMOUNT**

6/04/91 6000 SYSTEM 6710.00 Coverage Excludes: 26995-001 6000/50

BD., DATA MEMORY (AT CLONE) 0.00 AT CLONE Computer Systems

PUMP, VACUUM ONLY 0.00 X-ray tubes

50KV .35MA RH XRAY (4K,5K,6K 1701.00 Except where noted OR, ECD AL 0.00

Terms are NET 30

One PM Visit is included with each FULL SERVICE Agreement. PM visits must be scheduled prior to contract expiration

Quotation totals:

Acceptance of Agreement:

Title

Date

SUBTOTAL 8411.00 TAX 0.00 TOTAL 8411.00

PO#:

ACKLNOWLEGEMENT OF RECEIPT OF PO# FOR CONTRACT RENEWALL

Notes:



FULL SERVICE CONTRACT AGREEMENT

I. GENERAL INFORMATION

A Spectrace Instruments service agreement satisfies an ever increasing need for prompt efficient service after the standard warranty for the system has expired.

A FULL SERVICE AGREEMENT covers the following items:

- A. All parts costs;
- B. All freight costs;
- C. All labor costs for travel time and on-site time;
- D. All incurred costs such as round trip air fare, rental car, lodging, meals, parking, baggage handling, personal car use, etc.;
- E. Preventive Maintenance.

A service agreement allows the customer to accurately determine future repair costs for budgetary purposes, and know that service problems will be handled on a priority basis.

II. EQUIPMENT COVERED

The equipment covered under this Agreement is limited to the system components itemized on the service contract proposal.

III. REPAIR METHODS

If it is determined, after discussion of a problem with a Spectrace Instruments Service Engineer, that the problem is most likely confined to a particular board or module, the Service Engineer may send a replacement/loaner board or module to the customer to install in the system to correct the problem. If this initial attempt does not cure the problem, on-site service will be promptly accomplished. Equipment covered under this Agreement will normally be repaired at the customers site within two (2) working days after determining an on-site visit is required by Spectrace Instruments.

In certain situations it may be necessary to return system components to the Spectrace Instruments factory for repair if on-site service is not practical. In these isolated instances, all shipping costs will be paid by Spectrace Instruments. The customer must contact Spectrace Instruments Service Department for a Return Authorization number and method of shipment when the system must be returned. Spectrace Instruments will make a reasonable attempt to provide loaner equipment in these instances.

Page 1 of 4 April 1993

IV. RESPONSIBILITY AND EXCLUSIONS

SPECTRACE INSTRUMENTS EXCLUDES OR RESERVES THE RIGHT TO DETERMINE THE FOLLOWING AREAS OF RESPONSIBILITY.

This service Agreement does not cover, and Spectrace Instruments is not responsible for the following:

- A. Repair of instrument damage, replacement of parts or increase in service time caused by-
 - 1. failure to continually maintain a suitable environment as prescribed by Spectrace;
 - 2. accident, disaster, transportation, vandalism, neglect, misuse or abuse;
 - 3. another product or device not under Spectrace warranty or a Spectrace agreement;
 - 4. service of the instrument by other than Spectrace;
 - 5. a non-Spectrace modification;
 - 6. power line failures, fluctuations and/or transients:
 - 7. failure to maintain the SiLi detector at operating temperature, either by loss of liquid nitrogen, or loss of A/C for ECD systems;
 - 8. failure to maintain power or restore power to the ion pump in an ECD system;
 - 9. failure to operate the instrument as prescribed by operating and technical documentation:
- B. Any costs, service or repairs required due to damage or breakage of the detector window;
- C. Any X-ray tube costs, service, or repairs due to either damage to its window or decreased emission resulting from normal use;
- D. Service and parts for cathode ray tube repair when due to loss of display intensity, age, or tube breakage;
- E. Service or repairs required because of rearrangement or relocation. Spectrace Instruments service personnel will provide telephone assistance at no charge when help is needed concerning cabling of the system, special requirements of problems, etc.;
- F. Delays caused by suppliers in providing materials or service, strikes, delays in transportation, interruption in business by either party or other causes beyond the control of Spectrace Instruments;
- G. Any costs incurred by Spectrace Instruments to determine liability;
- H. Losses incurred by the customer due to instrument downtime;
- I. Consumable items paper for printers and plotters, ink ribbons, cassettes, diskettes, window material, etc.;
- J. Painting or refinishing instruments or furnishing material thereof;
- K. Electrical work done external to the instrument;
- L. Installation, maintenance, or removal of alterations or attachments to an instrument or any service which is impractical for Spectrace to render because of such alterations or attachments:
- M. Loss of files, data, or programs contained in storage media covered by this Agreement.

Service, repair, parts and freight charges associated with Section IV items A-M will be Billable at Spectrace Instruments service rates in effect at that time.

V. CUSTOMER ASSISTANCE

A reasonable effort on the part of the customer is expected when problems are encountered. This includes over the phone discussion of the problem, assistance in performing diagnostics' programs, replacement of boards/modules that have been shipped to the customer, simple voltage measurements of the DC power supplies, etc. all of which will be done under the guidance of a qualified service engineer. The customer has the right to refuse to help at the time but runs the risk of delaying repair time beyond normal and reasonable time.

Highly technical support or major assistance will not be expected or requested of the customer.

VI. SERVICE HOURS

Contact with our Service Department will be available between the hours of 8:00 AM and 5:00 PM, Monday through Friday, excluding Spectrace Instruments holidays.

VII. PREVENTIVE MAINTENANCE/EMERGENCY SERVICE

Spectrace will provide at the customers site one (1) scheduled Preventive Maintenance (PM) visit during the coverage period, and an unlimited number of emergency visits, excluding visits as described in section IV. The visits will include all labor, parts and materials Spectrace deems necessary to maintain the equipment in good operating condition (except those parts specifically excluded from this contract).

VIII. RENEWAL

This Agreement may be renewed for additional successive yearly periods by mutual consent at the rates currently in effect at time of renewal.

IX. PAYMENT TERMS

Standard terms are Net 30 days after date of invoice.

X. MULTIPLE SYSTEM DISCOUNT

Multiple system discounts are available when all systems are at the same location and covered under the same service agreement by the same purchase order number.

XI. MISCELLANEOUS

The customer signed acceptance and purchase order will constitute an offer in accordance with the terms hereof and such offer, upon endorsement of our acceptance by way of return invoice will constitute the contractual agreement.

Unless otherwise stated in writing, Spectrace Instruments' quoted prices do not include sales, use, excise or similar taxes. Consequently, the amount of any present or future tax shall be paid by the customer, or in lieu thereof, customer shall provide Spectrace Instruments with a tax exemption certificate.

Page 3 of 4 April 1993 Spectrace does not assure uninterrupted or error-free operation of the instrument/s. Spectrace is not responsible for failure to fulfill its obligations under this Agreement due to causes beyond its control.

XII. TERMS AND CONDITIONS

Any terms and/or conditions of the customer's order that are inconsistent with the terms and/or conditions of our Agreement shall not be binding on Spectrace Instruments and shall not be considered applicable to any sales made pursuant to this quotation. No waiver, alteration or modification of any of the provisions of this Agreement shall be binding unless in writing and signed by the Spectrace Instruments Service Manager.

You and Spectrace agree that the complete and exclusive statement of the agreement relating to the subject shall consist of the Agreement and its applicable Amendments and Supplements, including those effective in the future. This statement of the agreements supersedes all proposals or other agreements, oral or written, and all other communications between the parties relating to this subject. This Agreement may not be reassigned without the consent of Spectrace Instruments.

Neither party may bring an action, regardless of form, arising out of this Agreement more than 1 year after the cause of action has arisen. Spectrace may not bring an action for nonpayment more than two years after the date the last payment was due.

XIII. CUSTOMER RESPONSIBILITIES

Customer agrees to provide a suitable environment for the instrument as specified by Spectrace. Customer agrees to operate the instrument in accordance with Spectrace operating procedures and safety precautions. Customer will provide Spectrace full, free, and safe access to the instrument in the event a service call is necessary. Customer agrees to inform Spectrace of changes in location of the instrument prior to movement of the instrument.

Customer agrees to backup, remove, protect, and restore, as applicable, programs, files, data, and removable storage media contained in failing computers covered under this Agreement.

Customer agrees to remove all features, parts, options, alterations and attachments that are not subject to this Agreement before presenting a failed component for exchange or repair.

Parts sent to the customer for exchange are to be returned to Spectrace Instruments within 10 working days after the repair of the instrument. Spectrace Instruments will pay for the return shipping charges. Parts are to be returned via Federal Express Economy unless otherwise directed by a Spectrace Instruments representative. Failure to return the parts within the specified period may be cause for cancellation of this Agreement. The replacement cost of any part lost due to the customers negligence will be paid by the customer.

XIV. CANCELLATION

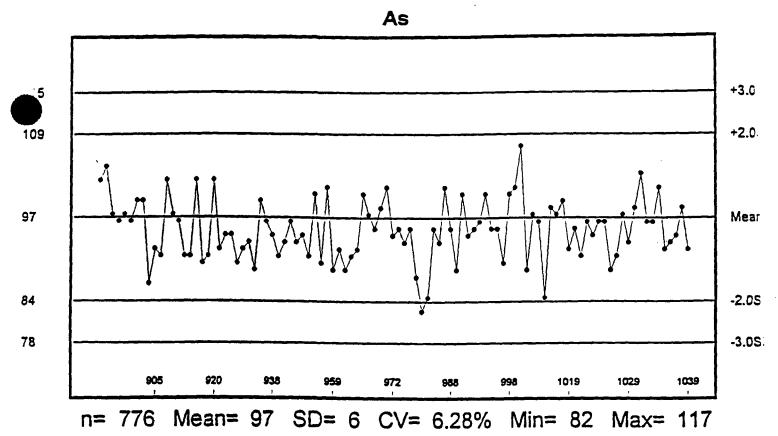
At any time without reason either Spectrace Instruments or the customer may cancel this Agreement. Written notification of cancellation is required. Spectrace Instruments will refund the unused prorated portion of the Agreement rounded to the end of the month cancellation takes place. This Agreement will be null and void at the end of the month cancellation takes place. Allow sixty (60) days for refund. Any refunded portion may be withheld to pay outstanding debt to Spectrace.

Page 4 of 4 April 1993

ATTACHMENT 9

QUALITY CONTROL CHARTS

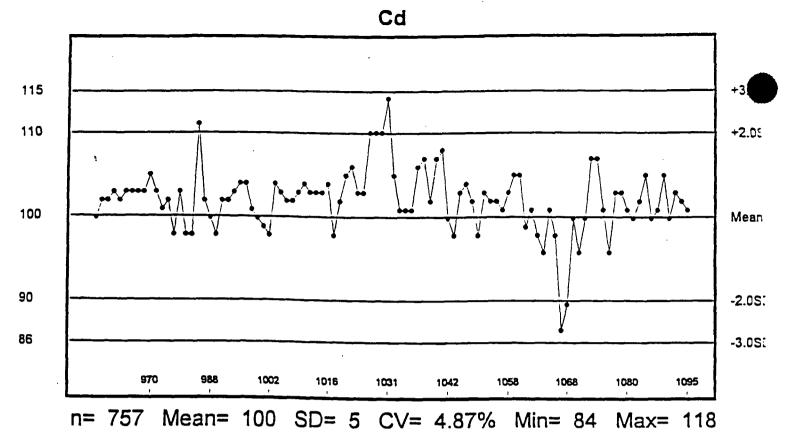
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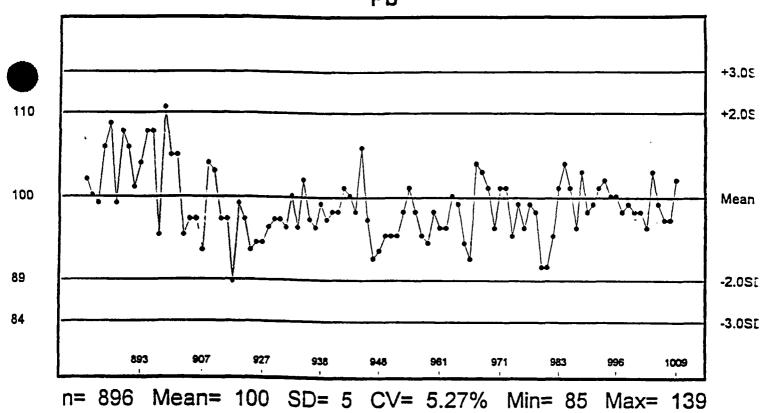
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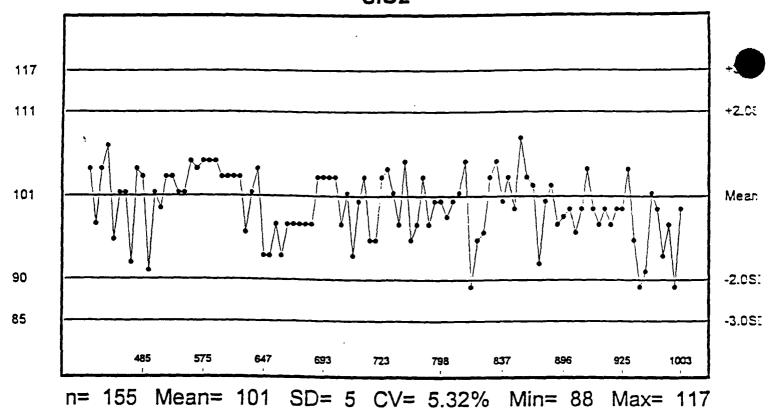
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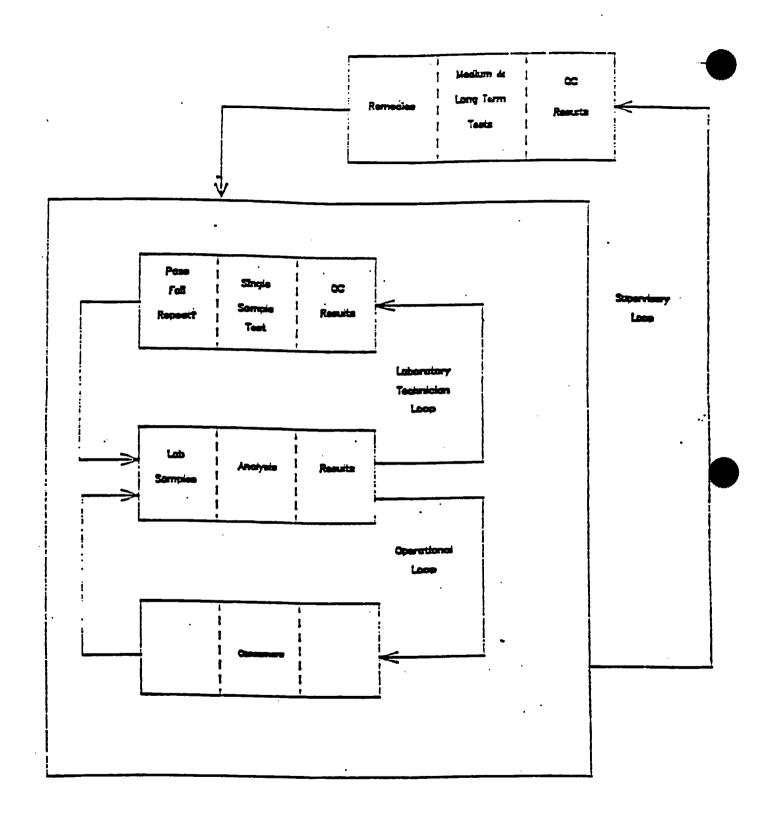
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ATTACHMENT 10

CONTROL LOOP

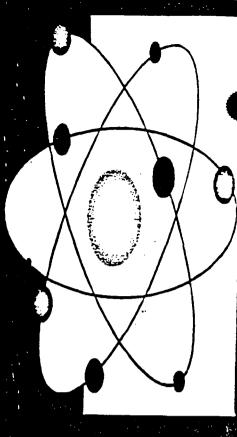


ATTACHMENT 11

STANDARDS LOG

MA 3000DV PERMINIMENT

ANALYSIS LOGBOOK 1



Optima 3000 DV Standards Log #26

Calibration Standard #1

Environmental Resource Associates ICP Calibration Standard #3

1.0 ppm Ba. Bc. Cd. Cr. Co, Cu, Mn, Mo, Ni, Pb, Sb, V and Zn

10 ppm Al. Fc, Mg and Se In 0.5%, v/v, HCl and 0.5%, v/v, HNO,

In use from December 3, 1997 to present.

Calibration Standard #2

50 ppm Ca

Mallinckrodt CaCO₃ (99.85%) lot #CXN. From 1000 ppm stock prepared April, 1995 by GJS. In 10%, v/v, HCl

In use from January 16, 1998 to present

Calibration Standard #3

1 ppm As⁺⁵

NBS 83c As₂O₃. From 1000 ppm stock prepared September, 1995 by GJS. In 10%, v/v, HCl

In use from February 27, 1998 to present

Calibration Standard #4

5 ppm Ag

J.T. Baker AgNO₃, lot #45088. From 1000 ppm stock prepared December, 1996 by GJS.

5 ppm T1

TiNO₃, lot #1-80. From 1000 ppm stock prepared April, 1995 by GJS.

In 10%, v/v, HCl

In use from October 16, 1997 to present

Calibration Standard #5

50 ppm K

J.T. Baker KCI, lot #58.0.0. From 1000 ppm stock prepared March, 1996 by GJS.

50 ppm Na

J.T. Baker NaCl, lot #57.0.0. From 1000 ppm stock prepared December, 1995 by GJS.

In 10%, v/v, HCl

In use from February 19, 1998 to present

Calibration Standard #6

5 ppm Au

ASARCO Central Research Au metal (99.99+%). From 1000 ppm stock prepared June, 1996 by GJS.

In 10%, v/v, HC1

In use from February 11, 1998 to present

ATTACHMENT 12

CERTIFICATIONS



DIVISION OF FPIDEMIOLOGY

AND LABORATORY SERVICES

State of Utah

Michael O. Leavitt Covernor

Rod L. Betit Executive Director

Charles D. Brokopp, Dr. P.H.

Director

January 4, 2001

Bureau of Laboratory Improvement

46 North Medical Drive Salt Lake City, Utali 84113-1105 Telephone: (801) 584-8169 Fax: (801) 584-8501

http://hlunix.hl.slate.ut.us/cls/labimp

ASARCO/AEC Laboratory Gary Stanga Director 3422 South 700 West Salt Lake City UT 84119

ID # ASAR Account # 2622459

Director,

On the basis of your most recent audit results and compliance with the ELCP requirements, the laboratory listed is certified for environmental monitoring under the Clean Water Act and authorized to perform the following analytes, or groups of analytes by method:

Inorganics and Metals	Potassium
5520 B	Selenium
Oil and Grease by Partition-Gravimetric	Silver
Method	Sodium
HACH Method	Thallium
COD	Vanadium
Method 120.1	Zinc
Conductance (Specific Conductance,	<u>Method 200.8</u>
umhos at 25-C)	Antimony
Method 150.1	Arsenic
pH (Electrometric)	Barium
Method 160.1	Beryllium
Residue, Filterable (Gravimetric)	Cadmium
Method 160.2	Chromium
Residue, Non-Filterable (Gravimetric,	Cobalt
TSS)	Lead
Method 180.1	Manganese
Turbidity (Nephelometric)	Molybdenum
Method 200.7	Nickel
Antimony	Selenium
Arsenic	Silver
Barium	Thallium
Beryllium	Vanadium
Cadmium	Zinc
Calcium	Method 204.2
Chromium	Antimony (Atomic Absorption, Furnace
Cobalt	Technique)
Iron	<u>Method 206.2</u>
Lead	Arsenic (Atomic Absorption, Furnace
Magnesium	Technique)
Manganese	Method 206.3
Molybdenum	Arsenic (Atomic Absorption, Gaseous-
Nickel	Hydride)

The expiration for the laboratory's certification is 03/31/2001. The Utah Environmental Laboratory Certification Program (ELCP) encourages clients and data user to verify the most current certification letter for the authorized method, Please call 801-584-8469.

ASARCO/AEC Laboratory Clean Water Act Page 2

Method 213.1

Cadmium (Atomic Absorption, Direct Aspiration)

Method 213.2

Cadmium (Atomic Absorption, Furnace

Technique)

Method 2258.1

Potassium (Atomic Absorption, Direct

Aspiration)

Method 2270.2

Selenium (Atomic Absorption, Furnace

Technique)

Method 236.1

Iron (Atomic Absorption, Direct

Aspiration)

Method 239.1

Lead (Atomic Absorption, Direct

Aspiration)

Method 239.2

Lead (Atomic Absorption, Furnace

Technique)

<u>Method 245.1</u>

Mercury (Manual Cold Vapor Technique)

Method 272.1

Silver (Atomic Absorption, Direct

Aspiration)

Method 272.2

Silver (Atomic Absorption, Furnace

Technique)

Method 273.1

Sodium (Atomic Absorption, Direct

Aspiration)

Method_289.1

Zinc (Atomic Absorption, Direct

Aspiration)

Method 310.1

Alkalinity (Titrimetric, pH 4.5)

Method 325.1

Chloride (Colorimetric, Automated

Ferricyanide AAI)

Method 335.2

Cyanide, Total (Titrimetric;

Spectrophotometric)

Method 340.2

Fluoride (Potentiometric, Ion Selective

Electrode)

Method 350.1

Nitrogen, Ammonia (Colorimetric,

Automated Phenate)

Method 351.1

Nitrogen, Kjeldahl, Total (Colorimetric,

Automated Phenate)

Method 353.1

Nitrogen, Nitrate-Nitrite (Colorimetric,

Automated, Hydrazine Reduction)

Method 365.1

Phosphorous, All Forms (Colorimetric,

Automated, Ascorbic Acid)

Method 365.4

Phosphorous, Total (Colorimetric,

Automated, Block Digester AA II)

Method 415.1

Organic Carbon, Total (Combustion Or

Oxidation)

Method 420.1

Phenolics (Spectrophotometric, Manual

4-AAP With Distillation)

This laboratory's certification date is effective: 10/01/2000.

The analytes or groups of analytes by method which a laboratory is authorized to perform at any given time will be those indicated in the most recent certificate letter. The most recent certification letter supersedes all previous certification or authorization letters. Any discrepancies must be documented and notice received by this Bureau within 15 days of receipt. The certification will be recalled in the event your laboratory's certification is revoked.

Respectfully,

Charles Brokopp, Dr. P.H.

The expiration for the laboratory's certification is 03/31/2001. The Utah Environmental Laboratory Certification Program (ELCP) encourages clients and data user to verify the most current certification letter for the authorized method. Please call 801-584-8469.



DIVISION OF EPIDEMIOLOGY

AND LABORATORY SERVICES.

State of Utah

Michael O. Leavitt Gavernor

Rod L. Betit

Executive Director

Charles D. Brokopp, Dr. P.H.

Director

January 4, 2001

Eureau of Laboratory Improvement

46 North Medical Drive Salt Lake City, Utah 84113-1105

Telephone: (801) 584-8469 Fax: (801) 554-5301

http://lilunix.hl.state.ut.ns/els/labinip

ID # ASAR Account # 2622459

ASARCO/AEC Laboratory Gary Stanga Director 3422 South 700 West Salt Lake City UT 84119

Director,

On the basis of your most recent audit results and compliance with the ELCP requirements, the laboratory listed is certified for environmental monitoring under the Resource Conservation and Recovery Act and authorized to perform the following analytes, or groups of analytes by method:

Cobalt Characteristics Lead 1311 Magnesium Toxicity Characteristic Leaching Procedure Metals Manganese Inorganic Molybdenum Nickel 9036 Potassium Sulfides (Colorimetric, Automated, Methylthymol Blue, AA II) Selenium Metal Digestion Silver 3005 Sodium Acid Digestion Total Recoverable or Thallium **Dissolved Metals** Vanadium 3010 Zinc Acid Digestion for Total Metals 6020 3020 Antimony Acid Digestion for Total Metals Arsenic 3050 Beryllium Acid Digestion of Sediments, Sludges Cadmium and Soils Chromium Metals Cobalt 6010 Lead Aluminum Manganese **Antimony** Nickel Arsenic Silver Barium Thallium Beryllium Zinc Cadmium 7470 Calcium Mercury Chromium

The expiration for the laboratory's certification is 03/31/2001. The Utah Environmental Laboratory Certification Program (ELCP) encourages clients and data user to verify the most current certification letter for the authorized melhod. Please call 801-584-8469.

ASARCO/AEC Laboratory
Resource Conservation and Recovery Act
Page 2

This laboratory's certification date is effective: 10/01/2000.

The analytes or groups of analytes by method which a laboratory is authorized to perform at any given time will be those indicated in the most recent certificate letter. The most recent certification letter supersedes all previous certification or authorization letters. Any discrepancies must be documented and notice received by this Bureau within 15 days of receipt. The certification will be recalled in the event your laboratory's certification is revoked.

Respectfully,

Charles Brokopp, Dr. P.H.

Arizona Department of Health Services

ENVIRONMENTAL LABORATORY LICENSE

Issued to:

Laboratory Director: Owner/Representative:

GARY STANGA

GARY STANGA, REPRESENTATIVE

ASARCO/AEC LABORATORY AZ0599

is in compliance with Environmental Laboratory's applicable standards for the State of Arizona and maintains on file a List of Parameters for which the laboratory is certified to perform analysis

PERIOD OF LICENSURE FROM: 01/20/2000 TO 01/20/2001



Wynand H. Nimmo, M.T., Chief Office of Laboratory Licensure,

Certification & Training



Office of Laboratory Licensure, Cerafication & Training

3443 N. Central Avenue, Suite 810

Phoenix, Arizona 85012-2208

JAMES L. SCHAMADAN, M.D., ACTING DIRECTOR
(602) 255-3454
(602) 255-3462 FAX

WATTS 1-800-952-0374

April 7, 2000

Gary Stanga ASARCO/AEC Laboratory 3422 South 700 West Salt Lake City, ut 84119

Dear Mr. Stanga:

This is to confirm that ASARCO/AEC LABORATORY has fulfilled all requirements for Arizona Environmental Laboratory Licensure under the Arizona Revised Statute §§ 36.495 et.sec. and rules.

Your Arizona Environmental Laboratory License number is AZ0599, which is the number you will need to use when reporting compliance results to ADEQ or the USEPA.

If there are any questions, please do not hesitate to call me at the letterhead telephone number.

Sincerely,

Steven D. Baker, Manager

Environmental Laboratory Licensure

SDB:rm

Leadership for a Healthy Arizona

June Luburatury

Environmental Laboratory Licensure Fees as of

Tuesday, December 28, 1999

Page #:

1

AZUcense AZ0599 A LabName

ASARCO/AEC LABORATORY

Entry Date 11/6/1998 | Gent Date

Division,

MailStreet::::3422 SOUTH 700 WEST

MailCity SALT LAKE CITY MainPhone: (801) 263-5251 EXT

MailST ÜT MailZip 84119 FAX (801) 264-9838

Owner/Rep GARY STANGA, REPRESENTATIVE

LabDirector : GARY STANGA Title: LABORATORY DIRECTOR

LICENSURE CATEGORIES	TOTAL LICENSED A	FEE
HAZARDOUS CHARACTERISTIC TESTING	2	\$362.00
INORGANIC CHEMISTRY - METALS	40	\$369.00
INORGANIC CHEMISTRY - NONMETALS	14	\$464.00
INSTRUMENTS	·3	\$181.00
MISCELLANEOUS COMPLIANCE TESTING	11	\$316.00
Non Billables	3	\$0.00

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Programs WAIR Air Wild Hazardous Waste SDW Safe Drinking Water WW Wastewater

State Laboratory

Environmental Laboratory Licensure Invoice Detail

Page 1-1

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Mail@iya SALT LAKE CITY		
MailZip 84119		
Owner	Rep GARY STANGA, REF	PRESENTATIVE
MainPhone (1(801) 263-5251 (37 / EXT.)	ector GARY STANGA	
EAX (801) 264-9838	LABORATORY DIRE	CTOR :
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CADMIUM	EPA 6010B EPA 6010B	11/6/1998 11/6/1998
CHROMIUM, TOTAL	EPA 6010B	11/6/1998
COPPER	EPA 6010B	11/6/1998
CORROSIVITY PH DETERMINATION	EPA 9040A	11/6/1998
EAD	EPA 6010B	11/6/1998
MERCURY	EPA 7470A	11/6/1998
SEDIMENTS, SLUDGES AND SOILS	EPA 3050B	11/6/1998
SELENIUM	EPA 6010B	11/6/1998
SILVER	EPA 6010B	11/6/1998
SULFATE	EPA 9036	11/6/1998
SYNTHETIC PRECIPITATION LEACHING PROCEDURE (SPLP)	EPA 1312	11/22/1999
TOTAL METALS	EPA 3010A	11/6/1998
TOTAL RECOVERABLE IN WATER	EPA 3005A	11/6/1998
TOXICITY CHARACTERISTICS LEACHING PROCEDURE	EPA 1311	11/6/1998
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ANTIMONY	EPA 200.7	11/6/1998
ANTIMONY	EPA 200.8	11/6/1998
ARSENIC	EPA 200.7	11/6/1998

Arizona Department of Health Services **State Laboratory**

Environmental Laboratory Licensure Invoice Detail Page 1-2

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PROGRAM: WW		The first state of the state of		
PARAMET	ÉR:		⊯METHOD	DATE
ARSENIC			EPA 200.8	11/6/1998
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BERYLLIUM			EPA 200.8	11/6/1998
CADMIUM			EPA 200.7	11/6/1998
CADMIUM			EPA 200.8	11/6/1998
CALCIUM			EPA 200.7	11/6/1998
CHLORIDE			EPA 325.1	11/6/1998
CHROMIUM TOTAL			EPA 200.7	11/6/1998
CHROMIUM TOTAL			EPA 200.8	11/6/1998
COBALT			EPA 200.7	11/6/1998
COBALT			EPA 200.8	11/6/1998
COPPER			EPA 200.7	11/6/1998
COPPER			EPA 200.8	11/6/1998
CYANIDE, TOTAL			EPA 335.2	11/6/1998
FLUORIDE	•		EPA 340.2	11/6/1998
HARDNESS			EPA 200.7,CA&M	11/6/1998
HYDROGEN ION (pH)			EPA 150.1	11/6/1998
IRON			EPA 200.7	11/6/1998
KJELDAHL NITROGEN			EPA 351.1	11/6/1998
LEAD			EPA 200.7	11/6/1998
LEAD			EPA 200.8	11/6/1998
MAGNESIUM			EPA 200.7	11/6/1998
MANGANESE			EPA 200.7	11/6/1998
MANGANESE			EPA 200.8	11/6/1998
MERCURY			EPA 245.1	11/6/1998
MOLYBDENUM			EPA 200.7	11/6/1998
MOLYBDENUM			EPA 200.8	11/6/1998
NICKEL			EPA 200.7	11/6/1998
NICKEL			EPA 200.8	11/6/1998
NITRATE			EPA 353.1	11/6/1998
OIL AND GREASE			SM 5520B	11/6/1998
ORTHOPHOSPHATE			EPA 365.1	11/6/1998
PHOSPHORUS TOTAL			EPA 365.4	11/6/1998
POTASSIUM			EPA 200.7	11/6/1998
RESIDUE FILTERABLE	(TDS)		EPA 160.1	11/6/1998

Programs : Programs :

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State Laboratory

Environmental Laboratory Licensure Invoice Detail

Page 1-3

AZUSense AZUS99 LabName	ASARCO/AEC LABORA	TORY
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RESIDUE NONFILTERABLE (TSS)	EPA 160.2	11/6/1998
SELENIUM	EPA 200.7	11/6/1998
SELENIUM	EPA 200.8	11/6/1998
SILVER	EPA 200.7	11/6/1998
SILVER	EPA 200.8	11/6/1998
SPECIFIC CONDUCTANCE	EPA 120.1	11/6/1998
THALLIUM	EPA 200.7	11/6/1998
THALLIUM ·	EPA 200.8	11/6/1998
VANADIUM	EPA 200.7	11/6/1998
VANADIUM	EPA 200.8	11/6/1998
ZINC	EPA 200.7	11/6/1998
ZINC	EPA 200.8	11/6/1998

Total Licensed Parameters for this Lab: /U

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ATTACHMENT 13

ANALYTICAL REPORTS



December 22, 1998

Mr.

Please find attached the analytical results for the HDS water sample collected December 14, 1998.

The sample was received by the laboratory on December 18, 1998.

Results were faxed to you on December 22, 1998.

Sinceyely

John H. Mokes

Laboratory/Environmental

Technician

cc: GRStanga (w/attach.)

JLloyd

Ramachandran

ASARCO TECHNICAL SERVICES CENTER

ANALYTICAL DATA REPORT

Water and Solid Waste (Project 3119)

Batch No: 1982521

LAB NO II

1.982521-1

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VALUE	.11 .001 .61 .61 .60 .005 .95 .95
PAHAMETER	AS(TR) CD(TR) CU(TR) FE(TR) HG PB(TR) SE(TR) TL(TR)
	NDS Effluent
DATE Hlected description	
o pest	14-DEC-98 HDS-E001
DATE COLLECTE	14-DEC-9

*SW846-3005 method of preparation for (TR) metals.

Approved

| S. | Fr. | D. |

Reviewer

ASARCO TECHNICAL SVICES CENTER
ANALYTICAL DATA REPORT

Water and Solid Waste (Project 3119)

Batch No: 46981540

MG961540-1	Matrix Spike	L982521-1	AS(TR) CO(TR) CO(TR) EE(TR) HG PB(TR) SE(TR) ZR(TR)	163. 97. 94. 103. 170. 170.	X RECOVERY BD X RECOVERY BD X RECOVERY JJT X RECOVERY VPK X RECOVERY BD X RECOVERY BD X RECOVERY BD X RECOVERY BD	21-0EC-98 21-0EC-98 21-0EC-98 18-0EC-98 21-0EC-98 21-0EC-98 18-0EC-98 18-0EC-98	6020 6020 6020 6010 245.1 6020 6020 6010 289.1
uc981540-2	Duplicate	L982521-1	AS(TR) CD(TR) CU(TR) FE(TR) HG PB(TR) SE(TR) TL(TR)	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	X RPD BB X R RPD BB X R RPD JJT X RPD VPK X RPD WPK X RPD BB X RPD BB X RPD BB	21.0EC-98 21.0EC-98 21.0EC-98 18.0EC-98 22.0EC-98 21.0EC-98 11.0EC-98 18.0EC-98	6020 6020 6020 6010 245.1 6020 6020 6010 289.1
MG981540-3	Laboratory Control	EE1CQ-50/FSHR	AS(TR) CO(TR) CU(TR) FE(TR) HG PHG SE(TR) TL(TR) ZN(TR)	107. 101. 95. 99. 105. 172. 97.	X RECOVERY BD X RECOVERY BD X RECOVERY JJT X RECOVERY VPK X RECOVERY BD	21-0EC-98 21-0EC-98 18-0EC-98 22-0EC-98 21-0EC-98 21-0EC-98 18-0EC-98	6020 6020 6020 6010 245.1 6020 6020 6010
MG981540-4	Prep Blank	12/18/98	AS(TR) CD(TR) CU(TR) FE(TR) HG PB(TR) SE(TR) ZN(TR)	.005.001.010.010.020.003.005.020.005.020.020	ppm 80 mcpq 80 ppm 80 ppm 80 ppm 90 p	21-0EC-98 21-0EC-98 21-0EC-98 18-0EC-98 22-0EC-98 21-0EC-98 18-0EC-98 18-0EC-98	6020 6020 6020 6010 245.1 6020 6020 6010

ASARCO TECHNICAL SERVICES CENTER

ANALYTICAL DATA REPORT

Water and Solid Waste (Project 3119)

Batch No: WG981540

IALYST AHALYZED DAYS METHOD	6020 6020 6020 6010 245.1 6020 6020 6010/6020 289.1
UNITS	w widd wedd dddddddddddddddddddddddddddd
VALUE	.005 .001 .010 .030 .003 .003 .020/.003
PARAMETER	AS(TR) CD(TR) CU(TR) FE(TR) HG PB(TR) SE(TR) TL(TR)
Description.	Practical Quantitation Limit

VG981540-5

Page 2

Standard Operating Procedure

Alkalinity - EPA Method 310.1

Parameter - Alkalinity

Range of Measurement - 1.0 mg/liter to 10,000 mg/liter

Method Detection Limit - 0.8 mg/liter. Calculated in accordance with 40 CFR part 136, Appendix B.

Sample Matrix - Water

<u>Principle, Scope & Application</u> - A sample is titrated to an end point of pH 4.5 with .02 N sulfuric acid. Do not filter, dilute, concentrate or alter sample. This method applies to drinking, surface and saline waters, domestic and industrial wastes.

<u>Interferences & Corrective Actions</u> - Soaps, oily matter and precipitates may cause sluggish electrode response. Allow additional time between titrant additions to let the electrode equilibrate.

<u>Safety Precautions</u> - Refer to the MSDS for sulfuric acid. Standard laboratory protective devices are to be utilized (safety glasses, lab coats and gloves).

<u>Sample Size, Collection, Preservation and Handling</u> - The sample is unpreserved and refrigerated at 4° C. The sample is allowed to equilibrate to room temperature prior to analysis.

<u>Apparatus</u> - pH meter with glass electrode Magnetic stirrer, pipets, flasks and other standard laboratory equipment. Buretts, Pyrex 50, 25 and 10 ml. <u>Routine Preventative Maintenance</u> - Store electrode as per manufacturers instructions.

Reagent & Calibration Standards - .02 N sulfuric acid made from Dilut-It brand primary standard ampule. pH 4, 7, and 10 buffer solutions

Calibration Procedures - Calibrate pH meter as per manufacturers instructions.

<u>Sample Preparation</u> - Allow sample to equilibrate to room temperature prior to analysis.

Analytical Measurement / Procedure - Sample size:

Use a sufficiently large volume of titrant (>20 ml in a 50 ml buret) to obtain a good precision while keeping volume low enough to permit sharp end point.

For <1000 mg CaCO₃/1 use 0.02 N titrant

For <1000 mg CaCO₃/1 use 0.1 N titrant

A preliminary titration is helpful.

Potentiometric titration:

Place sample in flask by pipetting with pipet tip near bottom of flask Measure pH of sample

Add standard acid, being careful to stir thoroughly but gently to allow needle to obtain equilibrium.

Titrate to pH 4.5. Record volume of titrant.

Potentiometric titration of low alkalinity:

For alkalinity of <20 mg/l titrate 100-200 ml as above using a 10 ml microburette and 0.02 N acid solution.

Stop titration at pH in range of 4.3-4.7, record volume and exact pH. Very carefully add titrant to lower pH exactly 0.3 pH units and record volume.

Data Treatment - Calculations:

Potentiometric titration to pH 4.5 Alkalinity, mg/1 CaCO₃ = $\underline{A \times N \times 50,000}$ ml of sample

where:

A = ml standard acid

N = normality standard acid Potentiometric titration of low alkalinity:

Total alkalinity, mg/1CaCO₃ = $(2B - C) \times N \times 50,000$ ml of sample

where:

B = ml titrant to first recorded pH

C = total ml titrant to reach pH 0.3 units lower

N = normality of acid

Data is recorded in the laboratory notebooks, utilizing standard practices, (i.e. Each page is dated and initialed). Data is logged out on the LIMS system. A hard copy of the data is kept on the file with the analysts copy of the report.

<u>Data Deliverables</u> -Data deliverables for this project include all raw data including laboratory notebooks.

<u>Quality Control Requirements</u> -Quality control includes blanks, duplicates, standard reference materials. Frequency is 1 per 20 samples.

References - Methods for Chemical Analysis of Water and Wastes.

Method Validation Data -Refer to Methods for Chemical Analysis of Water and Wastes.

Standard Operating Procedure

Chloride in Water - EPA 325.2

Parameter - Chloride

Range of Measurement - 1 to 100 ppm Chloride

Method Detection Limit - .77 ppm calculated in accordance with 40 CFR part 136, Appendix D.

Sample Matrix - Water

<u>Principle, Scope & Application</u> - Thiocyanate ion (SCN) is liberated from mercuric thiocyanate through sequestration of mercury by chloride ion to form un-ionized mercuric chloride. In the presence of ferric ion, the liberated SCN forms highly colored ferric thiocyanate in concentration proportional to the original chloride concentration. This method applies to drinking, surface, and saline waters, domestic and industrial wastes.

Interferences & Corrective Actions - No significant interferences.

<u>Safety Precautions</u> -Refer to MSDS's for mercuric thiocyanate solutions. Standard laboratory protective devices are to be utilized (safety glasses, lab coats and gloves).

<u>Sample Size, Collection, Preservation and Handling</u> - Samples are unpreserved and refrigerated at 4° C. The holding time for chloride is 28 days.

<u>Apparatus</u> - Alpkem model 3590 automated colorimetric segmented flow analyzer.

<u>Routine Preventative Maintenance</u> - Chloride is the simplest and most trouble free run on the instrument.

Replace pump tubes and ferrules as wear and tear requires.

The reagent must be filtered through .45 micrometer filter.

The reagent must be capped at all times.

Note: Collect and dispose of the waste as indicated in page 7 of the manual.

Reagent & Calibration Standards

- a. Stock Mercuric Thiocyanate: Dissolve 4.17g of mercuric thiocyanate (FW 316.75) in 500ml methanol (FW 32.04), mix and dilute solution to 1000ml with methanol. Filter.
- b. Stock Ferric Nitrate: Dissolve 202g of ferric nitrate (FW 404.0) in 500ml of deionized water, add 31.5ml nitric acid (FW 63.02), mix and dilute the solution to 1000ml with deionized water. Filter.
- c. Working Chloride Color Reagent: Add 30ml of Stock Mercuric Thiocyanite and 30ml of Stock Ferric Nitrate solution to 100 ml deionized water. Dilute the solution to 200ml with deionized water.
- d. Carrier: Add 2ml of Brij-35, 30% w/v to 1000ml of deionized water.
- -The stock calibrant is a 1000 ppm solution utilizing 1.64838 g of sodium chloride diluted to one liter with deionized water. Serial dilutions are made for standards at 2, 5, 10, 25, 50, 75 and 100 ppm.

Calibration Procedures - A calibration curve must be prepared each day for each sample run with a calibration blank and 2, 5, 10, 25, 50, 75 and 100 ppm standards. After calibration, the calibration curve must be verified by use of at least a calibration blank and a calibration check standard (made from a reference material or other independent standard material) at or near the mid-range. The calibration reference must be measured within 10% of it's true value for the curve to be considered valid.

<u>Sample Preparation</u> - Samples are allowed to equilibrate to room temperature prior to analysis. Turbid samples are filtered prior to analysis.

Analytical Measurement - SETTING UP THE RUN ON THE SAMPLER:

The Detector 510 and Sampler 501 should be adjusted according to the instructions on page 15 of the chloride section in the manual.

To set up the 502b pump turn to page 14.

Turn to page 7 for Operation Procedure.

BEGIN ANALYSIS.

Settings on the 510 detector should be the following: page 15

-wavelength

480

-range

0.2 AUFS

-rise time

10 sec.

The settings on the 501 sampler should be the following: page 15

-sample time

20 sec.

-wash time

60 sec.

-pecking

on

Procedure

- Set up manifold and reagents.
- Allow 30 minute warm up and pump reagents until a stable baseline is achieved.
- -Construct calibration curve
- -Analyze samples.

<u>Data Treatment</u> -A computer data printout is obtained from the Alpkem 3590 system. This includes calibration curves, correlation coefficients, concentrations and associated intensities. Data is logged out on the LIMS system. A hard copy of the data is kept on file with the analysts copy of the report.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements - In brief, the quality control includes; 1) the following four QC samples - Preparation Blank, Duplicate, Matrix Spike and Laboratory Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or less	< PQL (Practical Quantitation Limit)
Laboratory Control Sample	1 per batch of 20 or less	80-120%
Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL

Spike	1 per batch of 20 or less	75-125%
Initial Calibration Verification	after every calibration	90-110%
Initial Calibration Blank	after every calibration	< PQL
Continuing Calibration Verification	10%	90-110%
Continuing Calibration Blanks	10%	< PQL

References -Enviroflow 3500 Technical Information Perstorp Operators Manual.

The Flow Solution Operation Manual - Alpkem/Perstorp Analytical SoftPac (tm) Plus Manual - Perstorp Analytical

EPA Methods for Analysis of Water and Wastes

Method Validation Data - Refer to EPA Methods for Chemical Analysis of Water and Wastes - Method 325.2.

STANDARD OPERATION PROCEDURE FOR ENVIRONMENTAL DIGESTIONS (UPDATED 4/23/97)

Please review the following guidelines in order to insure a consistent and appropriate approach to sample digestion and reporting. Digestions will be conducted in laboratory clean glassware and covered throughout the course of the digestion.

1. WATER SAMPLES

For analysis of water samples, we reference <u>"EPA Methods for Chemical Analysis of Water and Wastes</u> (200 series numbers (ie: 239.1 for flame AA analysis of Pb). We also reference EPA SW 846, (ie: 6010 for ICP). The sample digestion for water samples should be consistent with these documents. The documents should be reviewed for details.

A. Total Metals

Measure an appropriate aliquot of sample (usually 50 mls), add 3 mls concentrated HNO₃ and evaporate to near dryness (no area of the beaker is to go dry). Add 5 mls of concentrated HNO₃ and reflux to near dryness. Add 10 mls of 1:1 HCL and 15 mls DI H₂O per 100 mls of final solution and warm for 15 minutes to dissolve any precipitate. Build to an appropriate final volume (usually 50 mls).

B. Total Recoverable Metals

To an appropriate volume of sample add 2 mls 1:1 HNO₃ and 10 mls 1:1 HCL and reduce volume on hot plate to approximately 25 mls. Build to an appropriate final volume.

C. Dissolved Metals

The filtered sample can be analyzed as received, or evaporated to a lesser volume for concentration purposes.

D. GFAA Analysis

HCL acid should not be employed if analysis by GFAA is required.

E. ICP MS Analysis - 6020 M

To a 100 ml aliquot add 2 ml of 1:1 HNO₃ and 1 ml of 1:1 HCl. Reduce the volume to approximately 25 mls. Build to 100 mls.

2. EXTRACTS

For extracts (eg: TCLP) reference EPA SW 846 methods (eg: 6010 for ICP). The appropriate method of digestion, which is 3010, should be referenced in a cover letter as well as the method number for the extraction (eg: TCLP-1311, SPLP-1312). This digestion method is, for all intents and purposes, identical to the "total metals" digestion for water samples.

3. SOIL AND VEGETATION SAMPLES

For soil analyses, reference EPA SW 846 methods. The method of digestion is 3050 and is summarized below.

A representative 1 to 2 gram sample is digested in 10 mls of 1:1 $\rm HNO_3$ for 10 to 15 minutes. Cool the sample and add 5 mls concentrated $\rm HNO_3$ and reflux for 30 minutes. Cool the sample, add 2 mls DI and 3 mls 30% $\rm H_2O_2$ and heat. Continue adding $\rm H_2O_2$ in 1 ml aliquots with warming up to a total of 10 mls. For flame AA and ICP, add 10 mls of 1:1 HCL per 100 mls final volume, reflux for 15 more minutes and build to volume (typically 1 gram to 50 mls). For GFAA analysis, do not add the HCL; reduce the sample to approximately 5 mls and build to volume.

cc: GRStanga VPKeller

Standard Operating Procedure

Hardness - EPA Method 200.7 Ca & Mg

Parameter - Hardness

Range of Measurement - 1 - 1000 mg/L Range can be extended by sample dilution.

Method Detection Limit - .10mg/l

Sample Matrix - Water and waste water

<u>Principle, Scope & Application</u> - Hardness as determined by calculation from the results of separate determinations of calcium and magnesium. This method applies to the determination of hardness in water at the range specified above.

Interferences & Corrective Actions - Refer to EPA Method 200.7

<u>Sample Size, Collection, Preservation and Handling</u> - Typically hardness is analyzed on an unpreserved sample, for purposes of comparison to the soluble anions, which are also to be analyzed on an unpreserved sample.

Apparatus - Perkin Elmer Model Optima 3000 DV

For: Routine Preventative Maintenance, Reagent & Calibration Standards, Calibration Procedures, Sample Preparation, Analytical Measurement, Procedure, Quality Control Requirements and Method Validation Data:

Reference 1) AMS Document TSC-430-L-PR-010, "Basic Operation and Maintenance of the Perkin Elmer Optima 3000 DV.

2) Standard Operating Procedure For Method 200.7.

<u>Data Treatment</u> - Data is logged out on the LIMS system. A hard copy of the data is kept on file with the analysts copy of the report. Hardness, Mg equivalent CaCO3/l - 2.497 [Ca, Mg/l] + 4.118 Mg, Mg/l]. <u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

References - EPA Method 200.7 Ca&Mg.

MICS. INC+131011 -. 2017.

Standard Operating Procedure

Mercury - EPA Method 245.1

Parameter - Mercury

Range of Measurement - .2 ppb - 5 ppb. Note; range can be extended, changing operating parameters and/or sample dilution.

Method Detection Limit - .03 ppb

Sample Matrix - Aqueous wastes, extracts, ground and surface waters

<u>Principle, Scope & Application</u> - The principle of measurement is cold vapor atomic absorption. This method applies to the analysis of the matrices listed above.

Interferences & Corrective Actions -

- 1) Potassium permanganate is added to eliminate possible interference from sulfide. Concentrations as high as 20 mg/L of sulfide as sodium sulfide do not interfere with spike recovery.
- 2) Copper has also been reported to interfere: however, copper concentrations as high as 10 mg/L had no effect on recovery of mercury from spiked samples.
- 3) Samples high in chlorides require additional permanganate (as much as 25 ml) because, during the oxidation step, chlorides are converted to free chlorine, which also absorbs radiation of 253.6 nm. Care must therefore be taken to ensure that free chlorine is absent before the mercury is reduced and swept into the cell. This may be accomplished by using an excess of hydroxylamine sulfate reagent (25 ml).

<u>Safety Precautions</u> - Mercury is toxic. Standard laboratory protective devices of safety glasses, gloves and lab coats should be utilized to prevent exposure. A tube containing potassium permanganate is utilized at the gas exit port to prevent exposure to mercury vapors.

Sample Size, Collection, Preservation and Handling - In accordance with Methods for Chemical Analysis of Water and Wastes, Chapter 1, the samples are to be acidified with HNO3 to pH <2 at the time of sample collection. The holding time for mercury is 28 days.

Apparatus - CETAC Model 6000A - Mercury Analyzer

Routine Preventative Maintenance - Refer to ASARCO AMS document #154 section 6.3. Also refer to CETAC M-6000A-Mercury Analyzer Operators Manual.

Reagent & Calibration Standards -

- 1) Sulfuric acid, (H2SO4), concentrated: Reagent grade.
- 2) Sulfuric acid, 0.5 N: Dilute 14.0 ml of concentrated sulfuric acid to 1.0 liter.
- 3) Nitric acid (HNO3), concentrated: Reagent grade of low mercury content.
- 4) Stannous Chloride: Add 100 g stannous chloride to 100 ml of concentrated HCl.
- 5) Sodium chloride-hydroxylamine sulfate solution: Dissolve 12 g of sodium chloride and 12 g of hydroxylamine sulfate in type II water and dilute to 100 ml. (Hydroxylamine hydrochloride may be used in place of hydroxylamine sulfate.)
- 6) Potassium permanganate, mercury-free, 5% solution (w/v): Dissolve 5 g of potassium permanganate in 100 ml of type II water.
- 7) Potassium persulfate, 5% solution (w/v): Dissolve 5 g of potassium persulfate in 100 ml of type II water.
- 8) Stock mercury solution: Dissolve 0.1354 g of mercuric chloride in 75 ml of type II water. Add 10 ml of concentrated HNO3 and adjust the volume to 100.0 ml (1 ml = 1 mg Hg).
- 9) Mercury working standard: Make successive dilutions of the stock mercury solution to obtain working standards, typically .2, .5, 1.0, 2.0 and 5.0 ppb Hg in 10% HCl.

<u>Calibration Procedures</u> - Calibrations are conducted prior to each analytical run and every 8 hours at a minimum. Routine samples are analyzed with a calibration blank and 5 standards ranging from .2 ppb to 5 ppb.

Sample Preparation - Samples are prepared in accordance with method 245.1 utilizing proportionately reduced volumes of sample and reagents (25 mls

sample, 1.25 mls H2SO4, .6 mls conc. HNO3, 4 mls KMNO4, 2 mls potassium persulfate and 2 mls hydroxylamine hydrochloride). Samples are heated at 95 deg. C in a hot water bath for 2 hours prior to the addition of hydroxalamine hydrochloride. Samples are built to a final volume of 50 mls.

Analytical Measurement / Procedure - The samples are placed in autosampler tubes and the analysis proceeds in accordance with the guidelines detailed in accordance with the ASARCO AMS document TSC-430-L-PR-003. This is a continuous flow analyzer with automated addition of stannous chloride.

<u>Procedure</u> - The procedure for analysis is detailed in the ASARCO AMS document "Basic Operation and Maintenance of the CETAC mercury Analyzer - Model 6000A."

<u>Data Treatment</u> - Data is kept on hard disk for one year. A hard copy of the data is kept on file with the analysts copy of the report. Data is logged out for analytical reports on the LIMS system.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements - In brief, the quality control includes; 1) the following four digestion QC samples- Preparation Blank, Duplicate, Matrix Spike and Laboratory Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or	< PQL (Practical

	less	Quantitation Limit)
Laboratory Control Sample	1 per batch of 20 or less	85-115%
Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL
Spike	1 per batch of 20 or less	75-125%
Initial Calibration Verification	after every calibration	95-105%
Initial Calibration Blank	after every calibration	< CRDL
Continuing Calibration Verification	10%	90-110%
Continuing Calibration Blanks	10%	< PQL

References - EPA Methods for Chemical Analysis of Water and Waste CETAC ASX-500-Auto Sampler Operators Manual CETAC M-6000A-Mercury Analyzer Software Manual CETAC M-6000A-Mercury Analyzer Operators Manual ASARCO AMS Document-TSC-430-L-PR-003-Basic Operation and Maintenance of the CETAC Mercury Analyzer-Model M6000A

Method Validation Data -Refer to EPA Methods for Chemical Analysis of Water and Waste.

Standard Operating Procedure

Mercury - Method 7470A

Parameter - Mercury

Range of Measurement - .2 ppb - 5 ppb Note; range can be extended, changing operating parameters and/or sample dilution.

Method Detection Limit - .03 ppb

Sample Matrix - Aqueous wastes, extracts, ground and surface waters

<u>Principle, Scope & Application</u> - The principle of measurement is cold vapor atomic absorption. This method applies to the analysis of the matrices listed above.

Interferences & Corrective Actions -

- 1) Potassium permanganate is added to eliminate possible interference from sulfide. Concentrations as high as 20 mg/L of sulfide as sodium sulfide do not interfere with spike recovery.
- 2) Copper has also been reported to interfere: however, copper concentrations as high as 10 mg/L had no effect on recovery of mercury from spiked samples.
- 3) Samples high in chlorides require additional permanganate (as much as 25 ml) because, during the oxidation step, chlorides are converted to free chlorine, which also absorbs radiation of 253.6 nm. Care must therefore be taken to ensure that free chlorine is absent before the mercury is reduced and swept into the cell. This may be accomplished by using an excess of hydroxylamine sulfate reagent (25 ml).

<u>Safety Precautions</u> - Mercury is toxic. Standard laboratory protective devices of safety glasses, gloves and lab coats should be utilized to prevent exposure. A tube containing potassium permanganate is utilized at the gas exit port to prevent exposure to mercury vapors.

Sample Size, Collection, Preservation and Handling - In accordance with SW846, Chapter 3, the samples are to be acidified with HNO3 to pH <2 at the time of sample collection. The holding time for mercury is 28 days.

Apparatus - CETAC Model 6000A - Mercury Analyzer

Routine Preventative Maintenance - Refer to ASARCO AMS document #154 section 6.3. Also refer to CETAC M-6000A-Mercury Analyzer Operators Manual.

Reagent & Calibration Standards -

- 1) Sulfuric acid, (H2SO4), concentrated: Reagent grade.
- 2) Sulfuric acid, 0.5 N: Dilute 14.0 ml of concentrated sulfuric acid to 1.0 liter.
- 3) Nitric acid (HNO3), concentrated: Reagent grade of low mercury content.
- 4) Stannous Chloride: Add 100 g stannous chloride to 100 ml of concentrated HCl.
- 5) Sodium chloride-hydroxylamine sulfate solution: Dissolve 12 g of sodium chloride and 12 g of hydroxylamine sulfate in type II water and dilute to 100 ml. (Hydroxylamine hydrochloride may be used in place of hydroxylamine sulfate.)
- 6) Potassium permanganate, mercury-free, 5% solution (w/v): Dissolve 5 g of potassium permanganate in 100 ml of type II water.
- 7) Potassium persulfate, 5% solution (w/v): Dissolve 5 g of potassium persulfate in 100 ml of type II water.
- 8) Stock mercury solution: Dissolve 0.1354 g of mercuric chloride in 75 ml of type II water. Add 10 ml of concentrated HNO3 and adjust the volume to 100.0 ml (1 ml = 1 mg Hg).
- 9) Mercury working standard: Make successive dilutions of the stock mercury solution to obtain working standards, typically .2, .5, 1.0, 2.0 and 5.0 ppb Hg in 10% HCl.

<u>Calibration Procedures</u> - Calibrations are conducted prior to each analytical run and every 8 hours at a minimum. Routine samples are analyzed with a calibration blank and 5 standards ranging from .2 ppb to 5 ppb.

<u>Sample Preparation</u> - Samples are prepared in accordance with method 7470A utilizing proportionately reduced volumes of sample and reagents (25 mls sample, 1.25 mls H2SO4, .6 mls conc. HNO3, 4 mls KMNO4, 2 mls potassium

persulfate and 2 mls hydroxylamine hydrochloride). Samples are heated for 2 hours in a hot water bath at 95 degrees C for 2 hours prior to the addition of hyroxylamine hydrochloride. Samples are built to a final volume of 50 mls.

Analytical Measurement / Procedure - The samples are placed in autosampler tubes and the analysis proceeds in accordance with the guidelines detailed in accordance with the ASARCO AMS document TSC-430-L-PR-003. This is a continuous flow analyzer with automated addition of stannous chloride.

<u>Procedure</u> - The procedure for analysis is detailed in the ASARCO AMS document "Basic Operation and Maintenance of the CETAC mercury Analyzer - Model 6000A."

<u>Data Treatment</u> - Data is kept on hard disk for one year. A hard copy of the data is kept on file with the analysts copy of the report. Data is logged out for analytical reports on the LIMS system.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements - A detail of quality control for this project can be found in SW846-method 7470A. In brief, the quality control includes; 1) the following four digestion QC samples- Preparation Blank, Duplicate, Matrix Spike and Laboratory Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or	< PQL (Practical

	less	Quantitation Limit)
Laboratory Control Sample	1 per batch of 20 or less	80-120%
Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL
Spike	1 per batch of 20 or less	85-115%
Initial Calibration Verification	after every calibration	90-110%
Initial Calibration Blank	after every calibration	< CRDL
Continuing Calibration Verification	10%	80-120%
Continuing Calibration Blanks	10%	< PQL

References - EPA SW846-Test Methods for Evaluating Solid Waste CETAC ASX-500-Auto Sampler Operators Manual CETAC M-6000A-Mercury Analyzer Software Manual CETAC M-6000A-Mercury Analyzer Operators Manual ASARCO AMS Document-TSC-430-L-PR-003-Basic Operation and Maintenance of the CETAC Mercury Analyzer-Model M6000A

<u>Method Validation Data</u> -Refer to data SW846 Method 7470A section 9.0-Method Performance.

Standard Operating Procedure Metals by Method 6010B-Inductively Coupled PlasmaAtomic Emission Spectroscopy

Parameter - Metals and Metalloids

Range of Measurement - Analyte specific

Method Detection Limit - Analyte specific - Refer to current MDL summaries.

<u>Sample Matrix</u> - Digestions of: soils, sediments, TCLP extractions and EP extracts, waters and aqueous samples.

<u>Principle, Scope & Application</u> - The principle of measurement is inductively coupled plasma emission spectroscopy. Refer to method 6010B section 2.2 for a synopsis of the principle. This method applies to the matrices listed above.

Interferences & Corrective Actions - Interferences and the associated corrections are detailed extensively in 1) section 3.0 of method 6010B, 2) "ICP WinLab Software Guide" The Perkin-Elmer Corporation, 1997, 3) ICAP 61E Operators Manual. -Thermo Jarrell Ash Corporation, 4) Basic Operation and Maintenance of the TJA ICAP 61-ASARCO AMS document TSC-430-L-PR-015.

The primary components of interference correction detailed in the references, 1) line selection, 2) off peak background correction, and, 3) interelement correction factors.

Safety Precautions -

- 1. Switch power to "off" before accessing any circuits.
- 2. Handle acidified solutions carefully. Address all spills immediately.
- 3. The instrument's ventilation fan(s) must be operating when the instrument is running.
- 4. Some components may remain at an elevated temperature long after the plasma has been extinguished.

Sample Size, Collection, Preservation and Handling -

1) Collect samples in accordance with SW846, Chapter 3. 2) Sample size is matrix specific. The final volume of aqueous digestate should be a minimum of 25 mls. 3) The holding time for metals is 6 months. 4) At the time of sampling, the sample must be preserved with nitric acid to pH <2.

Apparatus -Perkin-Elmer 3000 DV

Routine Preventative Maintenance -

- 1. Clean air filters monthly.
- 2. Change peristaltic pump tubing when a flat place develops between the wishbones.
- 3. Clean or change the torch when it is dirty. Reference AMS Document TSC-430-L-PR-010 for more detail regarding instrument maintenance.

Reagent & Calibration Standards - The rinse solution is 5%, v/v, HCl. The calibration solution(s) used are specific to the analyte(s) of interest. The blank solution is 10%, v/v HCl. Standards we purchase from a commercial vendor. Records of standard dilution and period of use are maintained in an analytical standards log book.

Calibration Procedures - A calibration curve must be prepared each day for each sample run with a minimum of a calibration blank and one standard per analyte of interest. After calibration, the calibration curve must be verified by use of at least a calibration blank and a calibration check standard, (made from a reference material or other independent standard material), at or near the midrange. The calibration reference must be measured within 10 % of its true value for the curve to be considered valid.

Sample Preparation - Samples are digested in accordance with SW846, chapter 3.

Analytical Measurement - Once the plasma has been established, allow the instrument to warm up under the analytical run conditions for at least an hour. Profile using the mercury lamp and in the case of the Optima, optimize the X and Y alignment of the torch to the optics.

Procedure

1) Select a method.

- 2) Calibrate and read the standards against the calibration curve.
- 3) Input the samples of interest, the quality controls and the relevant correction factor into a sample information file.
- 4) Select the analytical method, the sample information file and the data file for data storage.
- 5) Start the analytical run and periodically ascertain the quality of the data being generated.

Data Treatment -The data is automatically output to the printer connected to the computer that controls the instrument. Data is kept on hard disk for approximately one week. A hard copy of the data is kept on file with the analyst's copy of the report. Data is logged out for analytical reports on the LIMS system.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements - In brief, the quality control includes; 1) the following four digestion QC samples - Preparation Blank, Duplicate, Matrix Spike and Laboratory Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks, contract required detection limit standard, and interferents check samples A and A+B.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or less	< PQL (Practical Quantitation Limit)
Laboratory Control Sample	1 per batch of 20 or less	80-120%
Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL

Spike	1 per batch of 20 or less	75-125%
Initial Calibration Verification	after every calibration	90-110%
Initial Calibration Blank	after every calibration	< PQL
Continuing Calibration Verification	10%	90-110%
Continuing Calibration Blanks	10%	< PQL
Interferents Check Sample A (ICS A) Interferents Check Sample A+B (ICS A+B)	At the beginning of the first analytical run of the day, at the end of the last analytical run of the day but not less than every 8 hours.	80-120%

References -EPA SW846 - Test Methods for Evaluating Solid Waste ICAP 61 Operator's Manual-Thermo Jarrell Ash Corporation ICAP 61E Operator's Manual-Thermo Jarrell Ash Corporation ICAP 61 A General Scheme for Software Usage. Analytical Protocol and Method Development.

Optima 3000 Hardware Guide-The Perkin-Elmer Corporation ICP WinLab Software Guide-The Perkin-Elmer Corporation Optima 3000 DV Supplement-The Perkin-Elmer Corporation

<u>Method Validation Data</u> -Refer to data SW846 Method 6010B Section 9.0 - Method Performance

Standard Operating Procedure Metals by Method 6020-Inductively Coupled PlasmaMASS Spectrometry

Parameter - Metals and Metalloids

Range of Measurement - Analyte specific

Method Detection Limit - Analyte specific - Refer to current MDL summaries.

<u>Sample Matrix</u> - Dissolved elements and digestions of: soils, sediments and ground, surface and drinking water.

<u>Principle, Scope & Application</u> - The principle of measurement is inductively coupled plasma Mass spectrometry. Refer to method 6020 section 1.0 for a synopses of the principle. This method applies to the matrices listed above.

<u>Interferences & Corrective Actions</u> - Interferences and the associated corrections are detailed extensively in the following

- 1) Perkin Elmer SOP "Method 200.8 using Elan 6000",
- 2) Perkin Elmer SOP "Method 6020 using Elan 6000",
- 3) EPA Method 6020.

Safety Precautions -

General laboratory safety procedures are followed in the handling of chemical reagents. For more information, refer to EPA Method 200.8 Section 5.0.

Sample Size, Collection, Preservation and Handling -

Samples are collected in accordance with EPA SW 846. The final volume of aqueous digestates should be a minimum of 25 mls. The holding time for metals is six months.

<u>Apparatus</u> -Perkin Elmer Sciex Elan 6000 - Inductively Coupled Plasma-Mass Spectrometer.

Routine Preventative Maintenance - Preventive maintenance is covered extensively in;

1) Perkin Elmer Elan 6000 ICP-MS Service Manual.

2) ASARCO Document #TSC-430-L-PR-018 "Basic Operation and Maintenance of the PE Elan 6000 ICP-MS".

Reagent & Calibration Standards -

Calibration standard - Spex Claritas Instrument Calibration Standard 2 Quality Control Sample - Spex Claritas Instrument Check Standard 1 Internal Standard - Spex Claritas Multi-element Internal Standard Acids - J.T. Baker Instra-Analyzed©

<u>Calibration Procedures</u> - Standards are analyzed at the beginning of daily analysis, when sample matrices change or when sample sensitivity increases or decreases at a level to interfere with analysis.

Standard Concentrations:

0 ppb

10 ppb

20 ppb

200 ppb

Sample Preparation - For sample preparation refer to EPA SW 846.

The internal standard is added to all standards, quality control and samples in accordance with EPA Method 6020.

Analytical Measurement & Procedure -

For comprehensive instructions, refer to ASARCO AMS Document #TSC-430-L-PR-018.

"Basic Operation and Maintenance of the PE Elan 6000 ICP-MS"

Listed below is a summary

- 1) Tune the instrument
- 2) Perform a daily performance analysis to confirm that instrument performance is within specifications.
- 3) Calibrate the instrument
- 4) Start the analysis of the samples and associated quality control.

Data Treatment -The PE Elan 6000 has several report options. A general summary report is used for all analyses unless another type of report is requested. The report is delivered to the project chemist who was assigned the samples. Data is logged out on the LIMS system.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements - The quality control includes; 1) the following four digestion QC samples - Preparation Blank, Duplicate, Matrix Spike and Laboratory Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or less	< PQL (Practical Quantitation Limit)
Laboratory Control Sample	1 per batch of 20 or less	80-120%
Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL
Spike	1 per batch of 20 or less	75-125%
Initial Calibration Verification	after every calibration	90-110%
Initial Calibration Blank	after every calibration	< PQL
Continuing Calibration Verification	10%	90-110%
Continuing Calibration Blanks	10%	< PQL
Internal Standards	All inclusive	30-120%

Interference Check Standards A+B	Beginning of an analytical run or once ever 12 hrs.	None listed
Serial Dilution	1 per batch of 20 or less	± 10% of the undiluted sample value

References -EPA Methods for Chemical Analysis of Water and Waste Perkin Elmer SOP "Method 200.8 Using Elan 6000"
Perkin Elmer SOP "Method 6020 Using Elan 6000"
EPA Method 6020
Perkin Elmer Elan 6000 ICP-MS Service Manual
ASARCO AMS Document #TSC-430-L-PR-018 - "Basic Operation and Maintenance of the PE Elan 6000 ICP-MS

Method Validation Data -Refer to Method 6020 Section 9.0 Tables 1-5.

Standard Operating Procedure Orthophosphate in Water - EPA Method 365.1

Parameter - Orthophosphate

Range of Measurement - .10 - 1.5 ppm

Method Detection Limit - .005 ppm calculation in accordance with 40CFX part 136, Appendix B.

Sample Matrix - water

<u>Principle, Scope & Application</u> - This method is used for the determination of orthophosphate in drinking, surface and saline waters, domestic and industrial wastes. The applicable range of this method is 0.10 to 1.5 mg/L P. The range may be extended by changing the detector sensitivity or by sample dilution.

Orthophosphate reacts with molybdenum (VI) and antimony (III) in an acid medium to form an antimony-phosphomolybdate complex. This complex is subsequently reduced with ascorbic acid to form a blue color and the absorbance is measured at 660nm.

<u>Interferences & Corrective Actions</u> - Ferric iron up to 50 mg/L copper, up to 10 mg/L do not interfere. Filter turbid samples prior to analysis. Samples with background absorbance at the analytical wavelength may interfere.

<u>Safety Precautions</u> - Refer to MSDS's for Ammonium Molybdate, Antimony Potassium Tartrate, Ascorbic Acid, Potassium Dihydrogen Phosphate, Sulfuric Acid, concentrated. Standard protective devices are to be utilized (safety glasses, lab coats, gloves and vent hood).

Sample Size, Collection, Preservation and Handling - Determine orthophosphate in unpreserved samples immediately after they are collected. Unpreserved samples may be held for 48 hours when cooled immediately and stored at 4°C. Filter turbid samples prior to analysis.

Apparatus - Alpkem model 3590 automated colorimetric segmented flow analyzer

Routine Preventative Maintenance - Ascorbic Acid Stock Solution-Check color of the ascorbic acid reagent. It should be clear liquid, colorless to light amber in color; a darker color indicates poor reagent quality or aged solution. Either remake solution or change to new source of ascorbic acid.

When changing the sample pump tube, it is important to cut the tube close to the collars before reconnecting the transmission tubing. This helps prevent carryover.

When starting up, let system run on deionized water with 2.0 ml/L DowFax 2A1 for 10-15 minutes. Add the color reagent and wait for a stable baseline. The pH from the flowcell waste line, should approximate pH 1 when checked with pH paper.

If baseline is noisy, check:

- 1) Color of ascorbic acid stock solution.
- 2) Reagents should be filtered prior to use through a 0.45μ membrane filter.

Carryover can be solved by washing out entire system with a 1.0N sodium hydroxide solution for 5 minutes. For consistent results, flush system out every day before use and increase the sampler wash time.

Reagent & Calibration Standards-

1. DowFax Start-Up Solution (1L)

To a 1L volumetric flask containing 800 ml deionized water, add 2 ml of DowFax 2A1. Dilute to 1.000 ml with deionized water.

2. Stock Sulfuric Acid, 5N (1L)

Add 140 ml of concentrated sulfuric acid (FW 98.08) to 800 ml of deionized water contained in a 1L volumetric flask. Cool and dilute to 1,000 ml with deionized water.

- 3.Stock Antimony Potassium Tartrate Solution (1L)
 Dissolve 3.0 g of antimony potassium tartrate (FW 333.94) in 500 ml of
 deionized water contained in a 1L volumetric flask. Dilute to 1000 ml with
 deionized water. Store at 4°C in a dark bottle.
- 4. Stock Ammonium Molybdate Solution (1L)

Dissolve 40 g of ammonium molybdate (FW 1235.95) in 800 ml of deionized water contained in a 1 L volumetric flask. Dilute to 1000 ml with deionized water. Store at 4°C in a polyethylene bottle.

5. Stock Ascorbic Acid Solution (1L)

Dissolve 18 g of ascorbic acid (FW 176.13) in 700 ml of deionized water contained in a 1 L volumetric flask. Dilute to 1.000 ml with deionized water. This solution is stable for one week if stored at 4°C.

6. Stock Sulfuric Acid, 5N (1L)

Cautiously add 140 ml of concentrated sulfuric acid (FW 98.08) to 800 ml of deionized water contained in a 1 L volumetric flask. Cook to room temperature and dilute to 1000 ml with deionized water.

- 7. Stock Antimony Potassium Tartrate Solution (1L)
- Dissolve 3.0 g of antimony potassium tartrate (FW 333.94) in 500 ml of deionized water contained in a 1 L Volumetric flask. Dilute to 1000 ml with deionized water. Store at 4°C in a dark bottle.
- 8. Stock ammonium Molybdate Solution (1L)

Dissolve 40 g of ammonium molybdate (FW 1235.95) in 800 ml of deionized water contained in a 1 L volumetric flask. Dilute to 1000 ml with deionized water. Store at 4°C in a polyethylene bottle.

9. Stock Ascorbic Acid Solution (1L)

Dissolve 18 g of ascorbic acid (FW 176.13) in 700 ml of deionized water contained in a 1 L volumetric flask. Dilute to 1000 ml with deionized water. This solution is stable for one week if stored at 4°C.

- 10. Color Reagent (200 ml)
 - 100 ml Stock Sulfuric Acid, 5N
 - 10 ml Stock Antimony Potassium Tartrate Solution
 - 30 ml Stock Ammonium Molybdate Solution
 - 60 ml Stock Ascorbic Acid Solution

Add reagents in the order stated and mix after each addition. This will prevent the ascorbic acid from turning a darker color when the solution is first made. Prepare reagent daily.

The stock calibrant is a 1000 mg/l from 4.393 g of dry potassium dihydrogen phosphate (FW 136.09) diluted to 1 liter with deionized water. Serial dilutions are made for standards at .10, .20, .50, 1.0 and 1.5 ppm.

<u>Calibration Procedures</u> - A calibration curve must be prepared for each sample run with a minimum of calibration blank and .10, .20, .50, 1.0 and 1.5 ppm standards. After calibration, the calibration curve must be verified by use of at least a calibration blank and a calibration check standard (made from a

reference material or other independent standard material) at or near the midrange. The calibration reference must be measured within 10% of it's true value for the curve to be considered valid.

<u>Sample Preparation</u> - Samples are allowed to equilibrate to room temperature prior to analysis. Turbid samples are filtered prior to analysis.

Analytical Measurement - The detector 510 and sampler 501 should be adjusted according to the instructions on page 15 of the Orthophosphate section of the manual. To set up the 502b pump turn to page 14. Note; the helper line (black/black) is not used in this run. Attach the sample line #5 directly to 501 sampler line.

Turn to page 7 (paragraphs 3-7) for Operating Procedure

NOTE: The reagent is always made fresh

The heater is set to 37 degrees Celsius.

The surfactant solution for orthophosphate is DowFax.

The settings on the 510 detector should be the following; page 15

-wavelength

660;

-range

0.2 AUFS;

-rise time

3.0 sec

The settings on the 501 sampler should be the following; page 15

-sample time

15 sec

-wash time

55 sec

-pecking

off

If the Sync (highest standard) check is good, you can start the sampler run at this time.

Press Esc once

Press Alt and #1 simultaneously, then press Enter, followed by Y.

Press F3 which allows for monitoring and observing the analytical run.

Procedure -

- -Set up manifold
- -Pump Reagents until a stable baseline is obtained
- -Construct calibration curve
- -Analyze samples in accordance with the analytical measurement section of this document

<u>Data Treatment</u> - A computer data printout is obtained from the Alpkem 3590 system. This includes calibration curves, correlation coefficients, concentrations and associated intensities. Data is logged out on the LIMS system. A hard copy of the data is kept on file with the analysts copy of the report.

<u>Data Deliverables</u> - Standard data deliverables include sample date, date of analyses, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data, dependent o\upon the clients request.

Quality Control Requirements - The quality control includes; 1) the following four QC samples - Preparation Blank, Duplicate, Matrix Spike and Laboratory Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or less	< PQL (P ractical Q uantitation L imit)
Laboratory Control Sample	1 per batch of 20 or less	80-120%
Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL
Spike	1 per batch of 20 or less	75-125%
Initial Calibration Verification	after every calibration	90-110%
Initial Calibration Blank	after every calibration	< PQL

Continuing Calibration Verification	10%	90-110%
Continuing Calibration Blanks	10%	< PQL

References - Enviroflow 3500 Technical Information
Perstorp Operators Manual.
The Flow Solution Operation Manual - Alpekm/Perstorp
Analytical SoftPac (tm) Plus Manual-Perstorp
Analytical
EPA Method 365.1

Method Validation Data - Refer to EPA Method 375.2 of Methods for Chemical Analysis of Water and Wastes.

Standard Operating Procedure

pH in Water - EPA Method 150.1

Parameter - pH

<u>Principle, Scope & Application</u> - The pH of a sample is determined electrometrically using a combination electrode. This method is applicable to drinking, surface, and saline waters, domestic and industrial wastes.

Interferences & Corrective Actions - Coatings of oily material or particulate matter can impair electrode response. The corrective action is to clean the electrode. Temperature affects pH and is corrected for by temperature compensation.

<u>Sample Size</u>, <u>Collection</u>, <u>Preservation and Handling</u> - pH should be conducted on samples in the field at the time of sampling. Laboratory measurements, include notification of the aforementioned and are performed as soon as possible on an unpreserved sample.

Apparatus -

pH electrode
pH meter - Corning Model 130
Magnetic stirrer and teflon coated stirring bar

Routine Preventative Maintenance - Refer to pH electrode instruction manual.

Reagent & Calibration Standards

Commercially purchased standard pH buffer solution at pH 4.0, 7.0 and 10.0.

<u>Calibration Procedures</u> - Calibrate at a minimum of 2 points that bracket the expected pH of the samples and are 3 pH units apart. Repeat the adjustments in successive portions of the two buffer solutions until readings are within .05 pH units of the buffer solution value.

Sample Preparation - No sample preparation is required.

Analytical Measurement/Procedure -

Equilibrate buffer solutions and samples to within 2°C of each other. If the temperature differences are > 2°C, measure the temperature of the sample and utilize the temperature compensation dial of the pH meter. Transfer approximately 100 mls of sample to 150 ml beaker on a magnetic stirrer with teflon stirring bar. Rate of stirring should minimize the air transfer rate to the sample. Record the pH. Repeat the measurement until values differ by less than .1 pH unit.

Data Treatment - The Coming Model 130 does not have a data printout. Data is recorded in the laboratory notebooks, utilizing standard practices, (i.e. Each page is dated and initialed. Entries are in ink. Corrections are made with a single line through, dated and initialed). Data is logged out on the LIMS system. A hard copy of the data is kept on file with the analysts copy of the report.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements -

An outside source standard reference material is analyzed with each batch. The standard limits are the 95% confidence interval.

<u>References</u> -EPA Methods For Chemical Analysis of Water and Wastes Method 150.1.

Method Validation Data -Refer to EPA Method 150.1

Standard Operating Procedure

Specific Conductance in Water - EPA Method 120.1

Parameter - Conductivity

<u>Principle, Scope & Application</u> - The specific conductance of a sample is measured by use of a self contained conductivity meter. Results reflect conductivity at 25°C. This method is applicable to drinking, surface, and saline waters, domestic and industrial wastes.

<u>Interferences & Corrective Actions</u> - Samples are preferably analyzed at 25°C. Automatic temperature compensation is a function of the Orion Model 160 meter if the samples are not at 25°C.

<u>Sample Size</u>, <u>Collection</u>, <u>Preservation and Handling</u> -A minimum of 100 mls is preferable. Unpreserved samples are stored at 4°C until the time of analysis. The holding time for conductivity is 28 days.

Apparatus -

Conductivity Meter - Orion Model 160

Routine Preventative Maintenance - Refer to the Orion Model 160 instruction manual.

Sample Preparation - Allow the samples to equilibrate to 25°C if possible.

Analytical Measurement/Procedure -

After immersing the conductivity cell into a sample, the Model 160 shows the conductivity OF THAT SAMPLE.

For measurements in "natural waters" (e.g. drinking water, ground water and surface water) with the 015010 conductivity cell, the 160 is already adjusted (e.g. TC=nLF. C=0.609/cm). For ultrapure water, use the 014016 conductivity cell and adjust the cell constant. For other samples (e.g., acids, alkalies and salt

solutions) adjust the temperature coefficient (TC) and in case of special conductivity cells, adjust cell constant (C).

NOTE: If either the meter or cell maximum measuring range is exceeded, "OFL" will be displayed.

NOTE: Set the reference temperature to the appropriate setting for your application: 25°C

Data Treatment -The Orion Model 160 does not have a data printout. Data is recorded in the laboratory notebooks, utilizing standard practices, (i.e. Each page is dated and initialed. Entries are in ink. Corrections are made with a single line through, dated and initialed). Data is logged out on the LIMS system. A hard copy of the data is kept on file with the analysts copy of the report.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements -

An outside source standard reference material is analyzed with each batch. The standard limits are the 95% confidence interval.

References -EPA Methods For Chemical Analysis of Water and Wastes Method 120.1.

Orion Model 160 Conductivity Meter Manual.

Method Validation Data -Refer to EPA Method 120.1

Standard Operating Procedure

Sulfate in Water - EPA SW846 9036-

Parameter - Sulfate

Range of Measurement - 2.0 - 80. ppm

Method Detection Limit - .73 ppm calculated in accordance with 40CFX part 136, Appendix B.

Sample Matrix - Water

Principle, Scope & Application - The sample is first passed through a sodium form cation-exchange column to remove multivalent metal ions. The sample containing sulfate is then reacted with an alcohol solution of barium chloride and methyl thymol blue (MTB) at a pH of 2.5-3.0 to form barium sulfate. The combined solution is raised to a pH of 12.5-13.0 so that excess barium reacts with MTB. The uncomplexed MTB color is gray; if it is all chelated with barium, the color is blue. Initially, the barium and MTB are equimolar and equivalent to 300 mg SO₄/1; thus the amount of uncomplexed MTB is equal to the sulfate present.

<u>Interferences & Corrective Actions</u> - The ion exchange column eliminates interferences from multivalent cations. A fresh column is made prior to a daily run. Samples below pH 2 are neutralized to prevent elution of cations from the ion exchange resin. Turbid samples are filtered.

Safety Precautions -

Refer to MSDS's for barium chloride, sodium hydroxide and ammonium hydroxide. Standard laboratory protective devices are to be utilized (safety glasses, lab coats and gloves).

Sample Size, Collection, Preservation and Handling - Samples are unpreserved and refrigerated at 4° C. The holding time for sulfate is 28 days.

Apparatus - Alpkem model 3590 automated colorimetric segmented flow analyzer.

Routine Preventative Maintenance -

Carefully read information on page 11 of the Alpkem Flow Solution Operation Manual.

MTB reagent has a life span of 3 days. If your run malfunctions prior to that time focus on repacking the column and cleaning the system with buffered EDTA (page 5).

Sulfate cartridge has to be cleaned often due to Barium Sulfate precipitation and if necessary, use the syringe directly on the cartridge holes to force the buffer through the system at high pressure to clean up the clogged mass.

High carryover, loss of sensitivity immediately after calibration curve is run or a clog in the cartridge (indicated by sporadic, slow or whimsical solution flow) is usually an indication that the system must be cleaned; the cation exchange resin must be replaced, or the frit+ ferrule (filter that covers the ion exchange column) must be cleaned or replaced. If the filter is dirty it can cause a clog in the flow of reagents through the system and will result in an extreme decrease in sensitivity.

Replace the cation exchange column resin, making sure there is no air bubbles trapped in the column. Note: page 12, paragraph 10. The installation of the column is done with 502 pump running.

Once weekly clean the ferrule with the frit by cleaning it in buffered EDTA cleaning solution and placing the buffer solution with the frit+ferrule in sonicator for 20 minutes.

Reagent & Calibration Standards -

- a. Stock Barium Chloride: Dissolve 1.526g of Barium Chloride (FW 244.28) in 1000 ml deionized water.
- b. Methylthymol Blue Reagent: Dissolve .188g of MTB (FW 866.73) in 25 ml of stock barium chloride solution. Add 71ml deionized water and 4ml of 1N hydrochloric acid. Dilute to 500ml with ethanol 3A, anhydrous (FW 46.07). Mix this solution vigorously for 1hr of until it is thoroughly degassed. Filter the solution through .45 micrometer membrane filter. Add 2.5ml Brij-35 (30 %). Store in amber bottle.
- c. .18N Sodium Hydroxide: Add 9ml of 10N sodium hydroxide (40g NaOH to 100ml, FW 40.0) to 80ml of deionized water contained in a 500ml volumetric flask. Bring to volume of 500ml with deionized water.

- d. Diluent: Dilute .375ml of 1000ppm sulfate standard to 1L with deionized water. Add 4 drops Brij-35 (30%).
- e. Buffer pH 10.5: Dissolve 6.75g of ammonium chloride (FW 53.5) in 800ml deionized water, add 57ml ammonium hydroxide (FW 35.05), and dilute to 1000ml with deionized water.
- f. Cleaning solution, Buffered EDTA, 4%: Dissolve 40g of EDTA in 800ml of pH 10.5 buffer contained in a 1L volumetric flask. Dilute the solution to 1000ml with buffer.

The stock calibrant is a 1000 mg/l solution from 1.4787 g sodium sulfate diluted to 1 liter with deionized water. Serial dilutions are made for standards at 2, 4, 10, 20, 40, 60 and 80 ppm.

Calibration Procedures - A calibration curve must be prepared each day for each sample run with a minimum of a calibration blank and 2, 4, 10, 20, 60 and 80 ppm standards. After calibration, the calibration curve must be verified by use of at least a calibration blank and a calibration check standard (made from a reference material or other independent standard material) at or near the midrange. The calibration reference must be measured within 10% of it's true value for the curve to be considered valid.

<u>Sample Preparation</u> - Samples are allowed to equilibrate to room temperature prior to analysis. Samples below pH 2 are neutralized. Turbid samples are filtered prior to analysis.

Analytical Measurement - SETTING UP THE RUN ON THE SAMPLER: F3 allows for correlation between software and 501 sampler. Wash/sample times should be set at 25/85 accordingly. Vial type on 501 sampler is always set at 2, stop count is always set at 001.

The analyst should consult the Enviroflow 3500 on page 17 at this time. The Detector **510** and Sampler **501** should be adjusted according to the instructions on page 17.

To set up the 502b pump turn to page 16. Hint: Distinguish the 3 waste lines first and lay them over the 510 detector. This will allow you to set up the rest of the pump lines without confusion. Always set up in this order: Wash line (green), helper line (black), sample line (#4), Air line (#3), the rest of the lines, followed by setting up the three waste lines in direction counter to the pull of the pump.

The surfactant solution is 35% Brij solution.

Turn to page 8 for Operating procedure. Read page 11 for additional troubleshooting information.

BEGIN ANALYSIS:

If the reagents and column have been installed, press ZERO button on the 510 detector to ensure your baseline is equalized. You can begin analysis.

Press Esc which returns you to the main menu screen.

Press F5 and then F3. You will check the peak of you highest standard (Sync) at this time in order to see if your range setting on the 510 detector is correct. Press START button on the 501 sampler. Observe the peak height on the computer screen and adjust range on the 510 detector accordingly. If the peak is too high, lower the range to 1.0 setting. If the peak is too low, increase the range to the 0.2 setting.

Other setting on the 510 detector should be the following: page 17

-wavelength

460

-range

0.5 AUFS / .20 AUFS

-rise time

10 sec.

The settings on the 501 sampler should be the following: page 17

-sample time

25 sec. / 20

-wash time

85 sec. / 75

-pecking

on (peck cycle button located on back of sampler 501)

If the Sync (highest standard) check is good you can start the sampler run at this time.

Press Esc once

Press Alt and #1 simultaneously, then press Enter, followed by Y

Press F3 which allows for monitoring and observing the analytical run.

Procedure

- Set up manifold.
- Prepare ion exchange column
- -Allow 30 minute warm up and pump reagents until a stable baseline is achieved.
- -Construct calibration curve
- -Analyze samples in accordance with the analytical measurement section of this document.

Data Treatment -A computer data printout is obtained from the Alpkem 3590 system. This includes calibration curves, correlation coefficients, concentrations

and associated intensities. Data is logged out on the LIMS system. A hard copy of the data is kept on file with the analysts copy of the report.

<u>Data Deliverables</u> -Standard data deliverables include sample date, date of analysis, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data dependent upon the clients request.

Quality Control Requirements - In brief, the quality control includes; 1) the following four QC samples - Preparation Blank, Duplicate, Matrix Spike and Laboratory Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or less	< PQL (Practical Quantitation Limit)
Laboratory Control Sample	1 per batch of 20 or less	80-120%
Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL
Spike	1 per batch of 20 or less	75-125%
Initial Calibration Verification	after every calibration	90-110%
Initial Calibration Blank	after every calibration	< PQL
Continuing Calibration Verification	10%	90-110%

Continuing 10% < Calibration Blanks	10% < PQL
---	-----------

References -Enviroflow 3500 Technical Information Perstorp Operators Manual.

The Flow Solution Operation Manual - Alpkem/Perstorp Analytical SoftPac (tm) Plus Manual- Perstorp Analytical

EPA SW846 Test Methods For Evaluating Solid Waste - method 9036

Method Validation Data - Refer to EPA Method 375.2 of Methods for Chemical Analysis of Water and Wastes.

Standard Operating Procedure

Total Dissolved Solids - EPA 160.1

<u>Parameter</u> - Total Dissolved Solids - (TDS)

Range of Measurement - 10 mg/1 to 20,000 mg/l

Method Detection Limit - 7.95 mg/liter. Calculated in accordance with 40 CFR part 136, Appendix B.

Sample Matrix - Water

<u>Principle, Scope & Application</u> - A well mixed sample is filtered through a 0.45 micron filter. The filtrate is evaporated and dried to a constant weight at 180° C.

This method applies to drinking, surface and saline waters, domestic and industrial wastes.

<u>Interferences & Corrective Actions</u> - Samples high in dissolved solids may cause a positive interference.

<u>Safety Precautions</u> - Standard laboratory protective devices are to be utilized (safety glasses, lab coats and gloves).

<u>Sample Size, Collection, Preservation and Handling</u> - Samples are unpreserved and refrigerated at 4° C. Analysis should begin as soon as possible and within seven days of sample collection.

Apparatus - 0.45 micron syringe filters
Disposable 60 ml polyethylene syringes
Graduated cylinder
60 ml porcelain or platinum evaporating dishes
Drying oven, 180° C ± 2° C.
Analytical balance capable of weighing .01 mg.
Desiccator

<u>Routine Preventative Maintenance</u> - Calibrate drying oven thermometer against NBS certified thermometer every six months.

Annual calibration and servicing of analytical balance by certified technician.

<u>Reagent & Calibration Standards</u> - NBS traceable thermometer. Class "S" certified traceable weights.

<u>Calibration Procedures</u> - In addition to the annual calibration of the analytical balance, it is checked weekly against a certified weight and recorded in balance log book.

<u>Sample Preparation</u> - Evaporation dishes are cleaned in hot dilute hydrochloric acid, followed by hot deionized water rinses and dried in oven at 180° C \pm 2° C for one hour. Alternatively they are dried in a Bunsen burner flame. Dishes are stored in a desiccator until needed and weighed immediately before use.

Analytical Measurement - Filter an appropriate aliquot (25 to 50 mls) through a 0.45 micron filter into a pre-weighed evaporating dish. Bring to dryness in a drying oven at 90 to 100° C. Elevate the temperature to 180° C \pm 2° C for one hour. Cool in a desiccator and weigh. Repeat drying cycle until constant weight is obtained or weight loss is less than 0.5 mg.

Procedure

- Filter sample
- -Place measured aliquot in pre-weighed evaporating dish
- -Dry sample
- -Elevate temperature to 180° C for one hour
- -Desiccate sample
- -Weigh sample

<u>Data Treatment</u> -Calculate TDS as follows: TDS, mg/l = $(A - B) \times 1,000$

where:

A=weight of dried residue + dish in mg

B=weight of dish in mg

C=volume of sample used in ml.

The Mettler AT61 analytical balance is connected to a personal computer and weights and volumes are recorded on a hardcopy printout. Data is stored in the balance room and a hard copy is also kept on file with the analysts copy of the report. Data is logged out on a LIM's system.

<u>Data Deliverables</u> -Data deliverables for this project include all raw data. Results are reported in standard CLP format including all associated FORM I's and all associated quality control data.

Quality Control Requirements -Quality control includes a preparation blank, duplicate and standard reference material. Frequency is 1 per 20 samples.

References - Methods for Chemical Analysis of Water and Wastes.

<u>Method Validation Data</u> -Refer to Methods for Chemical Analysis of Water and Wastes.

Standard Operating Procedure

Total Suspended Solids - EPA 160.2

Parameter - TSS

Range of Measurement - 1.0 - 20,000 mg/liter

Method Detection Limit - .61 mg/liter. Calculated in accordance with 40 CFR part 136, Appendix B.

Sample Matrix - Water

<u>Principle, Scope & Application</u> - A well mixed sample is filtered through a 0.45 micron filter and the residue retained on the filter is dried to constant weight at 103-105° C.

<u>Interferences & Corrective Actions</u> - Samples high in dissolved solids may cause a positive interference. Pre-washing the filter of dissolved solids will minimize this potential interference.

<u>Safety Precautions</u> - Standard laboratory protective devices are to be utilized (safety glasses, lab coats and gloves).

<u>Sample Size, Collection, Preservation and Handling</u> - Samples are unpreserved and refrigerated at 4° C. Analysis should begin as soon as possible and within seven days of sample collection.

Apparatus - Sartorius brand cellulose nitrate filter - .45 micron pore size. Filtering apparatus with reservoir and disc support.
Suction flask and vacuum pump
Drying oven
Desiccator

Analytical balance capable of weighing .01 mg.

Routine Preventative Maintenance -

Utilize and empty water trap prior to vacuum pump. Calibrate drying oven thermometer against NBS certified thermometer every six months. Annual calibration and servicing of analytical balance by certified technician.

<u>Reagent & Calibration Standards</u> - NBS traceable thermometer. Class "S" certified traceable weights.

<u>Calibration Procedures</u> - In addition to the annual calibration of the analytical balance, it is checked weekly against a certified weight and recorded in balance log book.

Sample Preparation - PREPARATION OF THE FILTER:

Place the filter in the filtration apparatus and wash with approximately 20 mls of 3% nitric acid.

Wash with three successive 80 ml volumes of deionized water. Remove all traces of water with vacuum. Remove filter and dry in oven at 103-105° C for one hour. Remove to desiccator and store until needed. Repeat drying cycle until constant weight is achieved. Always handle filters with tweezers.

<u>Analytical Measurement</u> - Selection of sample volumes. 250 mls are typically utilized for the lowest achievable limit of detection. A smaller volume may be required for sample high in particulate.

Assemble the filtering apparatus and begin suction. Wet the filter with a small volume of distilled water to seat it against fitted support.

Shake the sample vigorously and quantitatively transfer an appropriate volume to the filter using a graduated cylinder. Remove all traces of water by continuing to apply vacuum after water has passed through. With suction on, wash the graduated cylinder, filter, non-filterable residue and filter funnel wall with three portions of deionized water allowing complete drainage between washing. Remove all traces of water by continuing to apply vacuum after water has passed through. Carefully remove the filter from the filter support. Dry for one hour at 103-105° C. Cool in a dedicator and weigh. Repeat the drying cycle until a constant weight is obtained.

Procedure

- Pretreat, dry, desiccate and weigh filters
- Vacuum filter an appropriate aliquot
- -Rinse apparatus and filter
- -Dry, desiccate and re-weigh filter

Data Treatment -Calculate TSS as follows:

TSS, $mg/l = (A - B) \times 1,000$

 C

where:

A=weight of filter (or filter and crucible) + residue in mg B=weight of filter (or filter and crucible) in mg

C=ml of sample filtered

The Mettler AT61 analytical balance is connected to a PC and weights and volumes are recorded on a hardcopy printout. Data is stored in the balance room and a hard copy is also kept on file with the analysts copy of the report. Data is logged out on a LIM's system.

<u>Data Deliverables</u> -Data deliverables for this project include all raw data. Results are reported in standard CLP format including all associated FORM I's and all associated quality control data.

<u>Quality Control Requirements</u> -Quality control includes a preparation blank, duplicate and standard reference material, at a frequency of 1 per 20.

References - Methods for Chemical Analysis of Water and Wastes.

<u>Method Validation Data</u> -Refer to Methods for Chemical Analysis of Water and Wastes.

Standard Operating Procedure Total Phosphorus in Water - EPA Method 365.4

Parameter - Total Phosphorus, (TP)

Range of Measurement - .20 - 2.0 ppm

Method Detection Limit - .023 ppm calculation in accordance with 40CFX part 136, Appendix B.

Sample Matrix - Water

<u>Principle, Scope & Application</u> - This method is used for the determination of total phosphorus in drinking, surface and saline waters, domestic and industrial wastes. The applicable range of this method is 0.2 to 2.0 mgP/L. The range may be extended by changing the detector sensitivity or by sample dilution.

Prior to analysis, samples are digested to hydrolyze phosphorus to orthophosphate. The acidic digestate is neutralized and analyzed for orthophosphate.

Orthophosphate reacts with molybdenum (VI) and antimony (III) in an acid medium to form an antimony-phosphomolybdate complex. This complex is subsequently reduced with ascorbic acid to form a blue color and the absorbance is measured at 660nm.

Interferences & Corrective Actions - Ferric iron up to 50 mg/L copper, up to 10 mg/L do not interfere. Filter turbid samples prior to analysis. Samples with background absorbance at the analytical wavelength may interfere.

<u>Safety Precautions</u> - Refer to MSDS's for Ammonium Molybdate, Antimony Potassium Tartrate, Ascorbic Acid, Chloroform, Potassium Dihydrogen Phosphate, Potassium Sulfate, Red Mercuric Oxide, Sulfuric Acid, concentrated. Standard protective devices are to be utilized (safety glasses, lab coats, gloves and vent hood).

Sample Size, Collection, Preservation and Handling - Determine total phosphorus in unpreserved samples immediately after they are collected. Unpreserved samples may be held for 48 hours when cooled immediately and stored at 4°C. Samples may be preserved by the addition of 2 ml concentrated sulfuric acid per liter of sample and cooled immediately. The holding time for acid preserved samples is 28 days.

Apparatus - Alpkem model 3590 automated colorimetric segmented flow analyzer

Routine Preventative Maintenance - When starting up, let the system run on deionized water with 0.5 ml/l Dowfax 2A1 for 10-15 minutes. Add all reagents on-line except for the molybdate/antimony solution, run 5 minutes more, then add. If precipitation occurs, check the pH from the flowcell waste line, (pH should approximate 1 when checked with pH paper).

If a high background or blank values are observed, carefully evaluate the source of water, cleaning procedure for the digestion glassware, and laboratory environment.

Problem carryover can be solved by washing out entire system with a 1.0N sodium hydroxide solution for 5 minutes. For consistent results, flush system out every day before use and whenever needed. For persistent carryover problems, add a wash cup between every sample using a blank, or increase the sampler wash time.

Reagents should be filtered prior to use through a 0.45 micron membrane filter.

4% Sulfuric Acid - The sampler wash and reagent sulfuric acid solution should be made up as a single lot. When samples are digested prior to analysis, deionized water blanks should also be digested.

Reagent & Calibration Standards-

Refer to Enviroflow 3500 Technical Informant Perstorp Operators Manual, (Total Phosphorus Section-Page 3-5)

The stock calibrant is a 1000 mg/l solution from 4.394 g of dry potassium dihydrogen phosphate (FW 136.09), diluted to 1 liter with deionized water. Preserve with 2 drops of chloroform (FW 119.38). Refrigerate at 2-6°C.

Calibration Procedures - A calibration curve must be prepared for each sample run with a minimum of a digested calibration blank and .20, .40, 1.0, 1.4 and 2.0 ppm digested standards. After calibration, the calibration curve must be verified by use of at least a calibration blank and a calibration check standard (made from a reference material or other independent standard material) at or near the mid-range. The calibration reference must be measured within 10% of it's true value for the curve to be considered valid.

<u>Sample Preparation</u> - Preserved samples are digested with blanks, calibration standards, and quality control samples. Samples are filtered post digestion if turbid. Samples are digested in accordance with EPA Method 365.4.

<u>Analytical Measurement</u> - Refer to AMS document #TSC-430-L-PR-019, "Basic Operation, Maintenance and troubleshooting of the Alpkem Automated Anion Analyzer, (Perstorp Model 3590)

Procedure -

- -Set up manifold and set heater to 37°C
- -Add all reagents except molybdate/antimony
- -Obtain a stable baseline
- -Add the molybdate/antimony and obtain a stable baseline
- -Construct calibration curve
- -Analyze samples in accordance with analytical measurement section of this document.

Data Treatment - A computer data printout is obtained from the Alpkem 3590 system. This includes calibration curves, correlation coefficients, concentrations and associated intensities. Data is logged out on the LIMS system. A hard copy of the data is kept on file with the analysts copy of the report.

<u>Data Deliverables</u> - Standard data deliverables include sample date, date of analyses, method of analysis and the analysts initials. Date of sample receipt is noted in the reporting cover letter. Data deliverables may or may not include quality control summaries and/or raw data, dependent upon the clients request.

<u>Quality Control Requirements</u> - The quality control includes; 1) the following four QC samples - Preparation Blank, Duplicate, Matrix Spike and Laboratory

Control Sample and 2) the following run QC; initial and continuing calibration verifications, initial and continuing calibration blanks.

The following table summarizes quality control limits and frequency requirements.

QUALITY CONTROL SAMPLES	FREQUENCY	LIMITS
Prep Blanks	1 per batch of 20 or less	< PQL (Practical Quantitation Limit)
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Duplicate	1 per batch of 20 or less	20% RPD, if <5 X PQL
Spike	1 per batch of 20 or less	75-125%
Initial Calibration Verification	after every calibration	90-110%
Initial Calibration Blank	after every calibration	< PQL
Continuing Calibration Verification	10%	90-110%
Continuing Calibration Blanks	10%	< PQL

References - Enviroflow 3500 Technical Information Perstorp Operators Manual.

The Flow Solution Operation Manual - Alpkem/Perstorp Analytical SoftPac (tm) Plus Manual-Perstorp Analytical

EPA Method 365.1

AMS Document #TSC-430-L-PR-019, "Basic Operation, Maintenance and Troubleshooting of the Alpkem

Automated Anion Analyzer (Perstorp Model 3590).

<u>Method Validation Data</u> -Refer to EPA Method 365.4 of Methods for Chemical Analysis of Water and Wastes.

Tabbed Page: Appendix G

RESPONSE ACTION CONTRACT FOR REMEDIAL, ENFORCEMENT OVERSIGHT, AND NON-TIME CRITICAL REMOVAL ACTIVITIES AT SITES OF RELEASE OR THREATENED RELEASE OF HAZARDOUS SUBSTANCES IN EPA REGION VIII

U.S. EPA CONTRACT NO. 68-W5-0022

DRAFT

FOR RICHARDSON FLAT TAILINGS SITE SUMMIT COUNTY, UTAH

Revision 0

Work Assignment No.: 088-RSBD-0894
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RESPONSE ACTION CONTRACT FOR REMEDIAL, ENFORCEMENT OVERSIGHT, AND NON-TIME CRITICAL REMOVAL ACTIVITIES AT SITES OF RELEASE OR THREATENED RELEASE OF HAZARDOUS SUBSTANCES IN EPA REGION VIII

U.S. EPA CONTRACT NO. 68-W5-0022

DRAFT
SAMPLING AND ANALYSIS PLAN ADDENDUM
FOR
RICHARDSON FLAT TAILINGS SITE
SUMMIT COUNTY, UTAH

Work Assignment No.: 088-RSBD-0894 Document Control No.: 3280-088-PP-SAMP-10418

Prepared for:
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February 2001

PHOTOGRAPHIC DOCUMENTATION OF FIELD ACTIVITIES

SOP 4-2

Revision: 3

Date: February 1

Page 9 of 9

- Original video tape footage shall be assigned a document control number. Video tape: have a document control number written on the tape case as well as the video tape cassett document control numbers will be written with black indelible ink. Original video footag not be edited but may be copied.
- 4. Completed pages of the appropriate logbook will be copied weekly and submitted to the pages.

6.0 RESTRICTIONS/LIMITATIONS

This document is designed to provide a set of guidelines for the field amateur or profes photographer to ensure that an effective and standardized program of visual documentat maintained.

It is not within the scope of this document to provide instruction in photographic procediwithin the scope of this document to set guidelines for presentation or "show" photography.

The procedures outlined herein are general by nature. The FTL is responsible for directly sq operational activity or procedure. Questions concerning specific procedures or requirements s be directed to the FTL.

7.0 REFERENCES

Sandia National Laboratories, Division 7723. Field Operating Procedure for Field Logbook C. and Control, PRO 92-04, Environmental Restoration Department, Albuquerque, New Mexico.

U.S. Department of Energy, Yucca Mountain Project. Branch Technical Procedure: Field Log Handling, and Documenting Borehole Samples, Las Vegas, Nevada, 1989.

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Acronyms

°C degree Celsius

CAR corrective action request

CDM Federal CDM Federal Programs Corporation

DQOs data quality objectives

EPA United States Environmental Protection Agency

GPS global positioning survey LCSS laboratory control samples

MSS matrix spikes

PARCC precision, accuracy, representativeness, completeness, and

comparability

PM Project Manager
QA quality assurance

QAPP quality assurance project plan

QC quality control

QMP quality management plan
RAC response action contract
Richardson Flat Richardson Flat Tailings Site

RI/FS remedial investigation/feasibility study
RMC Resource Management Consultants

RPDS relative percent differences RPM Remedial Project Manager

SAP Sampling and Analysis Plan, Remedial Investigation at

Richardson Flat

SOPS standard operating procedures
United Park United Park City Mines Company

List of Acronyms

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Section 1 Introduction

This document is an addendum to United Park City Mines Company's (United Park) "Sampling and Analysis Plan, Remedial Investigation at Richardson Flat," (SAP) prepared for the United States Environmental Protection Agency (EPA) Work Assignment No. 088-RSBD-0894 under the EPA Response Action Contract (RAC) No. 68-E5-0022. This addendum addresses field oversight activities CDM Federal Programs Corporation (CDM Federal) will provide during the focused remedial investigation and feasibility study (focused RI/FS) conducted by United Park at the Richardson Flat Tailings site near Park City, Utah. Activities during the oversight will include collecting and shipping split samples as well as validating and evaluating the analytical results of those samples. The intent of this addendum is to supplement the SAP, in so, describing those duties to be conducted by CDM Federal. This addendum is following the same format as the United Park's SAP and it will be referenced where applicable. The following sections are addressed in this addendum:

Section 1 - Introduction

Section 2 - Project Management

Section 3 - Measurement/Data Acquisition

Section 4 - Assessment/Oversight

Section 5 - Data Validation and Usability

Appendices

Appendix A - CDM Federal's applicable standard operating procedures (SOPs)

1.1 Project Objectives

CDM Federal's objective of this project is to provide oversight and documentation of field activities conducted by United Park as part of the focused RI/FS. Oversight activities will ensure that United Park performs field investigations according to the approved site work plan, SAP, SOPs, and quality assurance project plan (QAPP).

1.2 Project Schedule and Deliverables

Sampling in this addendum refers to five sampling events that will take place starting in 2001 in conjunction with United Parks SAP activities. The first event is scheduled to be performed during the spring of 2001. The remaining four events are scheduled to take place once every quarter following the spring sampling event. Deliverables for this project will include a technical oversight report following each sampling event. Also, one data validation and usability report, and one data evaluation technical memorandum will be provided following the last sampling event. All reports will be

submitted as described in CDM Federal's, "Original Work Plan for the RI/FS Oversight, Richard Flat Tailings Site, Park City, Utah, 2000."

Section 2 Project Management

All project personnel and management will conduct activities in accordance with the SAP and SAP addendum including CDM Federal's following additions.

2.1 Project Organization

CDM Federal will provide sampling oversight, including handling and shipment of split samples. CDM Federal also will document field activities conducted by both United Park and CDM Federal personnel. EPA's regional laboratory will provide analytical services.

2.1.1 Management Organization

The CDM Federal Project Manager (PM) for the Richardson Flat site is Mr. Jeff Montera. For this field investigation oversight, Mr. Montera is responsible for the following activities:

- Maintaining communications with EPA regarding the status of this project
- Supervising production and review of deliverables
- Preparing monthly status reports
- Reviewing analytical results
- Tracking work progress against planned budgets and schedules
- Incorporating and informing EPA of changes in the CDM Federal work plan, SAP addendum, and/or other CDM Federal project documents
- Notifying the CDM Federal Region VIII RAC Quality Assurance (QA) Manager, Regional QA Coordinator, or local QA Coordinator immediately of significant problems that affect the quality of data or the ability to meet project objectives
- Scheduling CDM Federal personnel and material resources
- Implementing the quality control (QC) measures specified in CDM Federal's QAPP (CDM Federal 1996a) for this contract, quality management plan (QMP) (CDM Federal 1996b) for this contract, this addendum, and other project documents
- Implementing corrective actions resulting from staff observations, QA/QC surveillance, and/or QA audits
- Providing oversight of data management

2.1.2 Quality Assurance Organization

The QA program is implemented by CDM Federal's Region VIII RAC QA Manager, Ms. RoseMary Gustin. She is independent of the technical staff and reports directly to the President of CDM Federal on QA matters. The QA Manager has the authority to objectively review projects and identify problems and the authority to use corporate resources, as necessary, to resolve any quality-related problems.

The Denver QA Coordinator for this project, Ms. Krista Lippoldt, and the RAC Region VIII Regional QA Coordinator, Mr. George Delullo, report to Ms. Gustin on all QA matters. Under Ms. Gustin's oversight they are responsible for the following:

- Reviewing and approving the project specific plans
- Directing the overall project QA program
- Maintaining QA oversight of the project
- Reviewing QA sections in project reports, as applicable
- Reviewing QA/QC procedures applicable to this project
- Auditing selected activities of this project, as necessary
- Initiating, reviewing, and following-up on response actions, as necessary
- Consulting with the CDM Federal QA Manager, as needed, on appropriate QA/QC measures and corrective actions
- Provideing monthly written reports on QA/QC activity to the CDM Federal QA Manager

2.2 Problem Definition/Background

Site background information for the Richardson Flat site is provided in Section 2.2 of the SAP (United Park 2001).

2.3 Project/Task Description

This addendum addresses the fieldwork that will be performed by CDM Federal for the RI/FS oversight at Richardson Flat. Soil, sediment, groundwater, surface water, and tailings will be collected from locations (including reference locations) on and around the site. All CDM Federal samples will be analyzed for all parameters listed in United Park's Table 2, per media. CDM Federal will not analyze only a percentage of samples for some of the parameters but will instead analyze each sample for all the parameters for that media.

1

2.4 Data Quality Objectives for Measurement Data

The data quality objectives (DQO) process is a series of planning steps based on the scientific method that are designed to ensure that the type, quantity, and quality of environmental data used in decision-making are appropriate for the intended purpose. EPA has issued guidelines to help data users develop site-specific DQOs (EPA 2000). The DQO process is intended to:

- Clarify the study objective.
- Define the most appropriate type of data to collect.
- Determine the most appropriate conditions from which to collect the data.
- Specify acceptable levels of decision errors that will be used as the basis for establishing the quantity and quality of data needed to support the design.

The goal of the DQO process is to "help assure that data of sufficient quality are obtained to support remedial response decisions, reduce overall costs of data sampling and analysis activities, and accelerate project planning and implementation (EPA 2000)."

The DQO process specifies project decisions, the data quality required to support those decisions, specific data types needed, data collection requirements, and analytical techniques necessary to generate the specified data quality. The process also ensures that the resources required to generate the data are justified. The DQO process consists of seven steps of which the output from each step influences the choices that will be made later in the process. These steps include:

- Step 1: State the problem.
- Step 2: Identify the decision.
- **Step 3: Identify the inputs** to the decision.
- Step 4: Define the study boundaries.
- Step 5: Develop a decision rule.
- **Step 6: Specify tolerable limits** on decision errors.
- Step 7: Optimize the design.

During the first six steps of the process, the planning team develops decision performance criteria (i.e., DQOs) that will be used to develop the data collection design. The final step of the process involves developing the data collection design based on the DQOs. A brief discussion of these steps and their application to this SAP addendum is provided below.

2.4.1 Step 1: State the Problem

The purpose of this step is to clearly define the problem so that the focus of the study will be unambiguous. The objective of this study is to ensure that the data collected by United Park during the focused RI/FS is representative of current site conditions.

2.4.2 Step 2: Identify the Decision

This step identifies what questions the study will attempt to resolve and what actions may result.

Primary Question:

Do the data that United Park collects during the focused RI/FS field investigation accurately represent current site conditions?

The following resolutions to the question and possible actions have been identified.

- The data United Park collects during the focused RI/FS field investigation accurately represent current site conditions.
 - o Alternative 1: No further action is conducted, unless data gaps become apparent.
- Any of the data United Park collects during their focused RI/FS field investigation do not accurately represent current site conditions.
 - o Alternative 1: United Park is tasked by EPA to re-conduct either deficient sections of or the entire field investigation.
 - o Alternative 2: EPA takes the lead role on Richardson Flats and conducts their own field investigation.
 - o Alternative 3: The dispute resolution process provided for in the Administrative Order on consent for the focused RI/FS is enacted.
 - o Alternative 4: No further action is conducted, unless data gaps become apparent.

2.4.3 Step 3: Identify the Inputs to the Decision

The purpose of this step is to identify the information that needs to be obtained and the measurements that need to be taken to resolve the decision statement. The information required to resolve the decision statement consists of the following:

 Concentrations of analytes listed in Table 2 (United Park 2001) in samples of surface water, groundwater, sediment, background soil, depth of cover soil, and test pit tailings collected by both CDM Federal and United Park.

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2.4.4 Step 4: Define the Boundaries of the Study

This step defines the spatial and temporal boundaries of this study. The horizontal spatial boundaries are identified in Figure 2.0 of United Park's work plan (United Park 2000). The vertical spatial boundaries are from the natural ground surface to 40 feet below ground surface. Temporal boundaries are from the time tailings were first placed on site (prior to 1950) to the present day.

2.4.5 Step 5: Develop a Decision Rule

The purpose of this step is to define the parameters of interest, specify the action level, and integrate previous DQO outputs into a single statement that describes a logical basis for choosing among the alternative actions.

For this investigation the actual concentration values of parameters listed in Table 2 (United Park 2001) will be used to make decisions. These concentrations in split samples of surface water, groundwater, sediment, soil, and tailings will be compared to concentrations of the same analytes in the original samples collected by United Park. The relative percent differences (RPDs) will be calculated if both results are positive and > 5 reporting limit. The action level for the RPDs are <50 percent for soil and <35 percent for water. If either result is < 5 reporting limits, then the action level is < 4 reporting limits for soil and < 2 reporting limits for water. If either concentration is non-detect, then there are no evaluation criteria. The methods for these analyses are listed in Section 3.4.2 and have been selected with the understanding of the DQOs. If all of the results for the split and original samples meet the action level criteria, then the original samples are considered representative of site conditions. If any of the results for the split and original samples do not meet the action level criteria, then the original samples may not be representative of site conditions.

2.4.6 Step 6: Specify Tolerable Limits on Decision Errors

Decision maker's tolerable limits on decision errors, which are used to establish performance goals for the data collection design, are specified in this step. Decision makers are interested in knowing the true value of the constituent concentrations. Since analytical data can only estimate these values, decisions that are based on measurement data could be in error (decision error). There are two reasons why the decision maker may not know the true value of the constituent concentration, these are:

- (1) Concentrations may vary over time and space. Limited sampling may miss some features of this natural variation because it is usually impossible or impractical to measure every point of a population. Sampling design error occurs when the sampling design is unable to capture the complete extent of natural variability that exists in the true state of the environment.
- (2) Analytical methods and instruments are never absolutely perfect, hence a measurement can only estimate the true value of an environmental sample. Measurement error refers to a combination of random and

systematic errors that inevitably arise during the various steps to the measurement process.

The combination of sampling design and measurement error is the total study error. Since it is impossible to completely eliminate total study error, basing decisions on sample concentrations may lead to a decision error. The probability of decision error is controlled by adopting a scientific approach in which the data are used to select between one condition (the null hypothesis) and another (the alternative hypothesis). The null hypothesis is presumed to be true (not rejected) in the absence of evidence to the contrary. The null and alternative hypotheses are as follows:

Null Hypothesis:

The data collected by United Park accurately represent site conditions.

Alternate Hypothesis:

The data collected by United Park do not accurately represent site conditions.

A false positive or "Type I" decision error refers to the type of error made when the null hypothesis is rejected when it is true. A false negative or "Type II" decision error refers to the type of error made when the null hypothesis is not rejected when it is false. In actual practice, it is impossible, using hypothesis-testing methods, to prove that the null hypothesis is true. Thus, in particular, if we accept the null, then actually it is more correct to say that we fail to reject the null hypothesis. For this project a Type I error would result in deciding that the United Park data do not accurately represent the site when they actually do. A Type II would result in deciding that the United Park data accurately represent the site, when they actually do not.

The closer the reported concentration is to the action level, the higher the probability that an incorrect decision will be made and, therefore, there is a "gray region" surrounding the action level. For this project no "gray region" has been established. However, tolerable decision limits have been established to allow decision-makers to use professional judgment when necessary. For a water matrix, the tolerable decision limit is \pm 10 percent (i.e., 25 percent – 45 percent) for RPDs and \pm 1 reporting limit for results < 5 reporting limits. For a soil matrix the tolerable decision limit is \pm 10 percent (i.e., 40 percent – 60 percent) for RPDs and \pm 1 reporting limits for results < 5 reporting limits.

2.4.7 Step 7: Optimize the Design for Obtaining Data

This step identifies a resource-effective data collection design for generating data that are expected to satisfy the DQOs. The sampling program described in this addendum is consistent with the DQOs and project objectives for the assessment. However, if during the period of sample collection, it becomes apparent that the quantity and/or distribution of samples is not sufficient for obtaining the data required to properly characterize water, soil, sediment, or tailings for this assessment, the number,

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distribution, or methods may be modified to reflect the specific needs of the project. EPA will approve any changes to this addendum prior to implementation. In addition any deviations from this addendum will be noted in the applicable field logbook and subsequent data summary reports.

2.5 Special Training Requirements/Certification

All training requirements are the same as the SAP with the following addition:

Region VIII laboratory chain-of-custody training

2.6 Documentation and Records

All documentation and records will be maintained in accordance with the SAP with the following additions.

CDM Federal's local administrative staff in Denver has the responsibility for maintenance of the document control system. This system includes a document inventory procedure and a filing system. Each document will be clearly identified (e.g., by title, contract number, project number, and/or date). RAC VIII requirements stipulate that each document also will be given an alphanumeric identifier number (i.e., a document control number) and entered in an index of project documents. Project personnel are responsible for project documents in their possession while working on a particular task. The contents and control of field logbook(s) will be conducted in accordance with CDM Federal SOP 4-1, Field Logbook Content and Control (Appendix A). Photographic documentation of field activities and the control of those photographs will be conducted in accordance with CDM Federal SOP 4-2, Photographic Documentation of Field Activities (Appendix A). Documentation describing changes to approved plans, if they occur, will be included in the logbooks and/or document control system.

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Section 3

Measurement/Data Acquisition

This section covers sample process design, sampling method requirements, handling and custody, analytical methods, QC, equipment maintenance, instrument calibration, supply acceptance, nondirect measurement, and data management.

3.1 Sample Process Design

The general goal of the field investigation oversight is to confirm that the work performed by United Park is in accordance with the applicable approved documents and therefore, representative of the site. Sediments, surface water, groundwater, and four different types of soils will be collected during this field investigation. These soil sample types include on-site cover soils, off-site cover soils, tailings, and reference soils.

CDM Federal will collect split samples at a rate of 20 percent of United Park's original samples with a minimum of one split per sampling media. In addition, CDM Federal will collect a duplicate sample at a rate of 20 percent of the split samples with a minimum of one duplicate per sampling media. For surface water and groundwater, five containers/aliquots will be collected for each split sample and each duplicate sample. For all other media, one container/aliquot will be collected for each split sample and each duplicate sample. Duplicate aliquots will be collected immediately following split sample aliquots. The number, type, location, and aliquot sequence of split samples and their respective duplicates collected by CDM Federal are presented in Table 3-1.

An example from Table 3-1 is surface water. For surface water it is estimated that United Park will collect one sample from each of 10 different sampling locations (i.e., a total of 10 samples). CDM Federal will collect a total of two surface water split samples. The locations of those split samples, determined by random number generation, are sample location number 2 and sample location number 3. At each split sample location, 10 aliquots will be collected (i.e., five for CDM Federal, and five for United Park). For the first split sample, collected at sample location 2, the five aliquots will be collected from the following sequence of aliquots: 2,3,4,6, and 9. Hence, United Park will keep the first aliquot, CDM the second, third, and fourth, United Park the fifth, and so on. For the second split sample, collected at sample location 3, the five aliquots will be collected from the following sequence of aliquots: 2,3,6,8, and 9. In addition, one duplicate sample will be collected. The location of that duplicate, chosen by random number generation, is sample location 3. Five aliquots will be collected immediately following the split sample aliquots.

3.1.1 Surface Water

Approximately 15 surface water samples will be collected on or near the site (United Park 2001, Figure 4), over a period of five sampling events (i.e., three per event). Actual sample locations will be logged with a global positioning survey (GPS) unit or

located with conventional survey methods. The number, location, and aliquot sequence of split samples and their respective duplicates collected for CDM Federal are presented in Table 3-1.

3.1.2 Groundwater

Approximately 15 groundwater samples will be collected on or near the site (United Park 2001, Figure 4), over a period of five sampling events (i.e., three per event). United Park will install two monitoring wells in Silver Creek's shallow alluvial aquifer (United Park 2001). Ground water will be sampled from these two locations, upon completion, in addition to up to two monitoring wells installed in the tailings south of the diversion ditch. Actual sample locations will be logged with a GPS unit or located with conventional survey methods. The number, location, and aliquot sequence of split samples and their respective duplicates collected for CDM Federal are presented in Table 3-1.

3.1.3 Soils/Sediments

3.1.3.1 Depth of Cover Sampling

Approximately 12 soil samples will be collected on the tailings impoundment during one sampling event (United Park 2001, Figure 5). Actual sample locations will be logged with a GPS unit or located with conventional survey methods. The number, location, and aliquot sequence of split samples and their respective duplicates collected for CDM Federal are presented in Table 3-1.

3.1.3.2 Off-Site Soils Cover Sampling

Approximately 14 soil samples will be collected along three transects, oriented perpendicular to the prevailing wind direction (United Park 2001, Figure 6). This sampling will occur during one sampling event. Actual sample locations will be logged with a GPS unit or located with conventional survey methods. The number, location, and aliquot sequence of split samples and their respective duplicates collected for CDM Federal are presented in Table 3-1.

3.1.3.3 Test Pit Tailings

Approximately six test pit tailings samples will be collected from locations within the impoundment on the site (United Park 2001, Figure 5). This sampling will occur during one sampling event. Actual sample locations will be logged with a GPS unit or located with conventional survey methods. The number, location, and aliquot sequence of split samples and their respective duplicates collected for CDM Federal are presented in Table 3-1.

3.1.3.4 Reference Soil Samples

Approximately seven reference soil samples will be collected from areas that most likely have not been affected by tailings deposition (United Park 2001, Figure 2). This sampling will occur during one sampling event. Actual sample locations will be logged with a GPS unit or located with conventional survey methods. The number,

location, and aliquot sequence of split samples and their respective duplicates collected for CDM Federal are presented in Table 3-1.

3.1.3.5 Sediment

Approximately three sediment samples will be collected in the south diversion ditch (United Park 2001, Figure 5). This sampling will occur during one sampling event. Actual sample locations will be logged with a GPS unit or located with conventional survey methods. The number, location, and aliquot sequence of split samples and their respective duplicates collected for CDM Federal are presented in Table 3-1.

3.2 Sampling Method Requirements

3.2.1 Surface Water

United Park field personnel will collect and preserve surface water split samples in accordance with Resource Management Consultants (RMC) SOP 1 for Surface Water Sampling (United Park 2001, Appendix C). Split samples will be collected using CDM Federal containers and submitted to CDM Federal personnel by United Park. Split samples, and their duplicates will be handled in accordance with Section 3.3.2.

3.2.2 Groundwater

United Park field personnel will collect and preserve groundwater split samples in accordance with RMC SOP 3c (United Park 2001, Appendix C). Split samples will be collected using CDM Federal containers and submitted to CDM Federal personnel by United Park. Split samples, and their duplicates will be handled in accordance with Section 3.3.2.

3.2.3 Soils/Sediments

3.2.3.1 On-site Soils Cover Sampling

United Park field personnel will collect on-site soil cover split samples in accordance with RMC's SOP 2a (United Park 2001, Appendix C). Split samples will be collected using CDM Federal containers and submitted to CDM Federal personnel by United Park. Split samples, and their duplicates will be handled in accordance with Section 3.3.2.

3.2.3.2 Off-site Soils Cover Sampling

United Park field personnel will collect off-site soil split samples in accordance with RMC's SOP 2a, 2b, and/or 2c (United Park 2001, Appendix C). Split samples will be collected using CDM Federal containers and submitted to CDM Federal personnel by United Park. Split samples, and their duplicates will be handled in accordance with Section 3.3.2.

3.2.3.3 Tailings

United Park field personnel will collect tailings split samples in accordance with RMC's SOP 2b and/or 2c (United Park 2001, Appendix C). Split samples will be collected using CDM Federal containers and submitted to CDM Federal personnel by

United Park. Split samples, and their duplicates will be handled in accordance with Section 3.3.2.

3.2.3.4 Reference Soil Samples

United Park field personnel will collect reference soil split samples in accordance with RMC's SOP 2a (United Park 2001, Appendix C). Split samples will be collected using CDM Federal containers and submitted to CDM Federal personnel by United Park. Split samples, and their duplicates will be handled in accordance with Section 3.3.2.

3.2.3.5 Sediment

United Park field personnel will collect sediment split samples in accordance with RMC's SOP 4 (United Park 2001, Appendix C). Split samples will be collected using CDM Federal containers and submitted to CDM Federal personnel by United Park. Split samples, and their duplicates will be handled in accordance with Section 3.3.2.

3.3 Sample Labeling, Handling, Custody, and Documentation

3.3.1 Labeling

An alpha-number coding system will uniquely identify each sample collected during the field investigation.

In addition to the field logbooks, the CDM Federal sampling team will use labels to track and identify samples, ensuring sample traceability. A sample label will be secured to each sample container at the time of collection. Sample numbers will begin with a two-digit identification of the year (i.e., 01 for 2001). The second character set defines which sampling event the sample was collected (i.e., E1 for event 1, E2 for event 2, etc.). The third character set of the sample identification will identify the media from which the sample is collected:

Surface Water	=	SW
Groundwater	=	GW
Sediment	=	SD
On-site Soils Cover	=	ON
Off-site Soils Cover	=	OF
Test Pit Tailings	=	TA
Reference Soil	=	RF

The fourth character set represents the sample location identification number. Sample location numbers for each media will be three digits starting at 001, and will begin with 3 if the sample is a duplicate, or a 5 if it is a rinsate blank (i.e., 301 is the duplicate

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sample for split sample location 001). Each media will begin with sample number 001 and that location number will identify that location which will be the same for sampling event (i.e., 001-E1-OF and 001-E2-OF is the same off-site soil location). The fifth character set represents end depths for soil, sediment, and tailings samples and will be 0 for water samples. End depths will be in inches for all media except test pit tailing, which will be in feet. The following are examples of sample numbers:

01-E1-OF-301-6

01 = year 2001

E1 = first sampling event

OF = Soil sample from an off-site soil cover

301 = Duplicate of sample from location 001

6 = 6 inch sample ending depth

01-E1-TA-001-3

01 = 2001

E1 = first sampling event

TA = Test pit tailings

001 = Sample from location 001

3 = 3 foot sample ending depth

Labels will be used in accordance with CDM Federal SOP 1-2, Sample Custody (Appendix A).

3.3.2 Sample Handling

Samples will be packaged and shipped in accordance with either CDM Federal SOP 2-3, Packaging and Shipping of Sulfuric Acid-Preserved Samples; SOP 2-4, Packaging and Shipping of Nitric Acid-Preserved Samples; and/or SOP 2-8, Packaging and Shipping of Environmental Samples (Appendix A), whichever is applicable. These split samples will be placed in a plastic bag and then in a cooler with ice and kept at 4° C (degrees Celsius) ± 2° C. All samples will be picked up by a courier, delivered to the laboratory, or shipped by an overnight delivery service to the designated laboratory.

3.3.3 Chain-of-Custody

Chain-of-custody procedures will follow the requirements stated in CDM Federal's SOP 1-2, Sample Custody included in Appendix A. The chain-of-custody record is

employed as physical evidence of sample custody and control. This record system provides the means to identify, track, and monitor each individual sample from the point of collection through final data reporting.

3.3.4 Documentation

The Field Team Leader will maintain field logbooks in accordance with CDM Federal's SOP 4-1, Field Logbook Content and Control (Appendix A). The log is an accounting of detailed activities at the site and will note problems or deviations from the governing plans and observations relating to the sampling and analysis program. The contents and control of field logbook(s) will be in accordance with CDM Federal SOP 4-1. Logbooks will be filed in Richardson Flats project files in the Denver office.

All deviations from the guiding documents will be recorded in the applicable field logbook. Any major deviations will be documented in accordance with CDM Federal's QMP (CDM Federal 1996b). Any modifications to the chain-of-custody forms will be made on all copies. The EPA Remedial Project Manager (RPM) will be notified of any major changes or deviations.

3.4 Analytical Method Requirements

The laboratory QA program and analytical methods are addressed below.

3.4.1 Laboratory Quality Assurance Program

Samples collected during this project will be analyzed in accordance with standard EPA and/or nationally recognized analytical procedures. The purpose of using standard procedures is to provide analytical data of known quality and consistency. Analytical laboratories will adhere to QC requirements as established by EPA methods.

3.4.2 Methods

The methods, holding times, and preservatives to be used for chemical analysis are located in Table 2 of United Park's SAP (United Park 2001). All of CDM Federal's split samples, split sample duplicates, and rinsate blanks will be analyzed for all of their respective parameters listed in Table 2.

3.5 Quality Control Requirements

Field duplicates are one of two types of QC samples that will be collected. A field duplicate will be collected with its parent sample and collected identically and consecutively over a minimum period of time. This type of duplicate measures the total system variability (field and laboratory variance), including the variability component resulting from the inherent heterogeneity of the sample media. Field duplicates are planned for a rate of 10 percent of the split samples, with a minimum of one, for each sampling type.

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The second type of QC sample to be collected and analyzed are rinsate blanks. If sampling equipment is decontaminated in the field, an equipment rinsate blank will be prepared and submitted for analysis at a minimum frequency of one per 20 split samples per media/sampling technique (5 percent) with a minimum of one rinsate blank per media. These rinsate blanks will consist of analyte-free water poured over the equipment used to collect the sample after equipment decontamination. The sample identification code will relate to the sample collected prior to blank collection.

3.5.1 Laboratory Quality Control

Analytical services will be provided through the EPA Region VIII Sample Broker by an EPA approved laboratory. The laboratory chosen will follow all laboratory QC checks, which may include laboratory duplicates, matrix spikes (MSS), laboratory control samples (LCSS), and/or laboratory blanks (i.e., preparation blanks).

3.5.1.1 Laboratory Internal Quality Control Samples

QC data are necessary to determine precision and accuracy and to demonstrate the absence of interferences and/or contamination. Each type of laboratory based QC sample will be analyzed at a rate of 5 percent of parent samples collected or once per batch (a batch is defined as a group of up to 20 samples analyzed together), whichever is more frequent. Results of QC analysis will be included in the QC package and QC samples may consist of laboratory duplicates, laboratory blanks, MSS, and LCSS, whichever is applicable, and any other method-required QC samples.

Laboratory blank samples will be analyzed to assess possible contamination so that corrective measures may be taken, if necessary. Laboratory duplicate samples are aliquots of a single sample that are split on arrival at the laboratory or upon analysis. Results obtained for two replicates that are split in a controlled laboratory environment are used to assess laboratory precision of the analysis. MS and LCS analyses may be used to determine accuracy.

3.5.1.2 Laboratory Quality Control Checks

The laboratory will perform the QC checks required by each analytical method.

3.5.2 Internal Quality Control Checks

Internal QC checks will be conducted throughout the project to evaluate the performance of the sampling team during data generation. All internal QC will be conducted in accordance with the applicable procedures listed below:

- All project deliverables will receive technical and QA reviews, where applicable, prior to being issued to EPA in any form.
- Completed review forms will be maintained in the project files.
- Corrective action of any deficiencies will be the responsibility of the CDM Federal Project Manager (PM), with assistance from the QA staff, if necessary.

3.6 Data Management

Sample results and QC data from the laboratories will be submitted to CDM Federal as both hard-copy and electronic data packages. Electronic copies of all project deliverables, including graphics, will be maintained by project number. Electronic files will be routinely backed up and archived.

CDM Federal's local administrative staff has the responsibility for maintaining the document control system. This system includes a document inventory procedure and a filing system. Project personnel are responsible for project documents in their possession while working on a particular task.

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	Estimated Number						Sample	
	70		Number of			Number	Location(s)	Omplicate
	United Park (UP)	Estimated Number	CDM Federal Solit Samples	Sample Location(s)	Split Sample	COM Endered Dublicates	of CDM	Alignot
Matirx	Sample Locations	Of HP Samoias	(20% of HD Samples)	COM Goderni Colli Camalani	5			
200	9		(2014)		Anduot Sequence	(soldwies mids to way)	Cupilcate	Sednence
Sunace water	OL .	10	2	2,3	[2,3,4,6,9], [2,3,6,8,9]	-	£	12.3.4.6.91
Ground Water	10	10	2	2.4	(4.5.6.7.8) (1.4.6.7.9)		2	114679
Sediment	0,	9	2	1.4	12	1		61,121,11
Beckground Soil	8	8	9	345171922	212122			•
Depth of Cover Soll****	950	SS	[8]. [2]	[1,5,6,7,8,9,16,25] [3, 8]	11.2.1.1.1.1.2.21.12.11	2	12, 63	121 [2]
							1.1.2.	5
Official Soil	۶	30 et 0-2"	6 84 0-2"	[4,6,11,16,19,22] et 0-2"	[2,2,1,1,2,2] at 0-2"		Location 4 at 0-2"	Location 3 at 0-2"
	3	30 at 1.6"	6 84 1-6"	[2,3,9,16,20,24] at 1.6"	[1,2,1,1,1,1] at 1-6"	2 - 1 al each depth	Location 20 et 1- 6"	
		5#0.1	1210.1		At location 4 4 2			
		5.041-2	12.1.2		Al Incalion 1 - 2.3			
Tailing Test Pit	u)	5612-3	1 21 2 3	4,1,2,3,5	At location 2 - 3-4"	-	-	۲,
		583-4	1834		At location 3 - 0-1"		•	
		584.5	1814-5		At location 5- 4-5			

COM Federal Sampling Event 2

	Estimated Number						Sample	
	5		Number of			Number of	Location(s)	Duplicate
	United Park (UP)	Estimated Number	CDM Federal Split Samples	Sample Location(s)	Split Sample	CDM Federal Duolicates	of CDM	Aliquot
Matirx	Sample Locations	of UP Samples	(20% of UP Samples)	of CDM Federal Solit Samples Allowot Seguence	Allanot Segmence ²	(20% of Solli Samples)	Continue	.,
Curdon Mator	•						Dupincate	Sectionics
Stillace Water	2	0.	2	9,	1.2.4.5.71 11.5.6.7.81	_	-	1246891
Ground Water	\$	10	2	27	1135671 1345691	-		0000
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CDM Federal Sampling Event 3

					THE REAL PROPERTY AND ADDRESS OF THE PERSON NAMED IN			
	Estimated Number						Sample	
	50		Number of			Number of	Location(s)	Duplicate
	United Park (UP)	Estimated Number	CDM Federal Split Samples	Sample Location(s)	Split Sample	CDM Federal Duolicates	of CDM	Aliquot
Matirx	Sample Locations	of UP Samples	(20% of UP Samples)	of CDM Federal Split Samples	Alignot Seguence	(20% of Split Samples)	Dunlicate	Secuence 4::
Surface Water	10	ů,	2	4.6		, , , , , , , , , , , , , , , , , , , ,		Salaman
				P	0.6.4.0.1 1.0.0.0.4.0	-	þ	8,9,6,8,1
Ground water	10	10	2	3,5	[4,5,6,7,8], [1,4,6,7,9]	-	3	(13569)
			VIII					

CDM Federal Sampling Event 4

	Estimated Number						Sample	
	ō		Number of			Number of	Location(s)	Duplicate
	United Park (UP)	Estimated Number	CDM Federal Split Samples	Sample Location(s)	Split Sample	CDM Federal Duniticates	of CDM	Aliquot
Matirx	Sample Locations	of UP Samples	(20% of UP Samples)	of COM Federal Split Samples	Alignot Sequence	(20% of Sulit Samples)	Oundinated	Connected
Surface Water	10	٤			ĺ,	Condition with the world	Capitale	Coductico
		2	7	4,0	3.4 6.8 9 [2.6.7,8.9]	_	80	[2,3,6,7,8]
Ground Water	10	10	2	6.7	11.2.3.5.61 [2.3.5.8.9]		7	1123561

CDM Federal Sampling Event 5

-									
		Estimated Number						Sample	
		50		Numberof			Number of	Location(s)	Duplicate
		United Park (UP)	Estimated Number	CDM Federal Split Samples	Sample Location(s)	Split Sample	CDM Federal Dunicates	of CDM	Allanot
	Matirx	Sample Locations	of UP Samples		of COM Federal Solit Samutes Atlanta Segments		Communications (1906)	0	
10	Andreas Interes			l	Saiding mid minor in the said	ı	(so w or spill samples)	Duplicate	Sednence
<u></u>	Illace water	01	2	7	4,6	[23457] [2457]	-	4	1146781
Č	Total Major	•						-	2
3	COULT VARIET	2	2	~	- 47	(23589) (12456)	_		1567801

1. Which, of the total number of samples locations chosen, will be used for CDM Federal split samples

Which, of the total number of aliquotes to be taken at each sample location, will be used for CDM Federal split samplesWhich, of the CDM Federal split sample location(s), will be used for duplicate sampling

4. Which, of the total number of aliquotes to be collected at each sample location, will be used for CDM Federal duplicate sample(s)

· Five total samples with a minimum of one at each depth, and one at each location

"All CDM duplicate samples will be taken consecutively following CDM Federalspilt samples
"Sample locations and aliquot sequences were determined using a random number generator
""Sample locations and aliquot sequences were determined using a random number generator
""Spproximately forry depth of cover samples will be taken at the surface (0-2"). All samples will be analyzed for Pb, and As, and 20% or all the samples will be analyzed for RCRA metals.

CDM Federal will take splits for 20% (i.e., 2) of these samples.

Section 4

Assessment/Oversight

Assessment and oversight reports to management are necessary to ensure that procedures are followed as required and that deviations from procedures are documented. These reports also serve to keep management current on field activities.

4.1 Assessments and Response Actions

Assessments and corresponding response actions are discussed below.

4.1.1 Assessments

Performance assessments are quantitative checks on the quality of a measurement system and are appropriate to analytical work. Performance assessments for the laboratory may be accomplished by submitting reference material as blind reference (or performance evaluation) samples. These assessment samples are samples with known concentrations that are submitted to the laboratory without informing the laboratory of the known concentration. Samples will be provided to the laboratory for performance assessment upon request from the EPA RPM. Laboratory audits may also be conducted upon request from the EPA RPM.

System assessments (field and office) are qualitative reviews of different aspects of project work to check on the use of appropriate QC measures and the functioning of the QA system. Any determination or changes for project assessments will be performed under the direction of the QA Manager, who reports directly to the CDM Federal President. Quality Procedure 6.2, as defined in the CDM Federal RAC Region VIII QMP (CDM Federal 1996b), defines CDM Federal's corporate assessments, procedures, and requirements. Due to the length of this work assignment (greater than one year)n one office audit will be conducted. Because the length of field work is less than five weeks total, no field audit is scheduled at this time. If the scope or length of any task changes, the audit requirements may be re-evaluated.

4.1.2 Response Actions

Response actions will be implemented on a case-by-case basis to correct quality problems. Minor response actions taken in the field to immediately correct a quality problem will be documented in the applicable field logbook, and a verbal report will be provided to the CDM Federal PM. For verbal reports, the CDM Federal Project Manager will complete a communication log to document that response actions were relayed to him. Major response actions taken in the field will be approved by the CDM Federal Project Manager and the EPA RPM prior to implementation of the change. Major response actions are those that may affect the quality or objective of the investigation. Quality problems that cannot be corrected quickly through routine procedures will require completion of a Corrective Action Request (CAR) Form (Figure 4-1).

All formal response actions will be submitted to either CDM Federal's RAC Region VIII QA Manager or RAC Regional QA Coordinator for review and issuance. CDM Federal's PM or local QA Coordinator will notify the QA Manager or Regional QA Coordinator when quality problems arise that may require a formal response action. CAR forms will be completed according to Quality Procedure 8.1 of the CDM Federal RAC Region VIII QMP (CDM Federal 1996b).

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FIGURE 4-1 CORRECTIVE ACTION REQUEST FORM

Project:
Contract/Project No.:
Project Manager:
Description of problem and date identified
Requested by:
Response Date:
Submit this form to the QA Manager or Regional QA Coordinator promptly.
Significance Condition Adverse to Quality? Yes/No
Responsible for Action:
Response Due:
Submit completed response to:
[To be completed by the responsible person. Attach additional pages as required. Include evidence that corrective action has been implemented.]
State cause of problem (if known or suspected):
Corrective Action(s) Taken to Correct Problem and Prevent Recurrence:
Signature:
Date:
Corrective Action Plan Accepted:
Date:
CA Verified By:
Date:
Corrective Action Accepted:
Date:

4.2 Reports to Management

QA reports will be provided to management whenever quality problems are encountered. Field staff will note any quality problems on field data sheets. CDM Federal's PM will inform the local QA Coordinator upon encountering quality issues that cannot be immediately corrected. Weekly reports and Change Request Forms are not required for this work assignment. Monthly QA reports will be submitted to CDM Federal's QA Manager by the local QA Coordinator and the RAC Regional QA Coordinator.

Topics to be summarized regularly may include, but not be limited to:

- Document technical and QA reviews that have been conducted
- Activities and general program status
- Project meetings
- Calibration data
- Unscheduled maintenance activities
- Corrective action activities
- Any unresolved problem
- Assessment of data deficiencies
- Any significant QA/QC problems not included above

P13280-RAC8\088\SAP Addendum\Sap Addendum Section 4-Spin

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Section 5 Data Validation and Usability

Laboratory results will be reviewed for compliance with project objectives. Data validation and evaluation are discussed in Sections 5.1 and 5.2, respectively.

5.1 Data Validation

CDM Federal will validate 100 percent of the analytical data for compliance with Organic and Inorganic National Functional Guidelines for Data Review (EPA 1994a, EPA 1994b) as discussed in United Park's SAP, Section 5.0. CDM Federal will qualify data according to the Organic and Inorganic National Functional Guidelines for Data Review (EPA 1994a, EPA 1994b). A data validation report will be prepared as required by the original work plan, different than as discussed in United Park's SAP, Section 5.

5.2 Data Evaluation

CDM Federal will evaluate 100 percent of the analytical data for compliance with precision, accuracy, representativeness, completeness, and comparability (PARCC) parameter criteria as discussed in United Park's SAP, Section 5.0. CDM Federal will qualify data according to the Organic and Inorganic National Functional Guidelines for Data Review (EPA 1994a, EPA 1994b), different than as discussed in United Park's SAP, Section 5. A data validation report will be prepared as required by the original work plan.

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Section 6 References

- CDM Federal Programs Corporation. 1996a. RAC Region VIII Quality Assurance Project Plan. August.
- CDM Federal Programs Corporation. 1996b. RAC Region VIII Quality Management Plan. August.
- United Park City Mines Company. 2001. Draft Sampling and Analysis Plan, Remedial Investigation Richardson Flat. Prepared by Environmental Resource Management Consultants. January.
- United Park City Mines Company. 2000. Statement of Work for Focused Remedial Investigation/Feasibility Study, Richardson Flat Tailings Site, Summit County, Utah. UT980952840.
- U.S. Environmental Protection Agency. 2000. Guidance for the Data Quality Objectives Process. EPA QA/G-4. August
- U.S. Environmental Protection Agency. 1994a. The National Functional Guidelines for Inorganic Data Review. February, with current revisions (Inorganic Guidelines).
- U.S. Environmental Protection Agency. 1994b. The National Functional Guidelines for Organic Data Review. February, with current revisions (Organic Guidelines).

Appendix A CDM Federal Standard Operating Procedures

SOP 1-2

Revision: 2

Date: December 18, 2000

Page 1 of 8

Prepared: Sharon Budney

Technical Review: Daw

Approved:

Signature/Date

OA Review: George DeLullo

Signature/Date

1.0 **OBJECTIVE**

Due to the evidentiary nature of samples collected during environmental investigations, possession must be traceable from the time the samples are collected until their derived data are introduced as evidence in legal proceedings. To maintain and document sample possession, sample custody procedures are followed. All paperwork associated with the sample custody procedures will be retained in CDM Federal Programs Corporation (CDM Federal) files unless the client requests that it be transferred to them for use in legal proceedings or at the completion of the contract.

2.0 BACKGROUND

2.1 **Definitions**

Sample - A material to be analyzed that is contained in single or multiple containers representing a unique sample identification number.

Sample Custody - A sample is under custody if:

- 1. It is in your possession.
- 2. It is in your view, after being in your possession.
- 3. It was in your possession and you locked it up.
- 4. It is in a designated secure area.

Chain-of-Custody Record - Form used to document the transfer of custody of samples from one individual to another.

Custody Seal - A custody seal is a tape-like seal that is part of the chain-of-custody process and is used to detect tampering with samples after they have been packed for shipping.

Sample Label - Adhesive label placed on sample containers to designate a sample identification number and other sampling information.

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Sample Tag - Tag attached with string to a sample container to designate a sample identification number and other sampling information. Tags may be used when it is difficult to physically place adhesive labels on the container (e.g., in the case of small air sampling tubes).

3.0 RESPONSIBILITIES

Sampler - The sampler is personally responsible for the care and custody of the samples collected until they are properly transferred or dispatched.

Field Team Leader - The Field Team Leader is responsible for ensuring that strict chain-of-custody procedures are maintained during all sampling events. The Field Team Leader is also responsible for coordinating with the subcontractor laboratory to ensure that adequate information is recorded on custody records.

4.0 REQUIRED SUPPLIES

- Chain-of-Custody Records (applicable CDM Federal forms)
- Custody seals
- Sample labels or tags
- Clear Tape

5.0 PROCEDURES

5.1 Chain-of-Custody Record

This procedure establishes a method for maintaining custody of samples through use of a Chain-of-Custody Record. This procedure will be followed for all samples collected or split samples accepted.

Field Custody

- 1. Collect only the number of samples needed to represent the media being sampled. To the extent possible, determine the quantity and types of samples and sample locations prior to the actual fieldwork. As few people as possible should handle samples.
- 2. The field sampler is personally responsible for the care and custody of the samples collected until they are properly transferred or dispatched.
- 3. Sample labels or tags shall be completed for each sample, using waterproof ink.

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4. The Field Team Leader determines whether proper custody procedures were followed during the fieldwork and decides if additional samples are required.

Transfer of Custody and Shipment

- 1. Samples are accompanied by a Chain-of-Custody Record (see Figure 1 for an example of a Chain-of-Custody Record). When transferring the possession of samples, the individuals relinquishing and receiving will sign, date, and note the time on the record. This record documents sample custody transfer from the sampler, often through another person, to the analyst in the appropriate laboratory.
 - The date/time will be the same for both signatures when custody is transferred directly to another person. When samples are shipped via common carrier (e.g., Federal Express), the date/time will not be the same for both signatures. Common carriers are not required to sign the form.
 - In all cases, it must be readily apparent that the person who received custody is the same person who relinquished custody to the next custodian.
 - If samples are left unattended or a person refuses to sign, this must be documented and explained on the Chain-of-Custody Record.
- 2. Samples will be packaged properly for shipment and dispatched to the appropriate laboratory for analysis, with a separate custody record accompanying each shipment.
- 3. All shipments will be accompanied by the Chain-of-Custody Record identifying its contents. The original record will accompany the shipment, and the copies will be retained by the Field Team Leader and if applicable, distributed to the appropriate sample coordinators. Freight bills will also be retained by the Field Team Leader as part of the permanent documentation. (Refer to Figure 1)

Procedure for Completing CDM Federal Chain-of-Custody Record (Refer to Figure 1.)

The following procedure is to be used to fill out the CDM Federal Chain-of Custody Record provided as an example chain-of-custody form. If another type of custody form (i.e., provided by the EPA Contract Laboratory Program or a subcontract laboratory) is utilized to track the custody of samples, the information recorded should be similar to the example CDM Federal Chain-of-Custody Record provided and the form should be filled out in its entirety.

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- 1. Record project number.
- 2. Record Field Team Leader for the project.
- 3. Record the name and address of the laboratory to which samples are being shipped.
- 4. Enter the project name/location or code number.
- 5. Record overnight courier's airbill number.
- 6. Record sample location number.
- 7. Record laboratory sample number.
- 8. Note preservatives type and reference number.
- 9. Note media type (matrix) and reference number
- 10. Note sample type.
- 11. Enter date of sample collection.
- 12. Enter time of sample collection in military time.
- 13. Samplers must enter their initials next to the samples they collected.
- 14. List parameters for analysis and the number of containers submitted for each analysis.
- 15. Enter MS/MSD (matrix spike/matrix spike duplicate) if sample is for laboratory quality control, or other remarks (e.g. sample depth).
- 16. All samplers must sign in the space provided.

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Figure 1 EXAMPLE CDM Federal Chain-of-Custody Record

CDM Federal Programs Corporation

125 Maiden Lane, 5th Fioor New York, NY 10038 (212) 785-9123 Fax: (212) 785-6114

CHAIN OF CUSTODY RECORD

PROJECT ID.	FI	ELD TE	AM LEAC	ER			BORATORY					ŀ	DATE	SHIPP	ED
PROJECT NAME/LO	CATION					1	ID ADDRES B CONTRAI						AIRBIL	⊥ NO.	
MEDIA TYPE	PI	RESER	VATIVES		SAI		TYPE	1	i	i	\neg		1 1		·
1. Surface Water	1.	HCI, p	H <2			Grai									
2. Groundwater			pH-<2		C=	Com	nposite		l	1 1	Į.	l			
3. Leachate	3.	NaOH.	pH >12				•	G. 0			i		1 1		
4. Field QC			4, pH <2					호를				-	1 1		
5. Soil/Sediment			cetate, ph	1.>9				프를	i	1 1	- {	ļ	1 1		
6. Oli		. Ice On						S			-				
7. Waste			eserved					SE			- 1	1			
8. Other	8.	. Other						ANALYSES (List no. of containers submitted)		1 1			1		
SAMPLE LOCATION	SAM	PLE	PRESER- VATIVES	MEDIA	SAMPLE TYPE	19_ DAT	TIME	1 - 1		-	Ì				MARKS
NO.	NUM	BER	ADDED											(Note	if MS/MS
1.						_	· ·								
2	<u> </u>			<u> </u>									 		
3.									.			!			
4,				ļ			!					1			
5															
6							i					1			
7															
8.										- 1			!		
9:							<u> </u>					<u> </u>	1		
10.															
SAMPLER SIGNATURES:															·
RELINQUISHED BY:	DATE/TIME	PRICE	VED BY:		DATE		RELINCUISHE	ED BY:		DATE/TIME	RECE	VED BY	·		DATE/TIM
36 44		(BIOPA)				[(BIGH)				(80 8)	-			
REUNQUISHED BY:	CATETIME	RECE	VED BY:		OATE/T		RELINOUISHE	D 6Y:		DATE/TIME	RECE	VED BY	:		DATE/TIM
(SECRE)		400			7	Ī	(SIGN)				(SIGN)				
COMMENTS:															

DISTRIBUTION: White and yellow copies addentipanty.earning shipment to leboratory; yellow copy retained by laboratory. Pink copy retained by samplers.

1/98

NOTE: If requested by the client, different Chain-of Custody records may be used. Copies of the template for this record may be obtained from the Fairfax Graphics Department.

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17. The originator checks information entered in items 1 through 16 and then signs the top left "Relinquished by" box, prints his/her name, and enters the current date and time (military).

- Upon completion of the custody record form, the top two copies (usually white and yellow) shall be sent with the samples to the laboratory; the bottom copy (usually pink) is retained for the project files. Additional copies will be retained for the project file or distributed as required to the appropriate sample coordinators.
- 18. The laboratory sample custodian receiving the samples checks the sample label information against the custody record form. He or she also checks sample condition and notes anything unusual under "Remarks" on the custody record form. The laboratory custodian receiving custody signs in the adjacent "Received by" box and keeps the pink copy. The white copy is returned to CDM Federal.

5.2 Sample Labels and Tags

Sample labels or tags will be utilized for all samples collected or accepted for CDM Federal projects.

- 1. Place adhesive labels directly on the sample containers. Place clear tape over the label to protect from moisture.
- 2. Sample tags will be securely attached to the sample bottle. On 80 oz. amber bottles, the tag string may be looped through the ring style handle and tied. On all other containers, it is recommended that the string be looped around the neck of the bottle, then twisted and re-looped around the neck until the slack in the string is removed.
- 3. One label or tag will be completed for each sample container collected. Each label or tag will be completed as follows (see Figure 2 for example of sample tag); labels are completed with the equivalent information:
 - Record the Project Code (i.e., project or task number).
 - Enter the Station Number if applicable.
 - Record the date to indicate the month, day, and year of sample collection.
 - Enter the time (military) of sample collection.
 - Place a check to indicate composite or grab sample.
 - Record the sample location.
 - Samplers must sign in the space provided.
 - Place a check next to "yes" or "no" to indicate if a preservative was added.
 - Under "analyses," place a check next to the parameters for which the sample is to be analyzed. If the desired analysis is not listed, write it in the empty slot. Note: Do not write in the box for "laboratory sample number."
 - Under "remarks," add additional, relevant information.

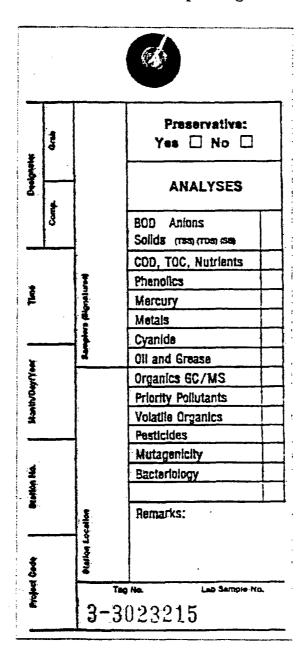
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Figure 2 EXAMPLE Sample Tag



NOTE: Equivalent sample labels or tags may be used.

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5.3 Custody Seals

Custody seals must be placed on the shipping containers prior to shipment. The seal should be signed and dated by a field team member.

Custody seals may also be placed on individual sample bottles. Check with the client or refer to EPA regional guidelines for direction.

5.4 Sample Shipping

CDM Federal's Standard Operating Procedures listed below define the requirements for packaging and shipping environmental samples.

- CDM Federal SOP 2-1, Packaging and Shipping of Hexane Preserved Samples
- CDM Federal SOP 2-2, Packaging and Shipping of Hydrochloric Acid Preserved Samples
- CDM Federal SOP 2-3, Packaging and Shipping of Sulfuric Acid Preserved Samples
- CDM Federal SOP 2-4, Packaging and Shipping of Nitric Acid Preserved Samples
- CDM Federal SOP 2-5, Packaging and Shipping of Sodium Hydroxide Preserved Samples
- CDM Federal SOP 2-6, Packaging and Shipping of Radioactive Materials in Limited Quantities

6.0 RESTRICTIONS/LIMITATIONS

For EPA Contract Laboratory Program (CLP) sampling events, combined chain-of-custody/traffic report forms or other EPA-specific records may be utilized. Refer to regional guidelines for completing these forms.

7.0 REFERENCES

U.S. Environmental Protection Agency, A Compendium of Superfund Field Operations Methods, EPA/540/P-87/001, December 1987.

U.S. Environmental Protection Agency, Sampiers Guide to the Contract Laboratory Program, EPA/540/P-90/006, December 1990.

CDM Federal Programs Corporation

SOP 2-3
Revision: 0

Revision: 0

Date: December 18, 2000

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Prepared: Arris Brownfield

Technical Review: _

Chuck Myets

QA Review: George Delullo

Approved:

Signature Date

Issued:

en Bustin 12/1/06

1.0 OBJECTIVE

This SOP provides guidance for the shipment of soil and water environmental samples regulated under the Department of Transportation (DOT) Hazardous Materials Regulations and the International Air Transportation Association/International Civil Aviation Organization (IATA/ICAO) Dangerous Goods Regulations for shipment by air. This SOP applies only to domestic shipments.

2.0 BACKGROUND

2.1 Definitions

<u>Environmental Sample</u> – An environmental sample is any sample that has less than reportable quantities for any hazardous constituents according to DOT regulations (49 CFR Part 172).

<u>Inner Packaging</u> – The container that the sample is shipped in (i.e., plastic overpackaging if liquid sample is collected in glass).

Exempted Quantity - The amount of hazardous material that does not fall under DOT/IATA/ICAO regulations. This exemption is very difficult to meet; most shipments will be made under Limited Quantity.

<u>Limited Quantity</u> - The maximum amount of a hazardous material for which there is a specific labeling or packaging exception.

<u>Performance Testing</u> - The required testing of outer packaging. These tests include the drop and stacking test.

<u>Qualified Shipper</u> – A person who has been adequately trained to perform the functions of shipping hazardous materials.

2.2 Transportation

This SOP was prepared for the shipment of sulfuric acid (H₂SO₄) preserved samples.

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Revision: 0

Date: December 13, 2000

: 1

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2.3 Containers

The inner packaging container (and amount of preservative) that may be used for these shipments includes:

Exempted quantities of preservatives

Preservative			in final	Quantity of preservative (ml) for specified container						
		рH	Conc.	40 ml	125 ml	250 ml	500 mi	l L		
H₂SO₄	37N	<2	0.35%	.1	.25	0.5	1	2.1		

5 drops = 1 ml

3.0 RESPONSIBILITY

It is the responsibility of the qualified shipper to determine the amount of preservative in each sample so that accurate determination of quantities can be made.

4.0 REQUIRED EQUIPMENT

- Outer packaging (for limited quantities) insulated cooler that has passed the performance test
- Garbage bags
- Clear waterproof tape
- Duct tape
- Strapping tape (optional)
- Ziploc®-type bags, large and small
- Vermiculite
- Bubble wrap
- Ice
- Chain-of-custody seals
- Chain-of-custody forms
- Survey documentation [if shipping from Department of Energy (DOE) or radiological sites]
- Class 8 corrosive labels
- Orientation labels
- Consignor/consignee labels

5.0 PACKAGING

Samples containing H_2SO_4 as a preservative that exceed the exempted concentration of 0.35% will be shipped as a limited quantity per packing instruction Y809 of the LATA/ICAO Dangerous Goods Regulations.

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The following steps are to be followed when packaging limited quantity samples shipments.

- Tape any interior opening in the cooler (drain plug) from the inside to ensure control of interior contents. Also tape the drain plug from the outside of the cooler.
- All sample containers will be properly labeled and the label protected with waterproof tape prior to sampling.
- At a minimum the label must contain:
 - Project name
 - Project number
 - Date and time of sample collection
 - Sample location
 - Sample identification number
 - Collector's initials
 - Preservative (note amount of preservative used in comments section of the chain-of-custody form)
- Wrap each glass container in bubble wrap (secure with waterproof tape) to prevent breakage.
- Place the bubble wrapped container into a 2.7 mil Ziploc®-type bag, once again removing trapped air.
- Place glass containers inside a polyethylene bottle filled with vermiculite. Seal the bottle.
- Place sufficient amount of vermiculite in the bottom of the cooler to absorb any leakage that may occur.
- Place a garbage bag in the cooler.
- Pack the samples appropriately inside the garbage bag (bottles placed upright) to prevent movement during shipment.
- Place a sufficient amount of double-bagged ice around the samples to maintain the required temperature during shipment.
- Seal the garbage bag by tying or taping.
- The maximum weight of the cooler shall not exceed 30 kg (66 lbs) for any limited quantity shipment of dangerous goods.
- Secure the chain-of-custody form (placed inside a Ziploc®-type bag) to the interior of the cooler lid.
- Wrap strapping tape or duct tape around both ends of the cooler and around the cooler lid.
- Affix custody seals to opposite sides of the cooler lid. Cover the custody seals with clear waterproof tape.
- Mark the outside of the cooler with the proper shipping name of the contents, corresponding UN number, and LTD. QTY. (as shown below).

SULFURIC ACID SOLUTION UN2796 LTD. QTY.

• Place a label with the company name, contact name, phone number, full street address, and state with zip code for both shipper and recipient to the front of the cooler.

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Affix a Corrosive label to the outside of the cooler.

Affix package orientation labels on two opposite sides of the cooler.

Secure the marking and labels to the surface of the cooler with clear waterproof tape to prevent their accidental removal during shipment.

An example of the cooler labeling/marking is shown in Fig. 1.

NOTE: Samples meeting the exemption concentration of 0.35% H₂SO₄ by weight will

be shipped as non-regulated or non-hazardous.

NOTE: No marking or labeling can be obscured by strapping or duct tape.

The inner packaging of dangerous goods may be placed into the designated NOTE:

cooler for shipment. Other non-regulated environmental samples may be added

to the cooler for shipment.

When shipping from a DOE facility the cooler will be surveyed by a qualified radiation control technician to ensure the exterior surfaces do not exceed 0.5 mrem/h on all sides. This survey will be documented and the results reviewed by the Qualified Shipper. A copy of the results of the radiation screen and cooler/sample survey will be placed with the chain-ofcustody form in the cooler.

- Complete the Dangerous Goods and Hazardous Materials Inspection Checklist for Shipping Limited Quantity (Appendix A).
- Complete a Dangerous Goods airbill.

Sample Packaging

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Date: December 18, 2000

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APPENDIX A Dangerous Goods and Hazardous Materials Inspection Checklist for Shipping Limited Quantity

Yes	No	N/A	
0	0	<u> </u>	The sample bottles are wrapped in bubble wrap and placed inside a Ziploc®-type bag. The sample bottles are placed into a polyethylene bottle, filled with vermiculite, and tightly sealed.
0	0	O · · ·	The drain plug is taped inside and outside to ensure control of interior contents. The samples have been placed inside garbage bags with sufficient bags of ice to
0	0 0	0	preserve samples at 4°C. The cooler does exceed the 66-pound limit for limited quantity shipment. The garbage bag has been sealed with tape (or tied) to prevent movement during shipment.
0	0	0 0 0	The chain-of-custody form has been secured to the interior of the cooler lid. The cooler lid and sides have been taped to ensure a seal. The custody seals have been placed on both the front and back hinge of the cooler, using clear waterproof tape.
<u>Air V</u>	Vavbill	Completi	<u>ion</u>
Yes	No []	N/A []	Section 1 has the Shippers name, the company & address, the account #, the date, the
٥	0	0	internal billing reference #, and the phone # where the Shipper can be reached. Section 2 has the recipient's name and company along with a phone number where they can be reached.
	0	0	Section 3 has the Bill Sender box checked. Section 4 has the Standard Overnight box checked.
0	0	0	Section 5 has the Deliver Weekday box checked. Section 6 has the number of packages and their weights filled out. Was the total of all packages & their weights figured up and added at the bottom of Section 6?
0	0	0	Under the Transport Details box, the Cargo Aircraft Only box is obliterated leaving only the Passenger and Cargo Aircraft box.
	0	0	Under the Shipment Type, the Radioactive box is obliterated leaving only the Non-Radioactive box.
0	0	0	Under the Nature and Quantity of Dangerous Goods box, the Proper Shipping Name, Class or Division, UN or ID No., Packing Group, Subsidiary Risk, Quantity And Type of Packing, Packing Instructions and Authorization has been filled out for the type of chemical being sent.
	0	0	The Name, Place & Date, Signature, and Emergency Telephone # appears at the bottom of the FedEx airbill.
0	0	0	The statement "In accordance with IATA/ICAO" appears in the Additional Handling Information box.

SOP 2-3 Revision: 0

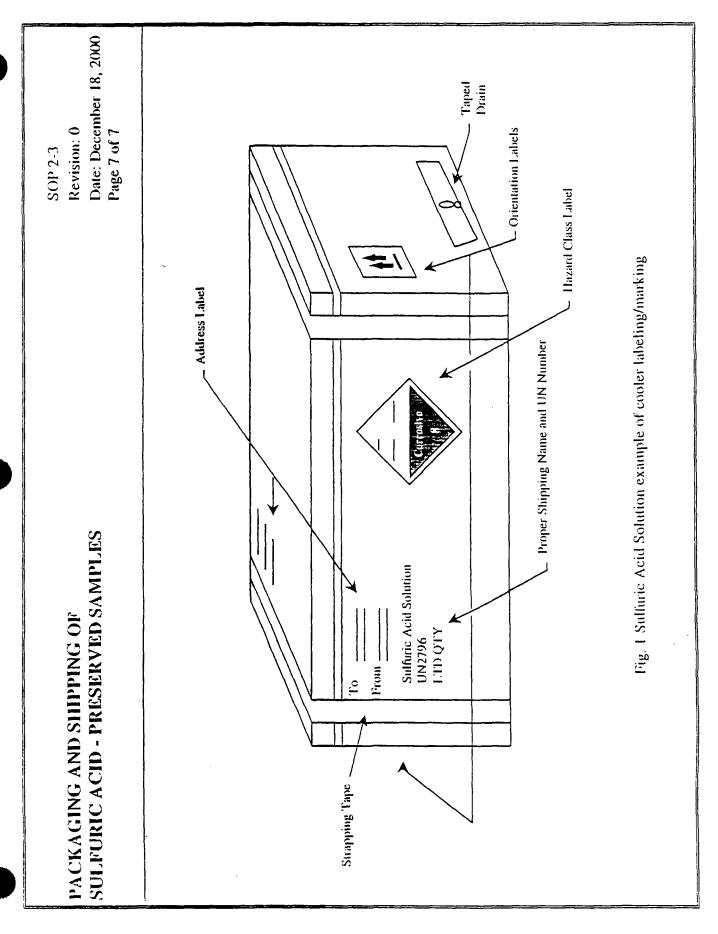
Date: December 18, 2000

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Proper shipping name	Class or Division	UN or ID No.	Packing Group	Sub Risk	Quantity	Packing Instruction	Authorization
Hydrochloric Acid Solution	3	UN1789	п		! plastic box × 0.5 L	Y809	LTD QTY
Nitric Acid Solution (with less than 20%)	8	UN2031	п		1 plastic box × 0.5 L	Y807	LTD QTY
Sodium Hydroxide Solution	8	UN1824	п		1 piastic box × 0.5 L	Y809	LID QTY
Sulfuric Acid Solution	8	UN2796	п		1 plastic box × 0.5 L	Y809	LTD QTY
Hexanes	3	UN1208	П		1 piastic box × ! L	Y305	LTD QTY

Sample Cooler Labeling

Yes	No	N/A	
Ū		0	The proper shipping name, UN number, and LTD. QTY, appears on the shipping container.
0		0	The corresponding hazard labels are affixed on the shipping container; the labels are not obscured by tape.
	0	0	The name and address of the shipper and receiver appears on the top and side of the shipping container.
			The Air Waybill is attached to the top of the shipping container.
0	0	0	Up Arrows have been attached to opposite sides of the shipping container.
U	П	0	Packaging tape does not obscure markings or labeling.



SOP 2-4 Revision: 0

Date: December 18, 2000

Page: 1 of 7

Prepared: Arris Brownfield

OA Review: George Delullo

Issued: Lase Mary Bustin 12/00 Signature/Date Technical Raview: Chuck My

Approved.

Signature/Date

1.0 OBJECTIVE

This SOP provides guidance for the shipment of soil and water environmental samples regulated under the Department of Transportation (DOT) Hazardous Materials Regulations and the International Air Transportation Association/International Civil Aviation Organization (IATA/ICAO) Dangerous Goods Regulations for shipment by air. This SOP applies only to domestic shipments.

2.0 BACKGROUND

2.1 Definitions

<u>Environmental Sample</u> – An environmental sample is any sample that has less than reportable quantities for any hazardous constituents according to DOT regulations (49 CFR Part 172).

<u>Inner Packaging</u> – The container that the sample is shipped in (i.e., plastic overpackaging if liquid sample is collected in glass).

Exempted Quantity - The amount of hazardous material that does not fall under DOT/IATA/ICAO regulations. This exemption is very difficult to meet; most shipments will be made under Limited Quantity.

<u>Limited Quantity</u> – The maximum amount of a hazardous material for which there is a specific labeling or packaging exception.

<u>Performance Testing</u> - The required testing of outer packaging. These tests include the drop and stacking test.

<u>Qualified Shipper</u> – A person who has been adequately trained to perform the functions of shipping hazardous materials.

2.2 Transportation

This SOP was prepared for the shipment of nitric acid (HNO₃) preserved samples.

SOP 2-4

Revision: 0

Date: December 18, 2000

. 1

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2.3 Containers

The inner packaging container (and amount of preservative) that may be used for these shipments includes:

Exempted quantities of preservatives

Preservative		1	d in final	Quantity of preservative (ml) for specified container						
		pH	Conc.	40 mi	125 ml	250 ml	500 ml	1 L		
HNO₃	6N	<2	0.15%	N/A	2	4	5	8		

5 drops = 1 ml

3.0 RESPONSIBILITY

It is the responsibility of the Qualified Shipper to determine the amount of preservative in each sample so that accurate determination of quantities can be made.

4.0 REQUIRED EQUIPMENT

- Outer packaging (for limited quantities) insulated cooler that has passed the performance test
- Garbage bags
- Clear waterproof tape
- Duct tape
- Strapping tape (optional)
- Ziploc®-type bags, large and small
- Vermiculite
- Bubble wrap (optional)
- Ice
- Chain-of-custody seals
- Chain-of-custody forms
- Survey documentation [if shipping from Department of Energy (DOE) or radiological sites]
- Class 8 corrosive labels
- Orientation labels
- Consignor/consignee labels

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5.0 PACKAGING

Samples containing HNO₃ as a preservative that exceed the exempted concentration of 0.15% HNO₃ will be shipped as a limited quantity per packing instruction Y807 of the IATA/ICAO Dangerous Goods Regulations.

The following steps are to be followed when packaging limited quantity samples shipments.

- Tape any interior opening in the cooler (drain plug) from the inside to ensure control of interior contents. Also tape the drain plug from the outside of the cooler.
- All sample containers will be properly labeled and the label protected with waterproof tape prior to sampling.
- At a minimum the label must contain:
 - Project name
 - Project number
 - Date and time of sample collection
 - Sample location
 - Sample identification number
 - Collector's initials
 - Preservative (note amount of preservative used in comments section of the chain-of-custody form)
- This step is optional. Wrap each container in bubble wrap (secure with waterproof tape) to prevent breakage.
- Place the bubble wrapped container into a 2.7 mil Ziploc®-type bag, removing trapped air.
- Place glass containers inside a polyethylene bottle filled with vermiculite. Seal the bottle.
- Place sufficient amount of vermiculite in the bottom of the cooler to absorb any leakage that may occur.
- Place a garbage bag in the cooler.
- Pack the samples appropriately inside the garbage bag (bottles placed upright) to prevent movement during shipment.
- Place a sufficient amount of double-bagged ice around the samples to maintain the required temperature during shipment.
- Seal the garbage bag by tying or taping.
- The maximum weight of the cooler shall not exceed 30 kg (66 lbs) for any limited quantity shipment of dangerous goods.
- Secure the chain-of-custody form (placed inside a Ziploc®-type bag) to the interior of the cooler lid.
- Wrap strapping tape or duct tape around both ends of the cooler and around the cooler lid.
- Affix custody seals to opposite sides of the cooler lid. Cover the custody seals with clear waterproof tape.

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• Mark the outside of the cooler with the proper shipping name of the contents, corresponding UN number, and LTD. QTY. (as shown below).

NITRIC ACID SOLUTION (with less than 20%) UN2031 LTD. OTY.

- Place a label with the company name, contact name, phone number, full street address, and state with zip code for both shipper and recipient to the front of the cooler.
- Affix a Corrosive label to the outside of the cooler.
- Affix package orientation labels on two opposite sides of the cooler.
- Secure the marking and labels to the surface of the cooler with clear waterproof tape to prevent their accidental removal during shipment.
- An example of the cooler labeling/marking is shown in Fig. 1.
 - **NOTE**: Samples meeting the exemption concentration of 0.15% HNO₃ by weight will be shipped as non-regulated or non-hazardous.
 - NOTE: No marking or labeling can be obscured by strapping or duct tape.
 - **NOTE:** The inner packaging of dangerous goods may be placed into the designated cooler for shipment. Other non-regulated environmental samples may be added to the cooler for shipment.
- When shipping from a DOE facility the cooler will be surveyed by a qualified radiation control technician to ensure the exterior surfaces do not exceed 0.5 mrem/h on all sides. This survey will be documented and the results reviewed by the Qualified Shipper. A copy of the results of the radiation screen and cooler/sample survey will be placed with the chain-of custody from in the cooler.
- Complete the Dangerous Goods and Hazardous Materials Inspection Checklist for Shipping Limited Quantity (Appendix A).
- Complete a Dangerous Goods airbill.

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APPENDIX A Dangerous Goods and Hazardous Materials Inspection Checklist for Shipping Limited Quantity

Yes	No	N/A	
0		0	The sample bottles are wrapped in bubble wrap and placed inside a Ziploc®-type bag.
	0	0	The sample bottles are placed into a polyethylene bottle, filled with vermiculite, and tightly sealed.
<u>Π</u>	0	0	The drain plug is taped inside and outside to ensure control of interior contents.
0	0	0	The samples have been placed inside garbage bags with sufficient bags of ice to preserve samples at 4°C.
0	0	0	The cooler does exceed the 66-pound limit for limited quantity shipment.
0	0	. 0	The garbage bag has been sealed with tape (or tied) to prevent movement during shipment.
	Ū	0	The chain-of-custody form has been secured to the interior of the cooler lid.
0 0	0	0	The cooler lid and sides have been taped to ensure a seal.
0	0	0	The custody seals have been placed on both the front and back hinge of the cooler, using clear waterproof tape.

Air Wavbill Completion

Sample Packaging

Section 1 has the Shippers name, the company & address, the account #, the date, the internal billing reference #, and the phone # where the Shipper can be reached. Section 2 has the recipient's name and company along with a phone number where they can be reached. Section 3 has the Bill Sender box checked. Section 4 has the Standard Overnight box checked. Section 5 has the Deliver Weekday box checked. Section 6 has the number of packages and their weights filled out. Was the total of all packages & their weights figured up and added at the bottom of Section 6? Under the Transport Details box, the Cargo Aircraft Only box is obliterated leaving only the Passenger and Cargo Aircraft box. Under the Shipment Type, the Radioactive box is obliterated leaving only the Non-Radioactive box. Under the Nature and Quantity of Dangerous Goods box, the Proper Shipping Name, Class or Division, UN or ID No., Packing Group, Subsidiary Risk, Quantity And Type of Packing, Packing Instructions and Authorization has been filled out for the type of chemical being sent. The Name, Place & Date, Signature, and Emergency Telephone # appears at the bottom of the FedEx airbill. The statement "In accordance with LATA/ICAO" appears in the Additional Handling Information box		Yes	No	N/A	
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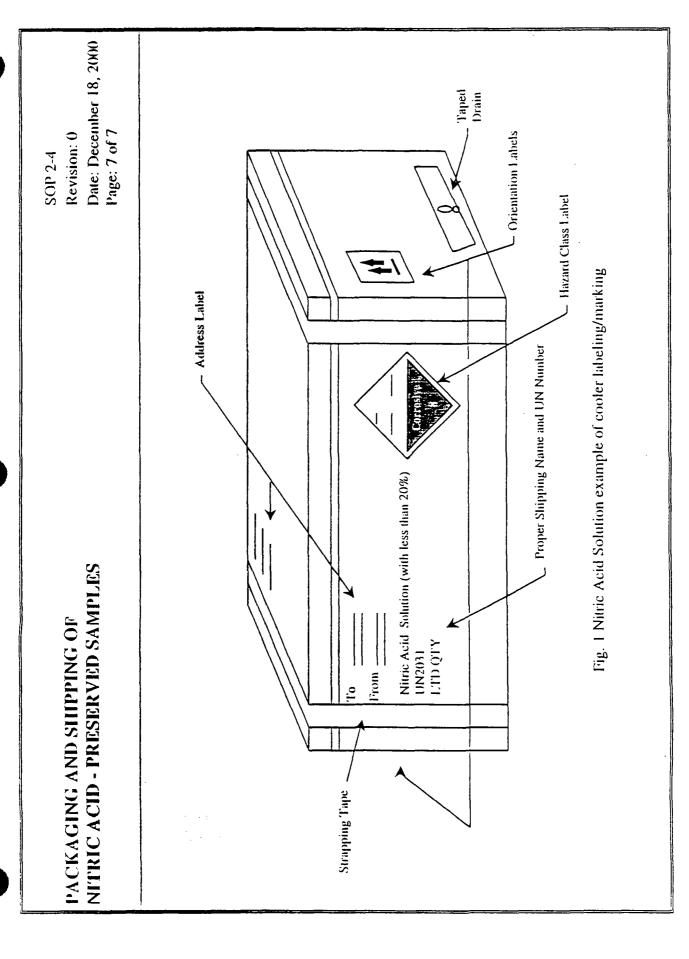
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Proper shipping name	Class or Division	UN or ID No.	Packing Group	Sub Risk	Quantity	Packing Instruction	Authorization
Hydrochioric Acid Solution	8	UN1789	п		1 plastic box × 0.5 L	Y809	LTD QTY
Nitric Acid Solution (with less than 20%)	8	UN2031	п		l plastic box × 0.5 L	Y807	LTD QTY
Sodium Hydroxide Solution	8 UN1824		п	1	l piasuc box × 0.5 L	Y309	LID QTY
Sulfuric Acid Solution	8	UN2796			1 plastic box × 0.5 L	Y809	LTD QTY
Hexanes	3	UN1208	П		l plastic box × l L	Y305	LTD QTY

Sample Cooler Labeling

Yes	No	N/A	
0	0	0	The proper shipping name, UN number, and LTD. QTY. appears on the shipping container.
0	0	0	The corresponding hazard labels are affixed on the shipping container; the labels are not obscured by tape.
0	0	0	The name and address of the shipper and receiver appears on the top and side of the shipping container.
	0		The Air Waybill is attached to the top of the shipping container.
	[]	0	Up Arrows have been attached to opposite sides of the shipping container.
			Packaging tape does not obscure markings or labeling.



PACKAGING AND SHIPPING OF ENVIRONMENTAL SAMPLES

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Prepared: Javil O. Alm 10/5/95	Technical Review:	10/19/15
Signature/Date		/ Signature/Date
QA Review: 11/104 a. Neli 10/20/95	Approved:	mt 11/8/9
Signature/Date	//	Signature/Date
James Case Harry Sturick Whyles	- //	

1.0 OBJECTIVE

The objective of this standard operating procedure (SOP) is to establish packaging and shipping requirements and guidelines for environmental sample shipping.

Signature/Date

2.0 BACKGROUND

2.1 Definitions

Environmental Sample - An Environmental Sample is any sample that has less than reportable quantities for any hazardous constituents according to Department of Transportation (DOT) regulations promulgated in 49 CFR - Part 172.

Chain of Custody Seal - Narrow strip of adhesive paper used to demonstrate that no tampering has occurred.

2.2 Discussion

Proper packaging and shipping is necessary to ensure the protection of the integrity of environmental samples shipped for analysis.

2.3 Associated Procedure

CDM Federal SOP 1-2, Sample Custody

3.0 RESPONSIBILITIES

Field Team Leader - The Field Team Leader is responsible for ensuring that packaging and sampling procedures are conducted in accordance with this SOP. The Field Team Leader is also responsible for ensuring that laboratory analysis of samples is properly coordinated by CDM Federal.

PACKAGING AND SHIPPING OF ENVIRONMENTAL SAMPLES

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4.0 REQUIRED EQUIPMENT

- · Coolers with return address of CDM Federal office
- Heavy-duty plastic garbage bags
- · Plastic zip-top bags, small and large
- Clear Tape
- Fiber tape
- Duct tape
- Vermiculite (or equivalent)*
- Bubble wrap (optional)
- Ice
- Chain-of-Custody seals
- Completed Chain-of-Custody record or CLP custody records if applicable
- Completed Bill of Lading
- "This End Up" and directional arrow labels
- * Check for any client-specific or laboratory requirements related to the use of absorbent packaging materials.

5.0 PROCEDURES

The following steps must be followed when packing sample bottles and jars for shipment:

- 1. Verify the samples undergoing shipment meet the definition of "Environmental Sample" and are not a hazardous material as defined by DOT. Professional judgement and/or consultation with the appropriate Health and Safety Coordinator or the Health and Safety Manager should be observed.
- 2. Select a sturdy cooler in good repair. Secure and tape the drain plug with fiber or duct tape. Line the cooler with a large heavy-duty plastic garbage bag.
- 3. Be sure the caps on all bottles are tight (will not leak); check to see that labels and chain-of-custody records are completed properly (SOP 1-2, Sample Custody).
- 4. Place all bottles in separate and appropriately sized plastic zip-top bags and close the bags. Up to three VOA vials may be packed in one bag. Bottles may be wrapped in bubble wrap. Optionally, place three to six VOA vials in a quart metal can and then fill the can with vermiculite or equivalent. Note: Trip blanks must be included in coolers containing VOA samples.

PACKAGING AND SHIPPING OF ENVIRONMENTAL SAMPLES

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5. Place two to four inches of vermiculite (or equivalent) into a cooler that has been lined with a trash bag, and then place the bottles and cans in the bag with sufficient space to allow for the addition of more packing material between the bottles and cans. It is preferable to place glass sample bottles and jars into the cooler vertically. Due to the strength properties of a glass container, there is much less chance for breakage when the container is packed vertically rather than horizontally.

- 6. Put ice in large plastic zip-top bags (double bagging the zip-tops is preferred) and properly seal. Place the ice bags on top of and/or between the samples. Several bags of ice are required (dependant on outdoor temperature, staging time, etc.) to maintain the cooler temperature at less than or equal to 4° Centigrade. Fill all remaining space between the bottles or cans with packing material. Securely fasten the top of the large garbage bag with fiber or duct tape.
- 7. Place the completed Chain-of-Custody Record or the CLP Traffic Report Form (if applicable) for the laboratory into a plastic zip-top bag, seal the bag, tape the bag to the inner side of the cooler's lid, and then close the cooler.
- 8. Fiber tape shall be wrapped around each end of the cooler two times, and completed Chain-of-Custody seals affixed to the top opposite sides of the cooler, half on the fiber tape so that the cooler cannot be opened without breaking the seal. Complete two more wraps around with fiber tape; place clear tape over custody seals.
- 9. The shipping container lid must be marked "THIS END UP" and arrow labels which indicate the proper upward position of the container should be affixed to the cooler. A label containing the name and address of the shipper (CDM Federal) shall be placed on the outside of the container. Labels used in the shipment of hazardous materials (such as Cargo Only Air Craft, Flammable Solids, etc.) are not permitted to be on the outside of the container used to transport environmental samples and shall not be used. The name and address of the laboratory shall be placed on the container, or when shipping by common courier, the Bill of Lading shall be completed and attached to the lid of the shipping container.

6.0 RESTRICTIONS/LIMITATIONS

The holding times for the samples packed for shipment must not be exceeded. It is recommended that samples be packed in time to be shipped nightly for overnight delivery. Use caution when shipping samples for weekend delivery; make arrangements with laboratory before sending samples.

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7.0 REFERENCES

U.S. Environmental Protection Agency, Sampler's Guide to the Contract Laboratory Program, EPA/540/P-90/006, December 1990.

U.S. Environmental Protection Agency, Region IV, Standard Operating Procedures and Quality Assurance Manual, February 1991.

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Prepared: Del Baird Technical Review: Jackie Mosher

OA Review: David O. Johnson

Approved:

1 2/3 3/99 Signature/Date

Issued: Lase Naw Bustin 264

1.0 OBJECTIVE

The objective of this standard operating procedure (SOP) is to set CDM Federal criteria for content entry and form of field logbooks.

2.0 BACKGROUND

2.1 Definitions

Biota - The flora and fauna of a region.

<u>Decontamination</u> - To remove contaminants from field sampling equipment that might bias analytical results.

Magnetic Declination Corrections - Compass adjustments to correct for the angle between magnetic north and geographical meridians.

2.2 Discussion

Information recorded in field logbooks include observations, data, calculations, time, weather, description of the data collection activity, methods, instruments, and results. Additionally, the logbook may contain descriptions of wastes, biota, geologic material, and site features including sketches, maps, or drawings as appropriate.

3.0 RESPONSIBILITIES

Field Team Leader (FTL) - The FTL is responsible for ensuring the nature and form of data entries are conducted in accordance with this procedure.

Site Personnel - All CDM Federal employees who make entries in field logbooks during on-site activities are required to read this procedure prior to engaging in this activity. The FTL will assign field logbooks to

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site personnel who will be responsible for their care and maintenance.

4.0 REQUIRED EQUIPMENT

- Site-specific plans
- Field notebook
- Indelible black or blue ink pen
- Ruler or similar scale (in some circumstances)

5.0 PROCEDURES

5.1 Preparation

In addition to this SOP, site personnel responsible for maintaining logbooks must be familiar with other pertinent CDM Federal and site SOPs. These should be consulted as necessary to obtain specific information about equipment and supplies, health and safety, sample collection, packaging, decontamination, and documentation. These procedures should be located at the field office.

Field logbooks shall be bound with lined, consecutively numbered pages. All pages must be numbered prior to initial use of the logbook. Prior to use in the field, each logbook will be marked with a specific document control number issued by the document control administrator, if required by the QIP. Not all contracts require document control numbers. The following information shall be recorded on the cover of the logbook:

- Field Logbook Document Control Number
- Activity (if the logbook is to be activity-specific)
- Name of CDM Federal contact and phone number(s)
- Start date

The first few (approximately five) pages of the logbook will be reserved for a table of contents. Mark the first page with the heading and enter the following:

TABLE OF CONTENTS

Date/Description Page

(Start Date)/Reserved for TOC 1-5

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The remaining pages of the Table of Contents will be designated as such with "TOC" written on the top center of each page.

5.2 Operation

The following is a list of requirements that must be followed when using a logbook:

- Record work, observations, quantities of materials, calculations, drawings, and related information directly in the logbook. If data collection forms are specified by an activity-specific plan, this information need not be duplicated in the logbook. However, any forms used to record site information must be referenced in the logbook.
- Do not start a new page until the previous one is full or has been marked with a single diagonal line so that additional entries cannot be made. Use both sides of each page.
- Do not erase or blot out any entry at any time. Indicate any deletion by a single line through the material to be deleted. Initial and date each deletion. Take care to not obliterate what was written previously.
- Do not remove any pages from the book.
- Record as much information as possible.

Specific requirements for field logbook entries include:

- Initial and date each page
- Sign and date the final page of entries for each day
- Initial and date all changes
- Multiple authors must sign out the logbook by inserting the following:

Above notes authored by:

- (Sign name)
- (Print name)
- (Date)
- A new author must sign and print his/her name before additional entries are made
- Draw a diagonal line through the remainder of the final page at the end of the day
- Record the following information on a daily basis:
 - Date and time
 - Name of individual making entry

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- Names of field team and other persons on-site
- Description of activity being conducted including station (i.e., well. boring, sampling location number) if appropriate
- Weather conditions (i.e., temperature, cloud cover, precipitation, wind direction, and speed) and other pertinent data
- Level of personal protection to be used
- Serial numbers of instruments
- Required calibration information
- Serial/tracking numbers on documentation (e.g., carrier airbills)

Entries into the field logbook shall be preceded with the time (written in military units) of the observation. The time should be recorded frequently and at the point of events or measurements that are critical to the activity being logged. All measurements made and samples collected must be recorded unless they are documented by automatic methods (e.g., data logger) or on a separate form required by an operating procedure. In these cases, the logbook must reference the automatic data record or form.

At each station where a sample is collected or an observation or measurement made, a detailed description of the location of the station is required. Use a compass (include a reference to magnetic declination corrections), scale, or nearby survey markers, as appropriate. A sketch of station location may be warranted. All maps or sketches made in the logbook should have descriptions of the features shown and a direction indicator. It is preferred that maps and sketches be oriented so that north is toward the top of the page.

Other events and observations that should be recorded include:

- Changes in weather that impact field activities
- Deviations from procedures outlined in any governing documents. Also record the reason for any noted deviation.
- Problems, downtime, or delays
- Upgrade or downgrade of personal protection equipment

5.3 Post-Operation

To guard against loss of data due to damage or disappearance of logbooks, completed pages shall be periodically photocopied (weekly, at a minimum) and forwarded to the field or project office. Other field records shall be photocopied and submitted regularly and as promptly as possible to the office. When possible, electronic media such as disks and tapes should be copied and forwarded to the office.

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At the conclusion of each activity or phase of site work, the individual responsible for the logbook will ensure that all entries have been appropriately signed and dated, and that corrections were made properly (single lines drawn through incorrect information, then initialed and dated). The completed logbook shall be submitted to the records file.

6.0 RESTRICTIONS/LIMITATIONS

Field logbooks constitute the official record of on-site technical work, investigations, and data collection activities. Their use, control, and ownership are restricted to activities pertaining to specific field operations carried out by CDM Federal personnel and their subcontractors. They are documents that may be used in court to indicate and defend dates, personnel, procedures, and techniques employed during site activities. Entries made in these notebooks should be factual, clear, precise, and as non-subjective as possible. Field logbooks, and entries within, are not to be utilized for personal use.

7.0 REFERENCES

Sandia National Laboratories, *Procedure for Preparing, Sampling and Analysis Plan. Site-Specific Sampling Plan, and Field Operating Procedures*, QA-02-03, Albuquerque Environmental Program Department 3220, Albuquerque, New Mexico, 1991.

Sandia National Laboratories, Division 7723, Field Operation Procedure for Field Logbook Content and Control, Environmental Restoration Department, Albuquerque, New Mexico, 1992.

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1.0 OBJECTIVE

The purpose of this standard operating procedure (SOP) is to provide standard guidelines and methods for photographic documentation, which include still photography and video tape recordings, of field activities and site features (geologic formations, core sections, lithologic samples, water samples, general site layout, etc.). This document shall provide guidelines designed for use by a professional or amateur photographer. This SOP is intended for circumstances when formal photographic documentation is required. Based on project requirements, it may not be applicable for all photographic activities.

2.0 BACKGROUND

2.1 Definitions

<u>Photographer</u> - Camera operator (professional or amateur) of either still photography or video tape recording whose primary function with regard to this SOP is to produce documentary or data-oriented visual media.

<u>Identifier Component</u> - Visual components used within a photograph such as visual slates, reference markers, pointers.

<u>Standard Reference Marker</u> - A reference marker that is used to indicate a feature size in the photograph and is a standard length of measure, such as a ruler, meter stick, etc. In limited instances, if a ruled marker is not available or its use is not feasible, it can be a common object of known size placed within the visual field and used for scale.

<u>Slates</u> - Blank, white, index cards used to write information pertaining to the subject/procedure being photographed. Letters and numbers on the slate will be bold and written with black, indelible marking pens.

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Arrows and Pointers - A marker used to indicate and/or draw attention to a special feature within the photograph.

<u>Contrasting Backgrounds</u> - Backdrop used to lay soil samples, borings, or other objects for clearer viewing and to delineate features.

<u>Data Recording Camera Back</u> - A camera attachment or built-in feature that will record, at the very least, frame numbers and dates directly on the film.

2.2 Discussion

Photographs and video tape recordings made during field investigations are used as an aid in documenting and describing site features, sample collection activities, equipment used and possible lithologic interpretation. This SOP is designed to illustrate the format and desired placement of identifier components, such as visual slates, standard reference markers, and pointers. These items shall become an integral part of the "visual media" that, for the purpose of this document, shall encompass both still photographs including digital photographs and video tape recordings (or video footage). The use of a photographic logbook and standardized entry procedures are outlined as well. These procedures and guidelines will minimize potential ambiguities that may arise when viewing the visual media and ensures the representative nature of the photographic documentation.

3.0 RESPONSIBILITIES

Field Team Leader (FTL) - The FTL is responsible for ensuring that the nature and form of data entries are conducted in accordance with this procedure.

Photographer - The photographer shall seek direction from the FTL and discuss the visual documentation requirements and schedule, regularly. The FTL is responsible for directing the photographer to specific situations, site features, or operations that the photographer will be responsible for documenting. The photographer is responsible for maintaining a logbook per Sections 5.1, 5.2.4, and 5.3.1 of this SOP.

4.0 REQUIRED EQUIPMENT

- 35mm camera or disposable single use camera (35mm or panoramic use)
- Digital Camera
- Video Camera

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- Logbook
- Indelible black ink pen
- Standard reference markers
- Slates
- Arrows or pointers
- Contrasting backgrounds
- Medium speed, or multi purpose fine-grain, color, 35-mm, negative film or slide film (project dependent)
- Data recording camera back (if available)
- Storage disk for digital camera

5.0 PROCEDURES

5.1 Documentation

A commercially available, bound logbook will be used to log and document photographic activities. Review the CDM Federal SOP 4-1 Field Logbook Content and Control and prepare all supplies needed for logbook entries.

Note: A separate photographic logbook is not required. A portion of the field logbook or site safety logbook may be designated as the photographic log section.

5.1.1 Field - Health and Safety Considerations

There are no hazards that an individual will be exposed to specific to photographic documentation. However, site-specific hazards may arise depending on location or operation. Personal protective equipment used in this operation will be site specific and dictated through requirements set by the Site Safety Officer, site Health and Safety Plan, and/or prescribed by the CDM Federal Corporate Health and Safety Program. The photographer should contact the Site Safety Officer for health and safety orientation prior to commencing field activities. The Site Health and Safety Plan must be read prior to entry to the site, and all individuals must sign the appropriate acknowledgement that this has been done.

The photographer should be aware of any potential physical hazardous while photographing the subject (i.e., low overhead hazard, edge of excavation).

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5.2 OPERATION

5.2.1 General Photographic Activities in the Field

The following sections (5.2.1 - 5.3.2) provide general guidelines that should be followed to visually document field activities and site features using still cameras and video equipment. The photographer shall seek direction from the FTL and discuss the visual documentation requirements and schedule, regularly. The FTL is responsible for directing the photographer to specific situations, site features, or operations which the photographer will be responsible for documenting. Listed below are general suggestions that the photographer should consider when performing activities under this SOP:

- The photographer should be prepared to make a variety of shots, from close-up to wide angle. Many shots will be repetitive in nature or format especially close-up site feature photographs. Consideration should therefore be given to designing a system or technique that will provide a reliable repetition of performance.
- All still photographs should be made using a medium speed, or multi purpose fine-grain, color negative film in the 35-mm format unless otherwise directed by the FTL.
- It is suggested that Kodak brand "Ektapress Gold Deluxe" film or equivalent be used as the standard film for the still photography requirements of the field activities. This film is stable at room temperature after exposure and will better survive the time lag between exposure and processing. It is suggested that film speed ASA 100 should be used for outdoor photographs in bright sunlight, ASA 200 film should be used in cloudy conditions, and ASA 400 film should be used indoors or for very low-light outdoor photographs.
- No preference of video tape brand is specified, and is left to the discretion of the photographer.
- The lighting for sample and feature photography should be oriented toward a flat condition with little or no shadow. If the ambient lighting conditions are inadequate, the photographer should be prepared to augment the light (perhaps with reflectors or electronic flash) to maintain the desired visual effect.
- Digital photography quality is determined by the camera. A camera that obtains a higher resolution has a higher number of pixels, and will store a fewer number of photographs per each disk.

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5.2.2 General Guidelines for Still Photography

Each new roll of film, digital storage disk and video tape shall contain upon the first usable frame (for film) or at the beginning of the tape (for video), a slate with the following information:

- A consecutively assigned control number (a consecutive, unique number that is assigned by the photographer to each roll of film or video tape, such as in sample numbers)
- The site location
- FTL's name
- Photographer's name
- Date

Caption Information

All still photographs will have a full caption permanently attached to the back containing the following information:

- Film roll control number and photograph document control number
- A brief site or scene description (What, Where, Who, Why)
- Date
- Time
- Direction
- Photographed by
- Witnessed by (if applicable)

A standard reference marker should be used in all documentary visual media. While the standard reference marker will predominantly be used in close-up feature documentation, inclusion in all scenes should be considered.

Close-up and Feature Photography

All close-up photographs will include a standard reference marker of appropriate size as an indication of the feature size and should contain a slate marked with the site name and any identifying label, such as a well number or core depth, that clearly communicates to the viewer the specific feature being photographed.

Feature samples, core pieces, and other lithologic media should be photographed as soon as possible after they have been removed from their in situ locations. This enables a more accurate record of their

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initial condition and color. Include a standard reference color strip (color chart such as Munsell Soil Color Chart or that available from Eastman Kodak Co.) within the scene. This is to be included for the benefit of the viewer of the photographic document and serves as a reference aide to the viewer for lithologic observations and interpretations.

The photographer will permanently attach a caption containing pertinent information to the photograph after it has been produced and prior to submitting it to the project files.

Site Photography

Site photography, in general, will consist predominantly of medium and wide angle shots. A standard reference marker should be placed adjacent to the feature, or when this is not possible, within the same focal plane.

While it is encouraged that a standard reference marker and caption/slate be included in the scene, it is understood that situations will arise that preclude their inclusion within the scene. This will be especially true of wide angle shots. In such a case, the film/tape control number shall be entered in the photographic logbook along with the frame number and all other information pertinent to the scene.

The photographer will permanently attach a caption containing this information to the photograph after it has been produced and prior to submitting it to the project files.

Panoramic

In situations where a wide angle lens does not provide sufficient subject detail, a single-use disposal panoramic camera is recommended. If this type of camera is not available, a panoramic series of two or three photos would be appropriate. Panoramics can provide greater detail while covering a wide subject, such as an overall shot of a site.

To shoot a panoramic series using a standard 35mm camera, the following procedure is recommended.

- Utilizing a stable surface or tripod to support the camera is recommended.
- 20 30% from overlap while maintaining a uniform horizon is suggested.
- 2 to 3 photos per series is recommended.

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5.2.3 General Photographic Documentation Using Video Cameras

As a reminder, it is <u>not</u> within the scope of this document to set guidelines for presentation or "show" video tape recording. The following guidelines are set for documentary video tape recordings only and should be implemented at the discretion of the FTL.

All documentary video tape recordings of field activities shall include an audio slate for all scenes. At the beginning of each video session an announcer will recite the following information: date, time (in military units), photographer, site ID number, and site location. This oral account should include any additional information clarifying the subject matter being recorded.

A standard reference marker should be used when taking close-up shots of site features with a video camera. The scene should also include a caption/slate. It shall be placed adjacent and parallel to the feature being photographed.

It is strongly encouraged that a standard reference marker and caption/slate be included in all scenes. The caption information is vital to the value of the documentary visual media and should be included. If it is not included within the scene, it should be placed before the scene.

Original video tape recordings will <u>not</u> be edited. This will maintain the integrity of the information contained on the video tape. If editing is desired, a working copy of the original video tape recording can be made. The original video tape will be archived in accordance with Section 5.3.2.

5.2.4 Photographic Documentation

Photographic activities must be documented in a photographic logbook or in a section of the field logbook or site safety logbook. Each logbook will be assigned a document control number. The photographer will be responsible for making proper entries.

In addition to following the technical standards for logbook entry as referenced in Section 5.1.2, the following information should be maintained in the appropriate logbook:

- An entry shall be made for each new roll/tape control number assigned. Include information that is recorded on the film/tape control slate (the site location, FTL's name, photographer name, and date).
- A description of the general setup, including approximate distance between the camera and the subject, will be recorded in the appropriate logbook. Include pertinent information that is

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necessary to provide a complete and full caption for each scene (refer to Sections 5.2.2 and 5.2.3).

- The laboratory used for photographic processing will be recorded (if known) in the appropriate logbook. Record the time and date that film is submitted to the photographic laboratory, as well as the time and date that the negatives and prints are received from the photographic laboratory.
- Document all situations encountered that cause an interruption in the progress of photographic activities under this SOP.
- Record as much other information as possible to assist in the identification of the photographic document.

5.3 Post Operation

All film will be sent for development and printing to a predetermined photographic laboratory (to be determined by the photographer). The photographer will be responsible for arranging transport of the film from the field to the photographic laboratory. The photographer shall also be responsible for delivery of the negatives and photographs or video tape to the project management.

5.3.1 Documentation

At the end of each day's photographic session, the photographer(s) will ensure that the appropriate logbook has been completely filled out and maintained as outlined in CDM Federal SOP 4-1.

3.5.3.2 Archive Procedures

- 1. All rolls of film, video tape recordings, and photographs should be considered controlled documents. Therefore, all photographs and the associated set of negatives and original unedited documentary video tape recordings will be submitted to the project files and handled accordingly. The FTL will ensure their proper distribution.
 - 2. Based on specific project requirements, the photographer will assign each roll of film or photograph a control number. Film control numbers may be photographed on the first usable frame of the film and permanently attached to the negative sleeve/holders. Control numbers for photographs may contain the control number of the film and the frame number from which the photograph was made and be written on the back of the photograph. All control numbers will be written with black indelible ink.

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- Original video tape footage shall be assigned a document control number. Video tapes must have a document control number written on the tape case as well as the video tape cassette. All document control numbers will be written with black indelible ink. Original video footage shall not be edited but may be copied.
- 4. Completed pages of the appropriate logbook will be copied weekly and submitted to the project files.

6.0 RESTRICTIONS/LIMITATIONS

This document is designed to provide a set of guidelines for the field amateur or professional photographer to ensure that an effective and standardized program of visual documentation is maintained.

It is not within the scope of this document to provide instruction in photographic procedures, nor is it within the scope of this document to set guidelines for presentation or "show" photography.

The procedures outlined herein are general by nature. The FTL is responsible for directly specific operational activity or procedure. Questions concerning specific procedures or requirements should be directed to the FTL.

7.0 REFERENCES

Sandia National Laboratories, Procedure for Preparing Sampling and Analysis Plans, Site-Specific Sampling Plans, and Field Operating Procedures, QA-02-03, Environmental Programs Department 3220, Albuquerque, New Mexico, 1991.

Sandia National Laboratories, Division 7723, Field Operating Procedure for Field Logbook Content, and Control, PRO 92-04, Environmental Restoration Department, Albuquerque, New Mexico, 1992.

U.S. Department of Energy, Yucca Mountain Project. Branch Technical Procedure: Field Logging, Handling, and Documenting Borehole Samples, Las Vegas, Nevada, 1989.